

Present Status of SHE Chemistry at JAEA

Tetsuya K. Sato
Advanced Science Research Center
Japan Atomic Energy Agency

Research Group for Heavy Element Nuclear Science

Group leader: A.Andreyev

Nucl. Phys.

K.Nishio (Manager, Exp.)
H.Koura (Theory)
Y.Utsuno (Theory)
K.Hirose (Exp.)
H.Makii (Exp.)
R.Orlandi (Exp.)

Adviser : Y. Nagame

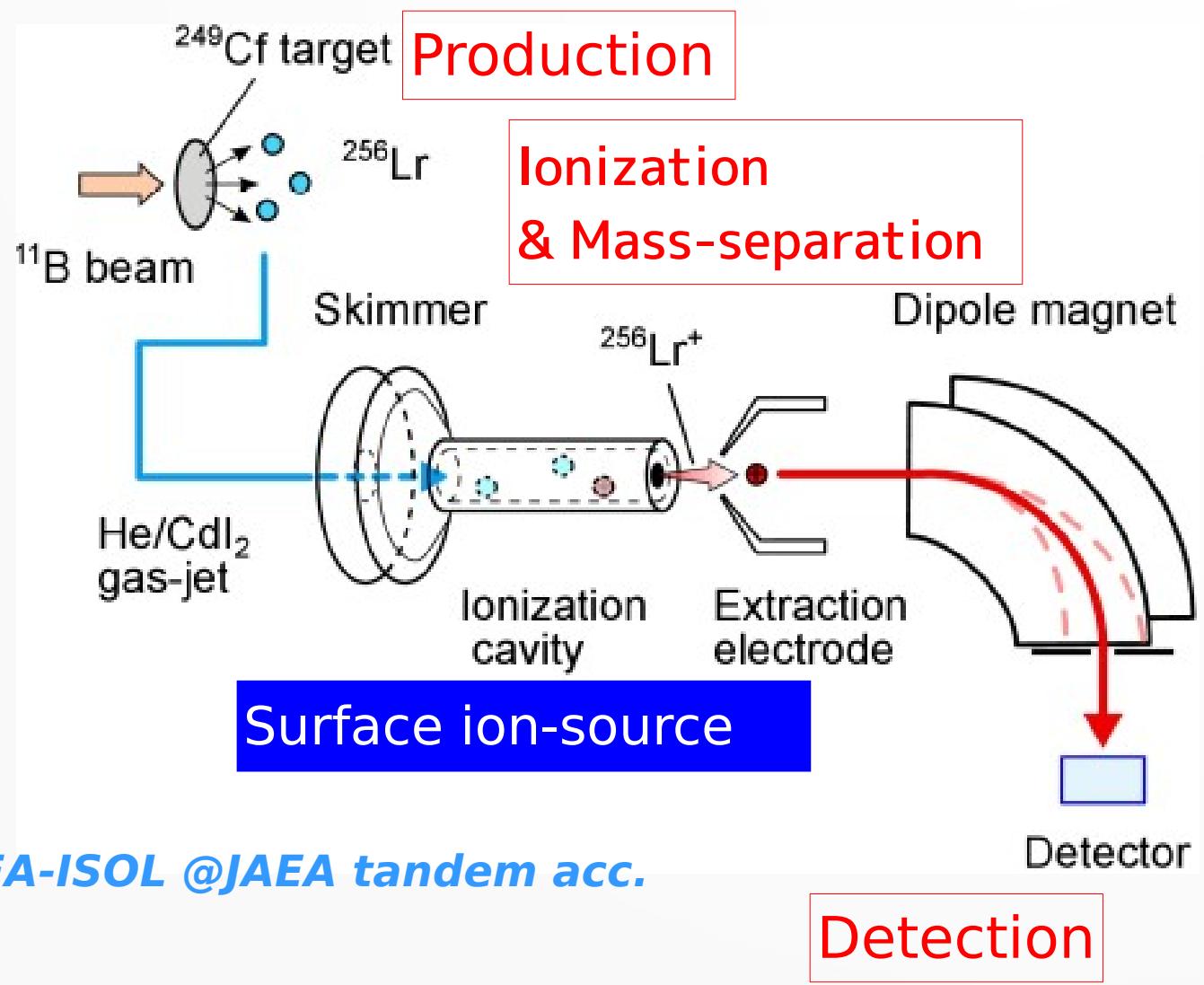
K.Ratha (Exp.)
S.Tanaka(Theory)
R.Yanagihara(Exp.)

