

R&D, innovation and physicist knowledge needs for radioactive waste characterization.

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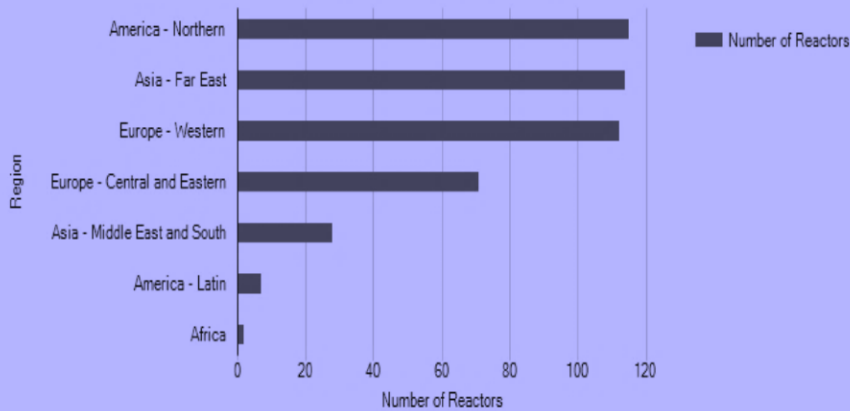
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Nuclear Power Reactors in the World

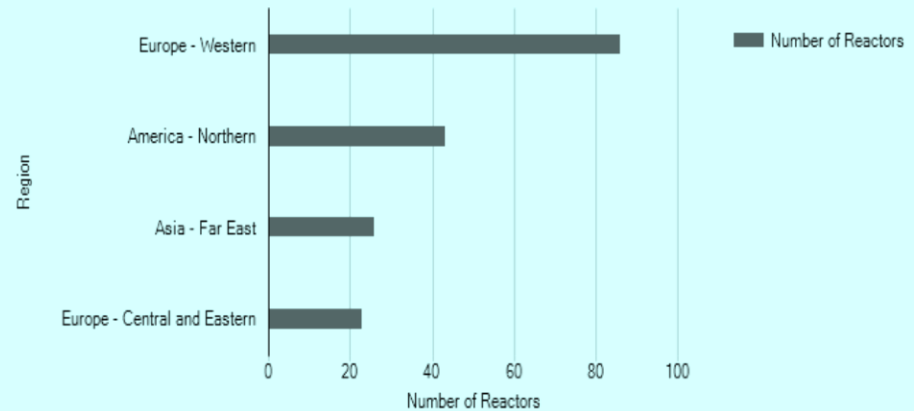
OPERATIONAL REACTORS

Total Number of Reactors: 449



PERMANENT SHUTDOWN REACTORS

Total Number of Reactors: 178

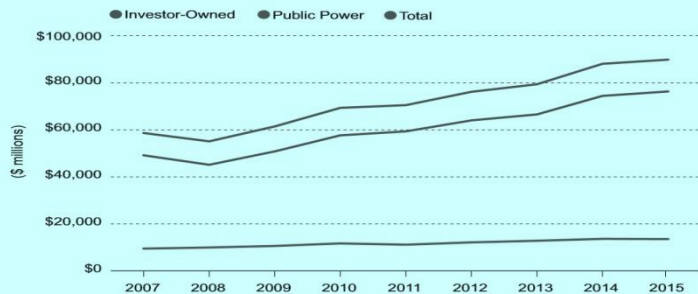


<https://pris.iaea.org/PRIS/WorldStatistics/>

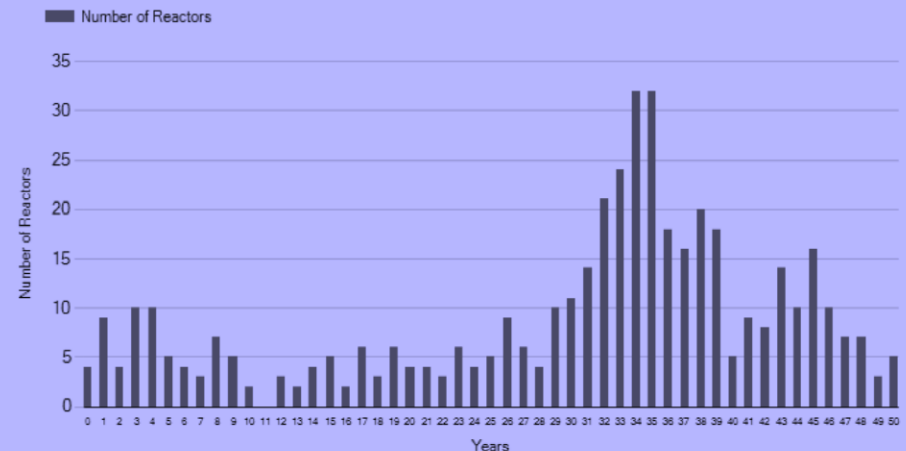
The largest number of nuclear power plants are either already closed or to be closed by 2025 is in Western Europe (>160).

