



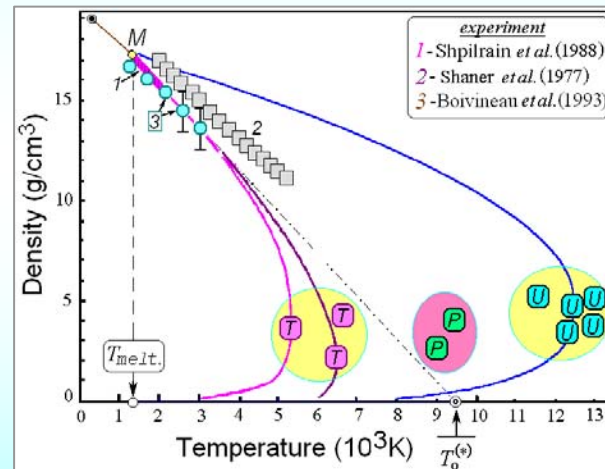
X-rays as a Tool for Probing Extreme States of Matter

EMMI Workshop, GSI, June 2010



Resolution of Uranium Critical Point Location Problem

as a goal for HIB and Laser heating and X-ray diagnostics



Igor Iosilevskiy

*Joint Institute for High Temperature (Russian Academy of Science)
Moscow Institute of Physics and Technology (State University)*

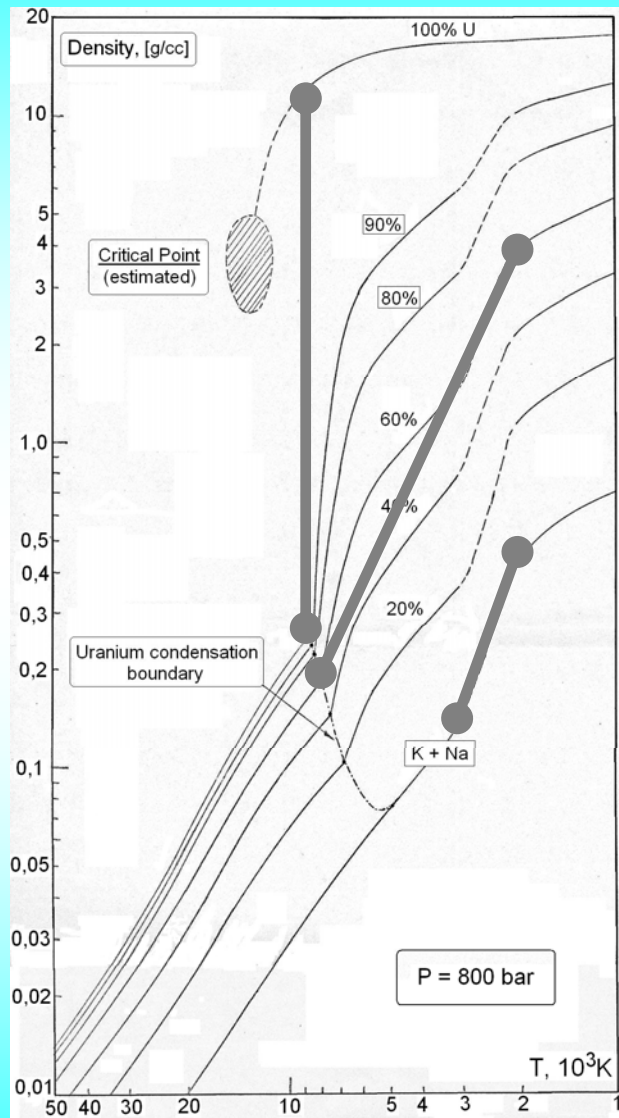
ExtreMe Matter Institute – EMMI



Developments of Gas-Core Nuclear Reactor

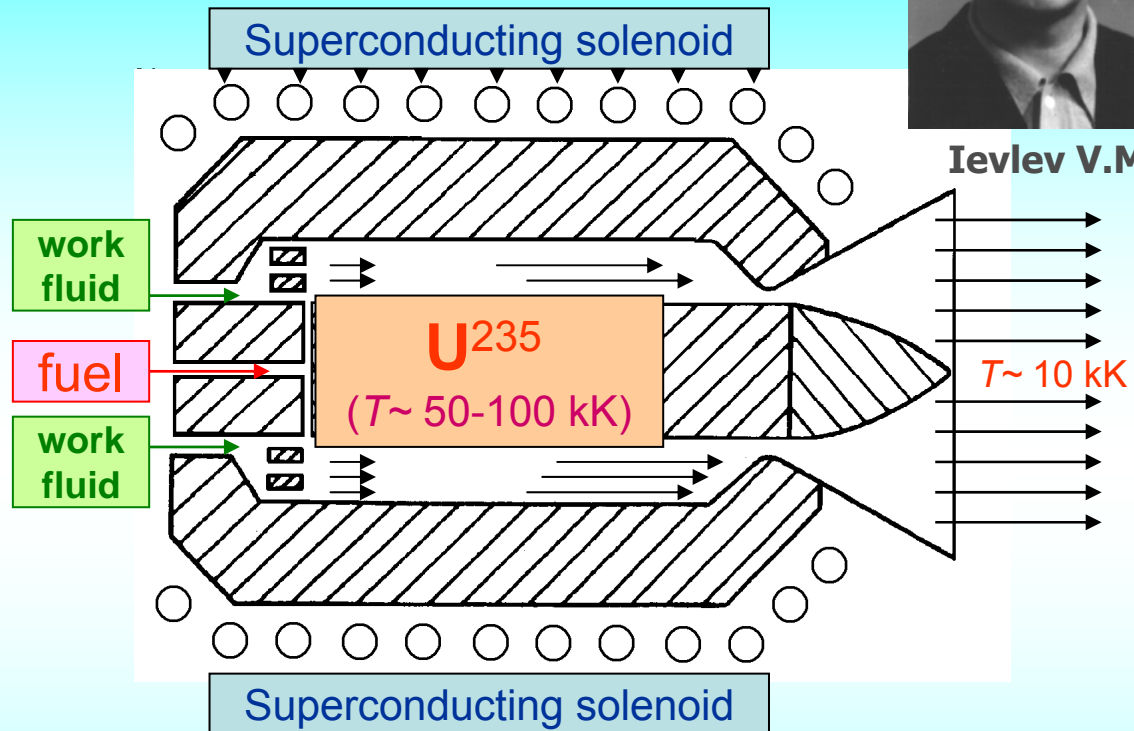


Ievlev V.M.



Phase diagram of mixture (U + K + Na)
Iosilevskiy et al. *ITPP Report*, 1972

(1950-1980)



High-temperature variant of GCNR

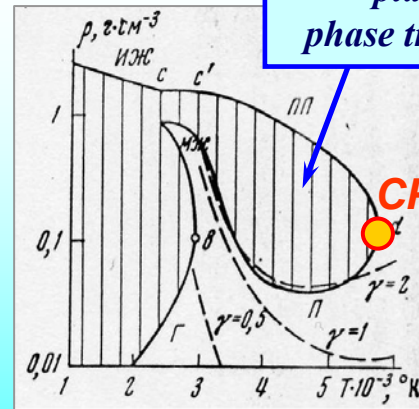
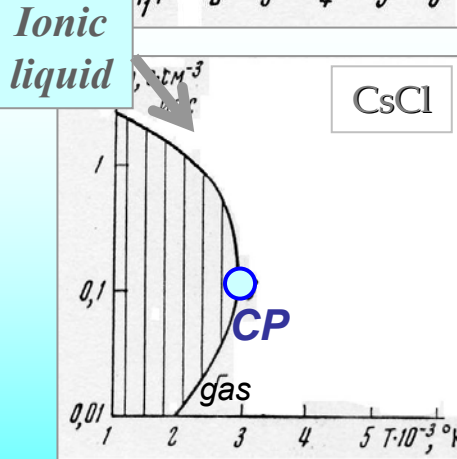
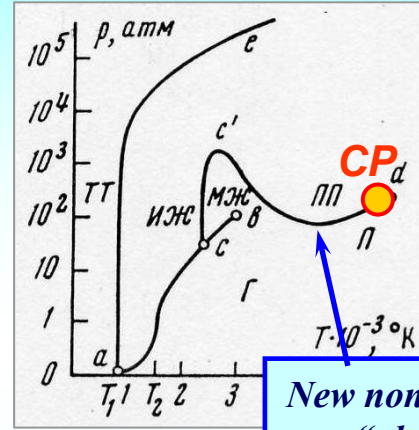
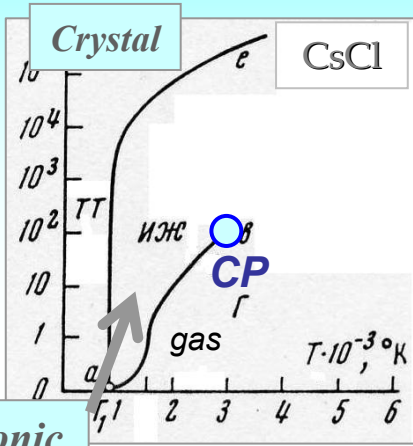
Ievlev V. *Bulletin of Russian Academy of Science (Izvestia RAS)*, № 6, (1977)

Gryaznov V, Iosilevskiy I, Fortov V, et al. "Thermophysics of gas-core nuclear reactor" /Ed. V. Ievlev (1980) (in Russian)

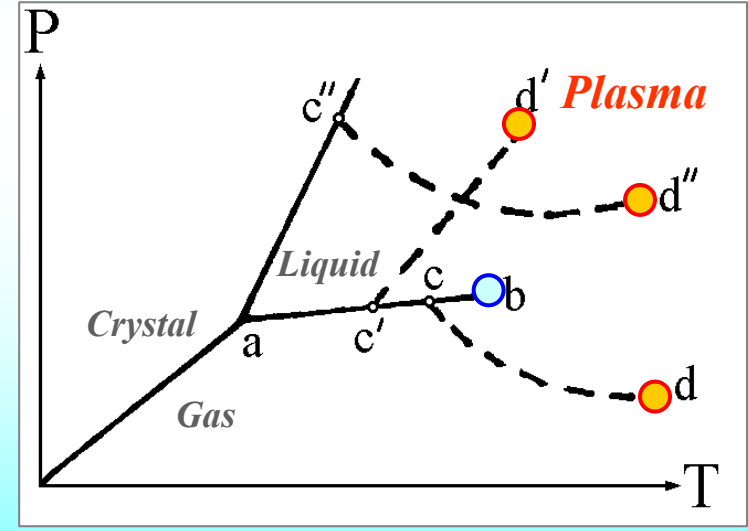
"Rocket engines and energy converters based on gas-core nuclear reactor", Ed. A. Koroteev, "Mashinostroeniye" Publishing, Moscow, (2002), (in Russian)

Hypothetical "Plasma Phase Transition" in Strongly Non-Ideal Plasma

Zeigarnik V., Kobzev G., Kurilenkov Yu., Norman G, *High-Temp.*, 10, (1972)



Plasma Phase Transition
 Landau & Zeldovitch – 1930-ths
 Norman & Starostin – 1970
 W.Ebeling et al



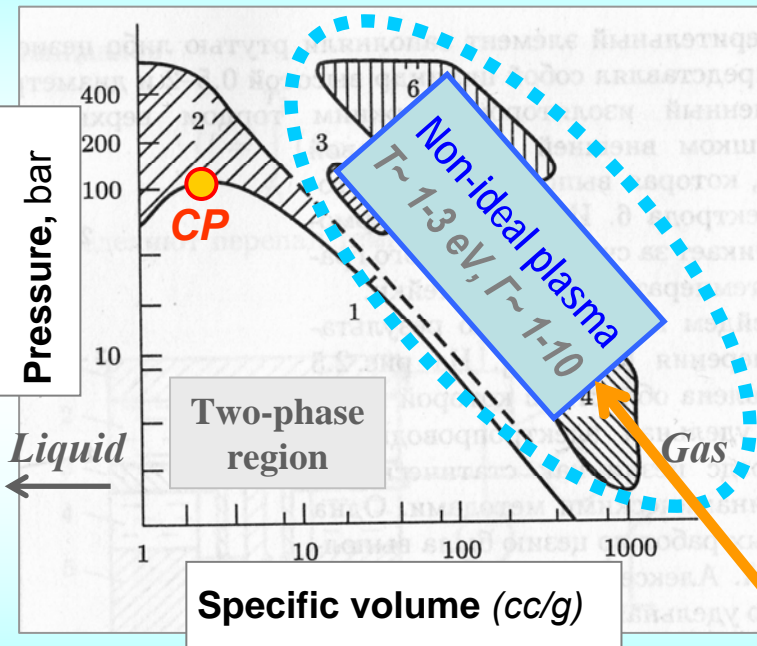
Phase diagram for non-ideal ionic mixture (CsCl)
 a,b,c,d, - triple and critical points; TT – solid phase;
 Г – gas; ИЖ – ionic liquid; МЖ – molecular liquid;
 П – plasma; ПП – warm dense plasma; T_1 – melting point

See for discussion:
 Iosilevskiy, in "Encyclopedia for Low-T Plasmas", Moscow, 2004

In search of "Plasma Phase Transition"

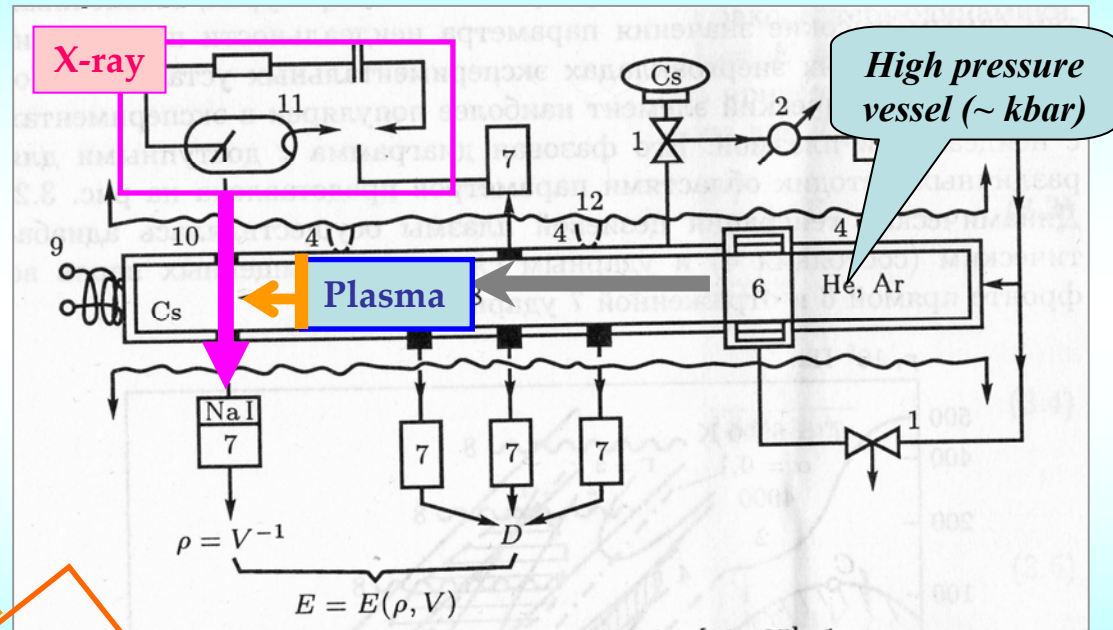
Strongly shock compressed cesium plasma

Phase diagram of cesium



Bushman A. Lomakin B. *et al.* Soviet JETP, 69 (1975)

Preheated cesium shock tube



X-ray for density measurement

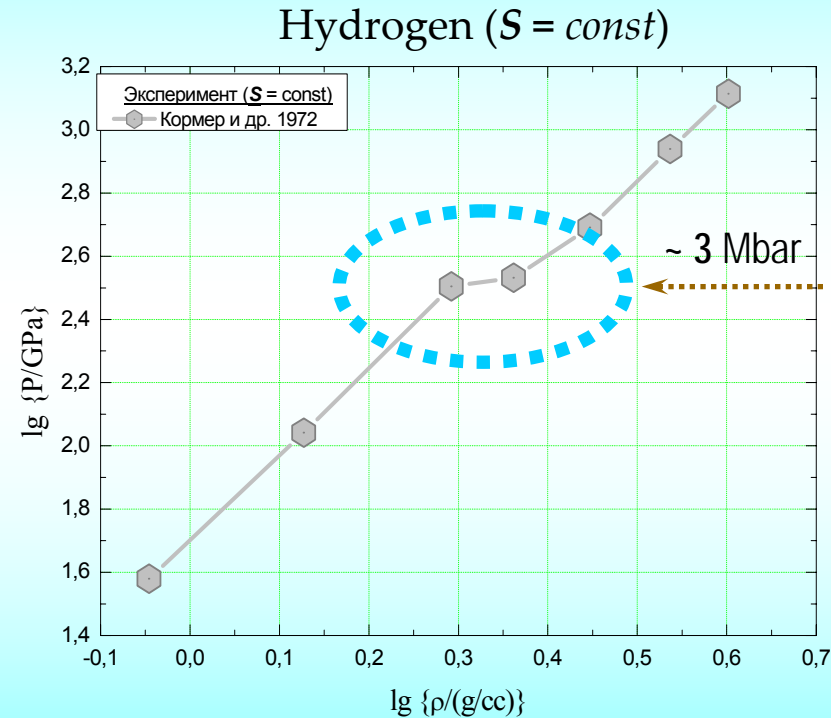
Equation of State for Non-Ideal Plasma
 $E(P, V)$

Anomalous Phase Transitions *in Hydrogen and Deuterium*

1972

*Anomalous density gap in isoentropically
compressed hydrogen ($P \sim 3$ Mbar)*
S. Kormer *et al.* (Russ. Nuclear Center, Sarov),

? - Plasma Phase Transition - ?