Nucl. Chem.

K.Tsukada(Exp.) T.Tomitsuka (Exp.)
M.Asai (Exp.) K.Tokoi (Exp.)
T.K.Sato (Exp.) H.Suzuki (Exp.)
~~A.Toyoshima~~
Y.Ito (Exp.)
N.M.Chiera (Exp.)

Staff: 11 (6+4)
PostDoc: 1
Student : 6 (3 + 3)
Total: 18

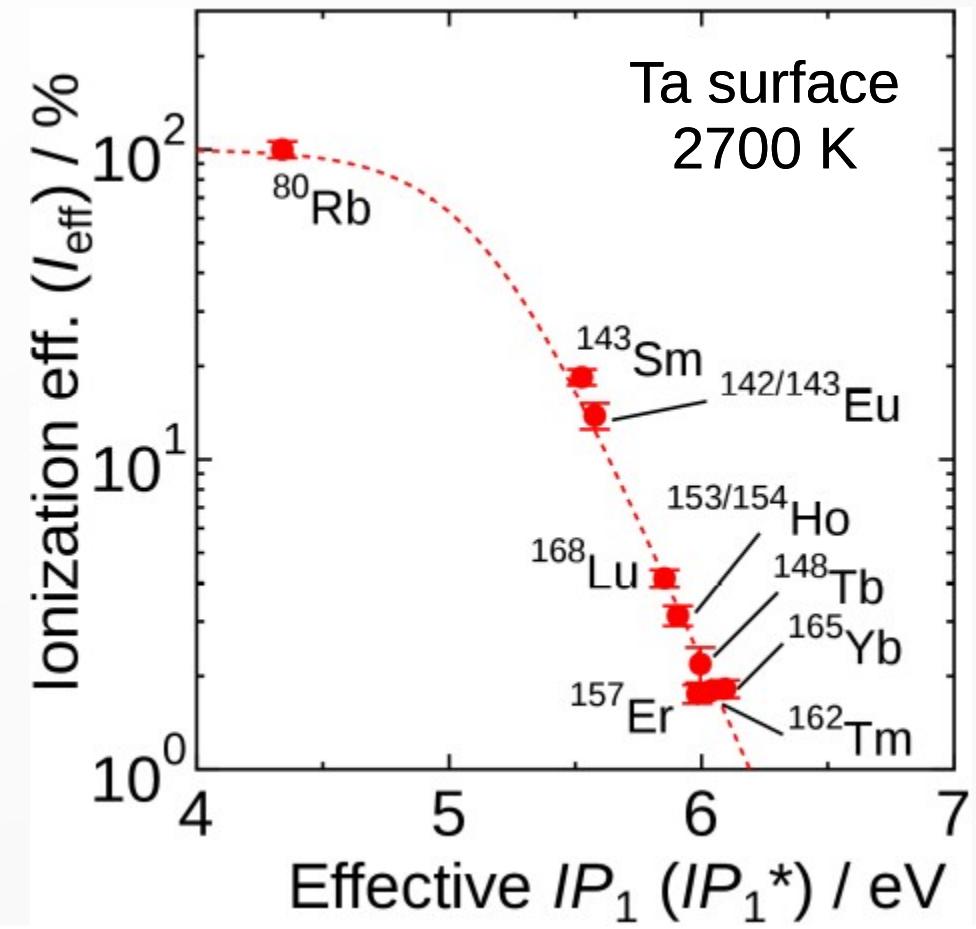
