



Numerical Studies of Ultra-Short Laser Interactions: Application for Nanotechnology

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Collaborations with

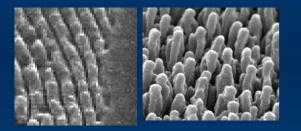
Laboratoire Hubert Curien, Saint-Etienne, France Laboratoire Lasers, Plasmas et Procédés Photoniques, Marseille, France

Rosnauka Kadri project, The Minister of Education and Science of the Russian Federation Academy of Sciences - CNRS of France

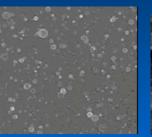
Laser-Assisted Nanoparticle and Nanostructure Production



Heiroth et al. JAP 2009



Courtesy of Torres et al., LP3



Deposited nanoparticles, Courtesy of Garrelie et al.

Nanoparticles produced on the target Kabashin et al.

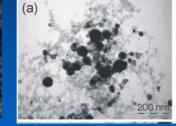
- Nanoparticles in the plume
- Deposition of nanoprticles on a substrate
- Particles can be produced on the target
- Colloidal nanoparticles
- Nanoparticles in solids (ex: sol-gel)
- Periodic and non-periodic surface structures
- Dendrides
- Nanoparticle-assisted structures (ex: nano-holes)
 Etc...

Q: Formation mechanisms ? Possibility of control ?

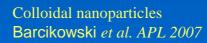
How to choose laser parameters? Ambient environment parameters ?

ps-pulses

1.38 J/cm 7.4 µg/s



200 nm 20

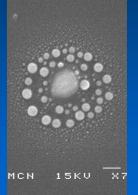


b 10 mm

Sol-gel nanoparticle arrays

J. Sol. State Chem. 2009

L. Bois et al.



Grojo et al.

LIBS application





Industrial applications

Pharmaceutical analysis



	Advantages	Disadvantages
	•Versatile sampling of solids, gases or liquids.	•Increased cost and system complexity.
;	 Little or no sample preparation is necessary. LIBS typically samples very small amounts of material and is non- destructive. 	•Large interference effects (including matrix interference and, in the case of LIBS in aerosols, the potential interference of particle size).
S	 Permits analysis of extremely hard materials. Possibility of simultaneous multi- 	•Detection limits are generally not as good as established solution techniques.
	 elemental analysis. Potential for direct detection in aerosols (a solid or liquid particle in a gaseous medium). Simple and rapid analysis (ablation and excitation processes are carried out 	•Poor precision - typically 5-10%, depending on the sample homogeneity, sample matrix, and excitation properties of the laser.

in a single step).

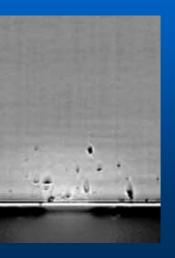
Physical Processes

- -<u>absorption</u> (Drude, interbande) for dielectrics: <u>ionization</u> (multiphotonic/tunnel, photo-abs, absorption on defects)
- -relaxation (el-el, el-ph, STE, recombination...)
- -photoemission
- -structural modifications (defects, melting, densification,...)
- -<u>thermal effets</u> (phase transitions: melting, evaporation, phase explosion)
- -<u>mechanical effects</u> (spallation, shock waves, rarefaction, fragmentation)
- -surface structuring («ripples», « spikes », «cones»)
- -material ejection (formation of laser plume, nanoparticles)

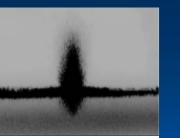
τ< relaxation time => no e-ph ou e-i equilibrium

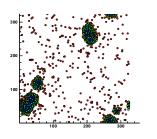
Laser Interactions in Vacuum

=>Plume temperature rises and mostly atoms are evaporated, then particulates are ejected et the end of the ablation process



Heiroth et al. JAP 2009





•Picosecond and shorter pulses

•Nanosecond and longer pulses

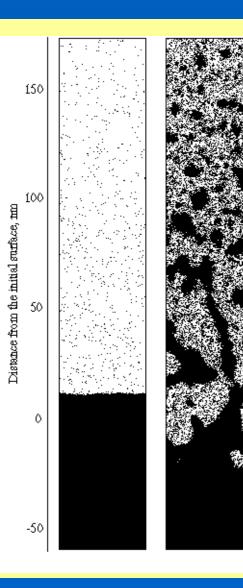
-Evaporation

-Radiation absorption

- -Fast explosive ablation mechanisms
- -No radiation absorption by the plume
- =>A mixture of monomers and clusters is ejected

Chem. Phys. Lett. , **452**, 129-132 (2008)

Molecular Dynamics Simulations



Targets: metals, molecular matrices

-Photo-thermal (phase explosion) and/or -Photo-mechanical (spallation)