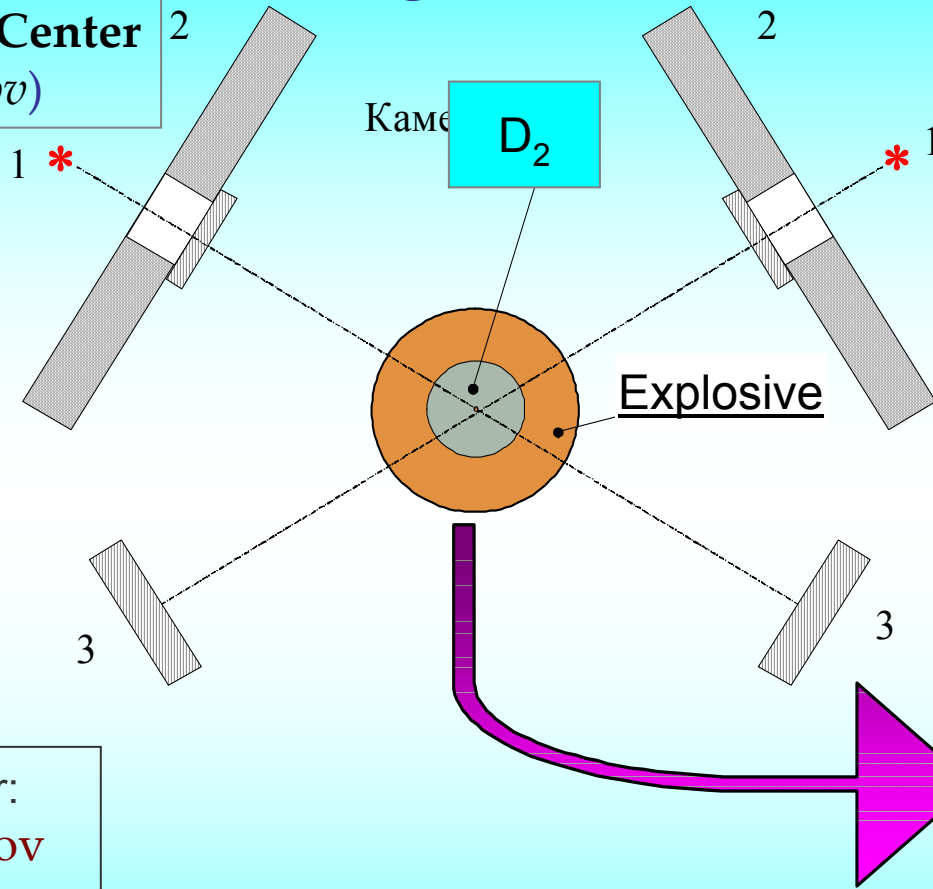


Norman & Starostin, *Plasma Phase Transition*, 1970



Study of quasi-isentropic compressibility of gaseous deuterium

Russian Nuclear Center
(VNIIEF, Sarov)



- 1 – Radiation sources,
- 2 – Micro casemates,
- 3 – Recorders

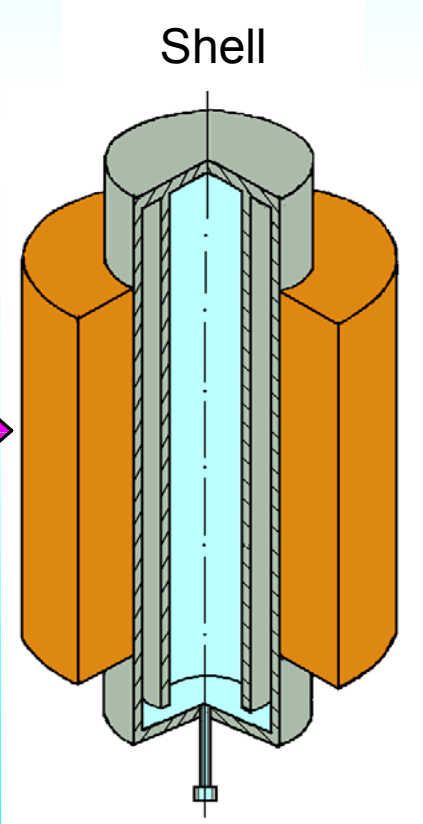


Figure after:
M. Mochalov
Et al.
SCCS-2005,
Moscow

Fortov V., Mochalov M.
Iosilevskiy I. *et al.*
(*Phys. Rev. Lett.* 99, 2007)

Quasi-isentropic compressibility of deuterium

$P \sim 100 - 300 \text{ GPa}$

Significant density gap on isentropic compression

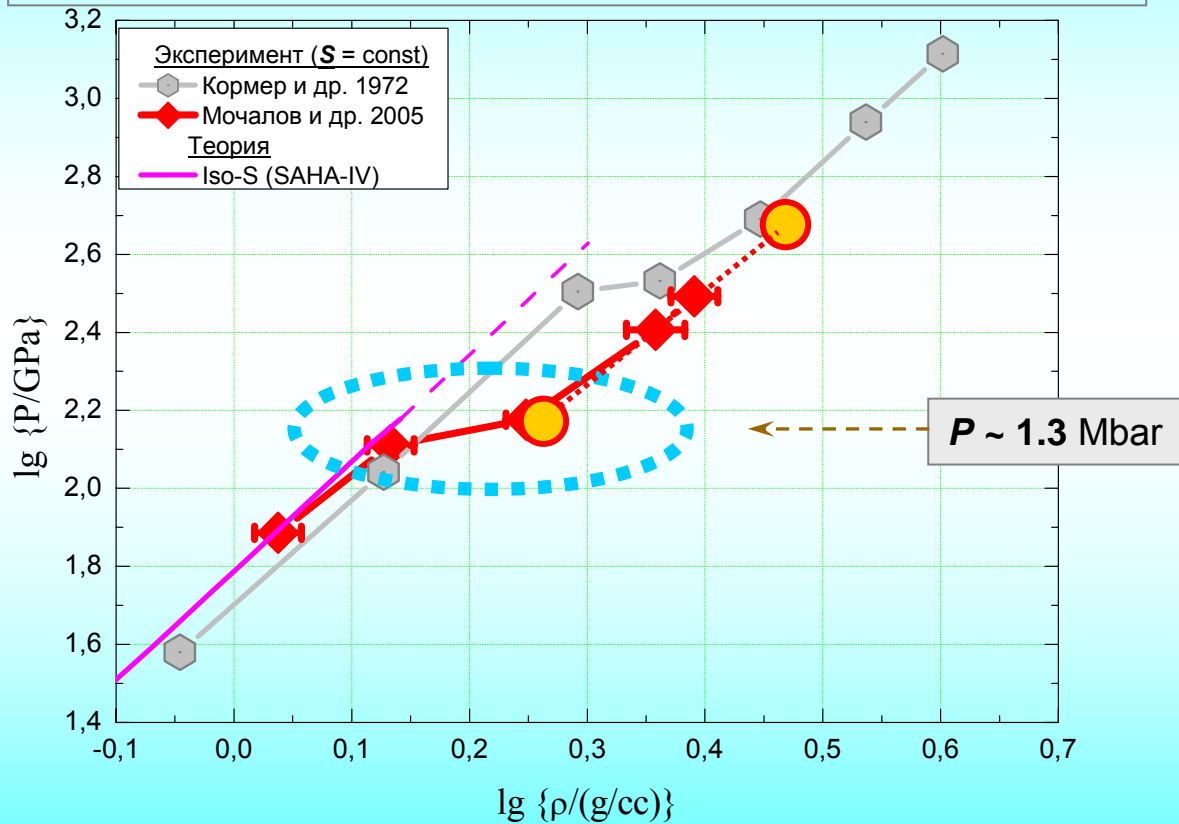


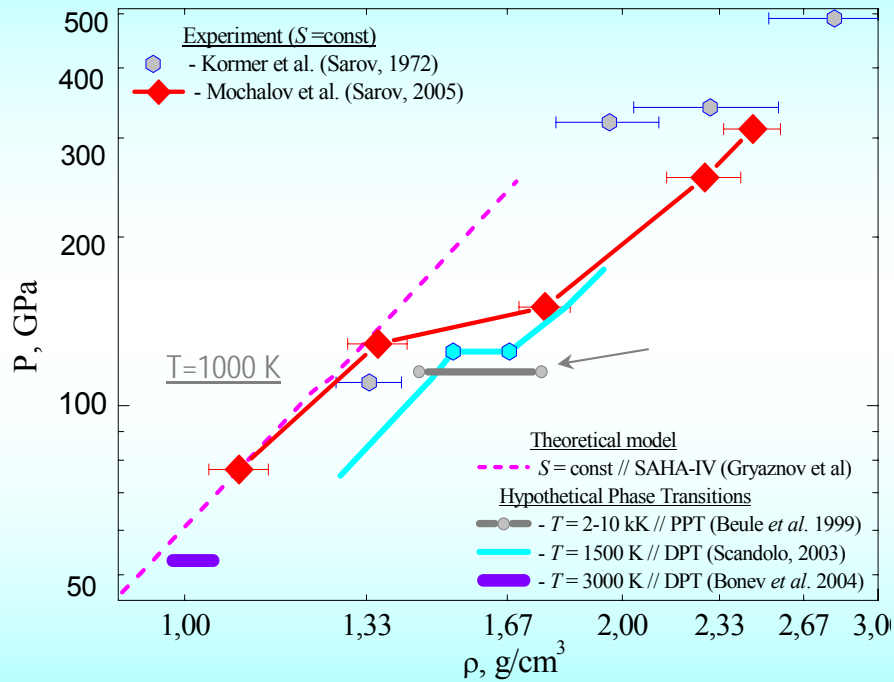
Figure after:
M. Mochalov et al.
SCCS-2005,
Moscow

Fortov V., Mochalov M.
Iosilevskiy I. *et al.*
(*Phys. Rev. Lett.* 99, 2007)

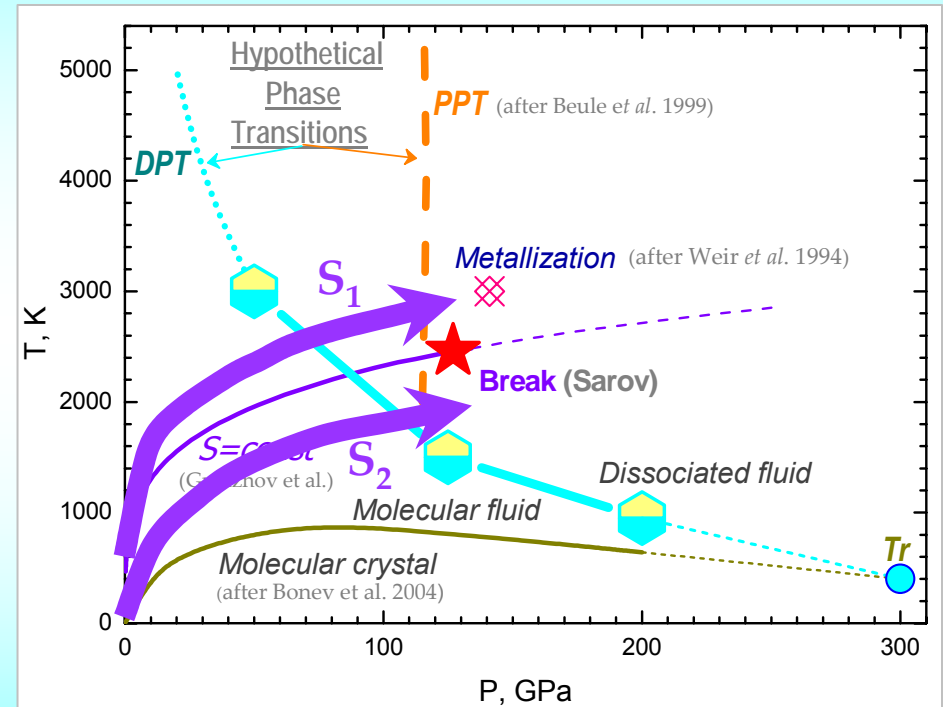
SAHA-IV: - Calculations
(*Gryaznov V. & Iosilevskiy I., 1972-2008*)

Density break in isentropic compression of gaseous deuterium \leftrightarrow hypothetical phase transition (?)

Pressure-Density Diagram



Pressure-Temperature Diagram



Historical comments - II

Boris Sharkov, Meeting in ITEPh, 1997; EMMI-Workshop, 20 May, 2010;

! - Brilliant perspectives with HIB energy deposition, $\Delta E \sim 100$ kJ/g. . . but now: $\Delta E \sim 1$ kJ/g

? - What could we do with $\Delta E \sim 1$ kJ/g and $t_{\text{HIB}} \sim 100$ ns ?

Igor Iosilevskiy, Meeting in ITEP, 1997

Igor Iosilevskiy, Meeting at HIF, 2002

Igor Iosilevskiy, Meeting in GSI, 2007

Igor Iosilevskiy, Meeting in GSI, 2009

? ? ? Meeting in GSI, 20..??

? - What could we do with $\Delta E \sim 1$ kJ/g and $t_{\text{HIB}} \sim 100$ ns ?

Study of thermophysical properties for WDM

- What to study
- How to arrange HIB energy deposition
- How to arrange measurements

HIB for thermophysical investigations

- What to study
- How to arrange HIB energy deposition
- How to arrange measurements

Basic point

- Careful choice of investigated substance and physical problem

Criteria

- great uncertainty
- great applied importance
- fundamental physics

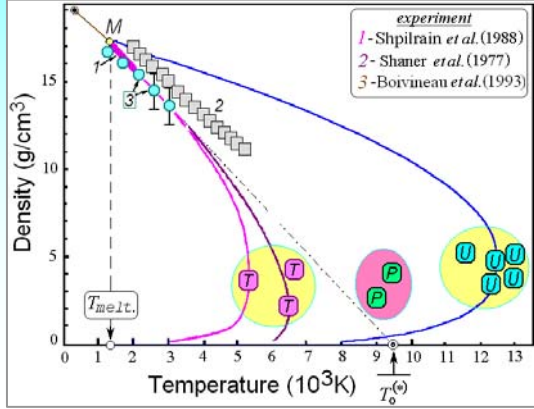
? - What substance - ? What property - ? What parameters - ?

Low energy deposition – what could we study via HIB ?