Estimated cost:

France (EDF, AREVA, CEA) 73 billion EUR
USA (US DOE):



Total Number of Reactors: 449



Waste management system (in Hungary)

SCOPE OF DUTIES



Collection

Minimization

Treatment

Conditioning

Characterization

Interim storage



Buffer storage

Overpacking

Disposal



Radioactive waste characterization

Package identification

- Five- or six-figure identification number
- Bar code

Packaging

- 200 litre drum
- 400 litre drum
- Four-cornered steel container
- Special container

Radiological parameters

- Dose rate
- Surface contamination
- Activity



Waste form properties

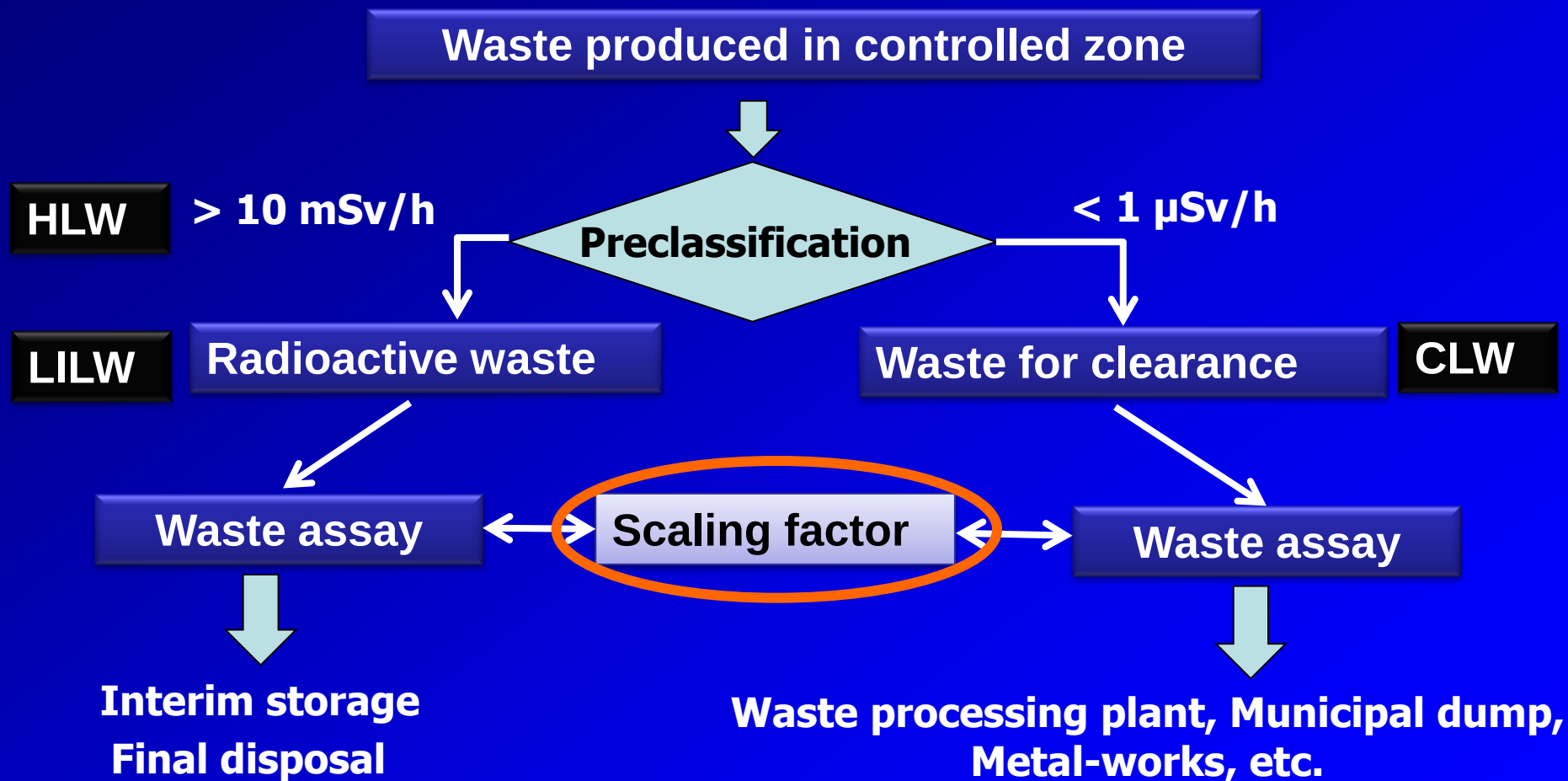
- Physical form
- Structural stability
- Impacts of heat and radiation
- Combustibility
- Forbidden components
- Gas generating
- Microbiological impact
- Free liquid
- Chelating and complexing agents
- Leaching
- Homogeneity

Process of Treatment

- Compacted
- Non-Compacted
- Solidified
- Special treatment

Classification of wastes (solid)

Classification of solid wastes:

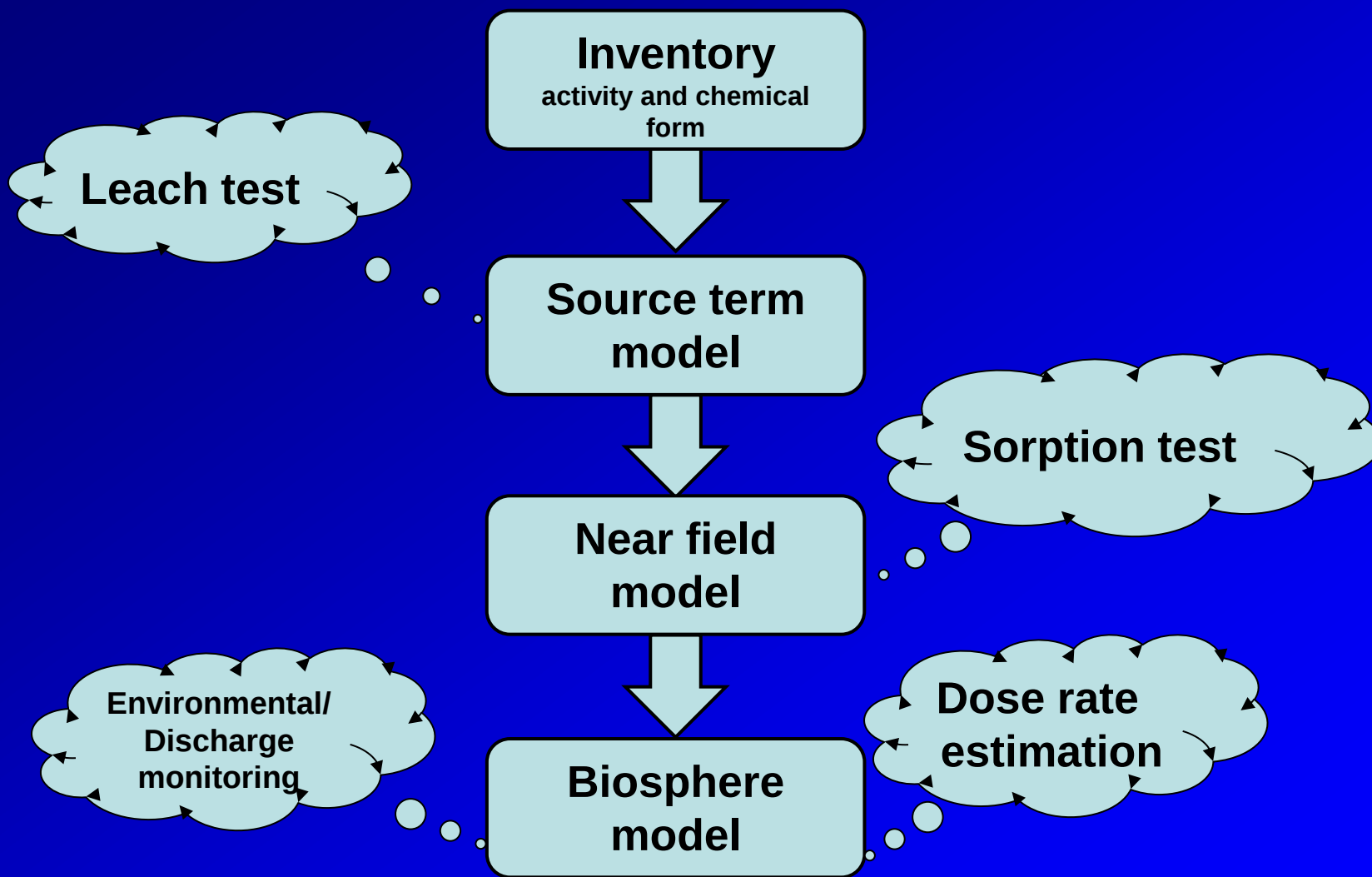


Scaling Factors

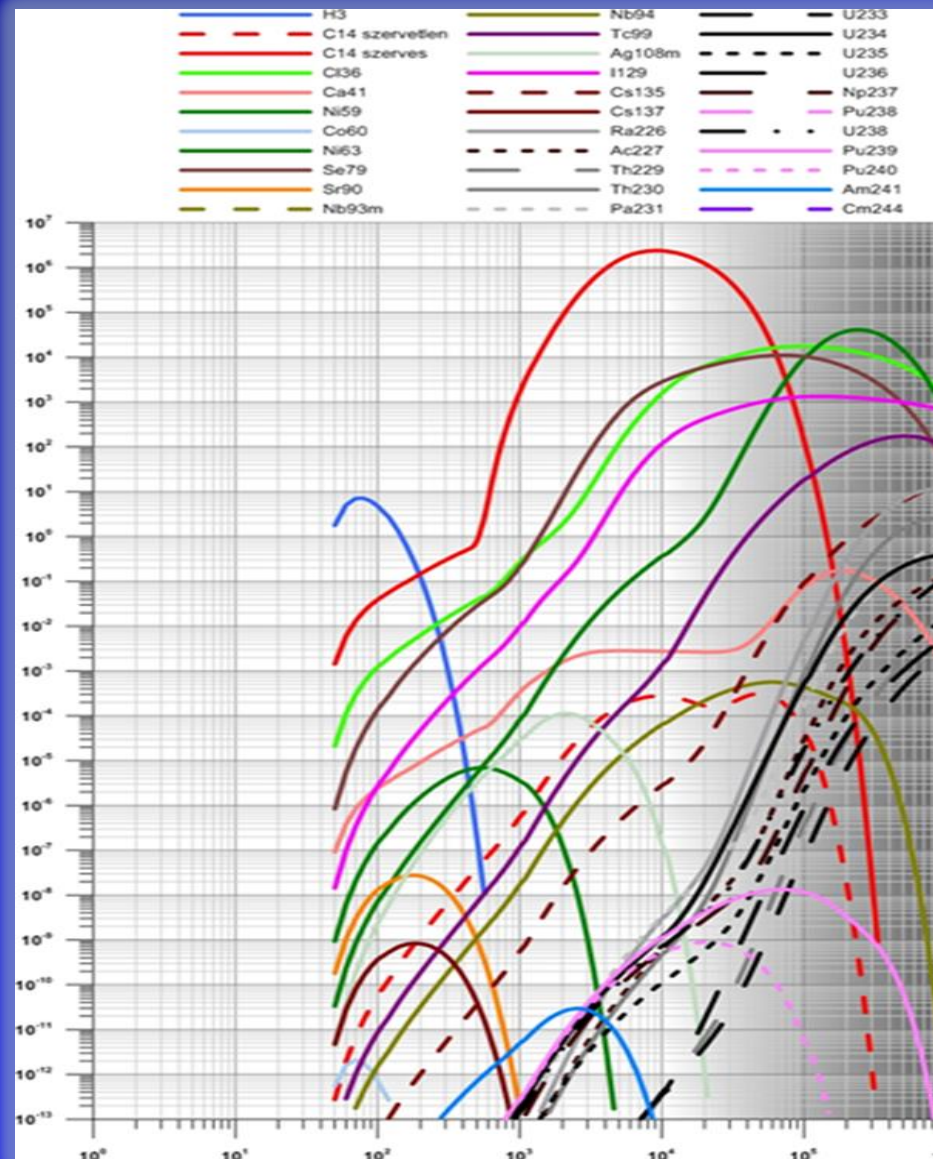
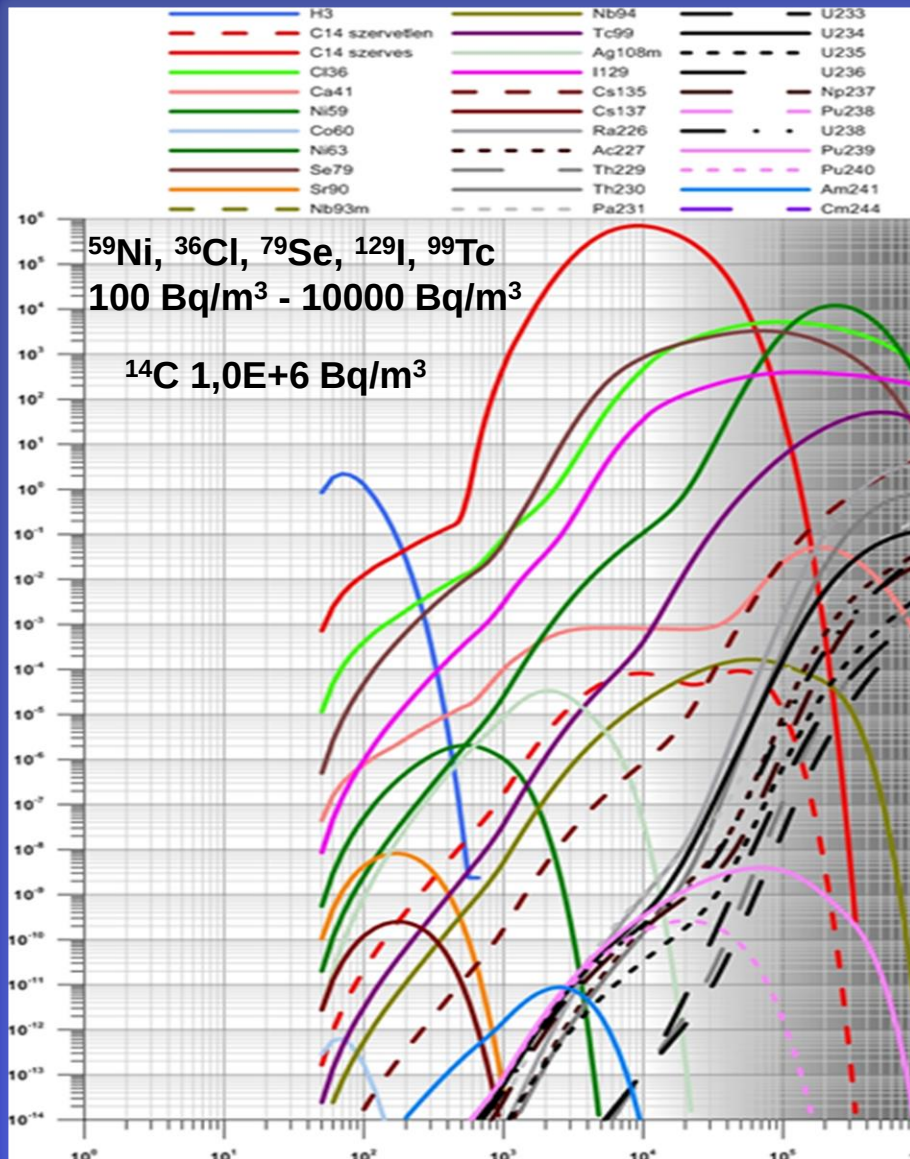
DETERMINATION OF NON-MEASURABLE NUCLIDES

- The radioactive content of every waste type is directly or indirectly originates from the primary coolant.
- Every waste type contains easily measurable isotopes. There is a correlation between certain easily measurable and difficult-to-measure (DTM) isotopes
 - Linear relationship (SF model): $C_{DTM} = SF * C_{KN}$
 - Non-linear relationship (LOG model): $C_{DTM} = a * C_{KN}^b$
- The key nuclide (KN) has to be relatively long-lived to be traceable during the characterization process. The chemical behaviour of the KN has to be similar to that of the related DTM. The gamma-intensity of the KN has to be high enough for low-error measurements.
- The DTM isotopes' activity concentration are usually several order of magnitude lower than the „bulk” (mostly $^{60}\text{Co} + ^{137}\text{Cs}$), due to
- Low formation probability (fission or activation); Low „escape” probability (from an intact fuel rod); Long half-life ► low specific activity; They are either pure alpha/beta emitters or have only very low energy and intensity X-ray or γ -ray lines.
- Result: The determination of their AC requires a radiochemical separation of various complexity and / or special measurement method.
 - Clearance Criterion: $\sum (A_i / A_{i \text{ ref. level}}) \leq 1$

Safety Assessment/Analysis



Safety Analysis



Activity concentration/time [Bq/m³_v_time]

(Excavation Damaged Zone)

Activity flux/time [Bq/year_v_time]