IP₁ of Heavy Actinides



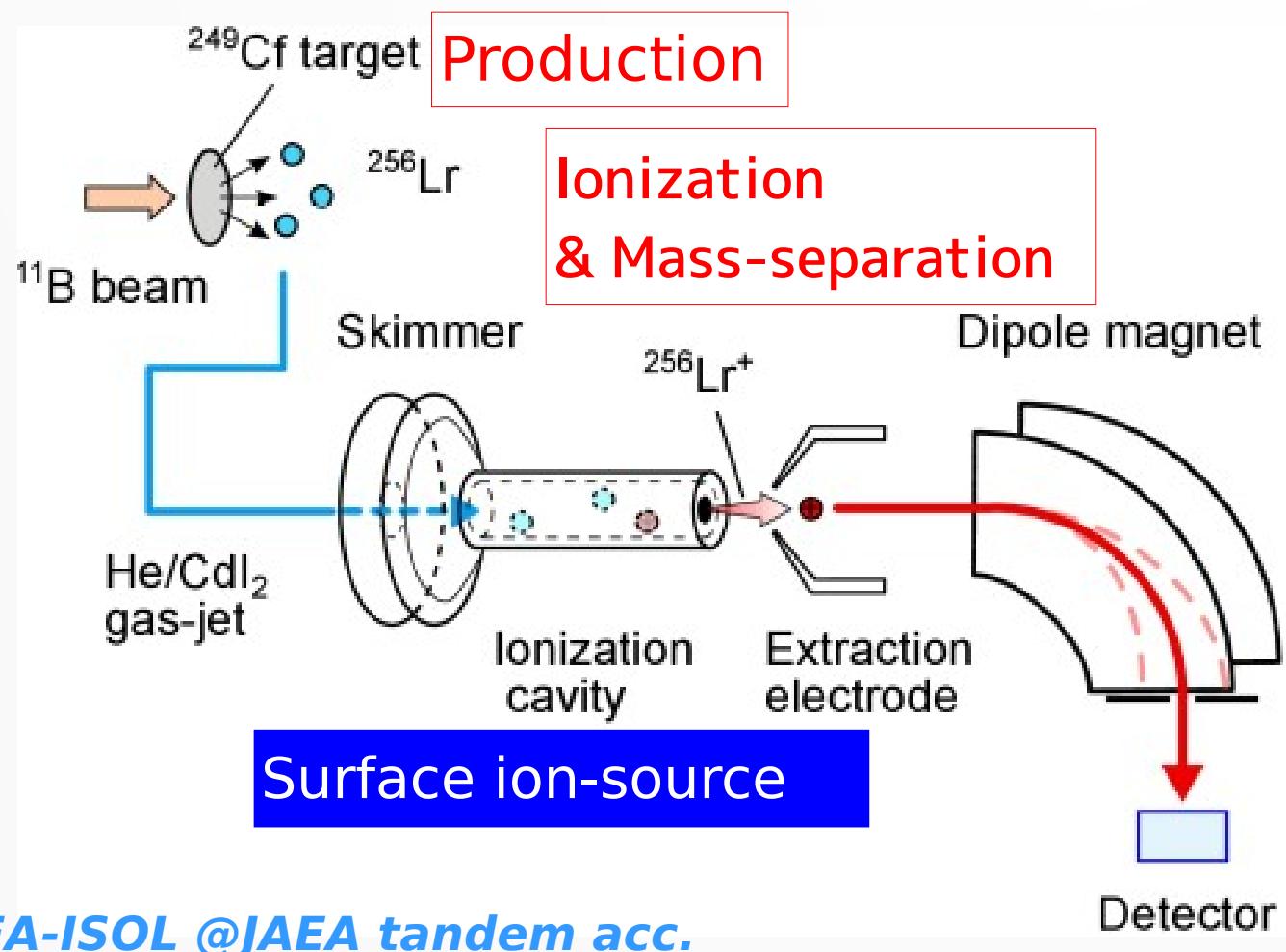
Saha-Langmuir eq.

$$I_{\text{eff}} = \frac{N \exp((\phi - IP_1^*)/kT)}{1 + N \exp((\phi - IP_1^*)/kT)}$$

Effective IP₁(IP₁^{*}) : IP₁^{*} = IP₁ - kT ln (Q_i/Q_o)