Two outstanding goals

– Uranium Critical Point Location Problem

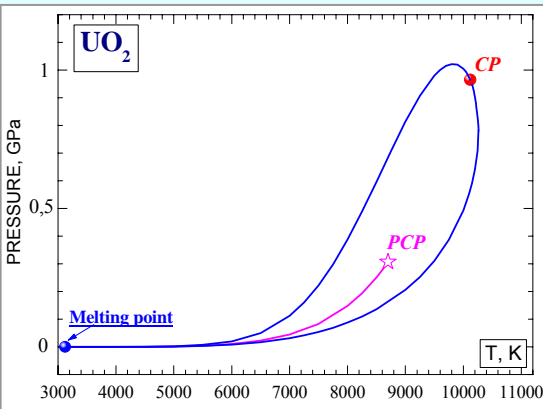
- applied importance
- phenomenology
- fundamental physical problem

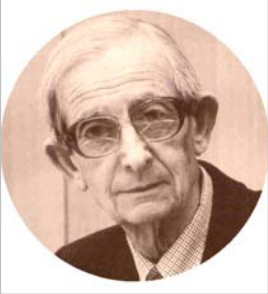


– Non-congruent Phase Transitions in High Energy Density Matter

(uranium-bearing fuels (UO₂, UC, UN ...) and other mixed substances)

- applied importance
- phenomenology
- fundamental physical problem





Yu. Khariton

«Extreme State of Matter»

Int. Conferences "Equation of State"

Russia, Elbrus, 1990 -2010

Int. Conferences "Khariton's Science Readings"

Russian Federal Nuclear Center, Sarov, 2005 -2010

Int. Conferences "Zababakhin's Science Readings"

Russian Federal Nuclear Center, Snezhinsk, 2005 -2010



Uranium Critical Point Location Problem

- Applied importance
- Uncertainty
- Fundamental physical problem

In cooperation with:

Victor Gryaznov (IPCP RAS) and Artem Ukrainets & Katya Romadinova (MIPT)

Critical Point Location Problem

Uncertainty

Recommendations of IVTAN-Database (1982)

Critical Point Parameters

Cr	9620	968	0.023
Mo	11150	546	0.0365
W	13400	337	0.043
V	12500	1078	0.027
Nb	19040	1252	0.030
Ta	20570	13500	0.036
Ti	11790	763	0.037
Zr	16250	752	0.051
Hf	18270	938	0.046
Sc	8350	408	0.048
Y	10800	374	0.068
La	11060	335	0.078
Th	14950	488	0.072
U	11630	611	0.045
UO ₂	7530	122.6	0.163
UF ₆	504.5	4.59	0.255
Pu	10000	324.2	0.081
PuO ₂	7620	101	0.202
Li	3680	60	0.055
Na	2503	25.64	0.111
NaCl	3400	35	0.266
K	2280	15.8	0.202
KCl	3200	22	0.415
Rb	2106	13.22	0.246
Cs	2043	11.75	0.308

Critical Point Location Problem

Приложение I. Критические параметры металлов

Элемент	T_c , К	p_c , 10^8 Па	ρ_c , г·см ⁻³	S_c , кал·моль ⁻¹ К ⁻¹
Zr	16250	7,52	1,79	40,45
Zn	3196	2,63	2,29	31,27
Y	10800	3,74	1,30	38,92
Yb	4280	1,38	2,36	36,92
U	11630	6,11	5,30	43,07
Ti	11790	7,63	1,31	37,92
Sn	8200	3,35	2,05	39,44
Th	14950	4,88	3,21	43,67
Tu	5910	2,65	3,22	38,52
Tl	4470	1,63	3,16	38,67
Tb	8060	3,08	2,57	42,34
Te	1850	0,75	2,21	35,22
Ta	20570	13,5	5,04	43,07
Sr	3860	0,90	0,86	35,23
Ag	7010	4,50	2,93	36,86
Se	8350	4,08	0,93	36,82
Sm	5340	2,10	2,51	40,62
Ru	15500	13,74	3,79	37,14
Rh	13510	11,23	3,62	39,66
Re	19600	15,7	6,32	43,42
Ra	3830	0,77	1,93	38,32
Pr	9160	2,85	1,86	41,39
Tc	15930	11,81	3,09	41,00
Pa	12650	4,82	3,72	43,34
Po	2050	0,62	2,67	37,52
Pt	14330	8,70	5,02	41,8
Pd	10760	7,64	3,20	36,64
Os	17110	14,49	6,83	42,60
Nb	19040	12,52	2,59	41,65
Ni	10330	9,12	2,19	36,50
Nd	7920	2,65	2,05	41,33
Mo	16140	12,63	3,18	41,21
Lu	7060	2,84	2,97	39,41
Fr	1810	0,12	0,65	39,50

Uncertainty

Fortov V., Dremin A.,
Leont'ev A.
High. Temp. 13, (1975)



Fortov V., Khrapak A.,
Yakubov I.
Non-Ideal Plasma Physics (2004)

Critical Point Parameters

Recommendations of IPCP RAS-GSI Database

	<i>Pressure</i> P_c kbar	<i>Temperature</i> T_c K	<i>Density</i> ρ_c g/cm ³	<i>Entropy</i> S_c J/g/K
<i>Be</i>	2.87	8877	0.398	13.18
<i>Mg</i>	2.46	3957	0.553	3.789
<i>Na</i>	0.47	2473	0.240	3.281
<i>Zr</i>	9.88	14860	1.634	1.693
<i>Hf</i>	11.74	15810	3.610	0.885
<i>V</i>	9.19	9915	1.631	2.718
<i>Nb</i>	11.06	19180	1.701	2.023
<i>Ta</i>	9.93	13530	4.263	0.923
<i>Cr</i>	9.91	7797	2.660	2.332
<i>Mo</i>	7.59	10180	3.690	1.520
<i>W</i>	11.80	15750	4.854	0.837
<i>Fe</i>	11.31	8787	2.183	2.496
<i>Co</i>	5.55	9157	1.890	2.458
<i>Ni</i>	10.42	7547	2.092	2.518
<i>Zn</i>	3.28	3079	2.381	1.468
<i>Cd</i>	0.87	2510	2.283	0.840
<i>Ag</i>	10.64	7053	3.279	1.118
<i>Au</i>	6.14	8515	6.061	0.624
<i>Re</i>	15.91	18710	6.024	0.824
<i>Ir</i>	13.40	16220	6.061	0.780
<i>Pt</i>	6.21	11430	5.236	0.807
<i>Sn</i>	2.39	8175	1.592	1.123
<i>Pb</i>	2.25	4869	2.937	0.529
<i>U</i>	7.70	9637	4.505	0.727

After
D. Varentsov
FAIR-Russia School
Moscow, 2009

Uranium Critical Point Location

fifty years old problem

$T_c \approx 12'000 \text{ K}$

(the early estimation)

Braut (1957)

List of Uranium
CP parameters
estimations

T_c [K]	p_c , bar	ρ_c g/cm ³	Z_c	ρ_s/ρ_c (*)	References
6'618	4160	4.12	0.437	4.60	Young D. (1977)
7'533	798	1.03	0.295	18.4	Gates D. et al. (1960)
6'200–7'663	-	-	-	-	Goldstein R.(1989) Hess H.(1995)
8'317–9'112	-	-	-	-	Guldberg C. (1890).
8'730	2360	5.17	0.150	3.67	Martynyuk M. (1989).
9'000 ($z_c = 3$)	5000	2.60	0.6	7.42	Likalter A. (1997)
9'400 ($z_c = 3$)	6000	2.59	0.706	7.32	Likalter A. (1985 - 1996)
11'630	6110	5.30	0.284	3.58	Fortov V. et al. (1975)
11'679–12'995	-	-	-	-	Kopp I. (1967) Lang G. (1977)
12'400	4800	3.55	0.312	5.34	Morris E. (1964)
12'434	4950	3.78	0.302	5.02	Gathers-Shaner-Young (1974)
12'500	-	-	-	-	Grosse A. (1961)
13'034	5136	4.03	0.280	4.71	Hornung K. (1975)
13'043	8'487	5.17	0.361	3.66	Young D. & Alder B. (1971)
9'636	7'700	4,50	Wide-range EOS		Bushman - Lomonosov
7'000	1'712	3,30	Extrapolation of Liquid $\rho(T)$		Apfelbaum - Vorob'ev (2009)
5'500– 6'500	100 – 1'000	(estimation – thermal EOS calibration)			Iosilevskiy (1990)
6'840	4'440	(Plasma model – thermal calibration)			Iosilevskiy & Gryaznov
12'800	8'450	(Plasma model – caloric calibration)			SAHA-code, JNM, (2005)

Table from:
H. Hess,
H.Schneidenbach
Z. Metallkd. (2001)
Vapor Pressure and
Critical Data for
Uranium

Additions IL

Uncertainty in high-temperature density-temperature diagram

Shpilrain *et al.* (1988)
 Static experiment: - ρ_{liquid} ($1400 < T < 2100$ K)
 high accuracy - 0.5 %

Thermal variants T

Yound & Shaner (1977)
Gathers et al. (1986)
Iosilevskiy (1991)
Iosilevskiy & Gryaznov (SAHA-T)

Correlation
 $T_c \Leftrightarrow$ Thermal expansion

Caloric variants U

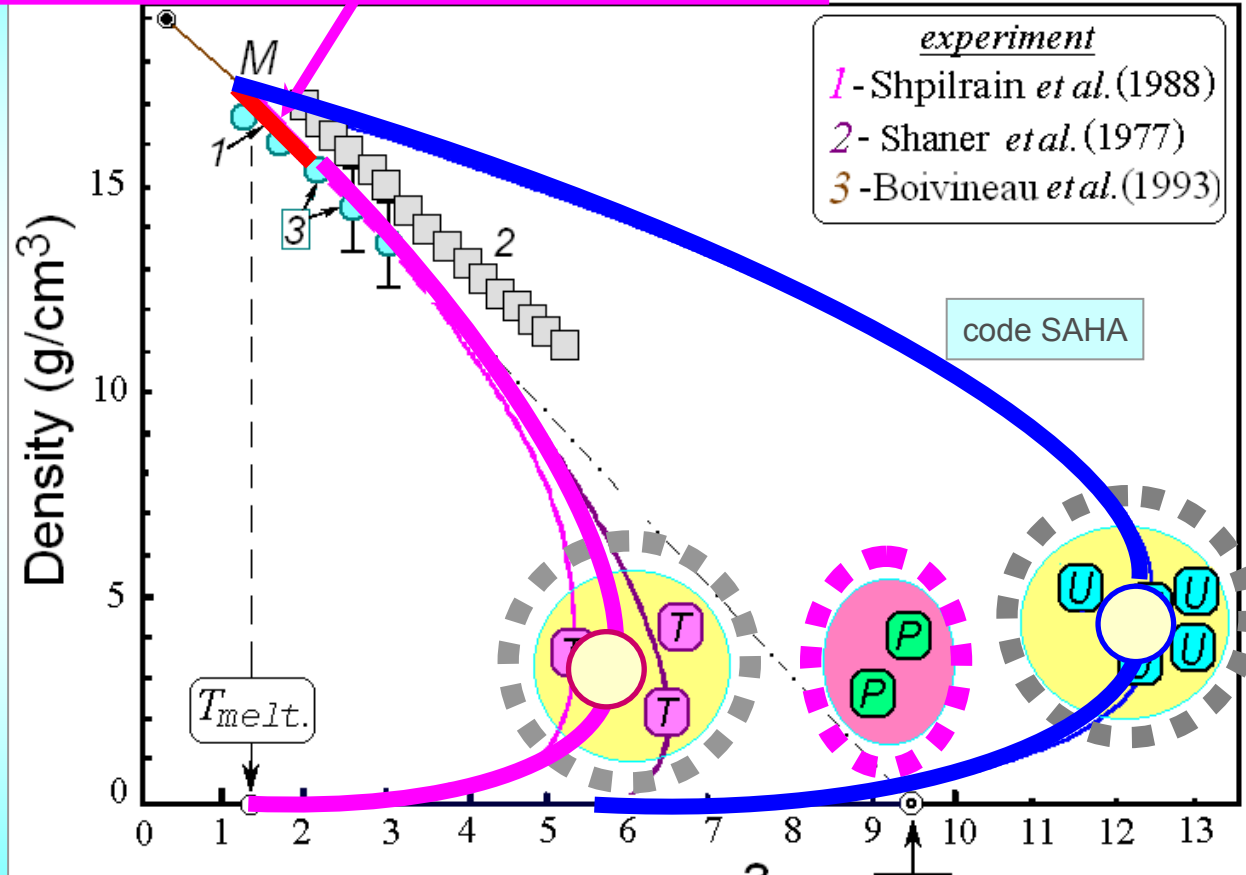
Brout (1957)
Grosse (1961)
Morris (1964)
Yound & Alder (1971)
Gathers et al. (1974)
Fortov et al. (1975)
Hornung (1975).....
Iosilevskiy & Gryaznov (code SAHA-E)

Correlation
 $T_c \Leftrightarrow$ Vaporization heat

Plasma Hypothesis P

Likalter (1981)
Likalter + Hess (1997)

Correlation
 $T_c \Leftrightarrow$ Ionization Potential



Extrapolation: (i) Guggenheim's formula, (ii) Law of Correspondent States (Iosilevskiy, 1990), (iii) SAHA-code (Iosilevskiy & Gryaznov, 2005)

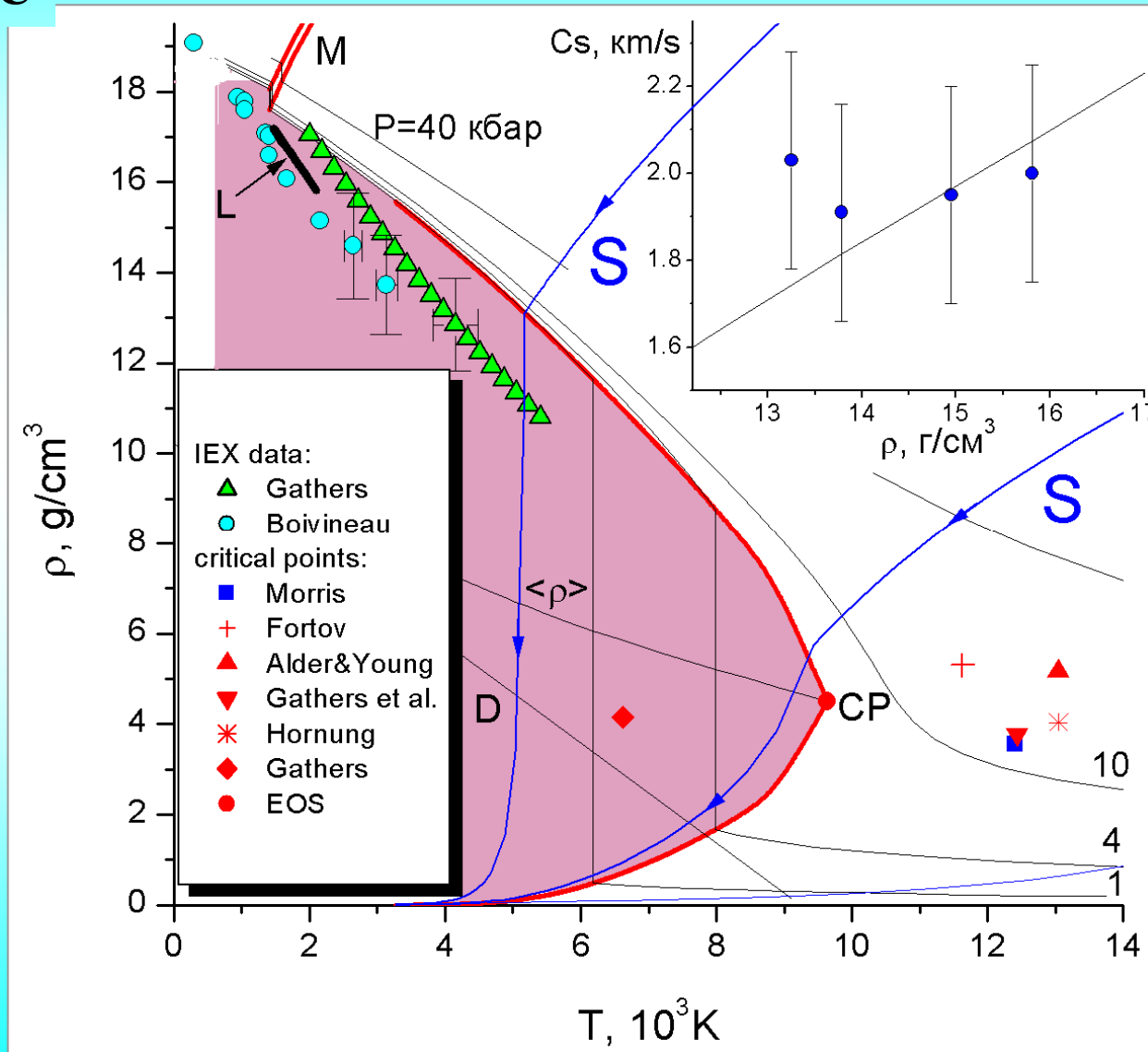
(i) $\frac{\rho_{l,v}}{\rho_c} = 1 + b_1(1 - \frac{T}{T_c}) \pm b_2(1 - \frac{T}{T_c})^\beta$