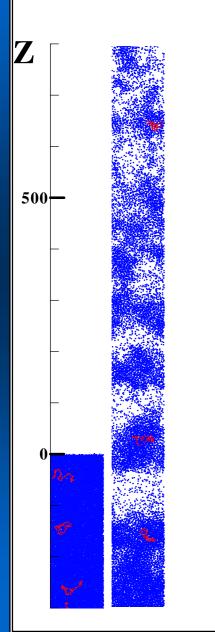
Mixture of gas and nanoparticles (size rises at the back side of the plume)

Non equilibrium processes =>difficult for classical modeling

T. E. Itina, L. V. Zhigilei, B. J. Garrison, J. Phys. Chem.B , **106**, 303-310 (2002)

MALDI and MAPLE Applications



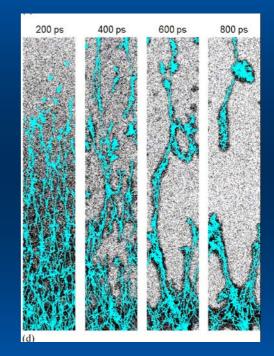


Method: Molecular Dynamics

New model : « breathing spheres » - « bead-and-spring»

Demonstrated

- molecular ejection
- nanoparticles formation
- polumers are sitting in matrix clusters



For MALDI and MAPLE

Appl. Surf. Sci., 203-204, 69-71(2003)

Two-temperature multi-material Eulerian hydrodynamics

Basic equations

Mixture model

$$\begin{split} \frac{\partial f^{\alpha}}{\partial t} + \nabla \cdot (f^{\alpha}\mathbf{u}) &= \frac{f^{\alpha}\overline{K}_{S}}{K_{S}^{\alpha}}\nabla \cdot \mathbf{u} \\ \frac{\partial (f^{\alpha}\rho^{\alpha}Z^{\alpha})}{\partial t} + \nabla \cdot (f^{\alpha}\rho^{\alpha}Z^{\alpha}\mathbf{u}) &= f^{\alpha}m_{i}^{\alpha}S^{\alpha} \\ \frac{\partial (f^{\alpha}\rho^{\alpha})}{\partial t} + \nabla \cdot (f^{\alpha}\rho^{\alpha}\mathbf{u}) &= 0 \\ \frac{\partial (\bar{\rho}\mathbf{u})}{\partial t} + \nabla \cdot (\bar{\rho}\mathbf{u}\otimes\mathbf{u}) + \nabla \overline{P} &= 0 \\ \frac{\partial (\bar{\rho}\mathbf{u})}{\partial t} + \nabla \cdot (\bar{\rho}\mathbf{u}\otimes\mathbf{u}) + \nabla \overline{P} &= 0 \\ \frac{\partial (\bar{\rho}\mathbf{u})}{\partial t} + \nabla \cdot (\bar{\rho}\mathbf{u}\otimes\mathbf{u}) + \nabla \overline{P} &= 0 \\ \frac{\partial (f^{\alpha}\rho^{\alpha}}{K_{S}^{\alpha}}\nabla \cdot \mathbf{u} - f^{\alpha}Q_{ei}^{\alpha} + Q_{L}^{\alpha} + \frac{f^{\alpha}\rho^{\alpha}C_{e}^{\alpha}}{\bar{\rho}\overline{C}_{e}}\nabla \cdot (\overline{\kappa}_{e}\nabla\overline{T}_{e}) + f^{\alpha}Q_{J}^{\alpha} + f^{\alpha}Q_{rad}^{\alpha} \\ \frac{\partial (f^{\alpha}\rho^{\alpha}E_{i}^{\alpha})}{\partial t} + \nabla \cdot (f^{\alpha}\rho^{\alpha}E_{i}^{\alpha}\mathbf{u}) &= -\overline{P}_{i}\frac{f^{\alpha}\overline{K}_{S}}{K_{S}^{\alpha}}\nabla \cdot \mathbf{u} + f^{\alpha}Q_{ei}^{\alpha} \\ \overline{F_{e}^{\alpha}}(\rho, T_{i}) &\Rightarrow E_{i}^{\alpha}, C_{i}^{\alpha}, P_{e}^{\alpha}, K_{eS}^{\alpha} \\ F_{e}^{\alpha}(\rho, T_{e}) &\Rightarrow E_{e}^{\alpha}, C_{e}^{\alpha}, P_{e}^{\alpha}, K_{eS}^{\alpha} \\ \end{split}$$

