

Activities related to SHE target production and aqueous chemistry of SHEs at RIKEN

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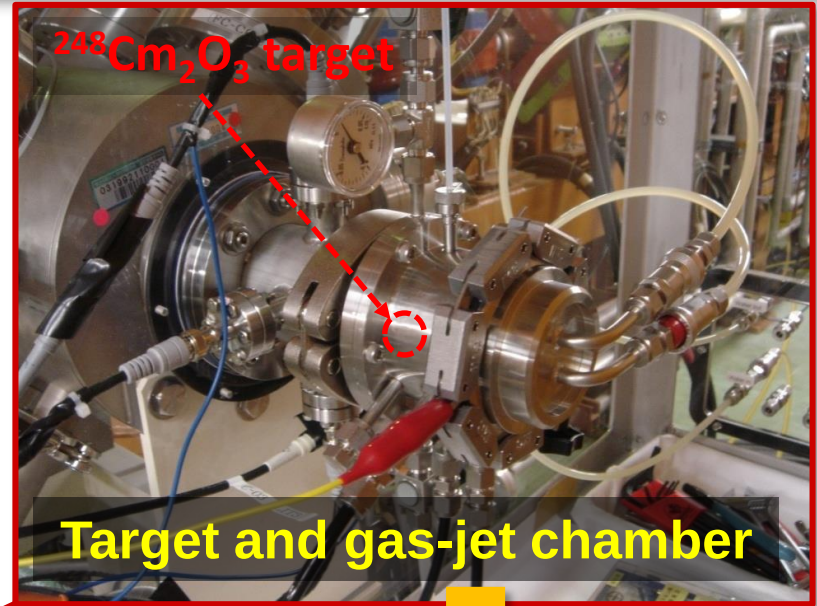
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Outline

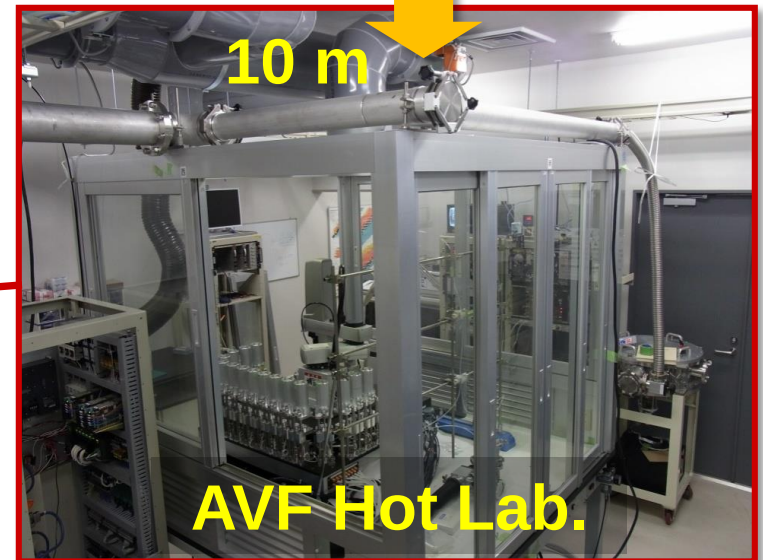
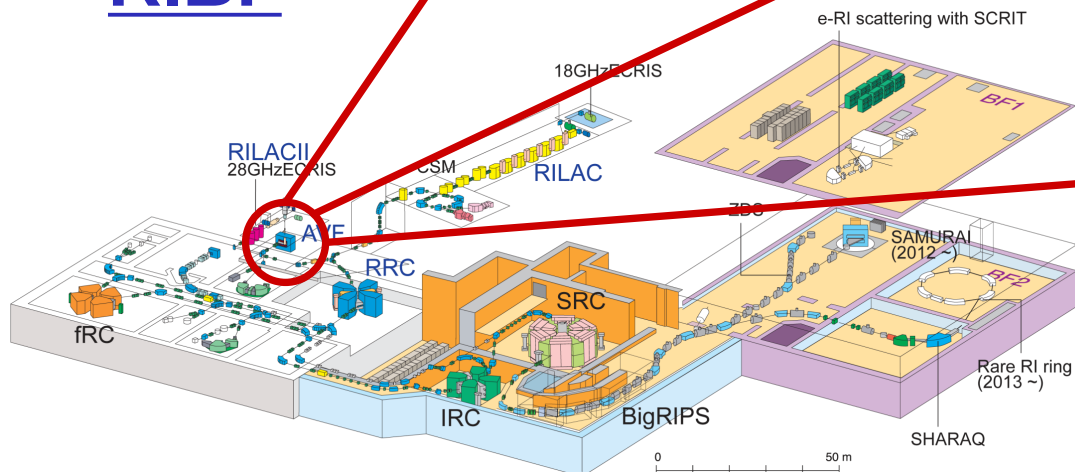
- 1. Target preparation and aqueous chemistry studies at AVF cyclotron**
- 2. Target preparation and SHE production for chemistry studies at RIKEN Linear Accelerator (RILAC)**
- 3. Summary and future perspectives**