(ii) $\left(\frac{\rho(T/T_{cr})}{\rho_{cr}}\right)_U = \left(\frac{\rho(T/T_{cr})}{\rho_{cr}}\right)_{Cs}$

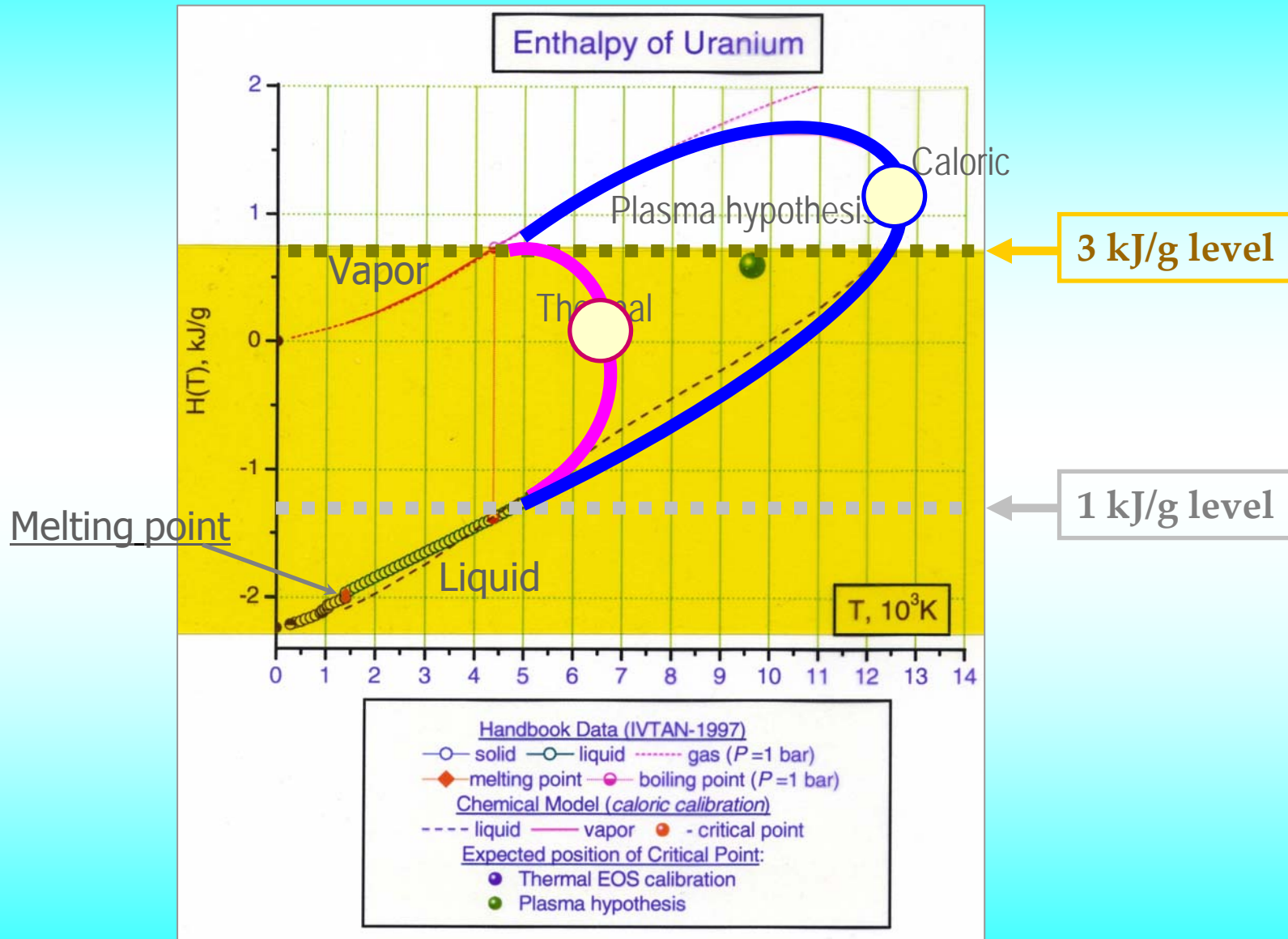
Wide-Range Equation of State

(after A. Bushman and I. Lomonosov)

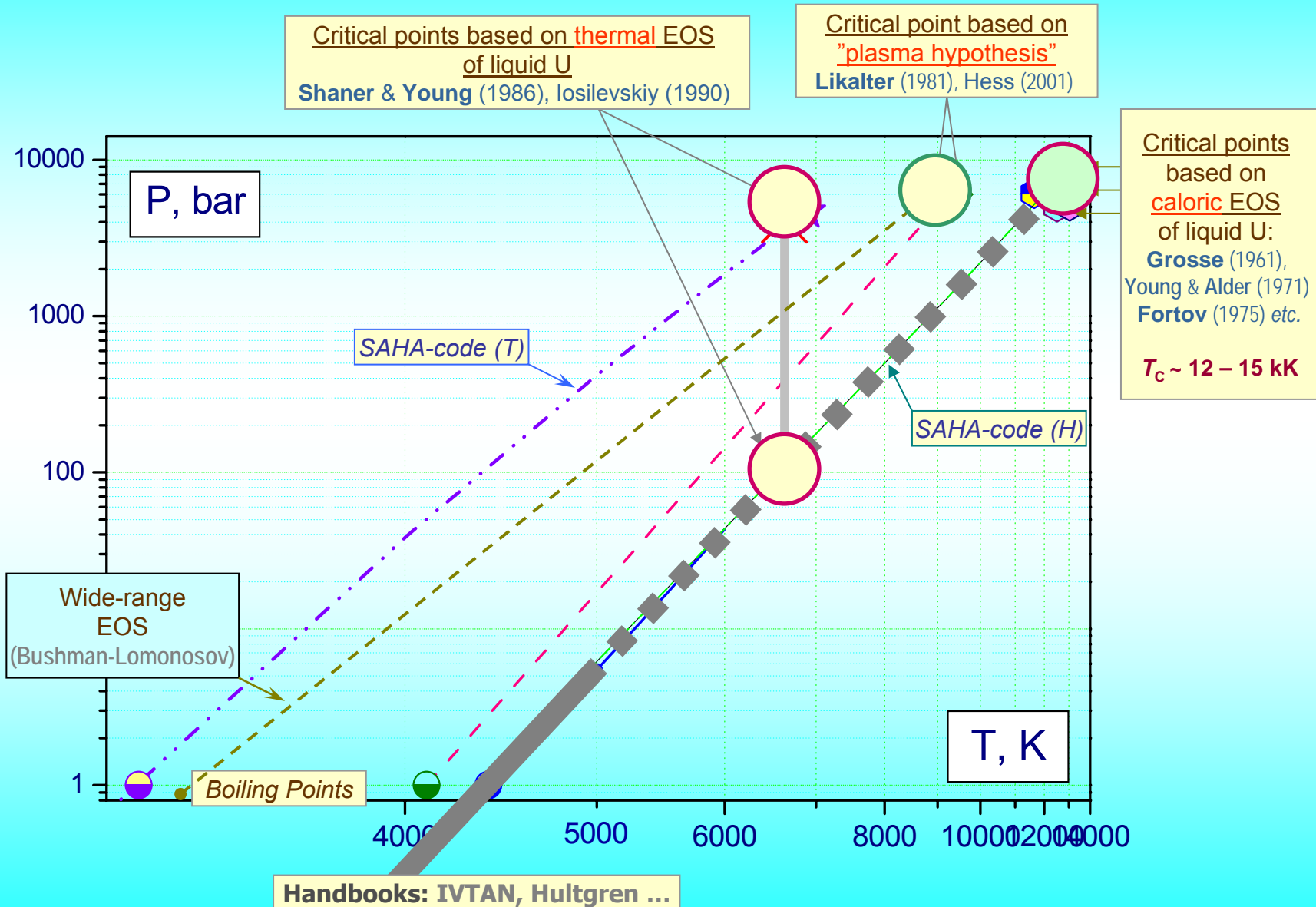
U



Uncertainty in high-temperature caloric phase diagram



Uncertainty in Uranium Critical Pressure



The Problem:

Experimental data:

Liquid density $\rho(T)$ - ($T < 5$ kK)
Liquid enthalpy $H(T)$ - ($T < 5$ kK)
Vapor pressure $P_s(T)$ - ($T < 5$ kK)

Semi-empirical rules:

Convexity of liquid density $\rho_L(T)$
Quasi-linear vapor pressure $\ln P_s(T^{-1})$
Universal evaporation enthalpy $\Delta H(T)$

Incompatible

Dilemma:

Access semi-empirical rules
and
Deny experimental data

or

Access experimental data
and
Violate semi-empirical rules

Hypothetical resolution ?

In Search *of* Resolution for Uranium Critical Point Location Problem

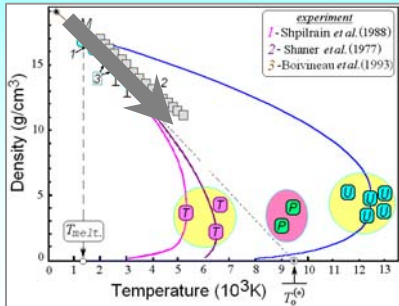
Low temperature

? Wrong experiment

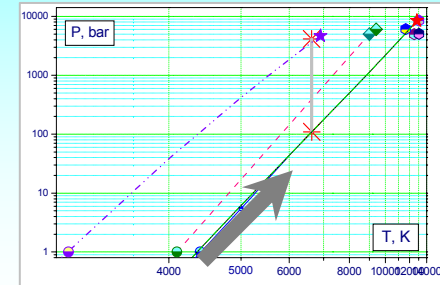


High temperature

wrong extrapolation ?



liquid density $\rho(T)$ - ?
liquid enthalpy $H(T)$ - ?
vapor pressure $P_s(T)$ - ?



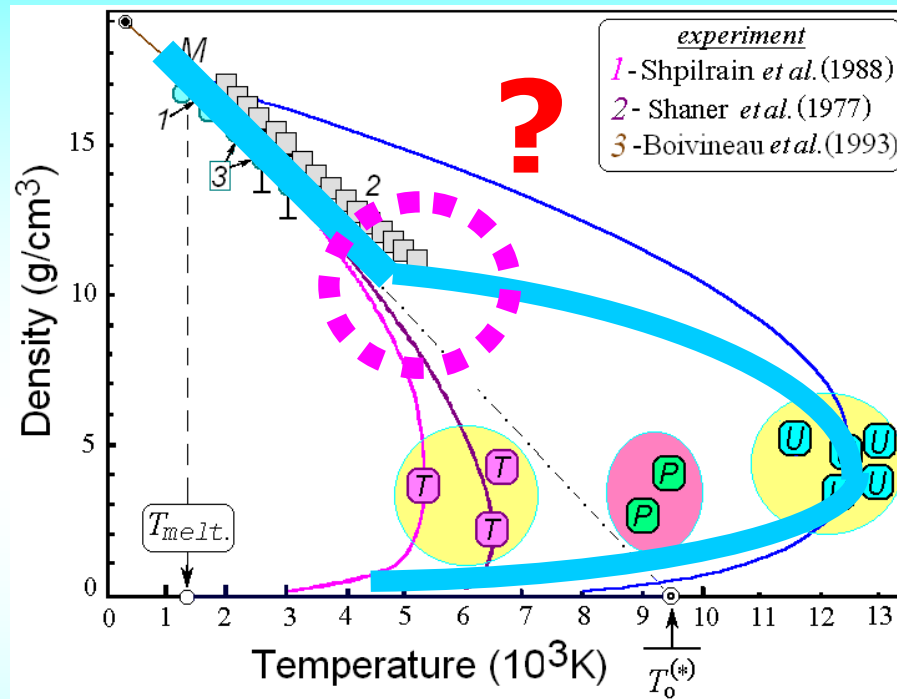
! unbelievable !

We have enough reason to expect

Violation of semi-empirical rule(s)

Hypothetical violation of semi-empirical rule(s)

Lost of convexity for $\rho_L(T)$?



? What physical reason can approve this violation ?

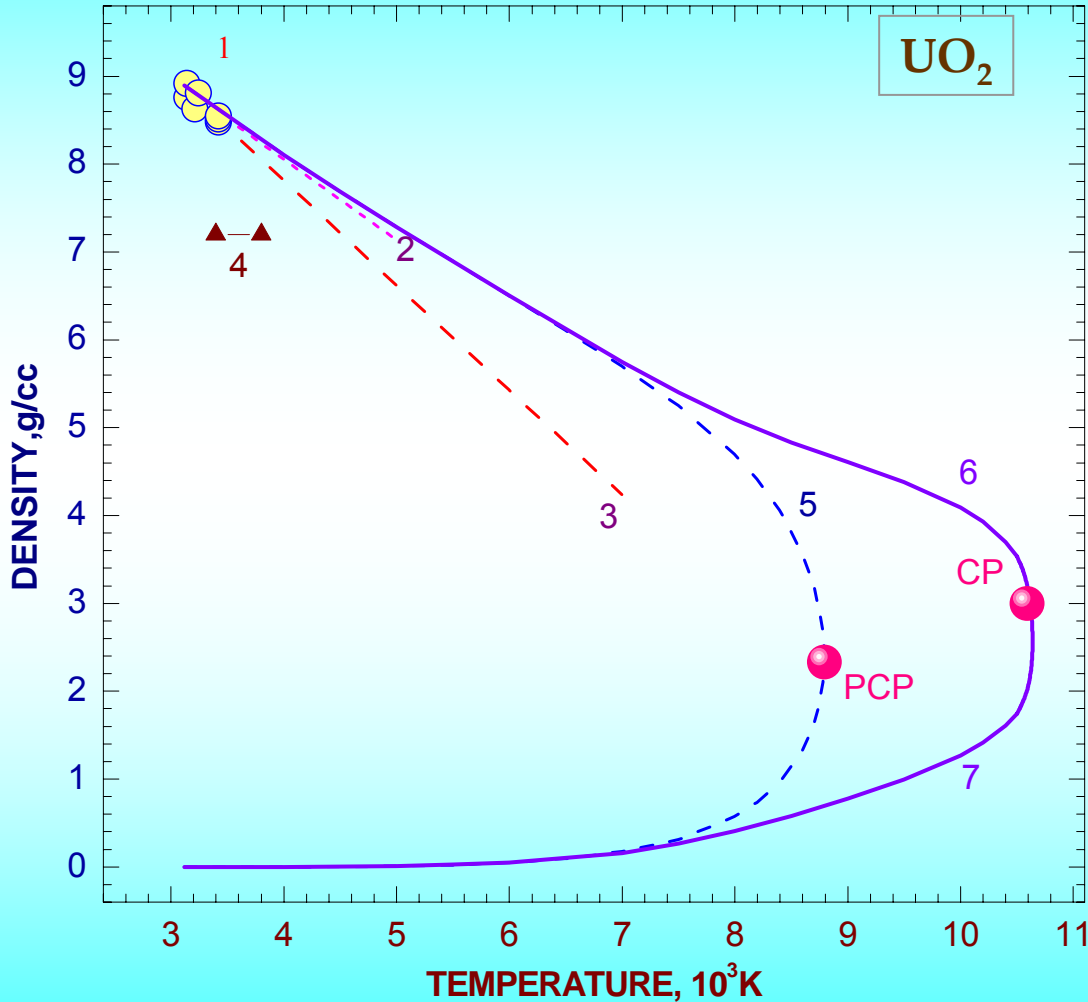


Exception

Non-congruent evaporation !