$$\begin{split} \sum_{\alpha} f^{\alpha} &= 1 \\ \overline{\rho} &= \sum_{\alpha} f^{\alpha} \rho^{\alpha} \\ \overline{C}_{e} &= \frac{1}{\overline{\rho}} \sum_{\alpha} (f^{\alpha} \rho^{\alpha} C_{e}^{\alpha}) \\ 1/\overline{K}_{S} &= \sum_{\alpha} (f^{\alpha} / K_{S}^{\alpha}) \\ \overline{P} &= \sum_{\alpha} \frac{f^{\alpha} P^{\alpha}}{K_{S}^{\alpha}} \Big/ \sum_{\alpha} \frac{f^{\alpha}}{K_{S}^{\alpha}} \\ \overline{\rho} \overline{C}_{e} / \overline{\kappa}_{e} &= \sum_{\alpha} (f^{\alpha} \rho^{\alpha} C_{e}^{\alpha} / \kappa_{e}^{\alpha}) \\ \overline{T} &= \sum_{\alpha} f^{\alpha} \rho^{\alpha} C^{\alpha} T^{\alpha} \Big/ \sum_{\alpha} f^{\alpha} \rho^{\alpha} C^{\alpha}. \end{split}$$

8

Povarnitsyn et al. Phys. Rev. B, 75, 235414 (2007)

Transport properties

$$\nu = \min(\nu_{met}, \nu_{pl}, \nu_{max})$$
$$\nu_{met} = \underline{A_1} \frac{k_B T_i}{\hbar} + \underline{A_2} \frac{k_B T_e^2}{\hbar T_F}$$
$$\nu_{pl} = \frac{4\sqrt{2\pi} n_e Z e^4}{3\sqrt{m_e} (k_B T_e)^{3/2}} \Lambda$$
$$\nu_{max} = \frac{\sqrt{v_F^2 + k_B T_e / m_e}}{r_0}$$

$$arepsilon = arepsilon_{bb} + 1 - rac{\omega_{pl}^2}{\omega_L(\omega_L + i
u)}$$

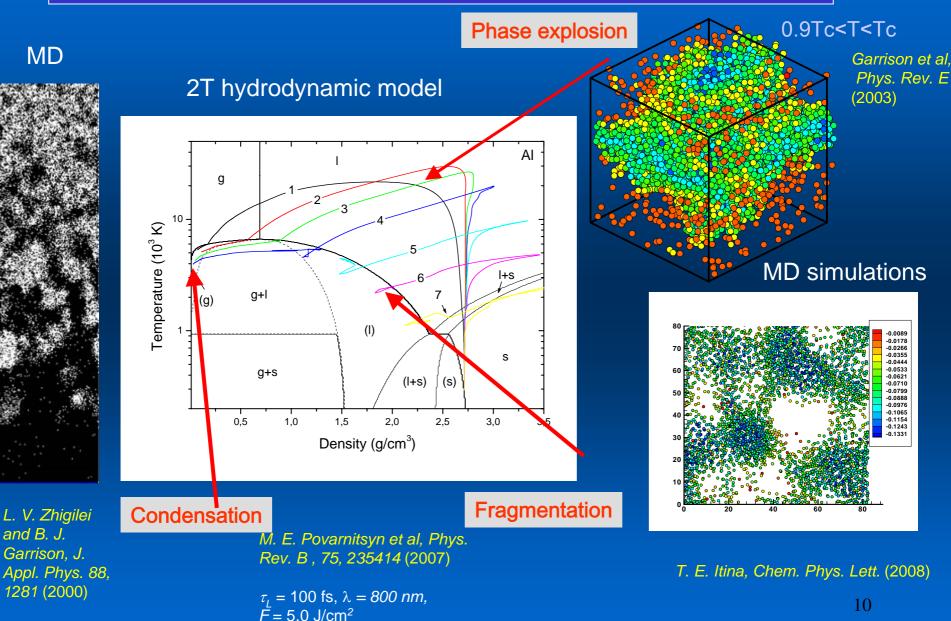
$$\gamma_{ei} = A_3 \frac{3m_e n_e \nu}{m_i}$$

$$\chi = A_4 \frac{k_B^2 n_e T_e}{m_e \nu}$$

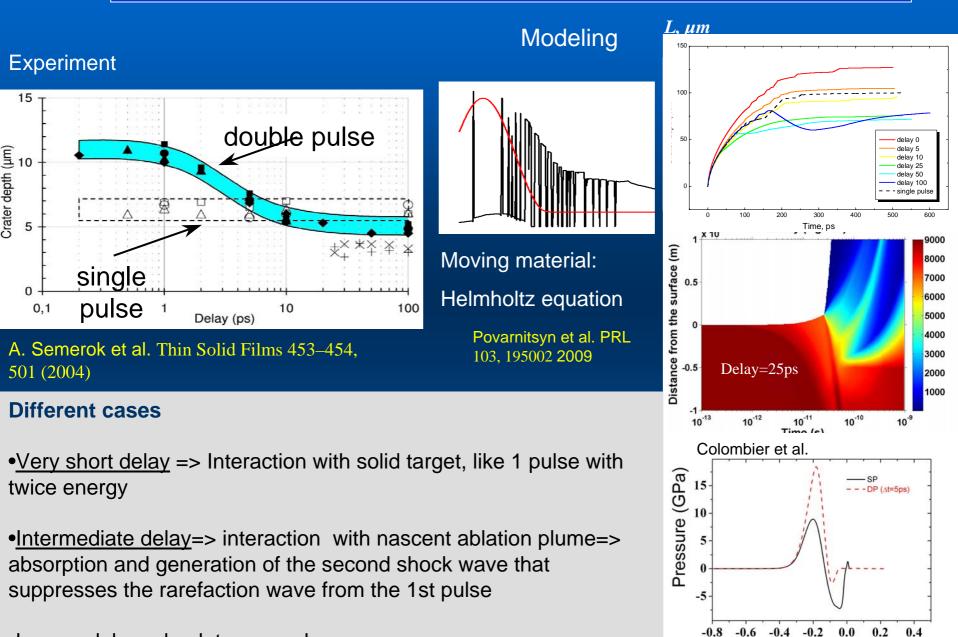
Handbook of optical constants of solids, E. Palik et al.