1. Target preparation and aqueous chemistry studies at AVF cyclotron

AVF cyclotron and gas-jet transport system



RIBF



Preparation of a ^{248}Cm target for AVF

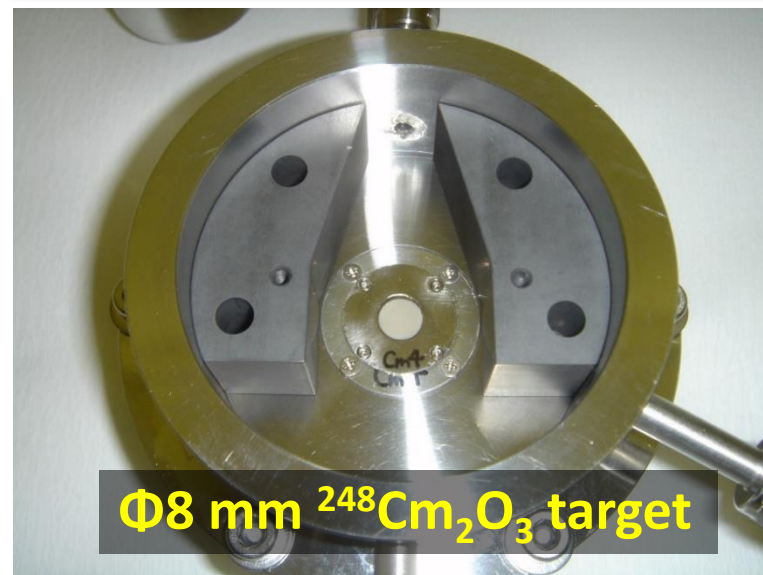
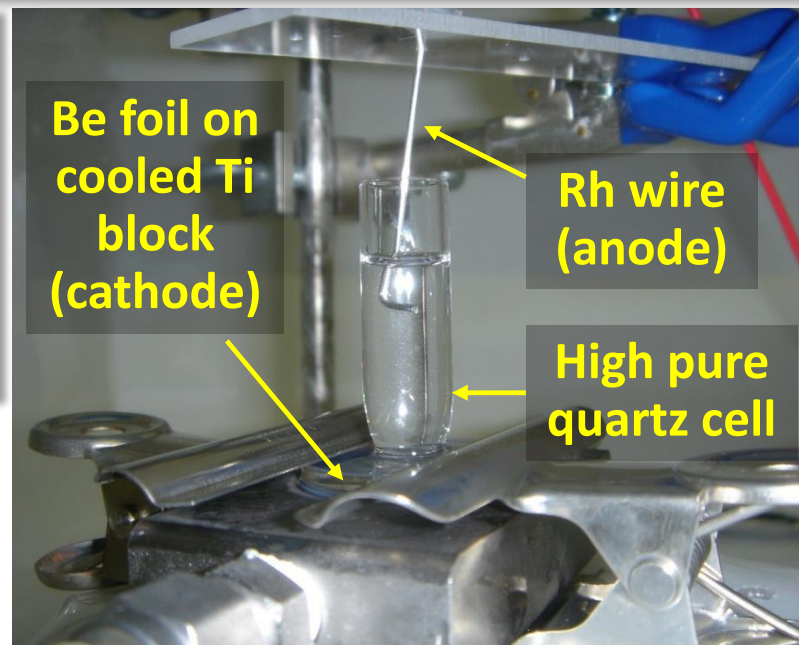
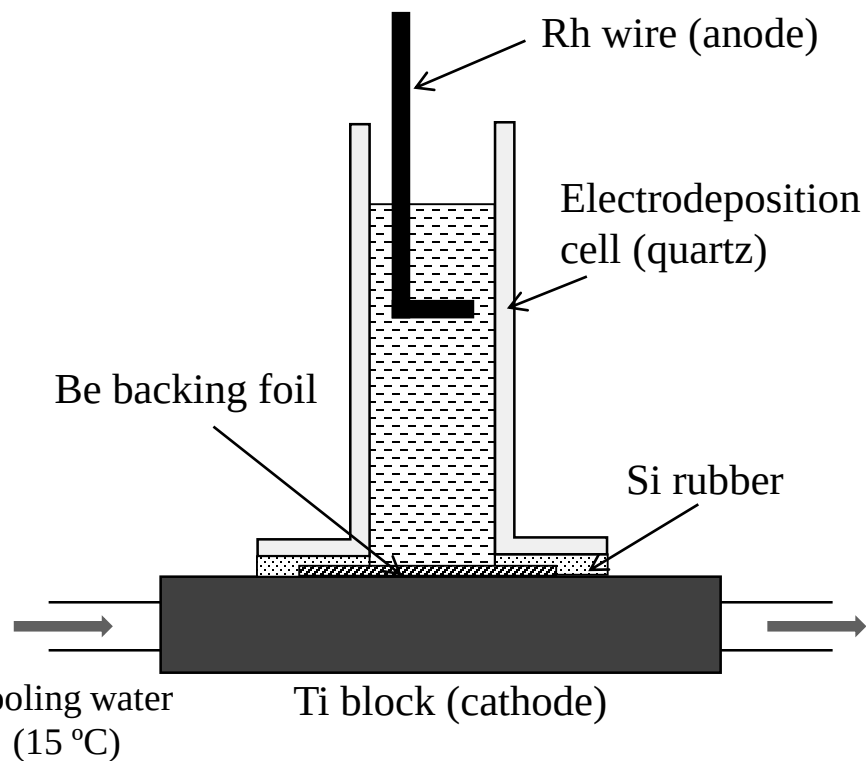
Electrodeposition conditions

0.15 mg of ^{248}Cm in 10 μL 0.2 M

HNO_3 + 2 mL 2-propanol

950 V x 4.7 mA/cm² for 20 min x 4

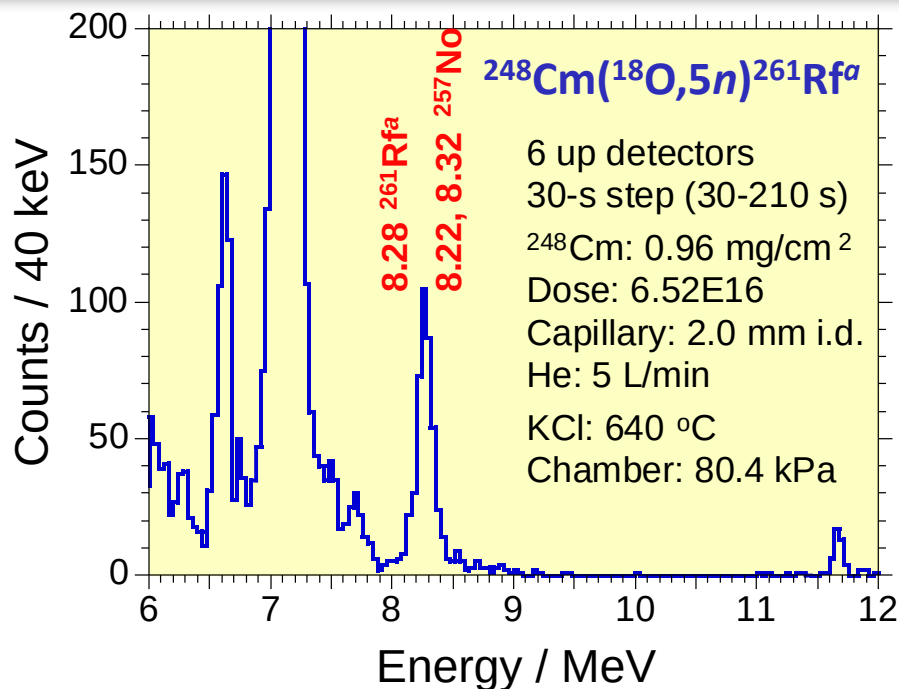
→ 850 $\mu\text{g}/\text{cm}^2$ $^{248}\text{Cm}_2\text{O}_3$ on 10- μm Be