Nuclides to determine

Nuclide	T _{1/2} (year)	MDA [Bq/l]	Type of measurement technique
³ H	12,32	1,0	LSC
¹⁴ C	5730	1,0	LSC, AMS
³⁶ Cl	301000	10,0	LSC
⁴¹ Ca	102000	10,0	AMS, LSC
⁵⁴ Mn	0,85	NR	gamma spectrometry
⁵⁵ Fe	2,73	1,0	X-ray spectrometry
⁶⁰ Co	5,27	10,0	gamma spectrometry
⁵⁹ Ni	75000	1,0	X-ray spectrometry
⁶³ Ni	100	10,0	LSC
⁷⁹ Se	327000	10,0	LSC/ICP-MS
⁹⁰ Sr	29,1	1,0	LSC
⁹³ Zr	1500000	1,0	ICP-MS
^{93m} Nb	16,1	10,0	X-ray spectrometry
⁹⁴ Nb	20300	1,0	gamma spectrometry
⁹⁹ Tc	211000	0,1	LSC
¹⁰⁶ Ru	1,01	NR	gamma spectrometry (¹⁰⁶ Rh)
¹⁰⁷ Pd	6500000	1,0	LSC
^{108m} Ag	418	1,0	
^{110m} Ag	0,68	NR	gamma spectrometry
¹²⁵ Sb	2,76	NR	
¹²⁹ I	15700000	0,1	LSC, TW-gamma spectrometry
¹³⁴ Cs	2,06	NR	gamma spectrometry
¹³⁵ Cs	2300000	1,0	NAA ICP-MS
¹³⁷ Cs	30,07	10,0	
¹⁴⁴ Ce	0,78	NR	gamma spectrometry
¹⁵⁴ Eu	8,8	NR	
²³⁴ U	244000	0,05	
²³⁵ U	704000000	0,05	
²³⁷ Np	2100000	0,05	alpha spectrometry
²³⁸ U	4470000000	0,05	
²³⁸ Pu	86,4	0,05	²³⁸ U, ²³⁵ U, ²³⁷ Np, ²³⁹ Pu
²³⁹ Pu	24400	0,05	ICP-MS
²⁴⁰ Pu	6540	0,05	
²⁴¹ Am	432	0,05	²⁴¹ Am, ²³⁵ U
²⁴³ Am	7380	0,05	gamma spectrometry
²⁴² Cm	0,45	0,05	
²⁴⁴ Cm	17,9	0,05	

NR: Not relevant for long term safety case

Radioactivity measurement



Liquid scintillation
counters



Alpha spectrometry



Gross alpha/beta measurement



Gamma spectrometry

Mass spectrometers



Environ MICADAS AMS



HELIX-SFT Noble gas mass spectrometer



QQQ-8800 ICP-MS



Q-MS



Delta Plus XP Stable Isotope Mass Spectrometer

Scaling Factors

DTM isotope	KN	Parameters		Coefficient of determination	Coefficient of correlation	Statistical test 95% confidence interval		Quality of fitting
Nuclide		m	b	r	R	F/F _{krit}	t/t _{krit}	
³ H	⁶⁰ Co	0,138	2,84E+04	0,192	0,438	12,663	3,556	proper
¹⁴ C	¹³⁷ Cs	-0,071	1,60E+04	0,034	-0,183	1,681	-1,296	bad*
		0*	1*					
³⁶ Cl	¹³⁷ Cs	0,398	3,31E-02	0,110	0,332	3,527	1,878	proper
⁴¹ Ca	¹³⁷ Cs	0,494	1,49E-03	0,231	0,481	4,657	2,158	good
⁵⁵ Fe	⁶⁰ Co	0,554	2,70E+00	0,350	0,592	19,687	4,437	good
⁵⁹ Ni	⁶⁰ Co	0,479	6,30E+00	0,444	0,667	34,564	5,879	very good
⁶³ Ni	⁶⁰ Co	0,712	1,48E+01	0,828	0,910	242,328	15,567	excellent
⁷⁹ Se	¹³⁷ Cs	0,553	7,20E-02	0,288	0,537	0,910	0,954	good
⁹⁰ Sr	¹³⁷ Cs	0,406	8,13E-01	0,136	0,401	8,620	2,936	proper
⁹⁴ Nb	⁶⁰ Co	0,172	1,37E+01	0,103	0,321	0,518	0,720	proper
⁹⁹ Tc	⁶⁰ Co	0,246	1,62E-01	0,193	0,440	8,917	2,986	proper
^{108m} Ag	⁶⁰ Co	0,888	1,89E-03	0,443	0,666	7,560	2,750	very good
¹²⁹ I	¹³⁷ Cs	0,701	2,33E-05	0,481	0,693	42,363	6,509	very good
²³⁴ U	²³⁸ Pu	0,266	2,02E-02	0,462	0,679	43,511	6,596	very good
²³⁵ U	²³⁸ Pu	0,311	2,70E-02	0,455	0,675	36,798	6,066	very good
²³⁷ Np	⁶⁰ Co	0,556	1,96E-02	0,203	0,451	1,084	1,041	good
²³⁸ U	²³⁸ Pu	0,306	2,22E-01	0,433	0,604	41,373	6,432	very good
²³⁸ Pu	⁶⁰ Co	0,415	1,27E-03	0,337	0,581	22,636	4,758	good
²³⁹⁺²⁴⁰ Pu	⁶⁰ Co	0,545	2,41E-04	0,347	0,589	24,180	4,917	good
²⁴¹ Am	⁶⁰ Co	0,696	1,10E-04	0,499	0,706	43,810	6,619	very good
²⁴² Cm	⁶⁰ Co	0,441	1,85E-04	0,247	0,497	11,460	3,385	good
²⁴⁴ Cm	⁶⁰ Co	0,605	7,68E-05	0,399	0,632	27,765	5,269	very good