IP₁ of Heavy Actinides

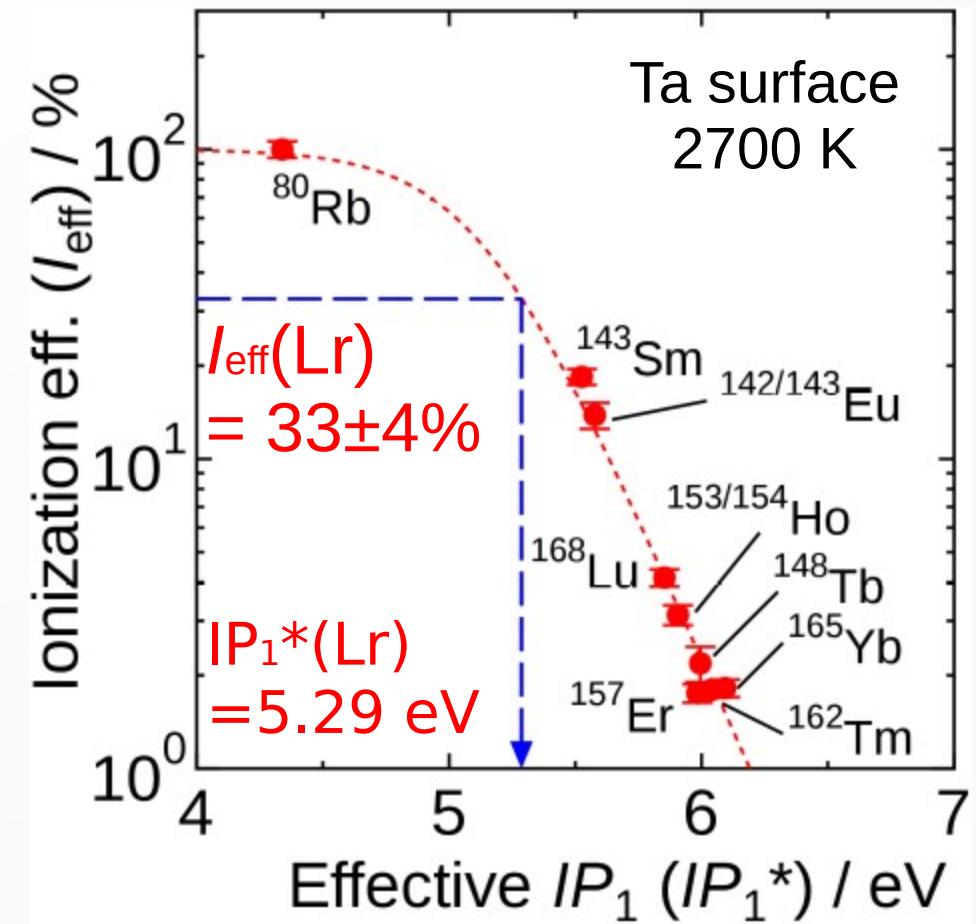


JAEA-ISOL @JAEA tandem acc.