Production of SHEs using AVF + gas-jet

- Heavy-ion beam (~ 1 pμA):
 ^{12}C , ^{18}O , ^{19}F , ^{22}Ne , ^{26}Mg ...
- High production yields without
a recoil separator

Z	Reaction	Half-life
102	$^{248}\text{Cm}(^{12}\text{C},5n)^{255}\text{No}$	3.52 min
104	$^{248}\text{Cm}(^{18}\text{O},5n)^{261}\text{Rf}^a$	68 s
105	$^{248}\text{Cm}(^{19}\text{F},5n)^{262}\text{Db}$	34 s



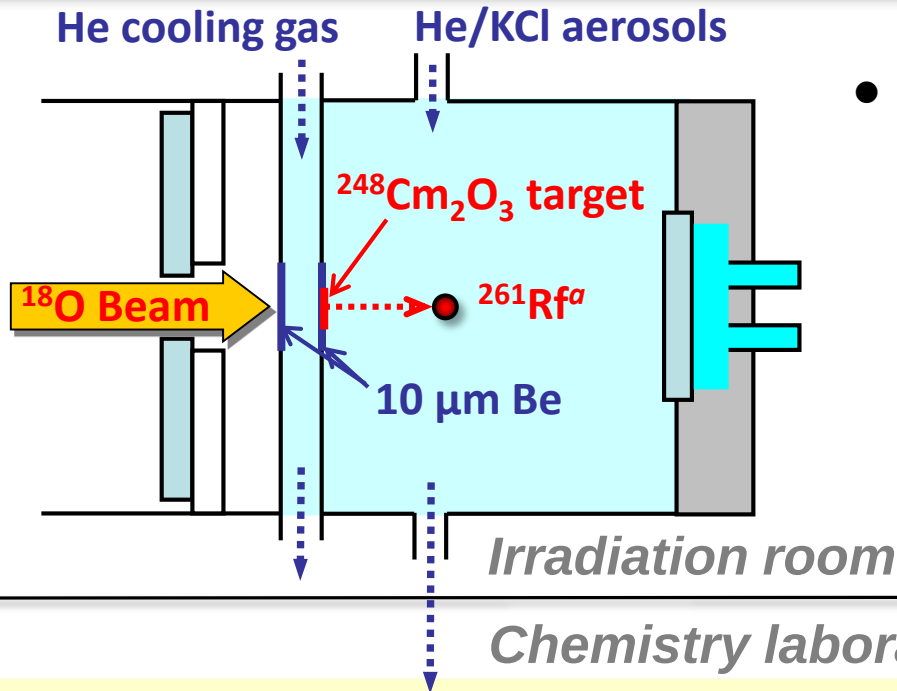
Exp.	Beam [MeV]	Irrad. [pμA]	Irrad. [h]	No. of α events	Gas-jet eff.** [%]	σ*** [nb]	Yield at AVF HL [atoms/min]
^{255}No	79.1	0.944	1.09	2689	46	900	200
$^{261}\text{Rf}^a$	96.4	0.870	3.34	865*	63	17	5
^{262}Db	103.1	0.870			63	2.1	0.6 (Calc.)

* Including α particles of ^{257}No .

** Estimated from the gas-jet efficiencies of ^{169}Hf produced in $^{nat}\text{Gd}(^{18}\text{O},xn)^{169}\text{Hf}$.

*** The target thicknesses were assumed to be 325, 569, and 569 μg/cm² for ^{255}No , $^{261}\text{Rf}^a$, and ^{262}Db , respectively.

Aqueous chemistry of SHEs using AVF + gas-jet



- Partition methods with single atoms
e.g. Ion-exchange chromatography,
solvent extraction, and reversed-
phase extraction

→ **Distribution coefficient (K_d)**
→ **Complex formation and
chemical species**

Aqueous chemistry apparatuses

- ARCA (liquid chromatography)
- CHIN (co-precipitation)
- AMBER (solid-liquid extraction
by batch method)
- ISE (flow solvent extraction)

+



Aqueous chemistry of Rf and Db using AVF + gas-jet

Kanazawa Univ.

- Reversed-phase TTA extraction chromatography of Rf from HF/HNO₃ (with ARCA) Yokoyama et al., Radiochim. Acta **107**, 27 (2019).
 $\Rightarrow K_d: \text{Rf} < \text{Zr} \approx \text{Hf}$

Osaka Univ.

- Solid-liquid extraction of Rf in Aliquat 336/HCl system (with AMBER*) **$\Rightarrow \text{Rf} > \text{Zr} > \text{Hf}$** Yokokita et al., Dalton Trans. **45**, 18827 (2016).
*Kasamatsu et al., Radiochim. Acta. **103**, 513 (2015).
- Hydroxide co-precipitation of Rf and No (with CHIN**) **$\Rightarrow \text{Rf} > \text{Zr} \approx \text{Hf}$** Kasamatsu et al., to be submitted.