$$A_{\text{DTM}} = m * A_{\text{K}}^b$$

A_{DTM} – activity of DTM

A_{KN} – activity of key nuclide

m,b – scaling factors

*suggested parameters

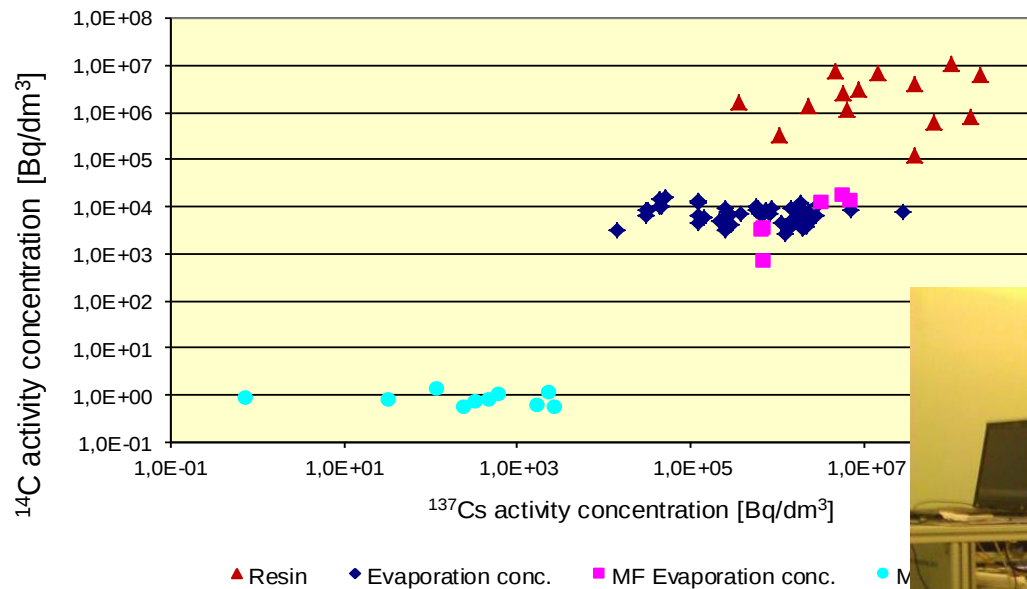
25-27 November 2019

R<0,3	→	bad
If 0,3≤R<0,45	→	proper
If 0,45≤R<0,6	→	good
If 0,6≤R<0,75	→	very good
If 0,75≤R	→	excellent

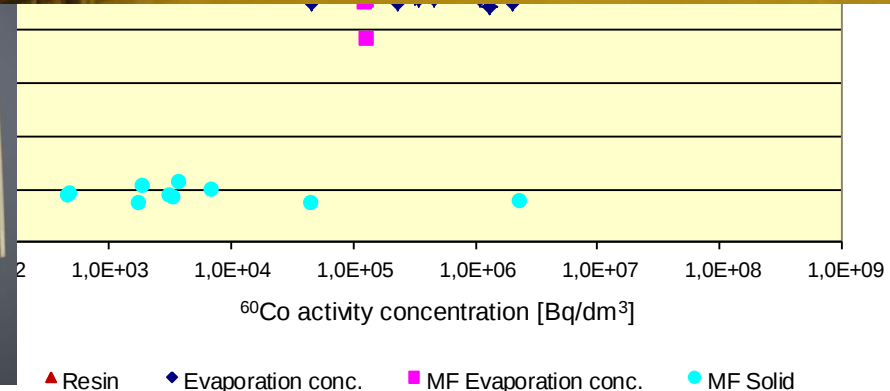
Scaling Factor (^{14}C)

How can we solve this problem?

- Individual sampling (difficult, not representative)
- Calculation (from what?)
- Gas generation?



Bq/dm³



^{14}C release by the gas generation of L/ILW

Interim storage building with 3000 LILW drums
controlled ventilation and C-14/H-3 sampling