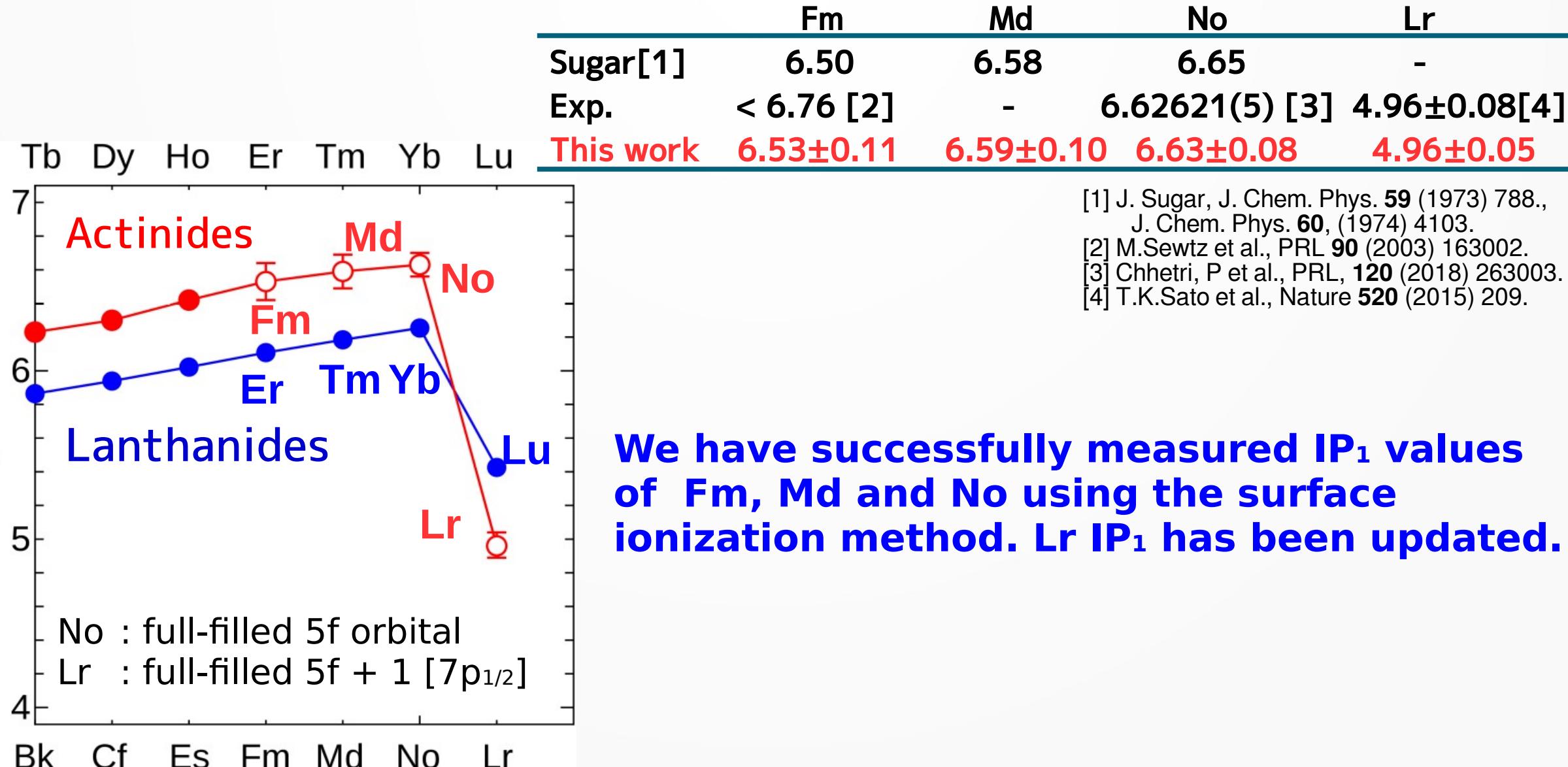
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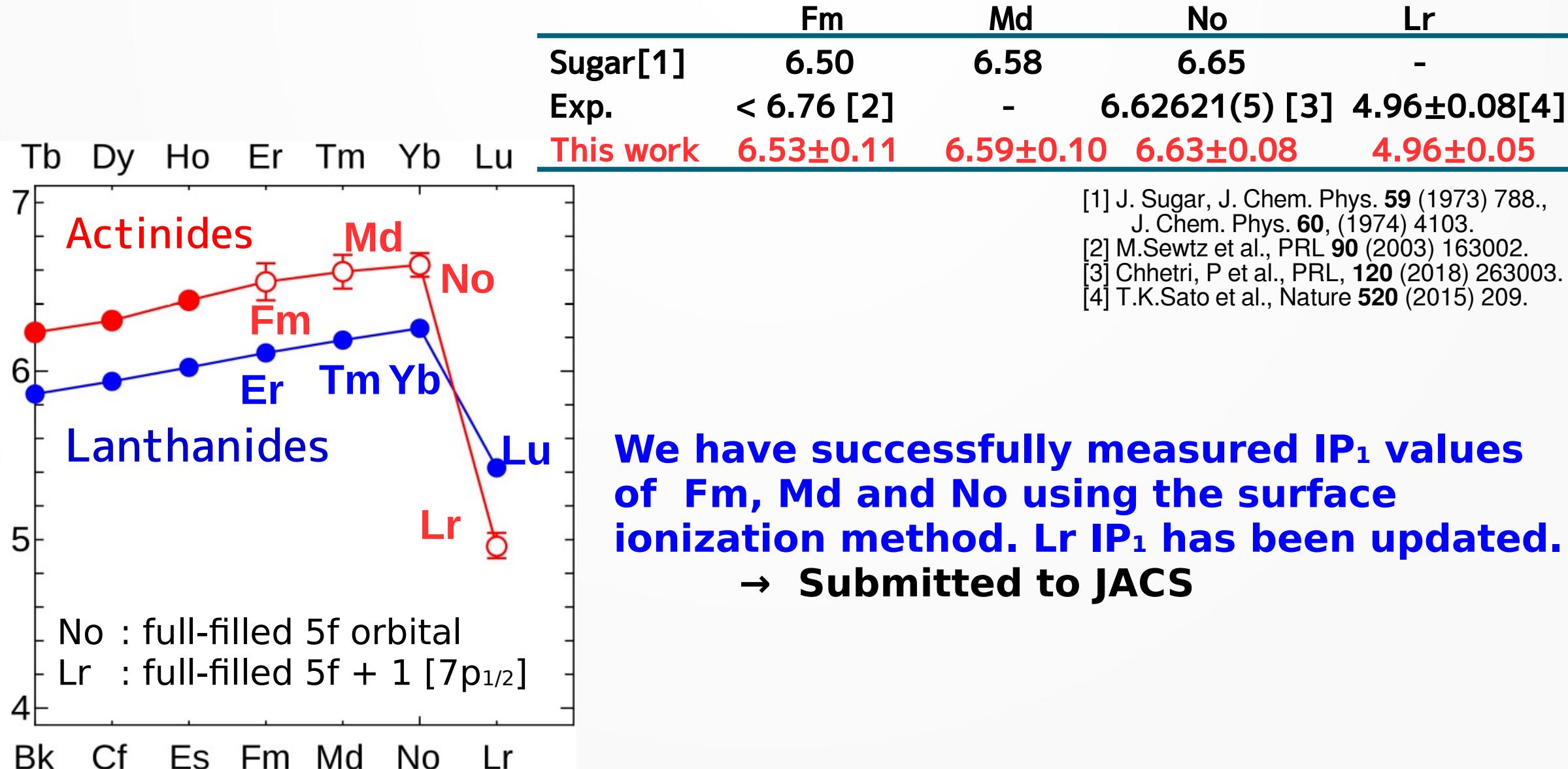