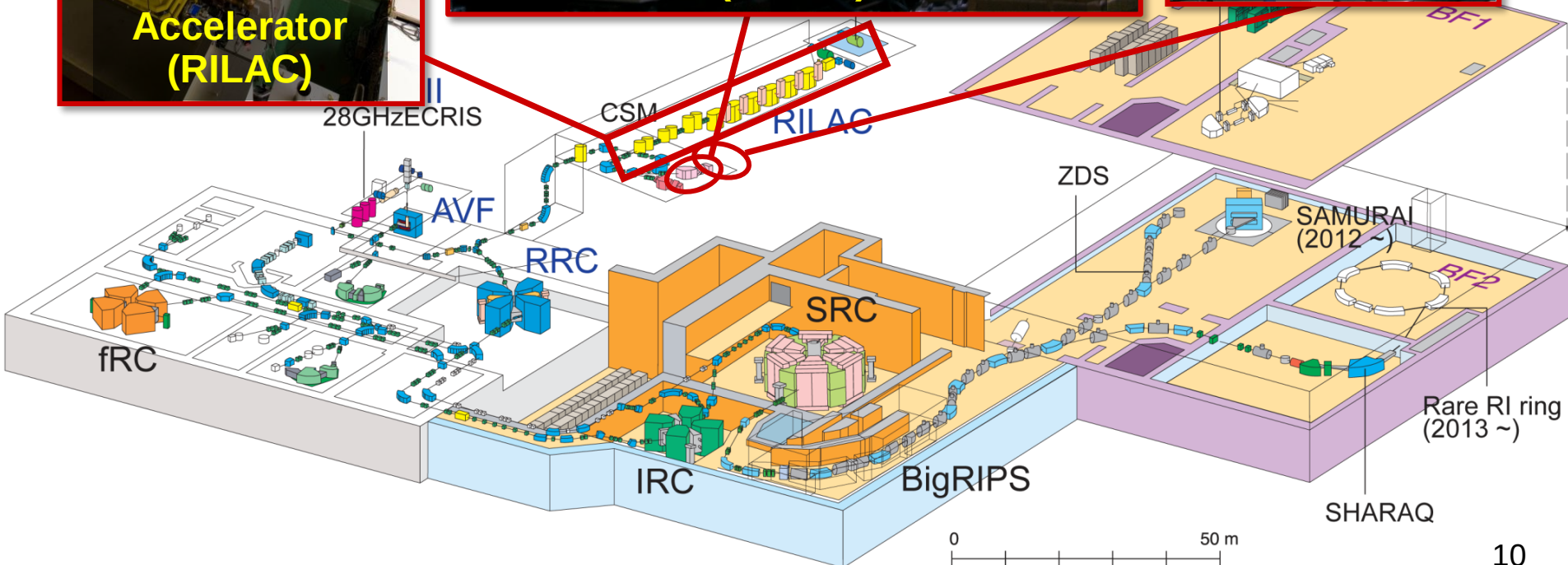
Kasamatsu et al., J. Nucl. Radiochem. Sci. **14, 7 (2014).

Niigata Univ.

- Reversed-phase TBP extraction chromatography of Db from HF (with ARCA) **$\Rightarrow \text{Db} \approx \text{Nb} < \text{Ta}$** Murakami et al., to be submitted.

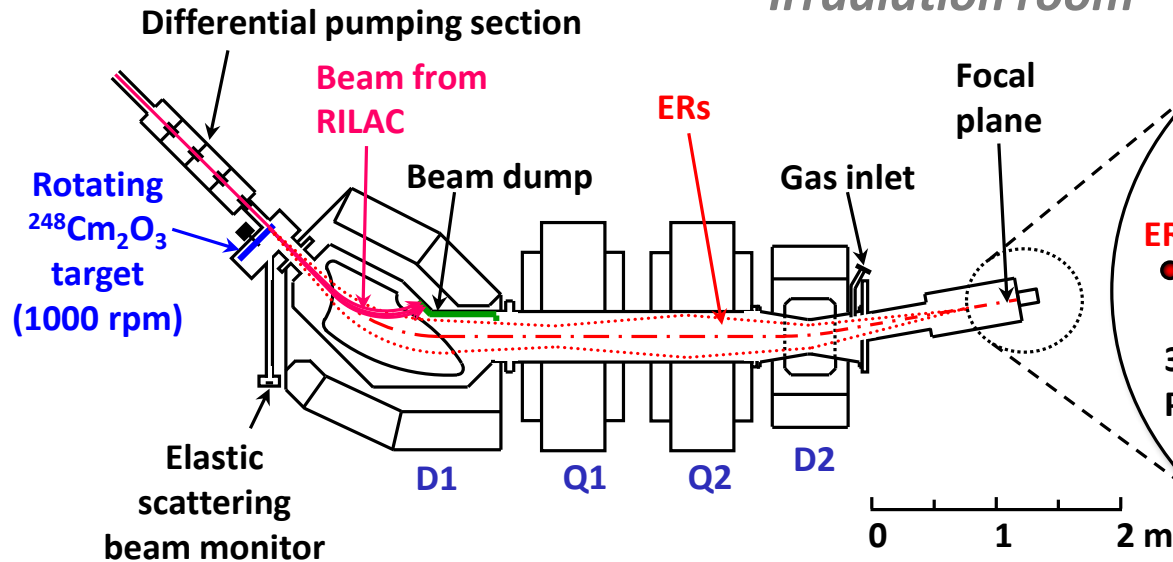
2. Target preparation and SHE production for chemistry studies at RILAC

RILAC and GARIS



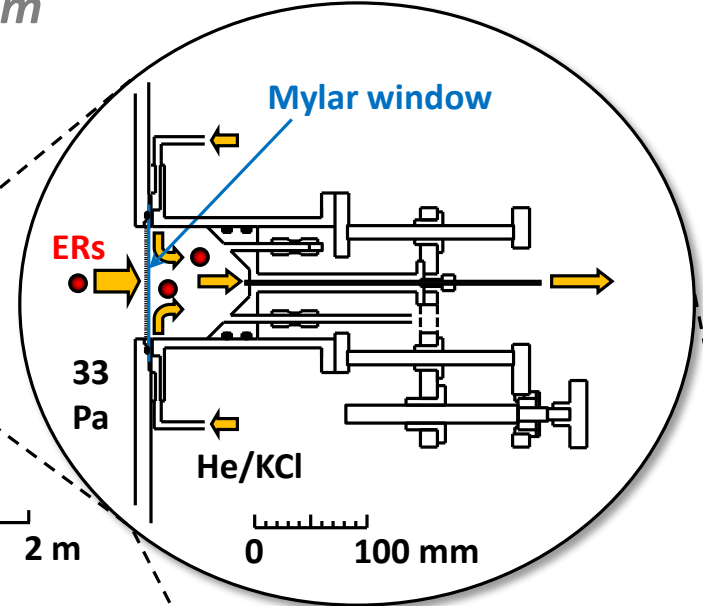
RIKEN GARIS gas-jet system for SHE chemistry

RIKEN GARIS



Haba et al., Chem. Lett. **38**, 426 (2009).