**7500 m³
air/day**

Used air

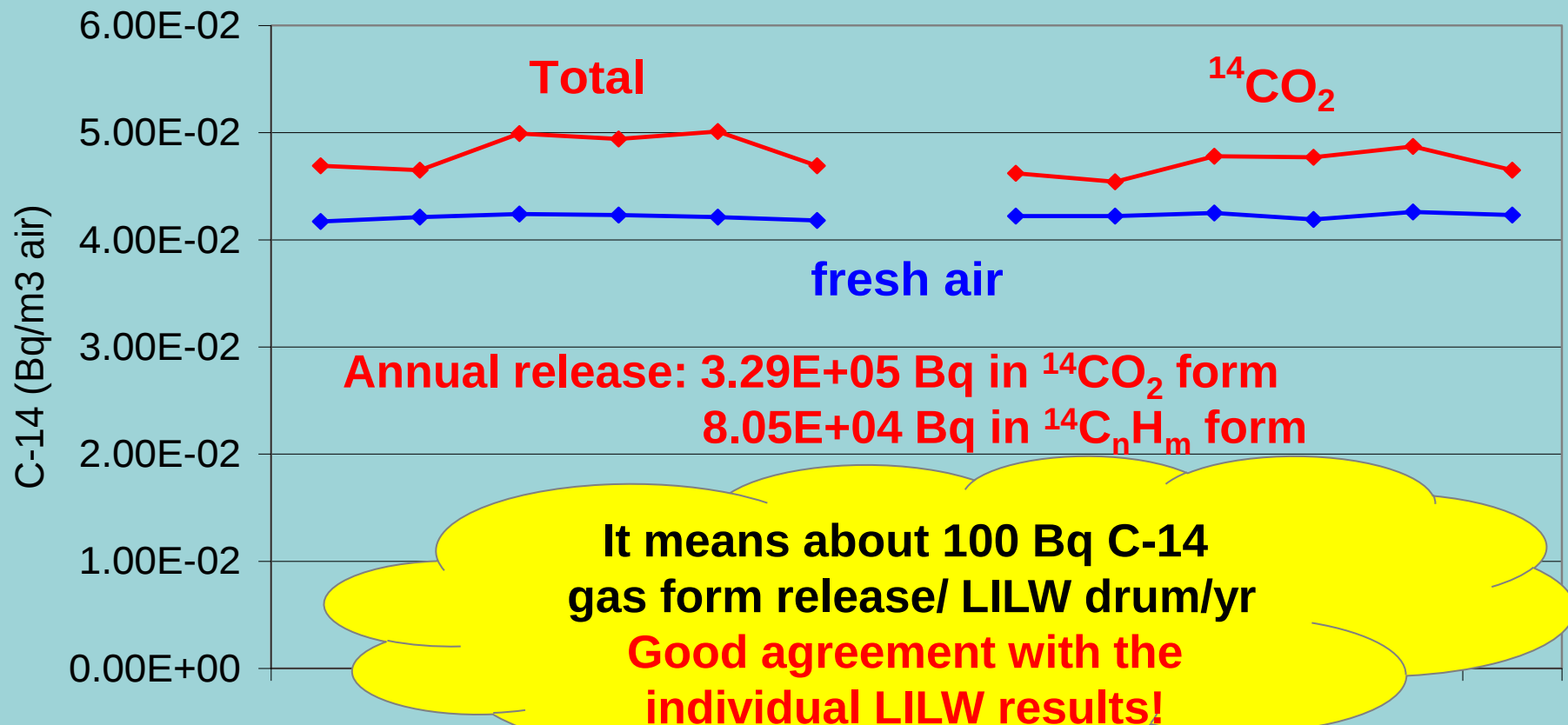
Fresh air

$^{14}\text{CO}_2$ $^{14}\text{C}_n\text{H}_m$
HTO HT

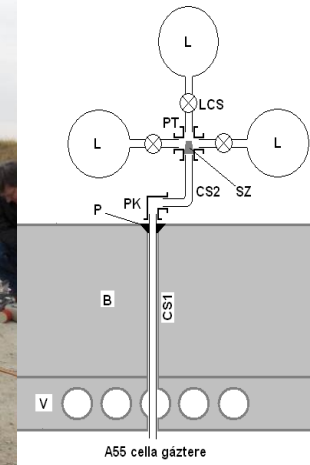
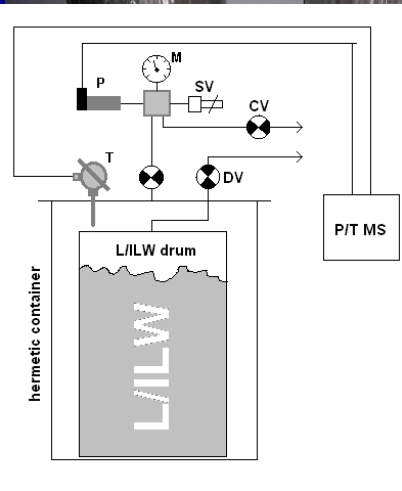
^{14}C release by the gas generation of L/ILW

Interim storage building with 3000 LILW drums controlled ventilation and C-14/H-3 sampling

C-14 in ventilated air of LILW storage building,
2 months integrated samples, 2012



^3H inventory estimation with $^3\text{H}/^3\text{He}$ method



$$N(^3\text{He}) = \Delta N(^3\text{H}) = N(t_1) - N(t_2) = N(t_1) \cdot (1 - e^{-\lambda \Delta t})$$