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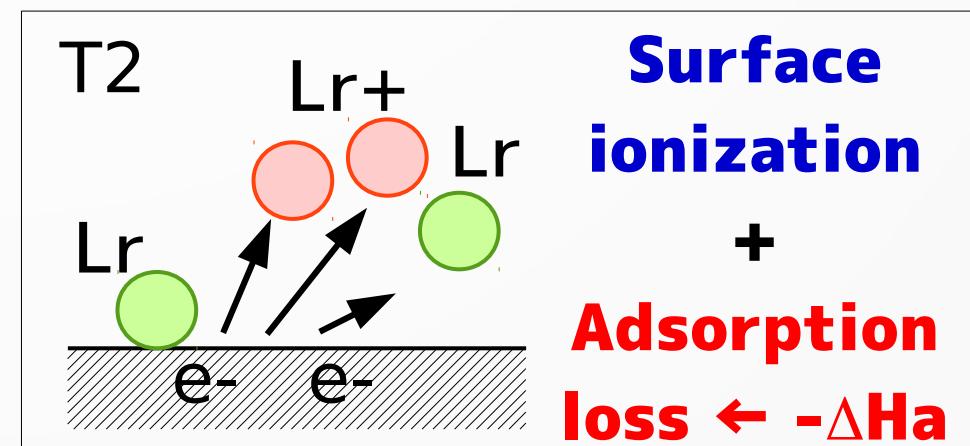