	λ_L , mkm	n	k	$\varepsilon_1 = n^2 - k^2$	$\varepsilon_2 = 2nk$	R
Cu	0.83	0.260	5.26	-27.60	2.74	0.964
Cu	1.24	0.433	8.46	-71.38	7.33	0.976
Au	0.83	0.188	5.39	-29.02	2.03	0.975
Au	1.24	0.372	8.77	-76.77	6.52	0.981

Analysis of Short-Pulse Laser Ablation



Ablation Suppression in Double Pulse Ablation



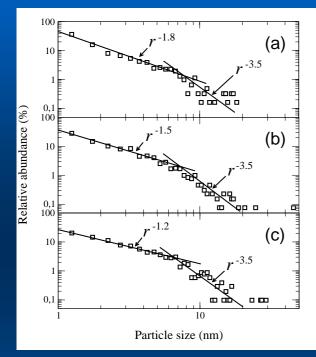
Position (µm)

Longer delay=>back to one pulse

Nanoparticle Size Distribution

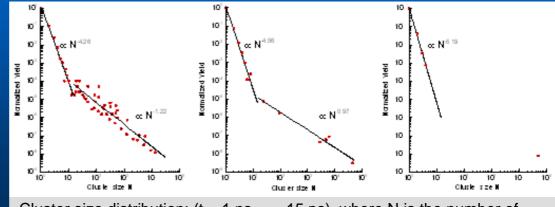
Expriments

100 fs F=4 J/cm2



EvaporationLonger scale time evolution?

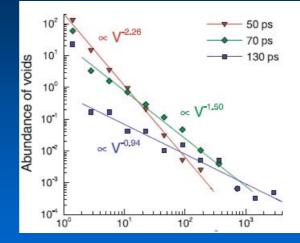
Molecular Dynamics (MD) simulations



Cluster size distribution: (t = 1 ns, τ = 15 ps), where N is the number of monomers. (a) - fluence F = 6 mJ/cm², (b)- F = 4 mJ/cm²; (c) - F = 3 mJ/cm²

Courtesy of Zhigilei et al.