Gas-jet transport system

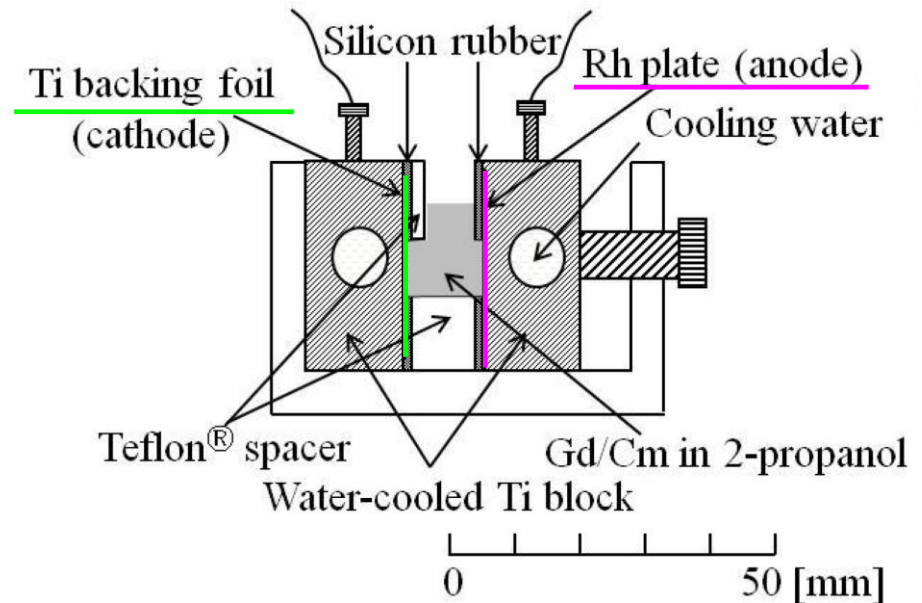
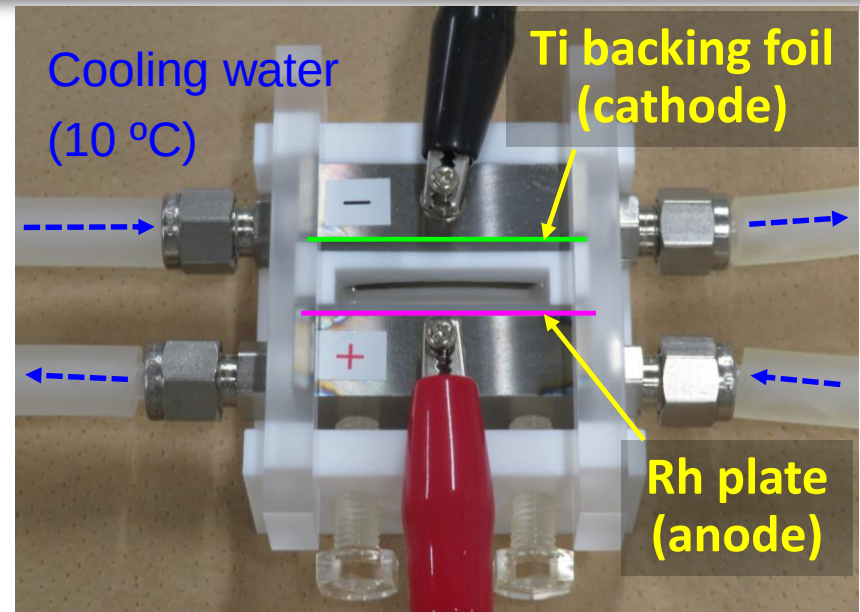
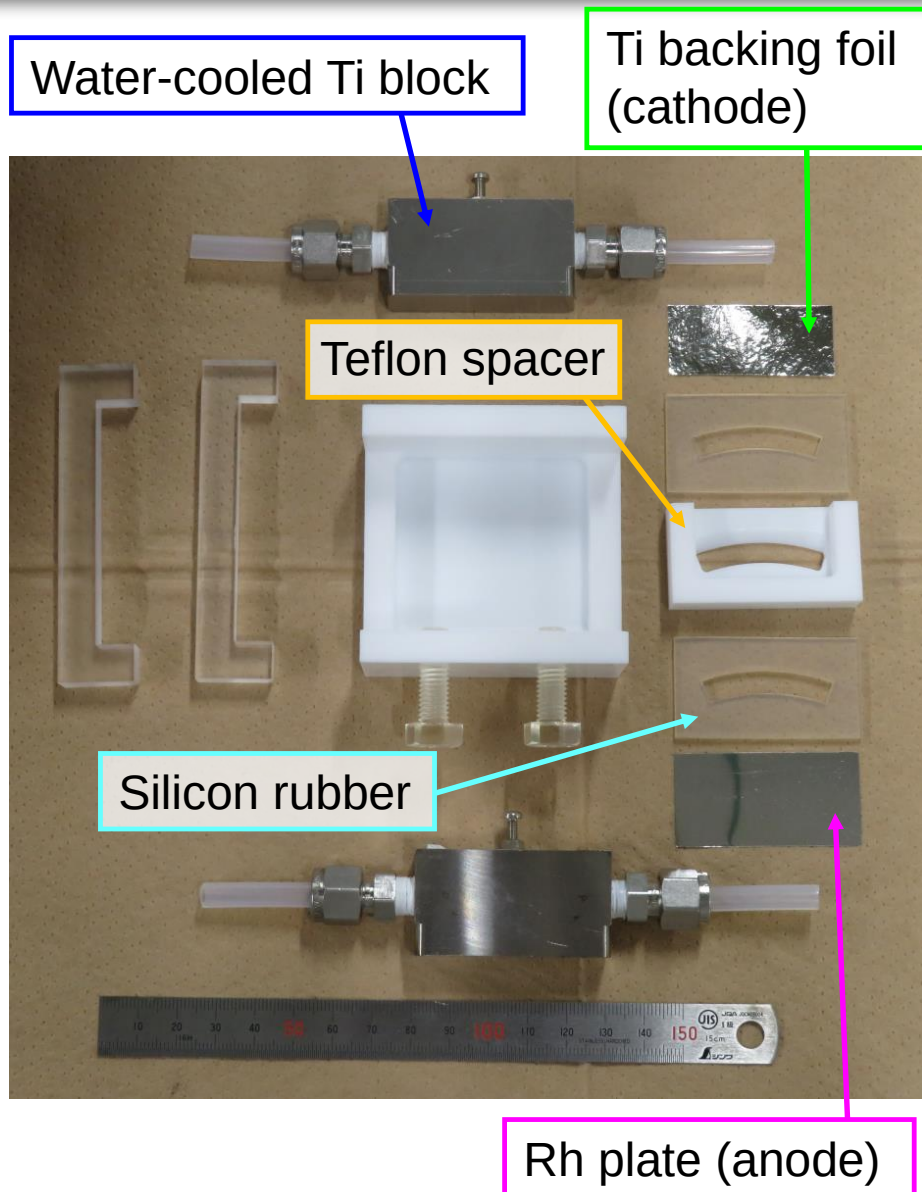