$N(^3\text{He})$ is the number of tritiogenic ^3He atoms

$\Delta N(^3\text{H})$ is number of tritium atoms that have already decayed

$N(t_1)$ and $N(t_2)$ are the numbers of tritium at time t_1 and t_2

Δt is the difference between t_1 and t_2

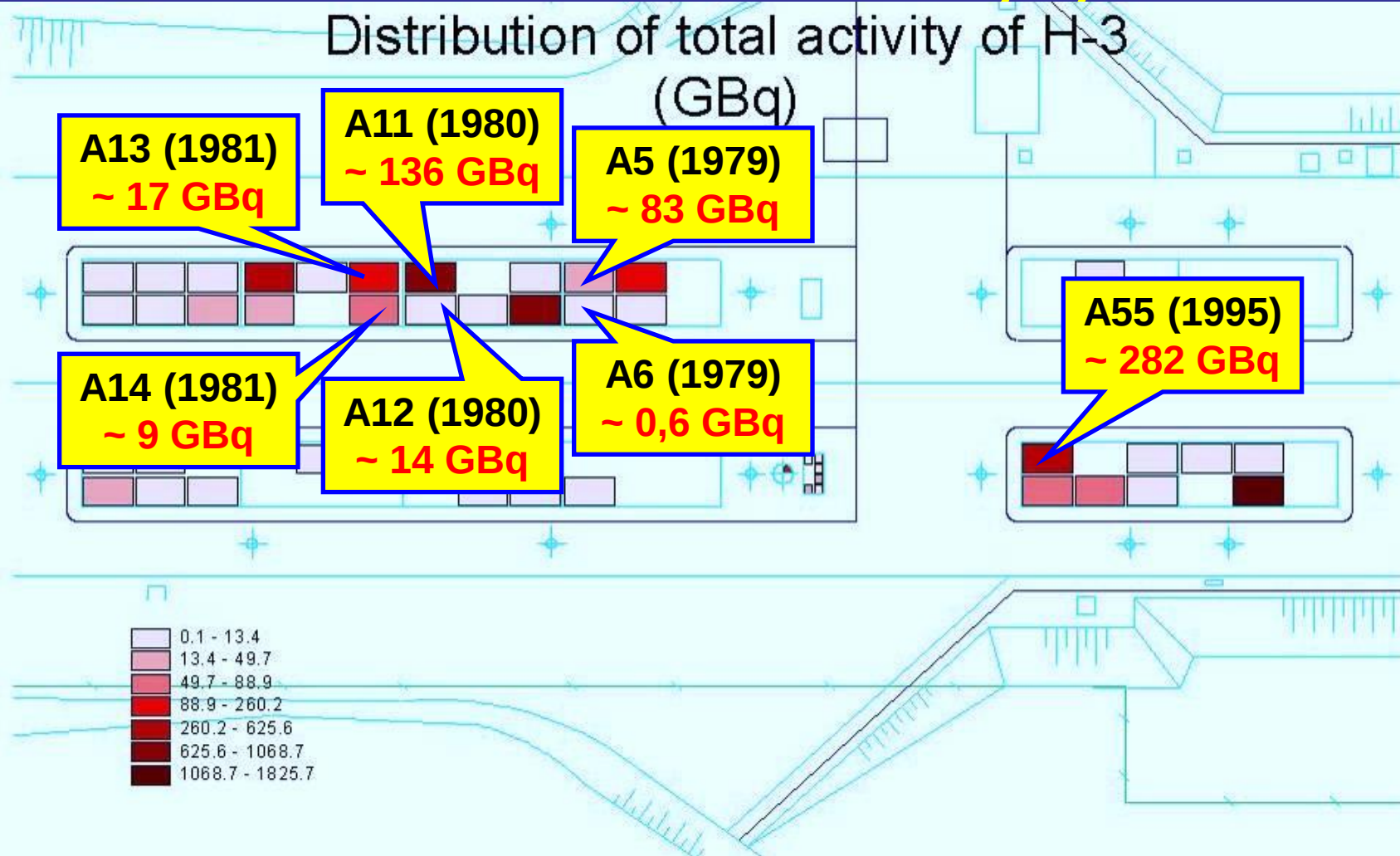
λ is the decay constant of tritium.



HELIX (Thermo Instruments)

^3H inventory estimation with $^3\text{H}/^3\text{He}$ method

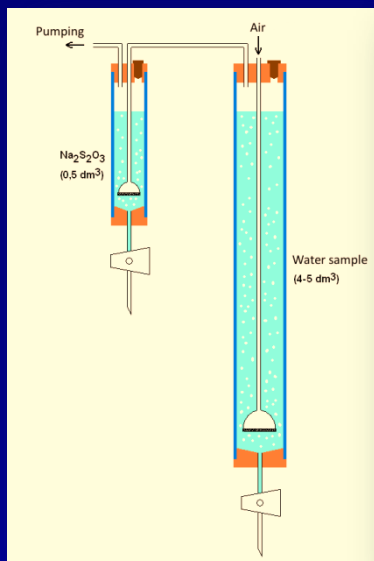
Estimation of total restored ^3H activity by ^3He results



$^3\text{H} \longrightarrow ^3\text{He}$, number of produced ^3He atoms is equal with number of decayed ^3H atoms, if the vault is closed (enough) for gases...

Palcsu L. et al. *Journal of Radioanalytical and Nucl. Chem.* 286 (2010)483-487

^{129}I activity measurement with $^{129}\text{I}/^{129}\text{Xe}$ method



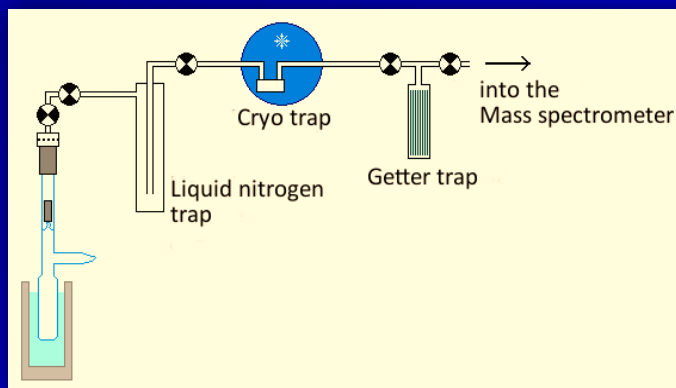
Carrier: Potassium-Iodide,-Iodate + H_2SO_4
 Purge with airflow
 Trapped in Sodium thiosulfate



Degassing, set aside



Xe freezing out



VG5400 (Fison Instruments)

$$C = N_{doug} \cdot \frac{e^{\lambda t_{SD}}}{(1 - e^{-\lambda t_{DM}})} \cdot \lambda \cdot \frac{1}{V_{samp}}$$

$$t_{SD}: t_{\text{sampling}} - t_{\text{degassing}}$$

$$t_{DM}: t_{\text{measurement}} - t_{\text{degassing}}$$

Summary

- We can measure most of the DTM isotopes with the proper detection limit in liquid radioactive waste
- New methods are welcome (faster, cheaper, no radiochemical pre-treatment are necessary to apply them).
- We have significant uncertainty (in some cases) when applying Scaling-Factor. In most cases it is impossible to check its uncertainty.
- R&D needed to develop fast, onsite measurement techniques (decommissioning, safety improvement programs, re-characterization of old disposal facilities, legacy waste)
- The success of minimization, treatment and conditioning depends largely upon this knowledge.
- The result of the „under” characterization that the cost of the disposal will increase! (more robust waste packages, engineering barriers)

Thank you for your attention!

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We really want to know what is inside!



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