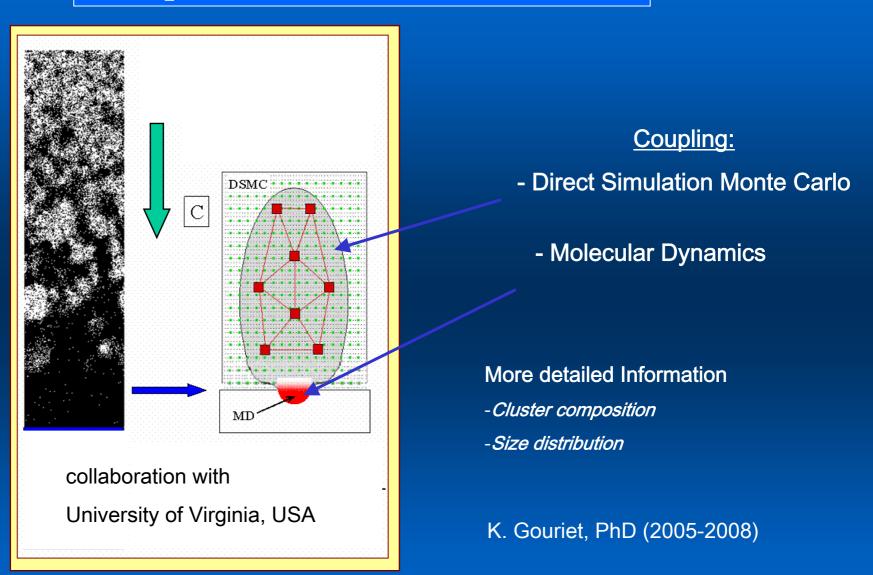
Void size distribution



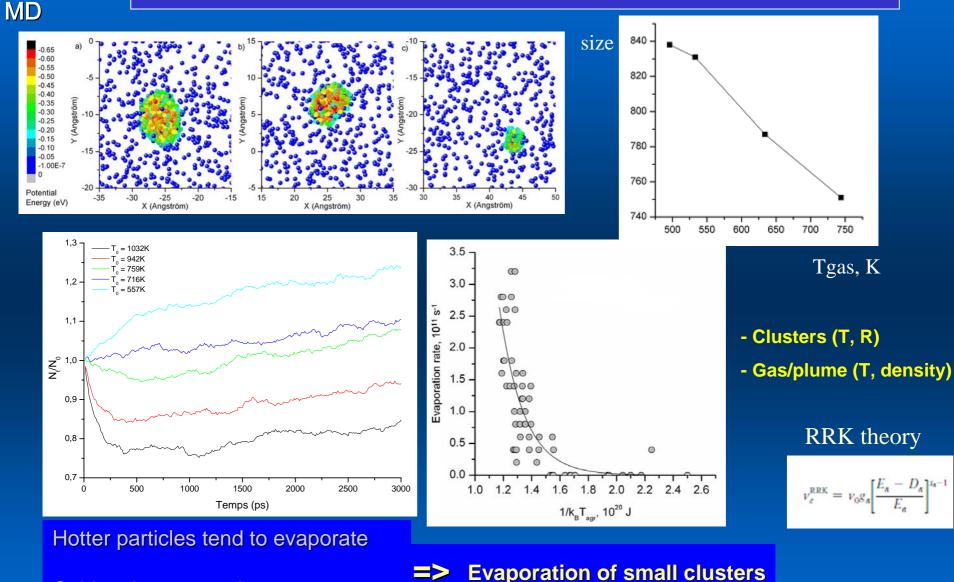
Ablation of molecular solids in vacuum

Leveugle et al. Appl. Phys. A 79, 1643–1655 (2004)

Nanoparticle formation mechanisms



Cluster Evolution

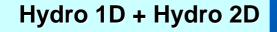


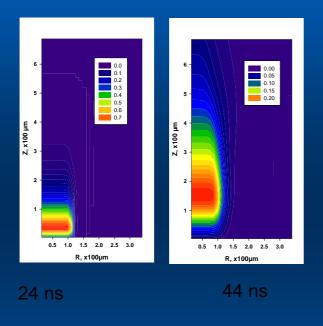
Colder clusters tend to grow

K. Gouriet, M. Sentis, T. E. Itina, J. Phys. Chem. C, 113(43), 18462-18467 (2009)

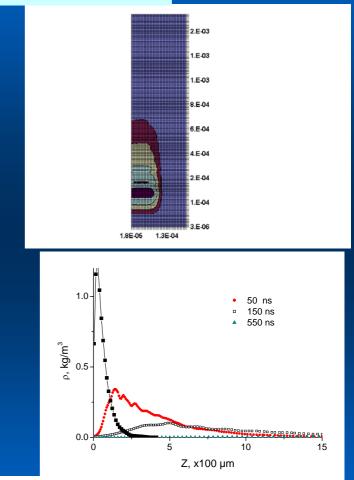
(metals and molecular solids)

Plume expansion





Hydro 1D +SDMC

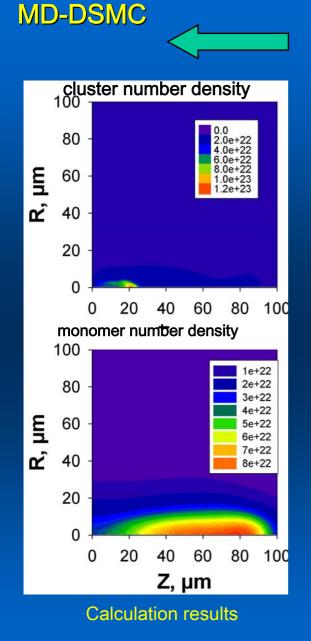


Calculated plume density pour λ = 800 nm

Laser pulse duration 100 fs, laser fluence Is 5 J /cm2, le , laser spot is 100 µm.