Advantages of gas-jet system coupled to GARIS

- Low background condition for α /SF spectrometry
- High gas-jet transport efficiency at intense beams
- New chemical reaction

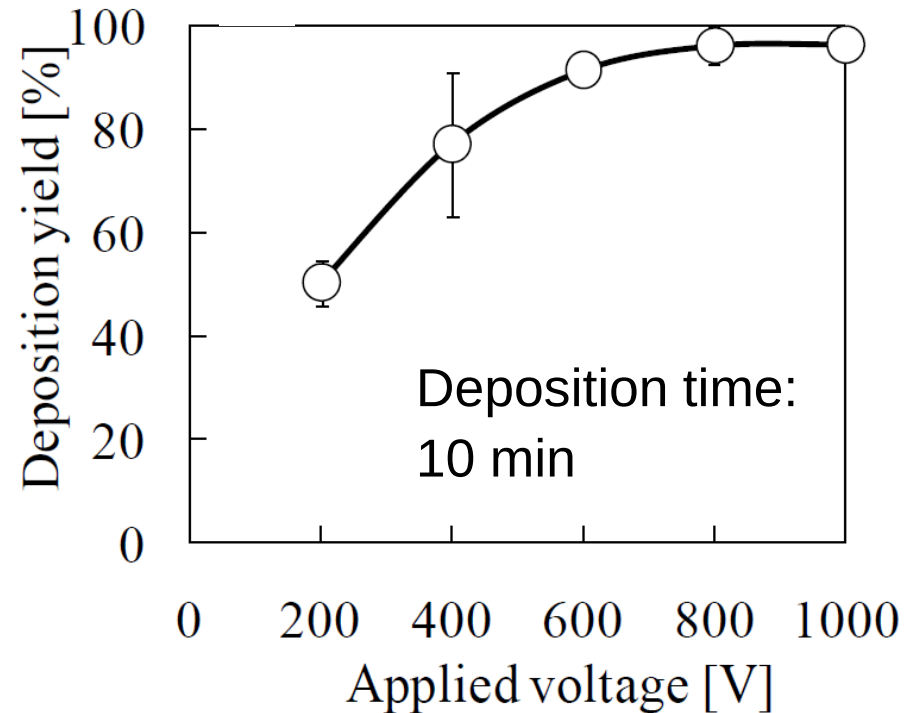
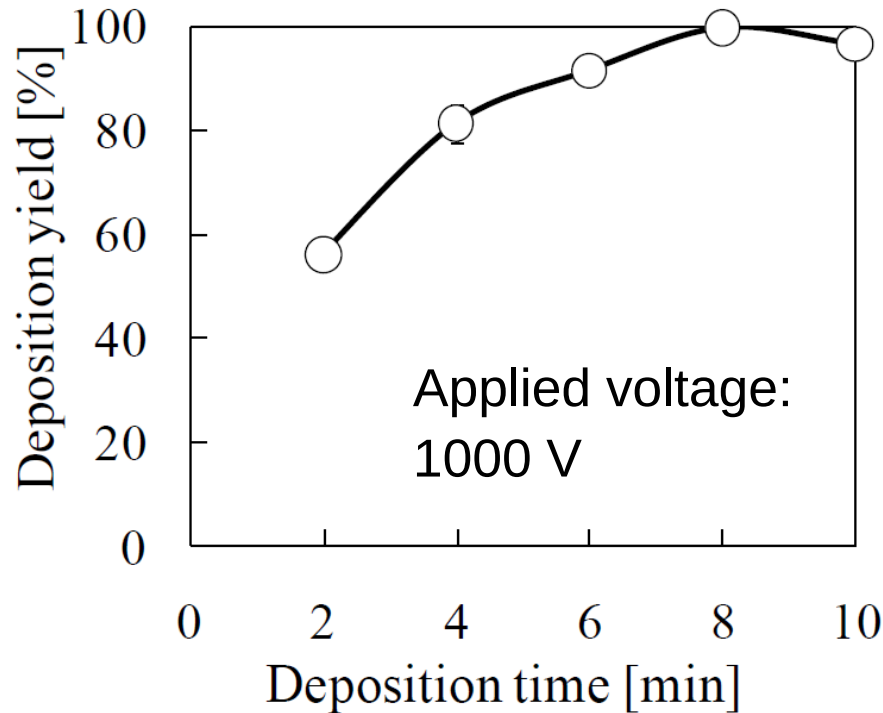


Preparation of rotating ^{248}Cm targets



Optimization of electrodeposition conditions

20 μL of 0.2 M HNO_3 containing 650 μg of $^{\text{nat}}\text{Gd} + ^{88}\text{Y}$ tracer was mixed with 5.5 mL of 2-propanol.



Kudou et al., RIKEN Accel. Prog. Rep. **42**, 265 (2009).