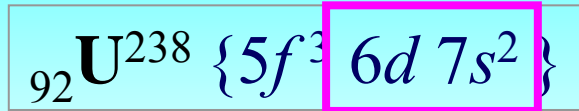


Elbrus_1990-2010



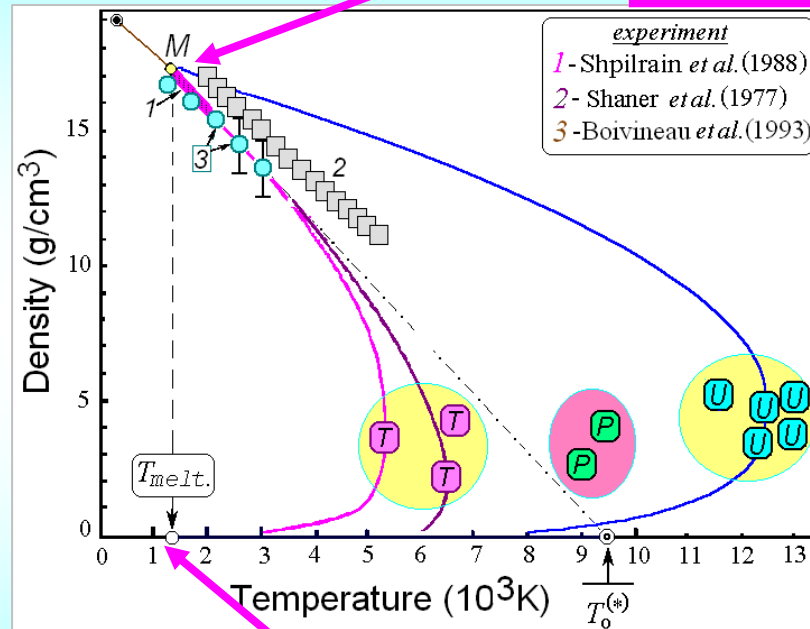
Iosilevskiy I, Hyland G., Ronchi C., Yakub E. "An Advanced Equation of State of UO₂ up to the Critical Point"
 - Trans. Amer. Nucl. Soc. **81** 122 (1999)// - Int. Journ. Thermophys. **22**, 1253 (2001)// Contrib. Plasma Phys. **43**, (2003)

What physical reason can approve the lost of convexity for $\rho(T)$ two-phase boundary in uranium ?



$Z = 3$

$\text{U}^{+3} + 3e^-$
 $\Gamma_D \approx 3300 // n_e \lambda_e^3 \approx 900$



$\text{U}^{+1} + e^-$
 $\Gamma_D \approx 80 // n_e \lambda_e^3 \approx 10$

$Z = 1$

$Z = 0$

NB !

Drastic **change** of **effective ion-ionic interaction** during thermal expansion of liquid uranium and decrease of electronic degeneracy

Drastic **change** of **phase behavior** of evaporating uranium (ρ - T diagram)

Uranium Critical Point Location Problem

General nature of the problem

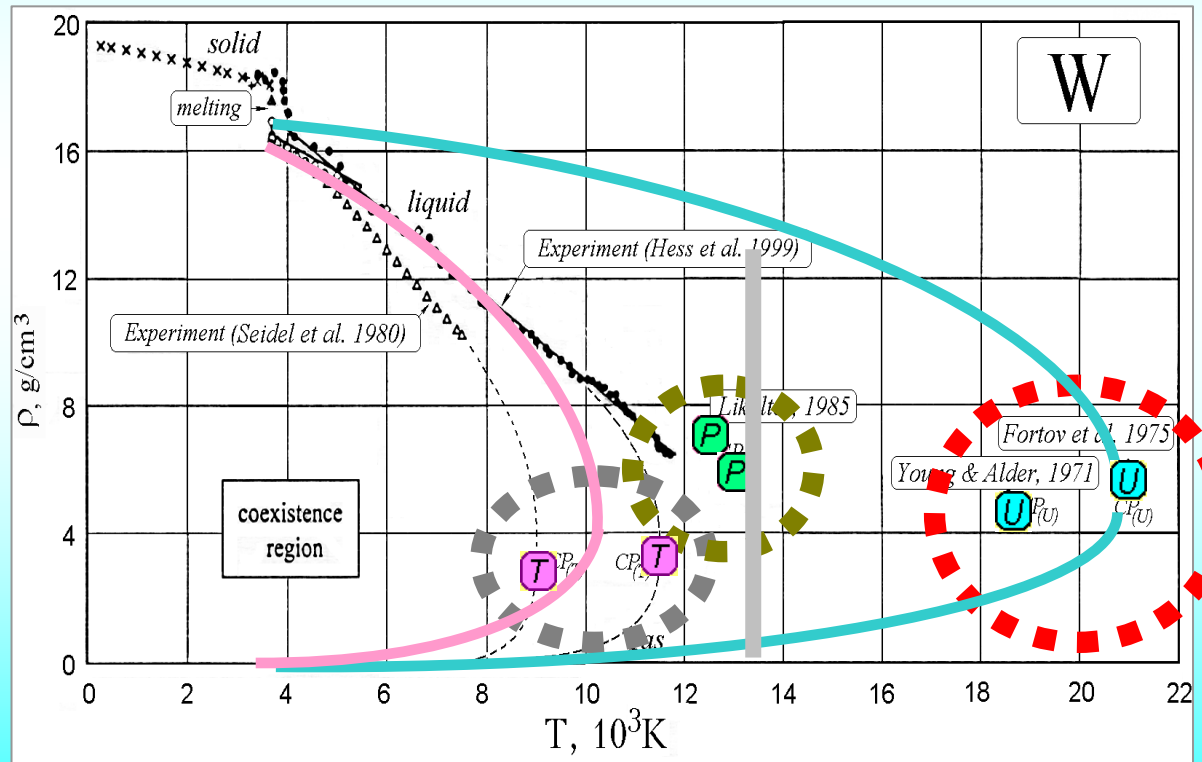
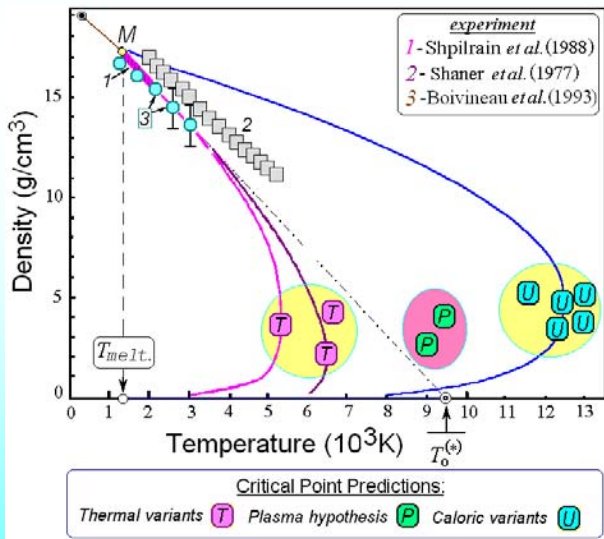
Discussion:

I. Iosilevskiy & Int. Conference: Subsecond Thermophysics, Moscow, 2008// FAIR-Russia School, Moscow, 2009

Uranium critical point problem – is **NOT** exception, it is a general rule !!

Density-Temperature Diagram of Tungsten

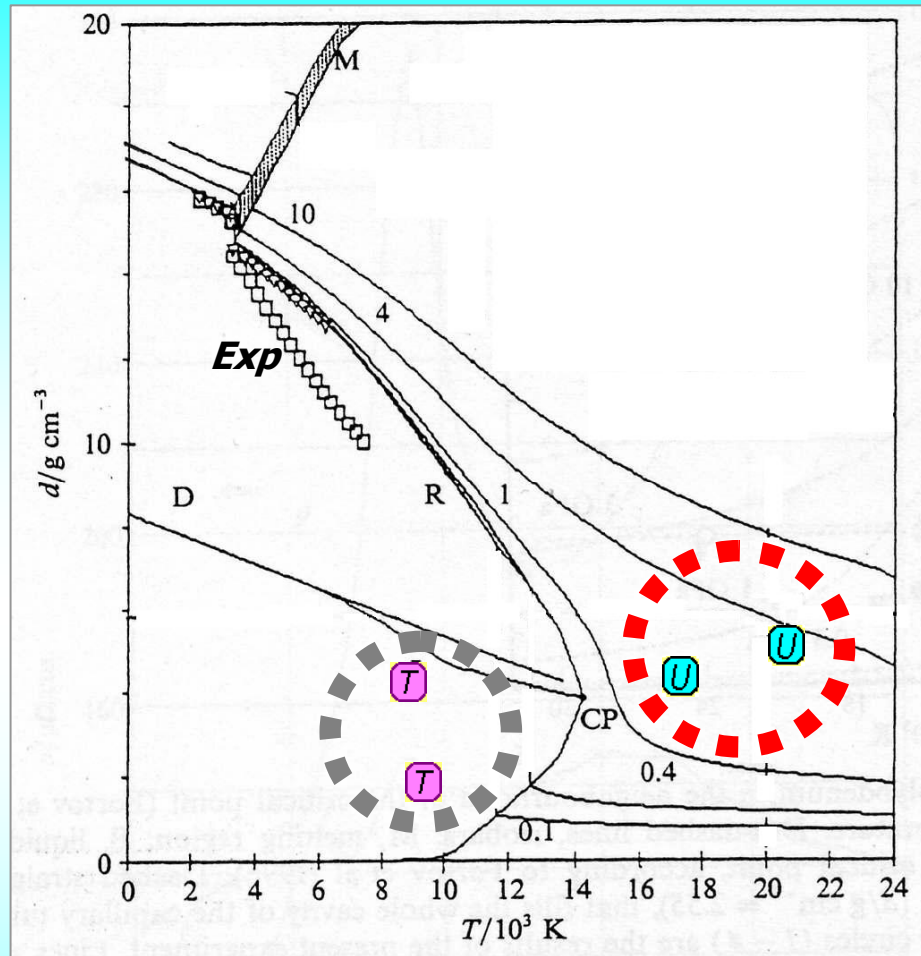
Uranium



Estimations of critical point of tungsten

– **Experimental** data on critical temperature [Ternovoy *et al.* High T-High P, V.34, (2002)] $T_C \sim 13'660 \pm 800\text{K}$

Density-Temperature Diagram of Tantalum



EOS of Tantalum [Fortov, Rakhel, Lomonosov et al. 1996]

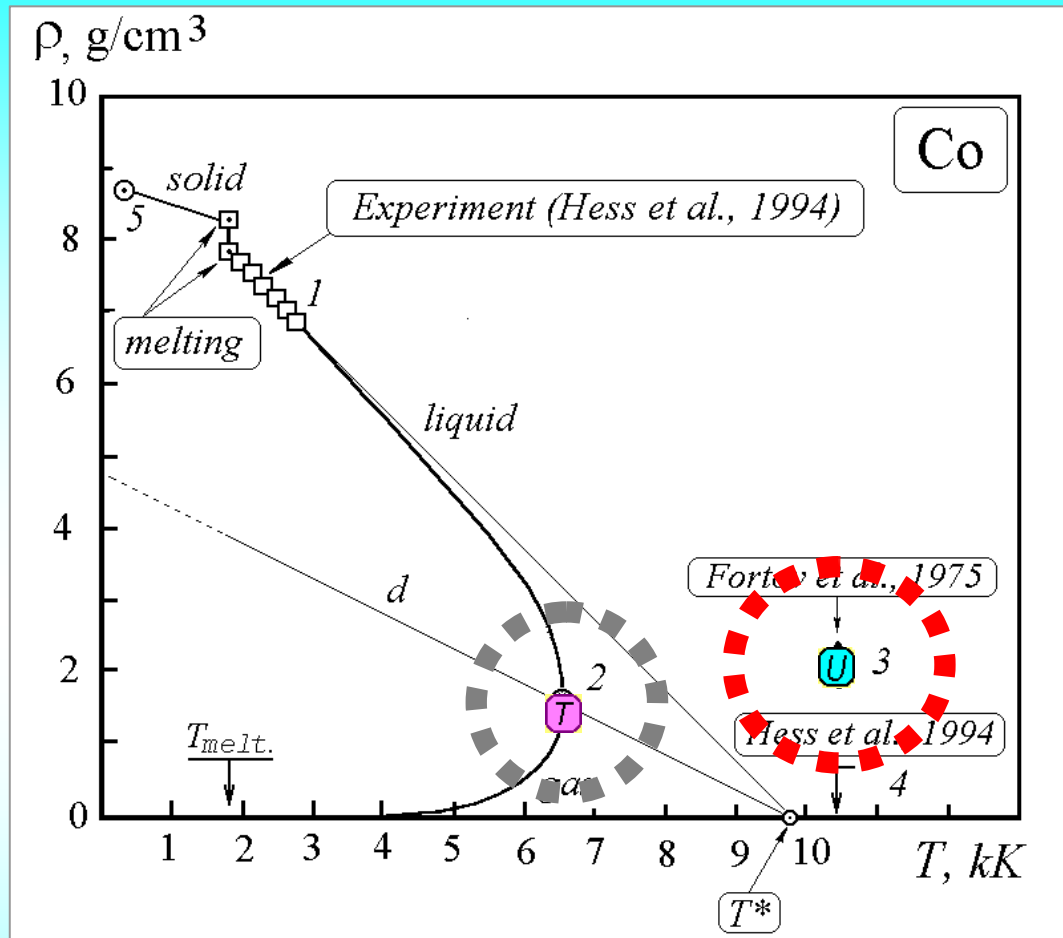
Exp – experimental data (exploded wire technique);

M – melting; *D* – “diameter of two-phase boundary (wide-range EOS);

U – critical point **estimations** based on **heat of vaporization** // Alder & Young (1971) // Fortov et al.(1975) //

T – critical point **estimations** based on **thermal** expansion of liquid Ta // Young et al.(1977) // Levashov ,*Ph.D Thesis*/

Density-Temperature Diagram of Cobalt



- 1 – **Experimental** data on density of liquid cobalt [Hess H., Kaschnitz E., Pottlacher G., (1994)];
 T_{M} – melting temperature of cobalt ($\cong 1768 \text{ K}$);
 T^* - linear extrapolation “to-zero” of liquid density $\{T^* \equiv T_{\text{M}} + \alpha^{-1}(T_{\text{m}})$; $\alpha(T)$ – thermal expansion coefficient};
 d – “diameter of two-phase boundary $\{d \equiv (\rho_{\text{liq}} + \rho_{\text{gas}})\}$ };
 2 – reconstruction of two-phase boundary and critical point estimation based on the data [1];
 3 – critical point estimation [2] based on **heat of vaporization**; 4 - critical temperature estimation of H. Hess *et al.* [1] from correlation of T_{C} with ionization potential (“Plasma” Hypothesis /A. Likalter [3]/);

Uranium Critical Point Location Problem

In search for possible resolution of the problem:

(A) **Wrong experiments** (!?)

- Shaner *et al.*-1977 // Shpilrain *et al.* -1987 // Mulford & Sheldon -1991 // Boiveneau *et al.*-1993....

(B) Anomalous (**non-convex**) form of density-temperature coexistence curve

(C) Anomalous large **upward non-linearity** of saturation curve in $(\ln p_S - 1/T_S)$ coordinates

(D) Anomalous **extra-low value** of the **critical compressibility factor** of uranium, $(pV/RT)_c \ll 1$.

(E) Anomalous **extra-low value** of the uranium **ratio** of **critical to normal** densities, $(\rho_c / \rho_0) \ll 1$.