Rarefaction wave in plume shape

Aluminum target (a) –2D density distribution ; (b) density as a function of distance for different delayes

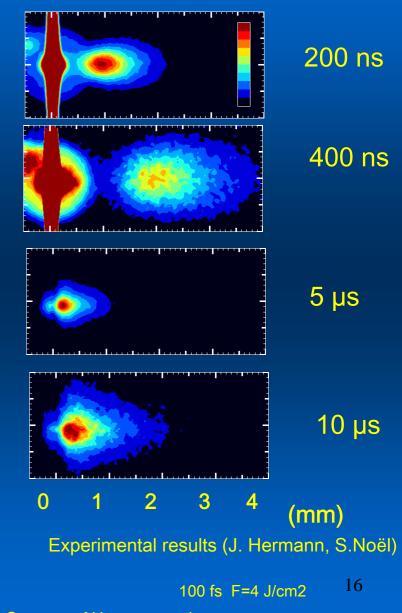


Laser

t= 50 ns, 15 ps, F=61 J/m2, R=10 μm

Appl. Surf. Sci. , 253, 7656-7661 (2007)

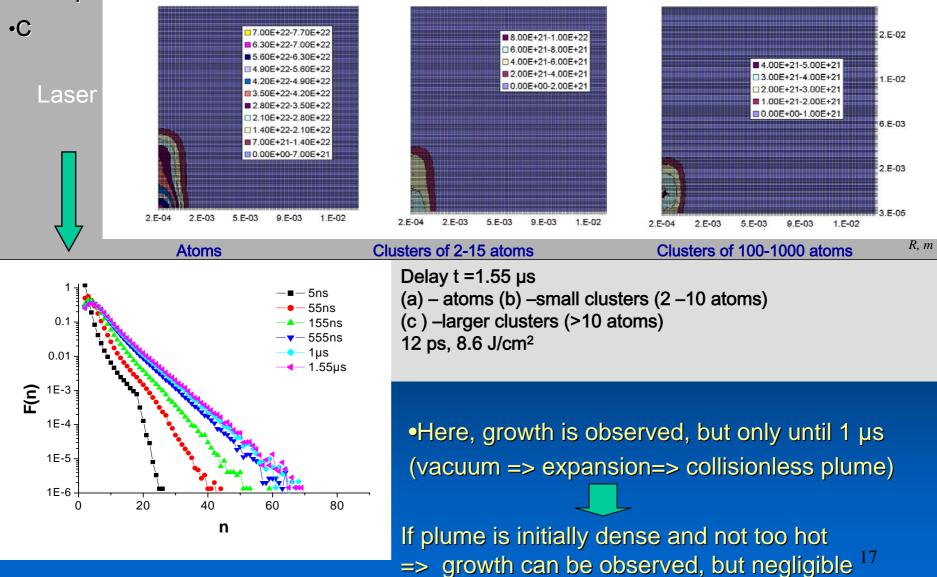
 $F_{las} = 4 \text{ Jcm}^{-2}$



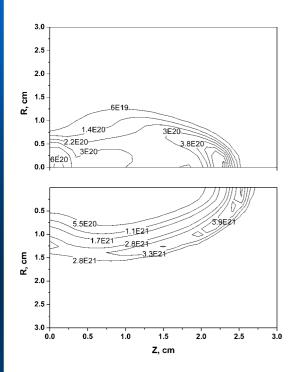
Courtesy of Hermann et al.

Plume Expansion in Vacuum: Monte Carlo Simulation

•Laser pico



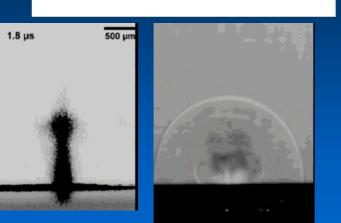
Laser Interactions in the Presence of a Gas



Plume confinement=> nucleation

Short pulses => mixture of atoms and clusters initially

+Additional collisions with the gas => condensation/evaporation, coalescence/aggregation/fragmentation

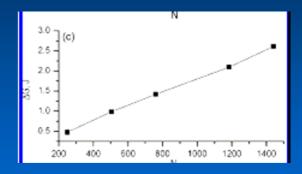


Heiroth et al. JAP 2009

Confinement in a gas

$$\mathbf{R} = \xi_0 \left(\frac{\mathbf{E}_0}{\rho_0}\right)^{1/5} t^q$$

Sedov, Zel'dovich & Raizer



Nucleation

Size distribution is controlled by free energy

$$\Delta G(n,c) = -nkT \ln(c/c_0) + 4\pi a^2 n^{2/3} \sigma$$

Supersaturation ratio

$$S = c / c_0$$

Rate of production of supercritical clusters -rises with S -rises with T

Cluster time-evolution=> master equation

$$\frac{dn_s}{dt} = \frac{1}{2} \sum_{i+j=s} [K(i,j)n_in_j - F(i,j)n_s]$$
$$- \sum_{j=1}^{\infty} [K(s,j)n_sn_j - F(s,j)n_{s+j}].$$

Sintes et al. PRA 1992

$$n_c = \left[\frac{8\pi a^2\sigma}{3kT\ln(c/c_0)}\right]^3$$

$$\rho(t) = K_c c^2 \exp\left[\frac{-\Delta G(n_c, c)}{kT}\right]$$

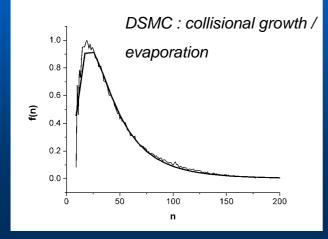
where Ks is attachment rate constant F is fragmentation rate

$$\frac{dN}{dt} = K_{S-1}N_1N_{S-1} - K_SN_1N_S \ (s \ge 3)$$

J. Park, V. Privman, E. Matijevic, 19 J. Chem. Phys. B 105, 11603-11635 (2001)

Nucleation+Growth or Condensation ?