$^{\text{nat}}\text{Gd}_2\text{O}_3$ target thickness: $350 \pm 10 \mu\text{g}/\text{cm}^2$ (10 min, 1000 V)

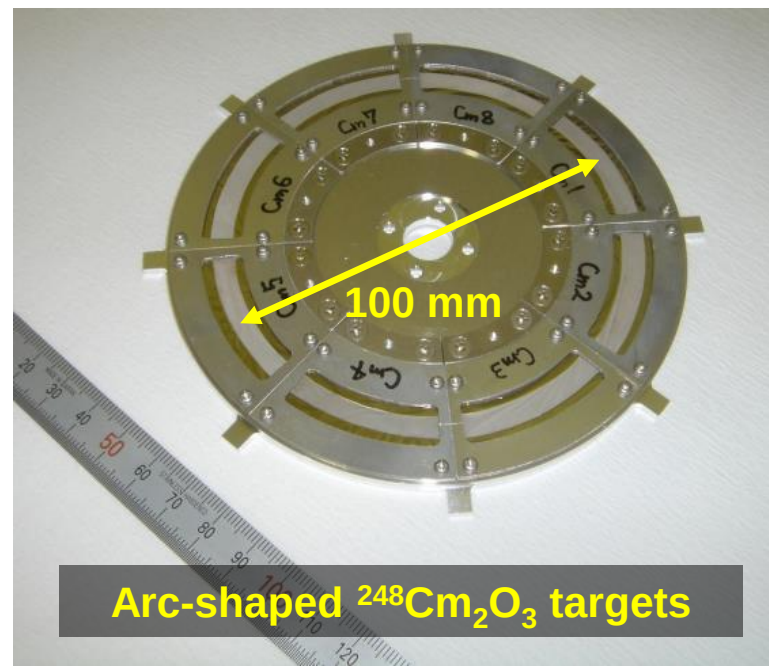
→ **Deposition time: 10 min, applied voltage: 1000 V for ^{248}Cm**

→ **Deposition yield: ~100%**

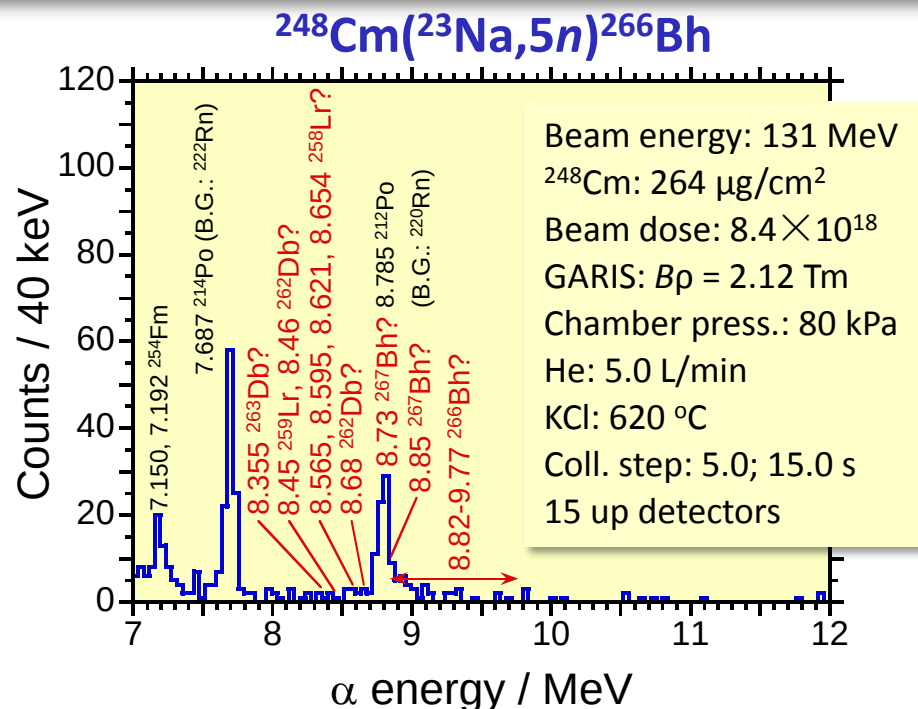
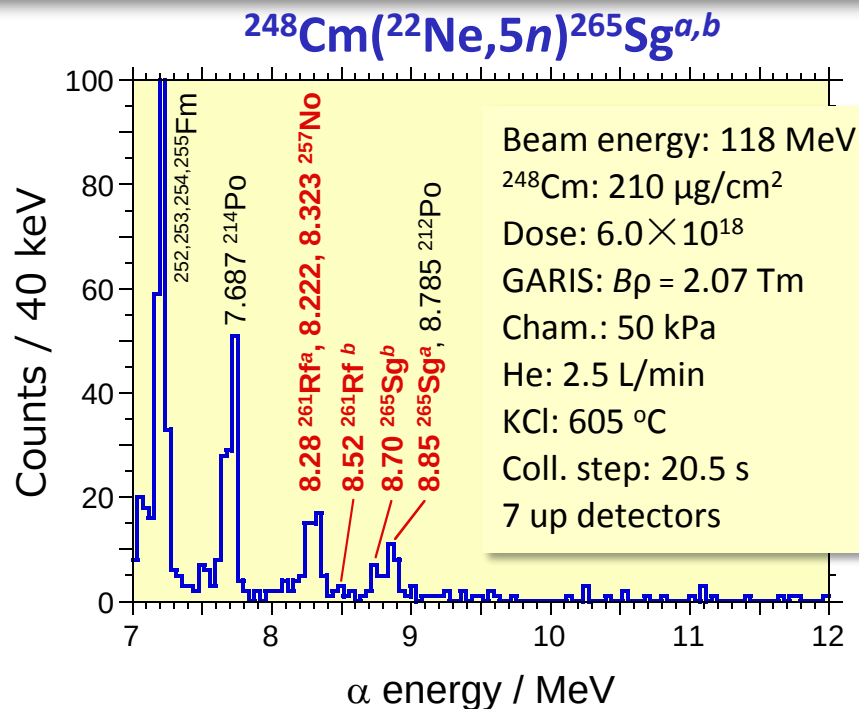
Electrodeposition of ^{248}Cm