Discussion:

I. Iosilevskiy & V. Gryaznov, Journ. of Nuclear Materials, 344 (2005)

Uranium Critical Point Location Problem

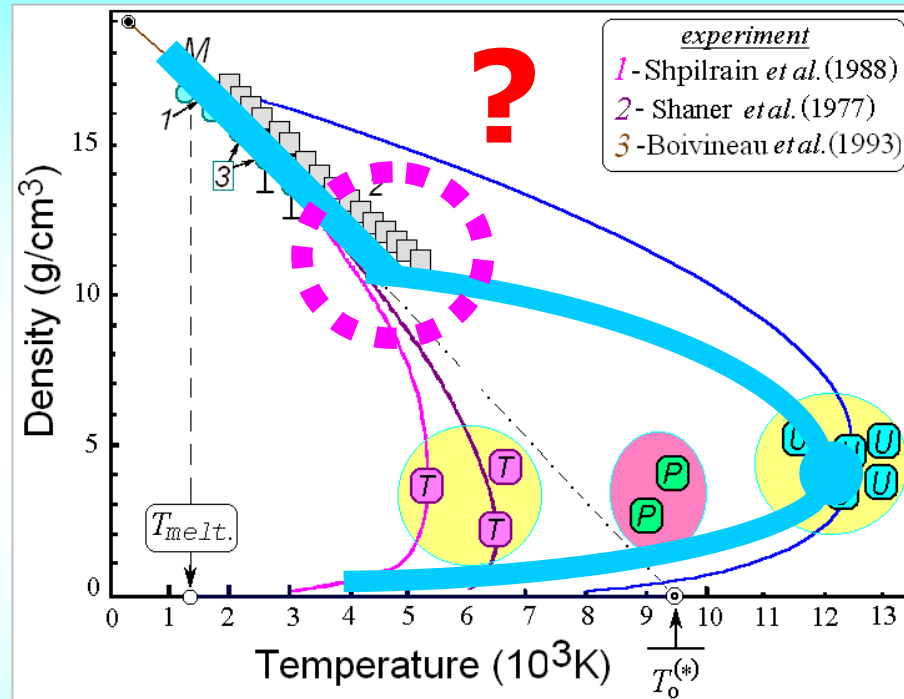
Hypothetical resolution - I

Discussion:

I. Iosilevskiy & Int. Conference: Subsecond Thermophysics, Moscow, 2008// FAIR-Russia School, Moscow, 2009

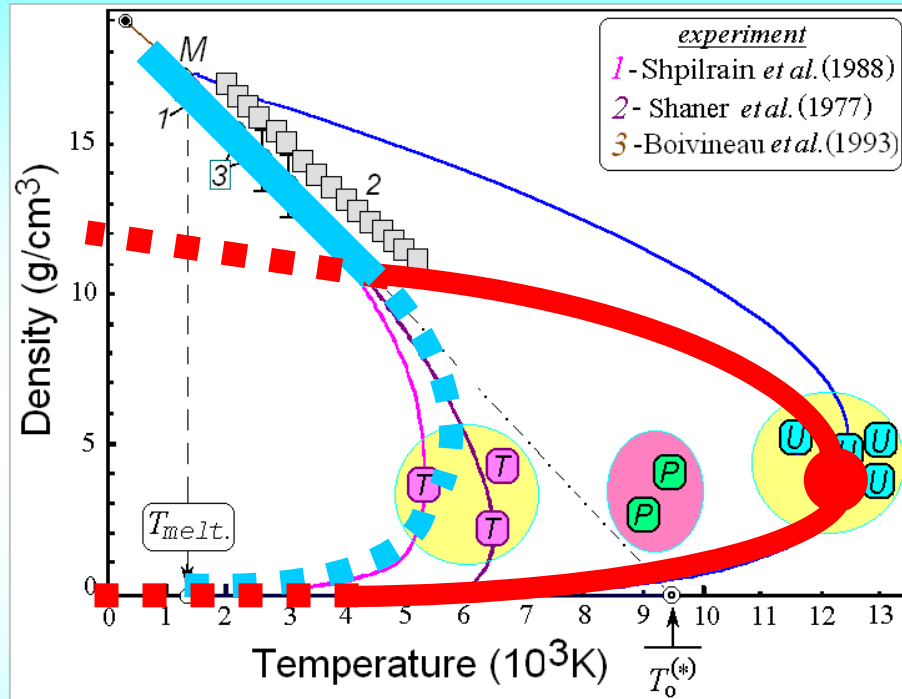
Hypothetical resolution of uranium critical point location problem

? ρ - T phase boundary consists of TWO fragments ?



Hypothetical resolution of uranium critical point location problem

? ρ - T phase boundary consists of TWO fragments ?



«high-density phase» \leftrightarrow «low-density phase»

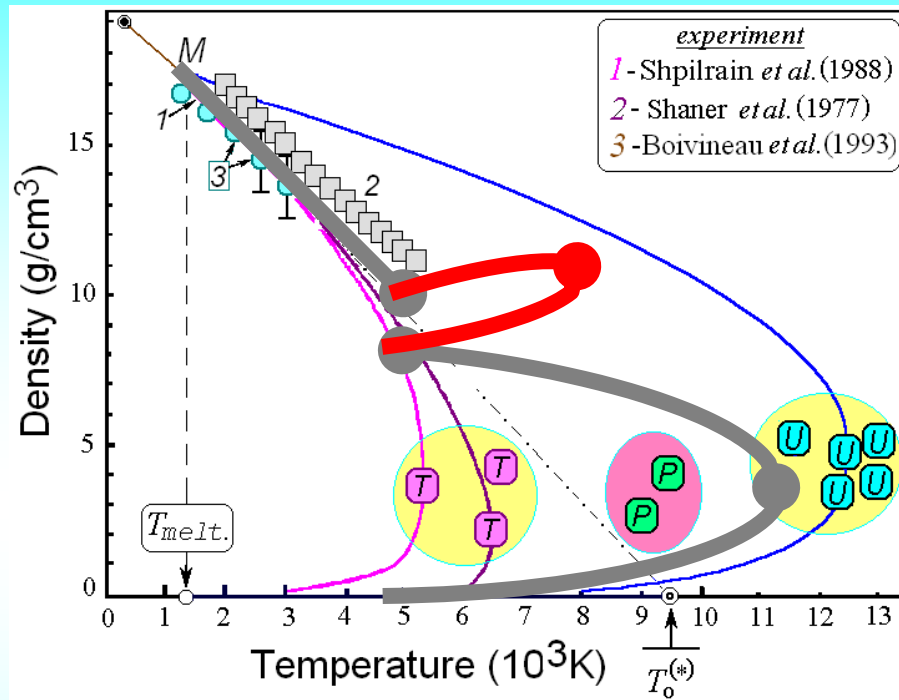
More exotic and hypothetical

Hypothetical resolution - II

Transition from the “high-density phase”
to the “low-density phase”
must not be continuous

Additional phase transition ?

Additional phase transition ?



Very exotic but not fantastic !

! Only 1 – 2 kJ/g is needed !

Fundamental Physical Problem

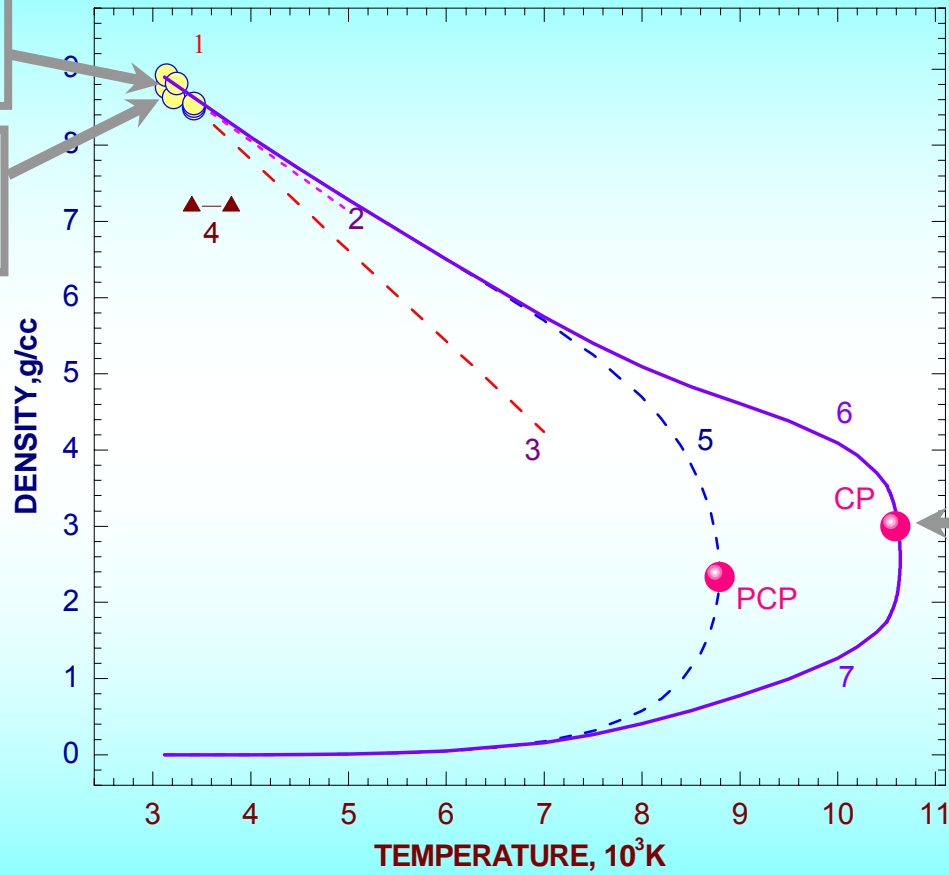
**Phase transition in a system
with
varying mean-particle interaction**

Non-congruent evaporation in $\text{UO}_2 \rightleftharpoons$ the same physical problem

Phase transition in the system with varying composition

Molecular model
 $\text{UO}_2 \rightleftharpoons \text{UO}_2$

Ionic model
 $\text{U}^{+4} \rightleftharpoons \text{O}^{-2}$



Ion-molecular mixture
 $\text{U} \rightleftharpoons \text{O} \rightleftharpoons \text{UO}$
 $\text{UO}^+ \rightleftharpoons \text{UO}_2^+$
 $\text{O}^- \rightleftharpoons \text{UO}_2^- \rightleftharpoons \text{UO}_3^-$

Phase transition in the system
 with varying mean-particle interaction !

Fundamental Physical Problem

What could we do ?

- Study via simplified analytical plasma models

One-component plasma model on uniformly-compressible compensating background {OCP(~)}

- Study via direct numerical simulation

FT-DFT_MD // Monte-Carlo //

- Experimental study:

Exploding wires, *etc*

Heavy Ion Beam

Surface Laser Heating

HIB for thermophysical investigations

– How to arrange HIB energy deposition

Priorities

- Uniformity of heated material
- Thermodynamic equilibrium

– How to arrange measurements

Priorities

- Direct measurement of thermodynamic parameters
without intermediate hydrodynamic re-calculations
- Energy deposition control

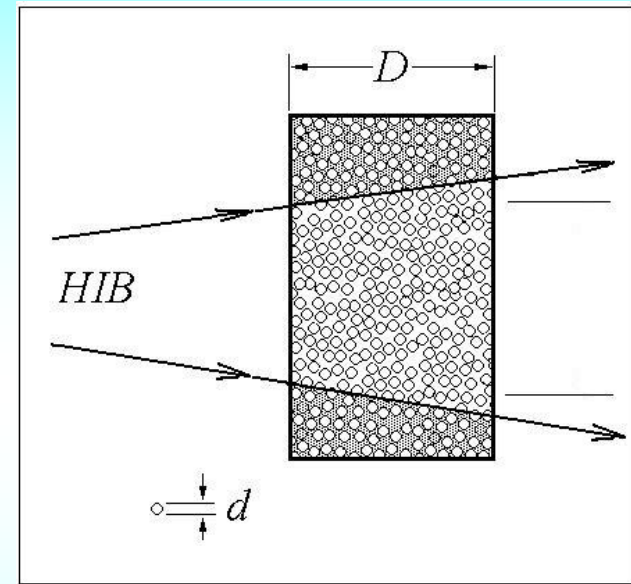
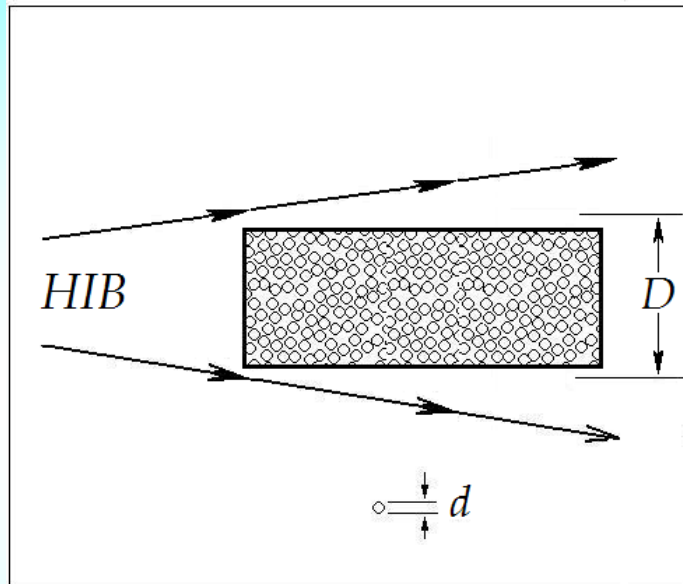
**HIB heating of highly dispersive materials –
– very promising for thermophysical investigations (*)**

* Iosilevskiy I. // Int. Conf. *Intense Ion Beam Interaction with Ionized Matter* // Moscow, ITEP Publishing (1999)

* Iosilevskiy I., Gryaznov V. // XIV Int. Conf. *Heavy Ion & Inertial Fusion* // Moscow, ITEP Publishing (2002)

How to arrange HIB energy deposition

- HIB heating of highly dispersive porous materials –**
– very promising for thermophysical investigations (*)



Advantages:

- uniform quasi-free equilibrium expansion of each grain
- no fast hydrodynamic movement
- surface thermodynamic parameters are equal to the bulk ones
- porosity (ρ_{00}/ρ_0) is well-controlled parameter

* Iosilevskiy I. // Int. Conf. *Intense Ion Beam Interaction with Ionized Matter* // Moscow, ITEP Publishing (1999)

* Iosilevskiy I., Gryaznov V. // XIV Int. Conf. *Heavy Ion & Inertial Fusion* // Moscow, ITEP Publishing (2002)

Moment "X"

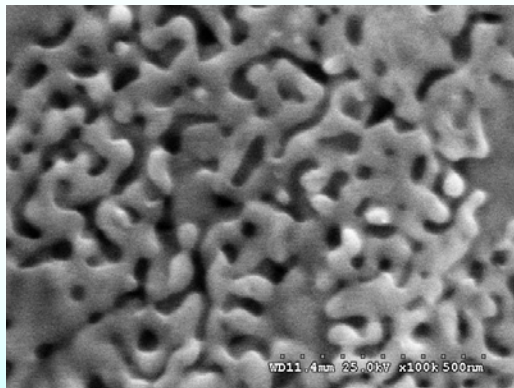
* * * * *

Basic idea

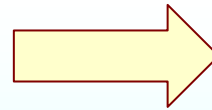
Critical event – Exhausting of free volume for grain's thermal expansion

!! In this moment we obtain:

- **Uniform** and **homogeneous** state of investigated material
- **Known density** (*due to porosity and initial density control*)



|← 1 μm →|



!! It means :

- **End** of free quasi-**isobaric** expansion
- Fast **increasing** of bulk **pressure**
- **Start** of stressed quasi-**isochoric** expansion

Pressure Jump

Moment "X"

* * * * *

Pressure Jump

!! If we catch this moment

and if we know:

- temperature (*surface*)
- energy deposition (*beam control*)
- density (*porosity control*)

!! We obtain:

- Density of expanded liquid $\rho(T)_{liquid}$ (*or* $\rho(H)_{liquid}$)
- Thermal (*or caloric*) expansion coefficient