Nucleation+Growth

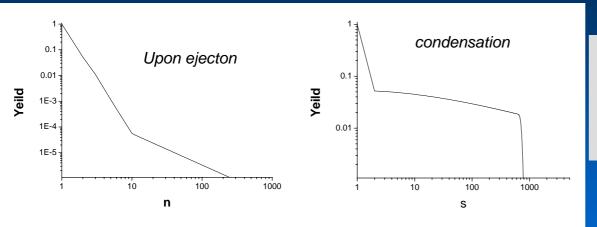


Growth without nucleation

•Typical for nanosecond ablation in a gas

Log-normal distribution

Gold ablation in water



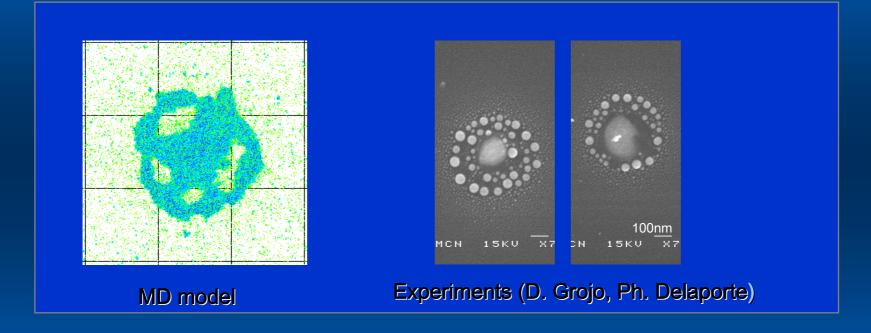
•Typical for femtosecond and picosecond ablation in vacuum (cluster precursors are ejected)

•No log-normal distribution

Gold in a gas

Laser – Particle Interactions

=>structure on the surface

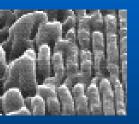


Gold nanoparticle on Si surface, 400 ps after laser irradiation

MEB

(simulation, laser picoseconde)

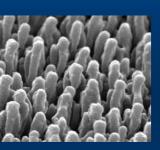
Periodic Structure Formation



Periodic (resonant) surface structures (« ripples »)
 Interference

-Surface plasmons - under certain conditions ?

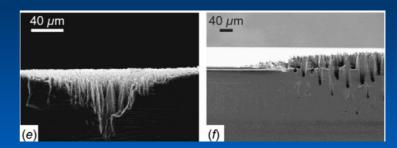
-Surface quality (diffraction)



•Non-periodic (non-resonant) structures -melting -instabilities -defects, chemical reactions, etc

Ex: « black silicon » production for sollar cells (E. Mazur et al., Th. Sarnet et al.)

- Femto => in depth
- Nano => on the surface



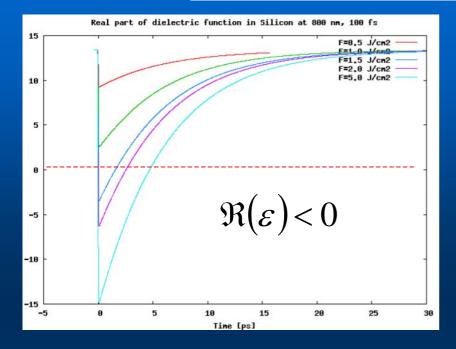
CH Crouch, APL 84, 11 (2004)

Femtosecond

Nanosecond₂₂

 $\Re(\varepsilon) < 0$

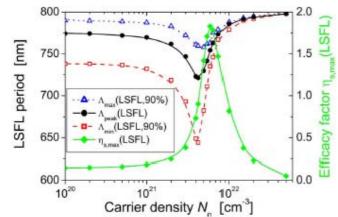
Periodic Surface Structures (Ripples) on Si

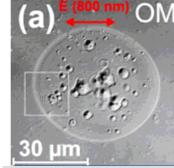


•surface EM wave (plasmons) ?

$$\varepsilon^*(n_{\rm e}) \cong 1 + (\varepsilon_{\rm g} - 1) \left(1 - \frac{n_{\rm e}}{n_0}\right) - \frac{n_{\rm e}}{n_{\rm cr}} \frac{1}{1 + i\frac{1}{\omega\tau}}$$

Derrien et al.