Electrodeposition conditions

Solution	0.53 mg of ^{248}Cm in 20 μL 0.2 M HNO_3 + 5.5 mL 2-propanol
Applied voltage	1000 V
Current density	8–11 mA/cm^2
Deposition time	10 min
Backing material	2- μm Ti
Deposition area	2.04 cm^2
Target thickness*	$280 \pm 10 \mu\text{g}/\text{cm}^2$ $^{248}\text{Cm}_2\text{O}_3$
Deposition yield	$98 \pm 2\%$



Production of $^{265}\text{Sg}^{a,b}$ and ^{266}Bh using GARIS gas-jet



Nuclide	$T_{1/2}^*$ [s]	σ^* [pb]	Target [$\mu\text{g}/\text{cm}^2$]	Beam [pA]	GARIS eff. [%]	Grid [%]	Gas-jet eff. [s]	Yield at Chem. Lab. [atoms/d]
$^{265}\text{Sg}^a$	8.5	180	300	4	13	78	50	11
$^{265}\text{Sg}^b$	14.4	200	300	4	13	78	50	14
^{266}Bh	10.0	57	300	4	15	78	50	4

*Haba et al., Phys. Rev. C **85**, 024611 (2012).;

EPJ Web of conferences **131**, 07006 (2016).

Assumption: gas-jet transport time: 2.7 s

Towards aqueous chemistry of Sg and the heavier SHEs

Pioneering cation-exchange studies of Sg in HNO_3/HF and HNO_3

Schädel et al., Radiochim. Acta **77**, 149 (1997).; Radiochim. Acta **83**, 163 (1998).

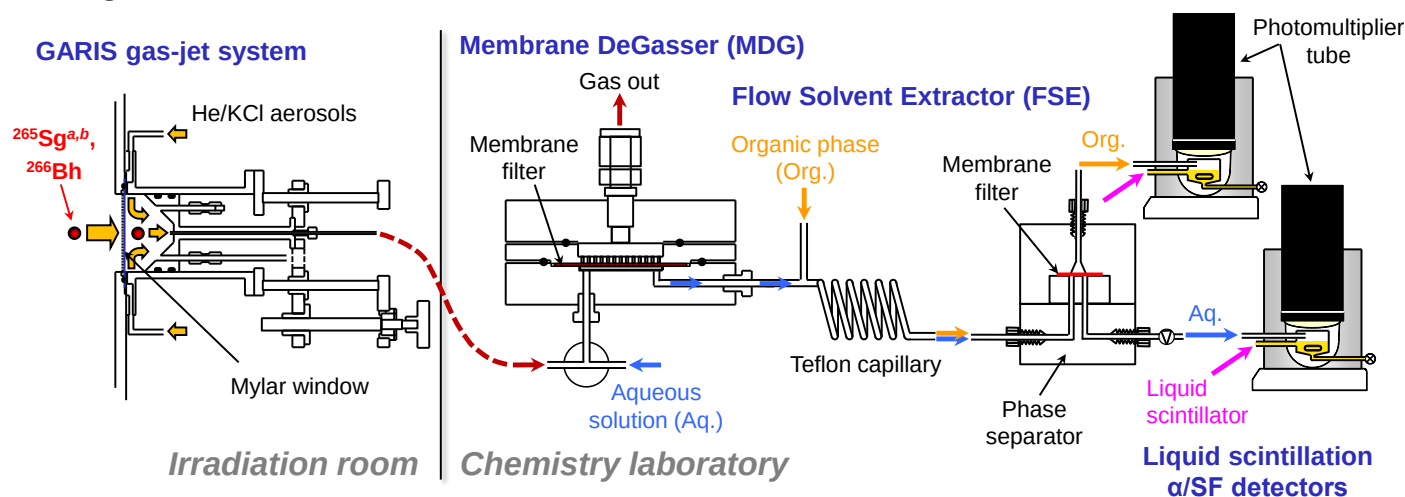
- Batch-wise column chromatography apparatus (ARCA) with Si detectors

4 α - α correlations of $^{261}\text{Rf} \rightarrow ^{257}\text{No} \rightarrow$

- No aqueous chemistry studies of Sg and the heavier elements have been reported since 1998.

Development of a continuous rapid solvent extraction apparatus

RIKEN–Niigata Univ.–JAEA–Osaka Univ.–Tohoku Univ.–IMP–Univ. Oslo collaboration



Continuous dissolution, solvent extraction, and α /SF detection with LS

→ **Event rates of $^{265}\text{Sg}^{a,b}$ and ^{266}Bh can be increased by 50-100 times**

3. Summary and future perspectives

Summary

(1) AVF + gas-jet

- Aqueous chemistry studies of Rf and Db are ongoing using the stationary $^{248}\text{Cm}_2\text{O}_3$ target and gas-jet transport system.

(2) RILAC + GARIS + gas-jet

- A rotating $^{248}\text{Cm}_2\text{O}_3$ target for GARIS was prepared by a molecular plating method.
- We successfully produced $^{265}\text{Sg}^{a,b}$ and ^{266}Bh and extracted them to the chemistry laboratory using the GARIS gas-jet system.
- A continuous rapid solvent extraction apparatus is under development for the future aqueous chemistry of Sg and Bh.

Future perspectives

(1) Increase of production yields of SHEs

Upgrade of RILAC: 28 GHz ECR & SC-QWR

- **Beam intensity of RILAC: X 5**
 - **Thickness of target: X 1.5**
 - **GARIS III: X 1.5**
- Production yields of SHEs: X10**

⇒ Precision chemical data based on enough counting statistics

(2) Development of target preparation techniques

- Development of durable targets for long-term and highly intense beam irradiations
- Preparation of other actinide targets (e.g. ^{238}U , ^{243}Am , ^{244}Pu) for Cn, Nh, and Fl chemistry

(3) Development of a new aqueous chemistry apparatus and experimental methods

- Development of a continuous rapid extraction apparatus

Collaborators

RIKEN	H. Haba, T. Yokokita, N. Sato, Y. Wang, K. Morimoto, D. Kaji, T. Tanaka, K. Morita, D. Mori, A. Yoneda, M. Osanai
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Niigata Univ.	S. Goto, H. Kudo
Kanazawa Univ.	A. Yokoyama
IMP	F. Fan, Z. Qin
Univ. Oslo	J. P. Omtvedt

Thank you for your kind attention.