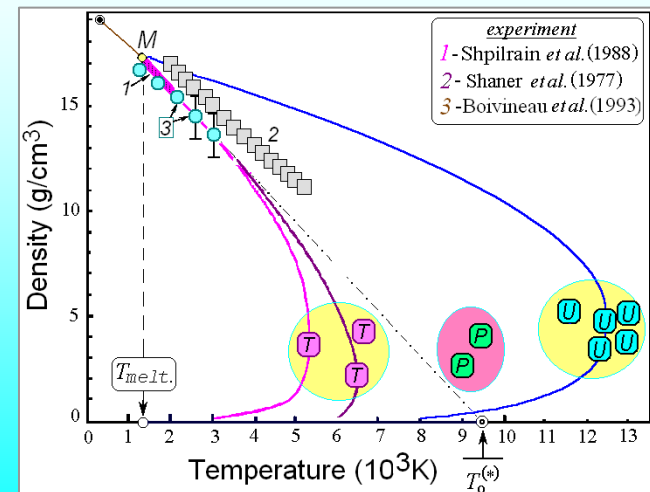
$$\alpha_P = (\partial\rho/\partial T)_P \quad \alpha_P^* = (\partial\rho/\partial U)_P$$

* * * * *

- Heat capacity $C_P = (\partial U/\partial T)_P$

!! Hypothetically:

- sound speed,
- vapor pressure,
- electro-conductivity ... *etc.*



Quasi-static heating of a stack target

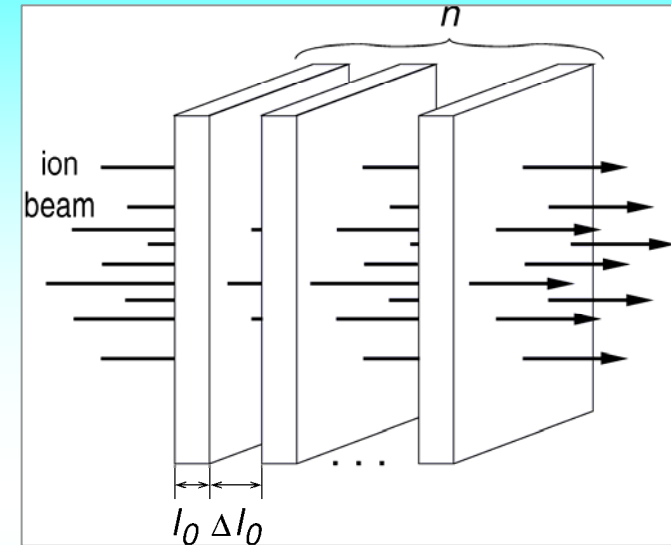
$t < t_x$: quasi-static heating of a stack of foils

• ρ, P, ε (and T) are spatially uniform $u_l = \frac{1}{2} \alpha l q$

• expansion velocity

l – foil thickness, q – heating rate,
$$\alpha = \left(\frac{\partial P}{\partial \varepsilon} \right) \left(\frac{\partial P}{\partial \ln \rho} \right)^{-1}$$

• kinetic energy $E_{kin} \ll \varepsilon$



Thermal (not hydrodynamic) expansion if the foils are thin:
 $\Delta\rho/\rho \ll 1$ over $t_s = l/c_s$ (sound propagation time)

mean density:
$$\rho_{00} = \rho_0 \frac{l_0}{l_0 + \Delta l_0}$$

at $t = t_x$: $l = l_0 + \Delta l_0$

$t = t_x$: the foils merge, weak shocks are generated

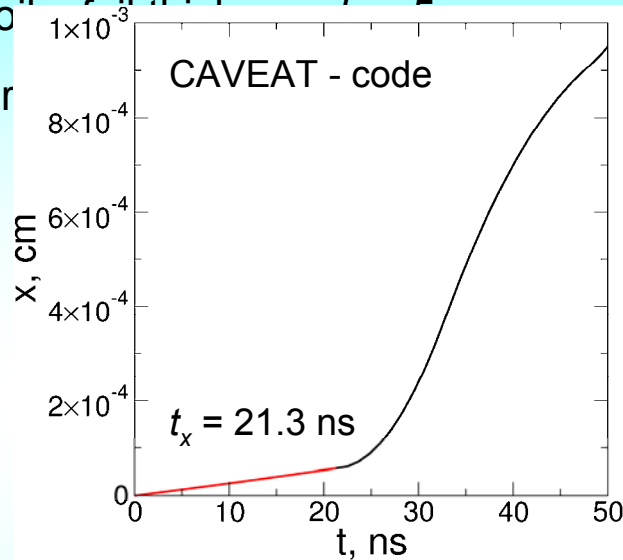
$t > t_x$: expansion velocity is determined by shock hydrodynamics

1D target expansion in planar geometry; t_x can be detected by measuring the surface velocity

Hydrodynamic simulation

foils made of Al: $\rho_0 = 2.7 \text{ g/cm}^3$, $c_s = 5 \cdot 10^5 \text{ cm/s}$, $\Gamma = 2.5$

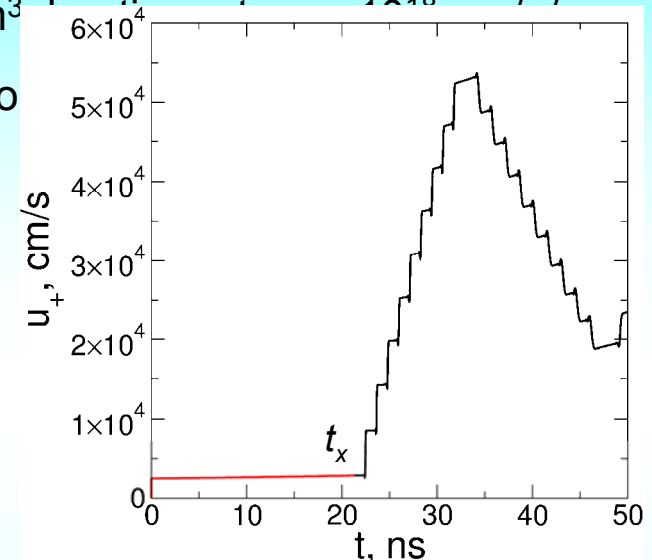
$n = 10$ foils
sound pressure



Position of the target surface

2.2 g/cm^3
expansion

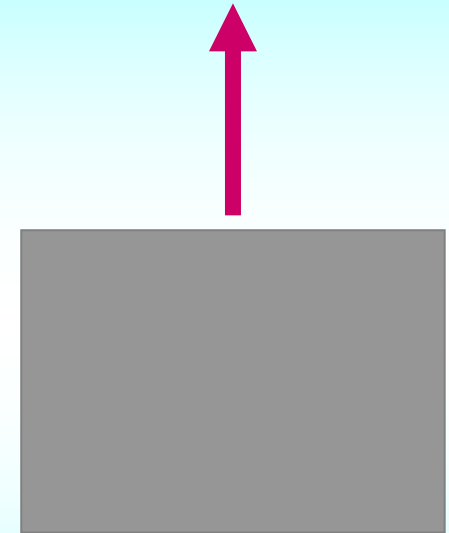
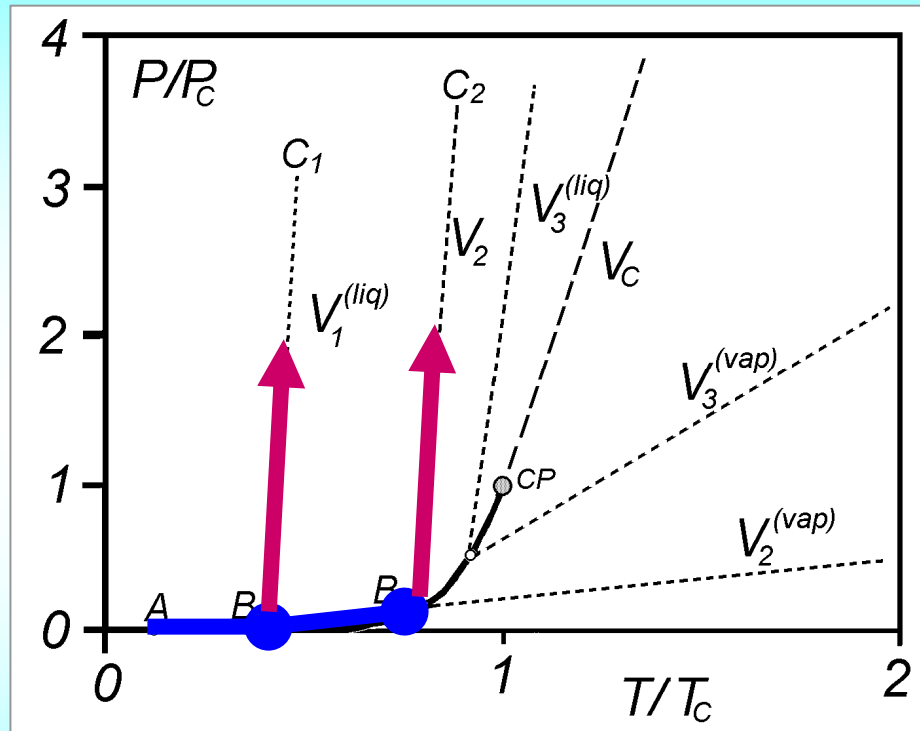
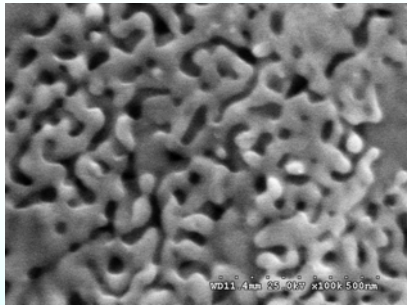
Expansion velocity



The “homogenization” time t_x can be detected by measuring the surface velocity

Thermophysical investigations via HIB
(*novel regimes*)

Tracing of the Boiling Curve



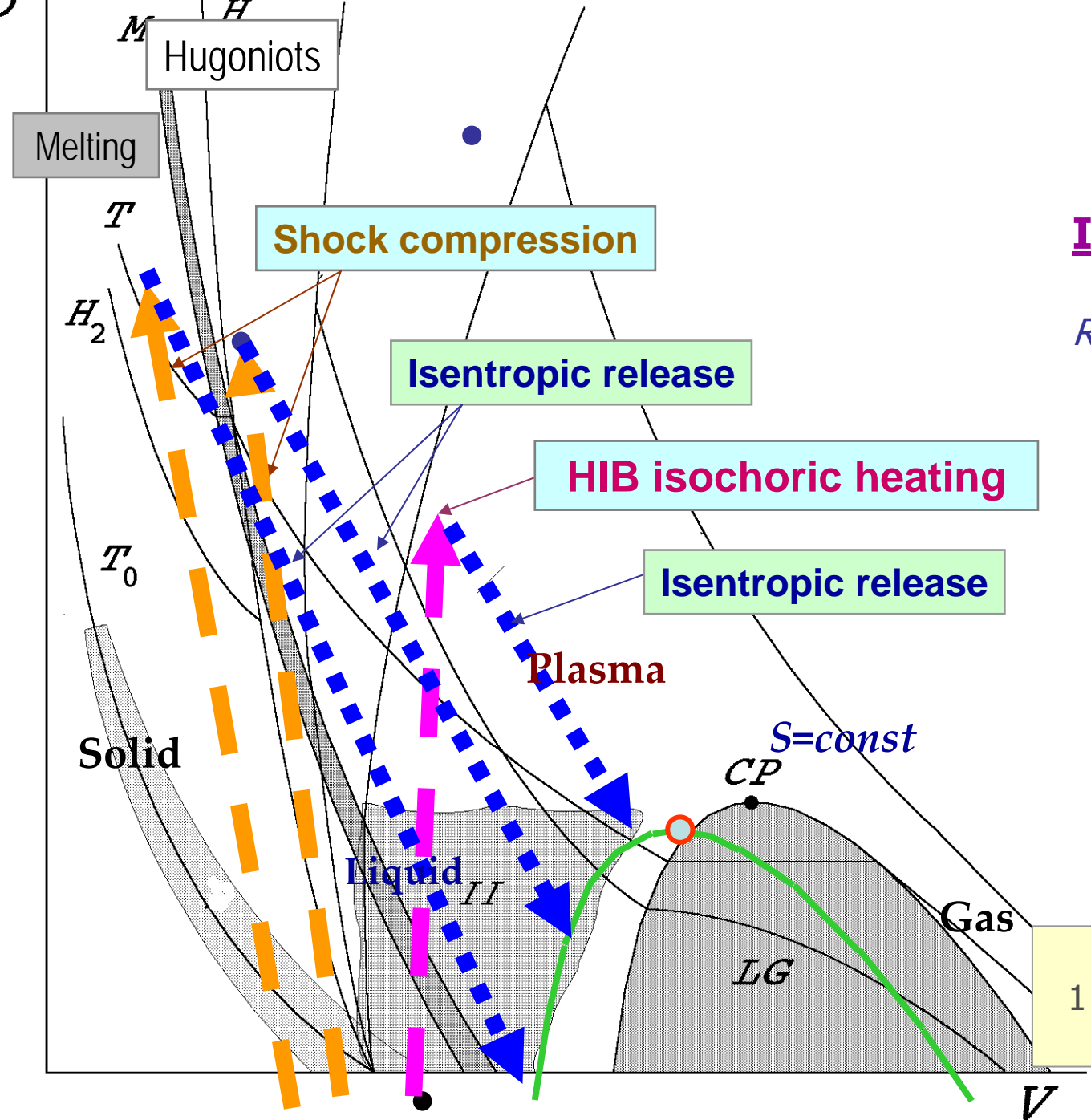
$0 < t < t^*$ – **Quasi-free “isobaric” expansion**



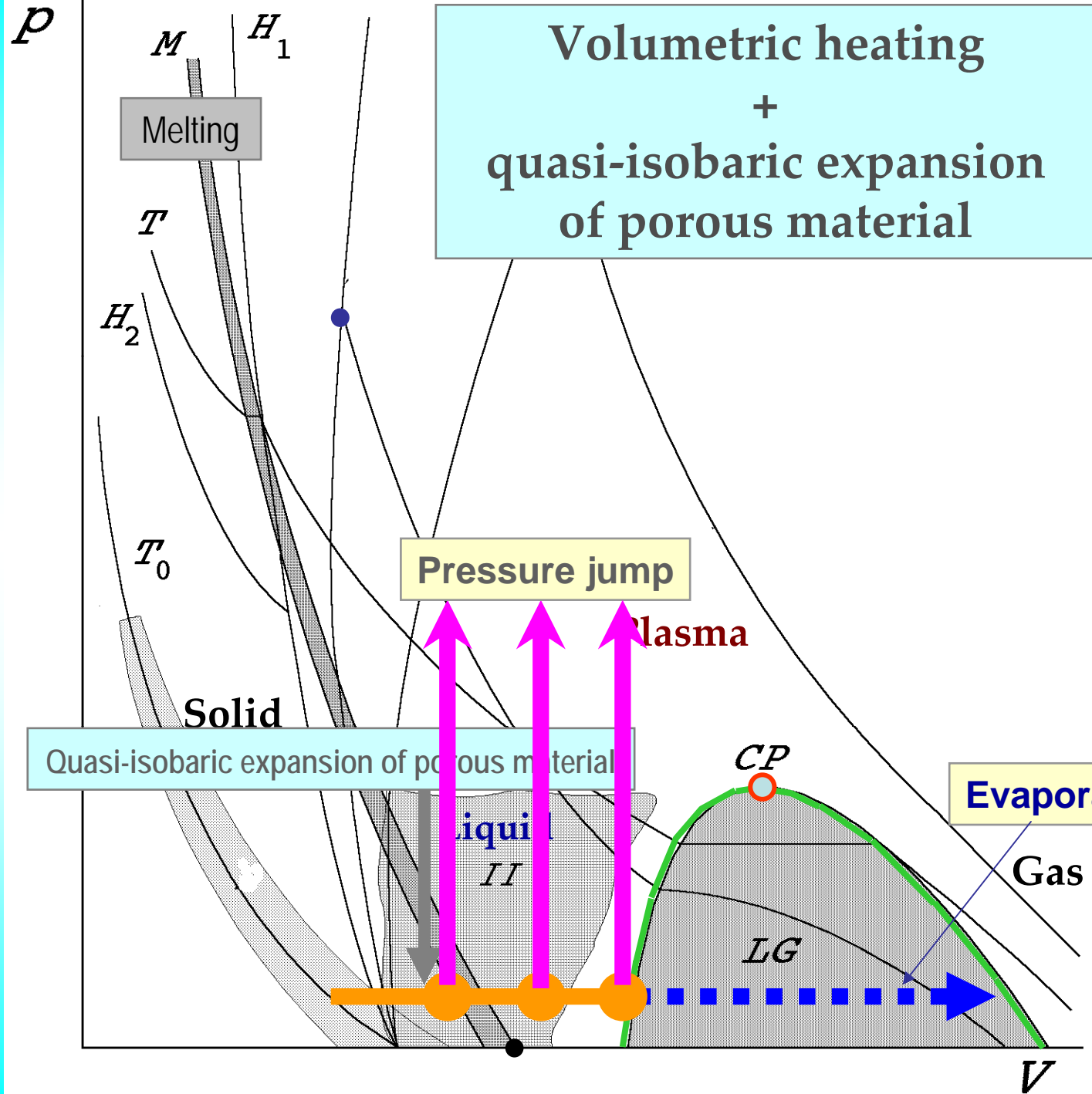
Moment “X”: $\rho = \rho^*$

$t > t^*$ – **Isochoric Heating + Hydrodynamic Expansion**

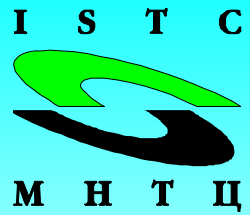
Goal for experimenters: – to catch the *pressure jump moment!*