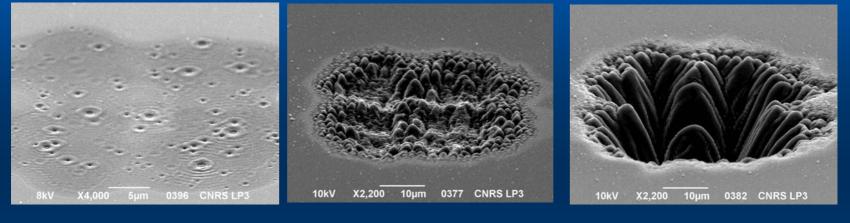
A minimum in the period of ~720 nm is observed around the critical carrier density

Bonse et al JAP 106, 2009

Structure Formation: Number of Pulses

F=900 mJ/cm², τ =100 fs, λ =800 nm

Si For different number of shots, N



Stochastic

N=1000

Large structures

Courtesy of R. Torres, LP3 CNRS, Marseille

=>Ablation on randomly-distributed sites

N=2

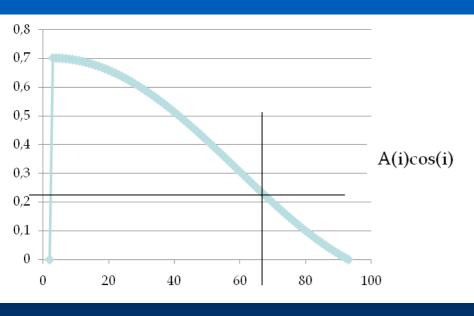
=>Melting, instabilities, capillary waves => «beads»

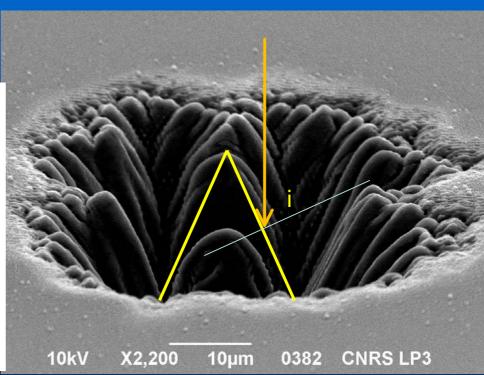
=>Further ablation and re-deposition => conical structures

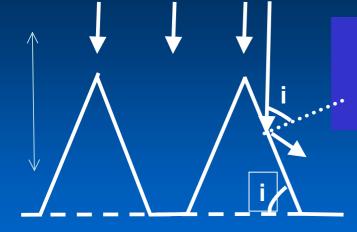
♦ Saturation of structures after a certain number of laser pulses

N=5

Structure Saturation Effect







 $F_a = FA(i)cos(i) = F_{th}$

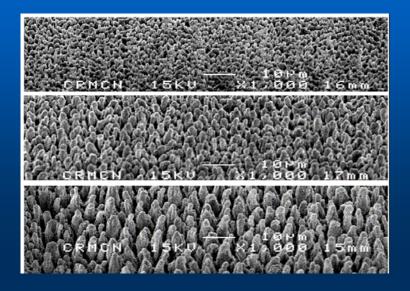
Fth=0.2 J/cm² Fexp = 0.9 J/cm² → Fth/F ~ 0.22

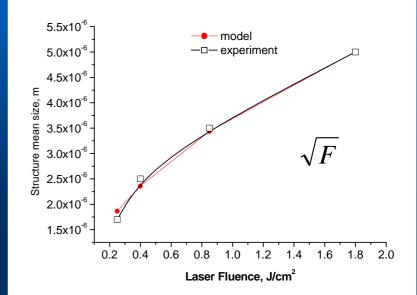
→good agreement

Ablation threshold depends on the angle !

Dyer et al. Appl. Phys. B 48 489 (1989) courtesy of V. Tokarev

Structure Formation: Laser Fluence





Surface structure formation on Si by femtosecond laser pulses: SEM pictures of laser treated surfaces (tilt 50°, same magnification) and different laser fluences

(a)- 250, (b)- 400, (c)- 850 mJ/cm²

Sarnet at al. SPIE Vol. 6881, 688119 (2008)

Mean diameter of surface structures as a function of laser fluence, where black line corresponds to the experimental results and red one to

d=3.73 10⁻⁶
$$\sqrt{F}$$
 [m]

Summary

- Modeling helps to explain
 - nanocluster formation in laser ablation
 - parameters required for nucleation
 - role of cluster ejection, nucleation, condensation, aggregation
- Femto regime=> no energy absorption by the plume
- Plume confinement is required to induce nucleation and growth
- Larger ablated mass, optimum temperature, and high collision rate favor nanoparticle formation
- Non-periodic structures have been attributed to

-Surface EM waves -> <u>periodic ripples</u> -random absorption on defects -> <u>non-periodic</u> structures -melting

-ablation and re-deposition with angle-dependent ablation threshold

Thank you !