Uranium
 1 kJ/g ⇔ 5000 K
Crit. Point



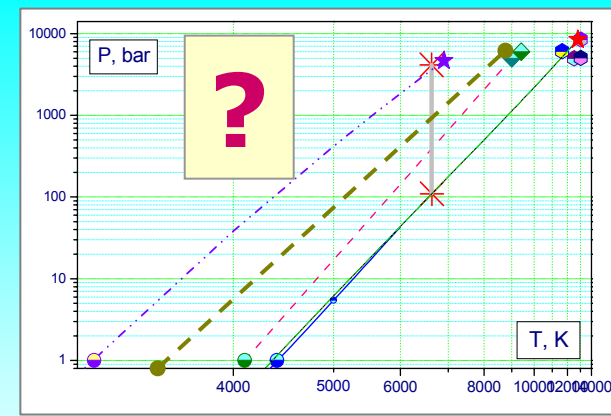
Volumetric heating
+
quasi-isobaric expansion
of porous material



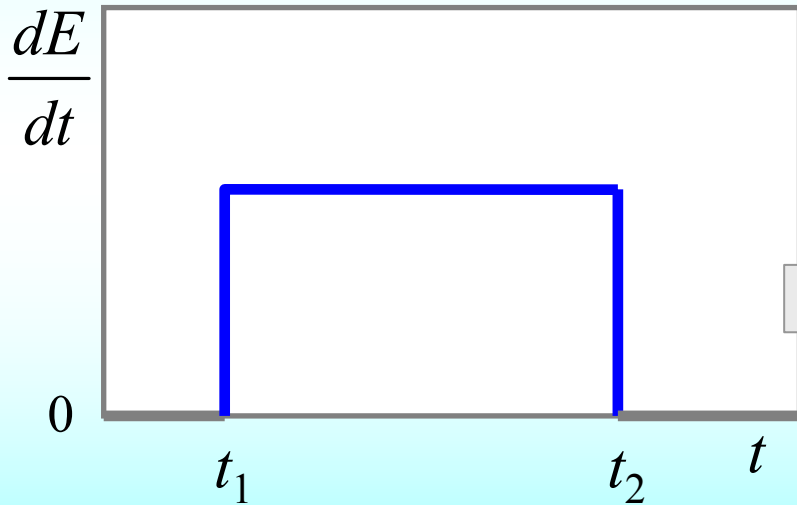
ISTC Project
(perspectives)

Measurement of Uranium Vapor Pressure in Experiment with Surface Laser Heating

$P = \text{const}$ (in buffer gas)

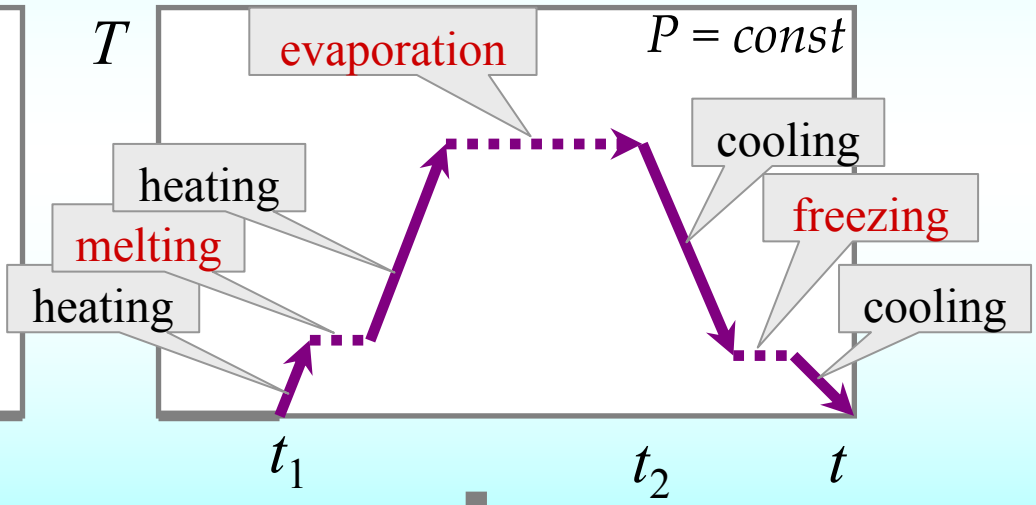


Laser impulse



$P = 1 - 1000 \text{ bar}$

Schematic thermogram



$T_{\text{melt}}(P)$ $T_{\text{boil}}(P)$



Conclusions *and* **Perspectives**

In the case of **uranium** we meet **fundamental physical problem**:

- *Phase transition in a system with mean-particle interaction strongly dependent on density (and temperature)*

- It is **promising** to investigate this problem analytically via **simplified plasma models**

- It is **promising** to investigate this problem in **direct numerical simulation** in frames of ***ab initio*** approaches

- It is **promising** to investigate non-congruent phase transitions **experimentally**. in particular with **intense laser** and **heavy ion** heating

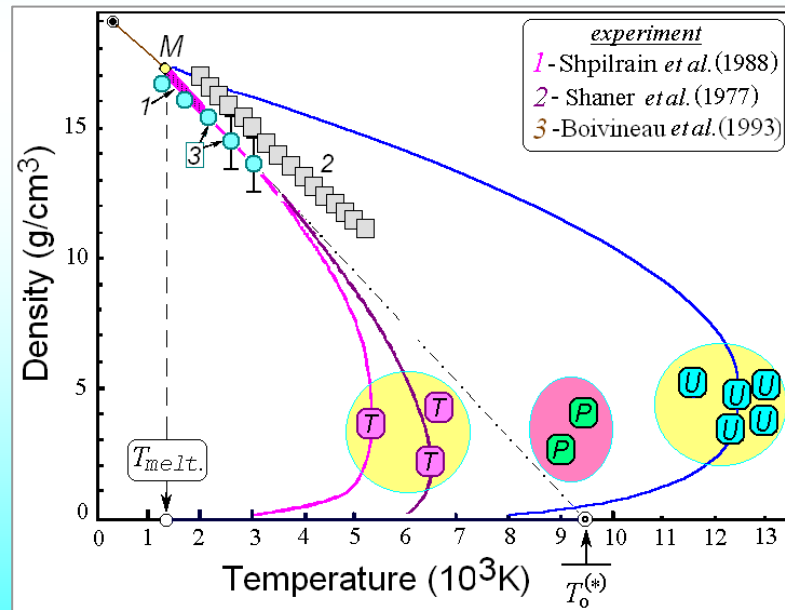
! Only 1 – 3 kJ/g is needed !



Features of phase transitions in cosmic matter and laboratory



Thank you!



Support: RAS Scientific Program "Physics of Extreme State of Matter"
MIPT Research-Education Center "Physics of Extreme State of Matter"
ExtreMe Matter Institute - EMMI

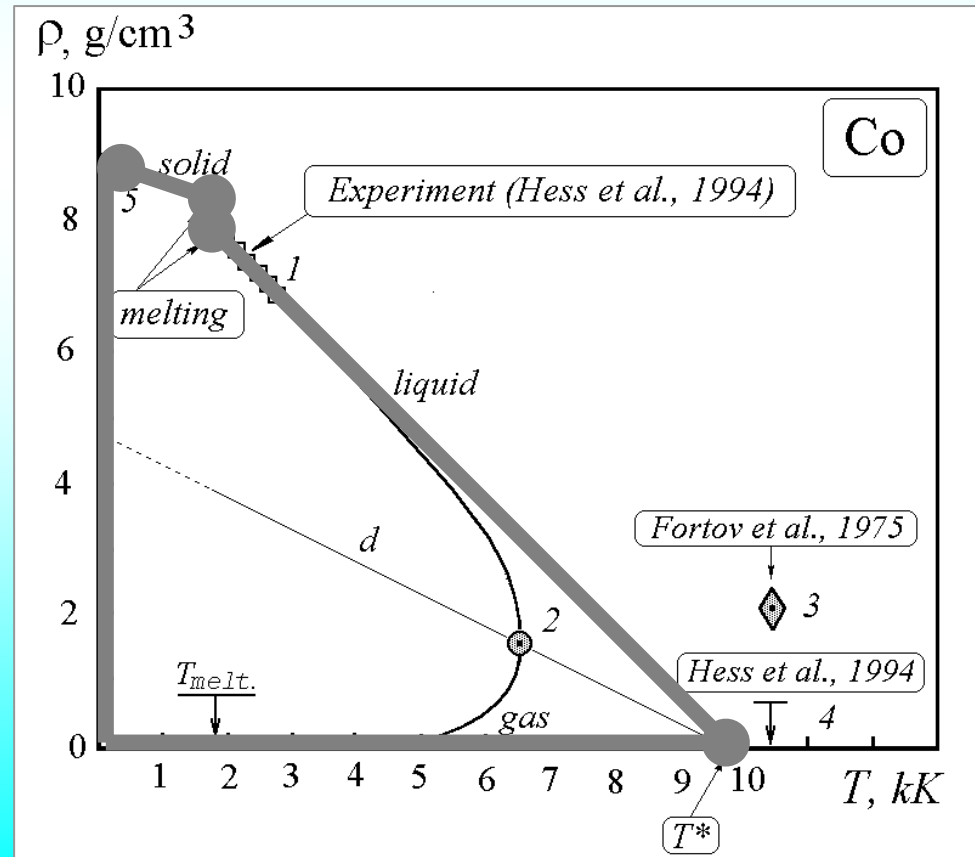
Выпуклость фазовой границы $\rho(T)$

(правило «Бермудского» треугольника)

- **Плотность** конденсированного состояния (кристалла и жидкости), как функция температуры, **линейна** на большей части границы сублимации (tv) и кипения ($ж$)

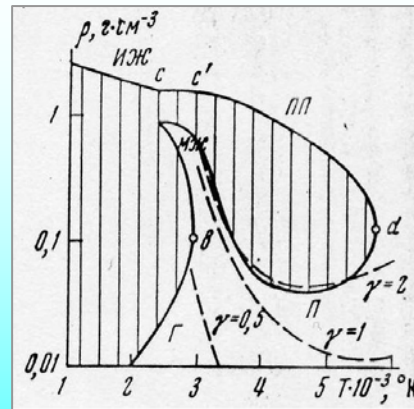
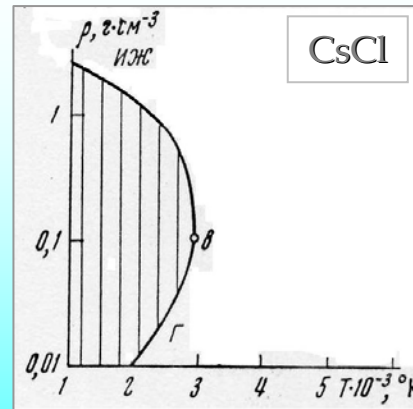
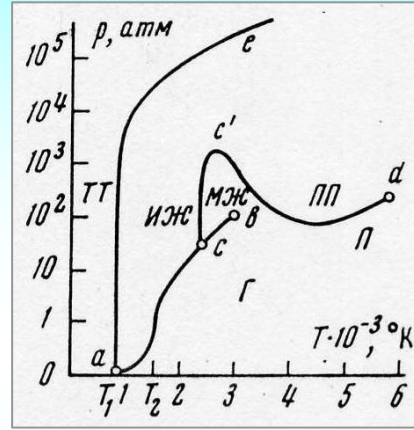
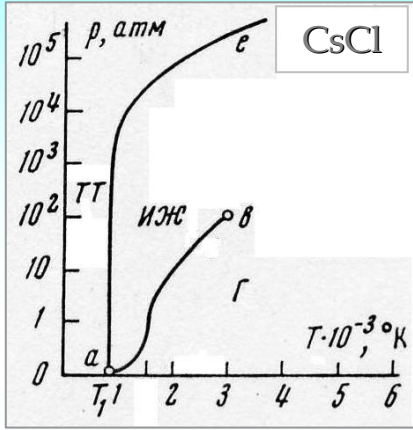
$\rho(T)$ -граница двухфазной области всегда выпукла (эмпирическая закономерность)

Эта граница всегда лежит **внутри треугольника** (эмпирическая закономерность)

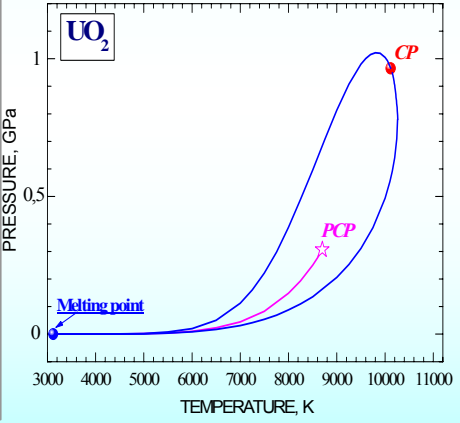
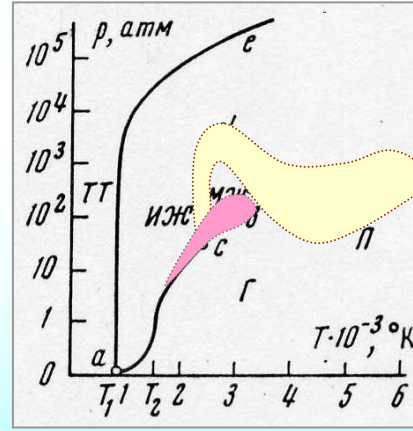


Hypothetical "Plasma Phase Transition" in Strongly Non-Ideal Ionic Plasma

Zejgarnik V., Kobzev G., Kurilenkov Yu., Norman G, *High-Temp.*, 10, (1972)



Plasma Phase Transition
Landau & Zeldovitch – 1930-ths
Norman & Starostin – 1970
W.Ebeling et al



"Standard" gas-liquid phase transition in CsCl must be **non-congruent**

Hypothetical "plasma phase transition" in CsCl even more must be **non-congruent**

Phase diagram for non-ideal ionic mixture (CsCl)
a,b,c,d, - triple and critical points; TT – solid phase;
 Г – gas; ИЖ – ionic liquid; МЖ – molecular liquid;
 П – plasma; ПП – warm dense plasma; T_1 – melting point

Iosilevskiy, 2004

Anomalous Phase Transitions in Hydrogen and Deuterium

1968 - 1970

Норман и Старостин, *Плазменный Фазовый Переход*

1972

С.Б. Кормер с сотр. (ВНИИЭФ, Саров),
*Аномальный скачок плотности при изоэнтропическом
сжатии водорода до давлений $P \sim 3$ Мбар*

? - Plasma Phase Transition - ?