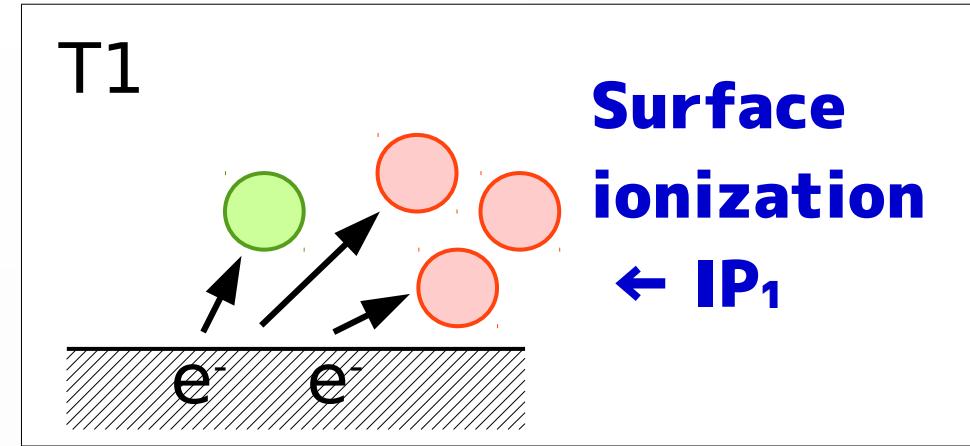
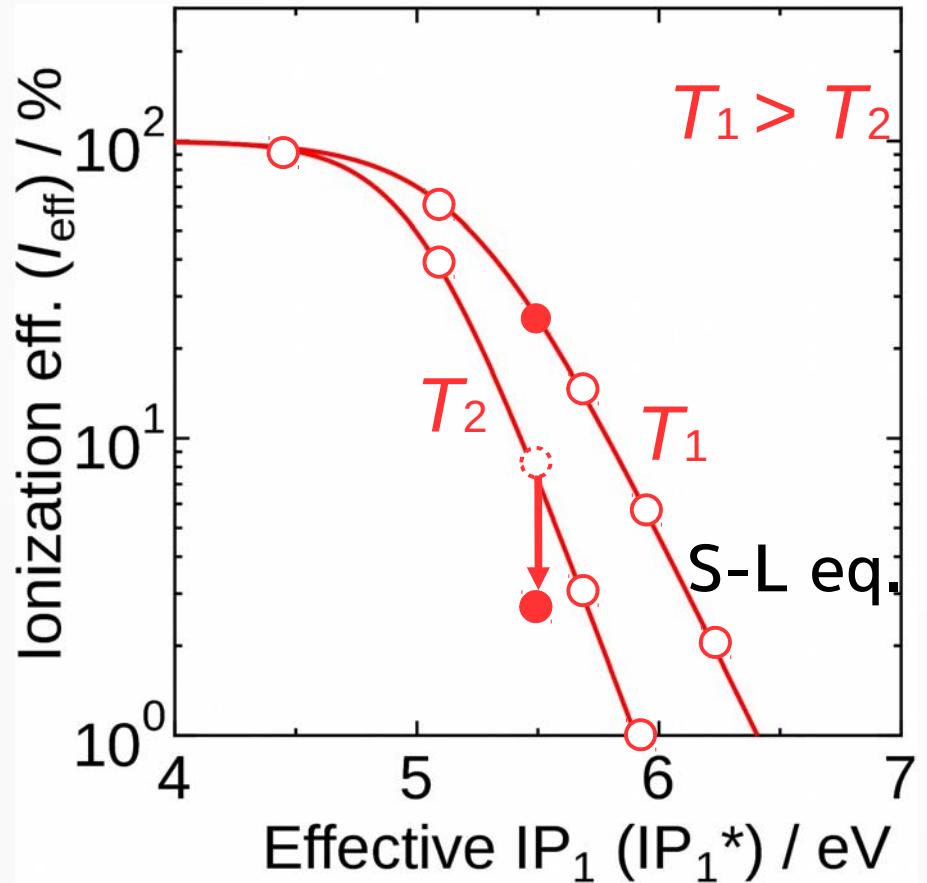
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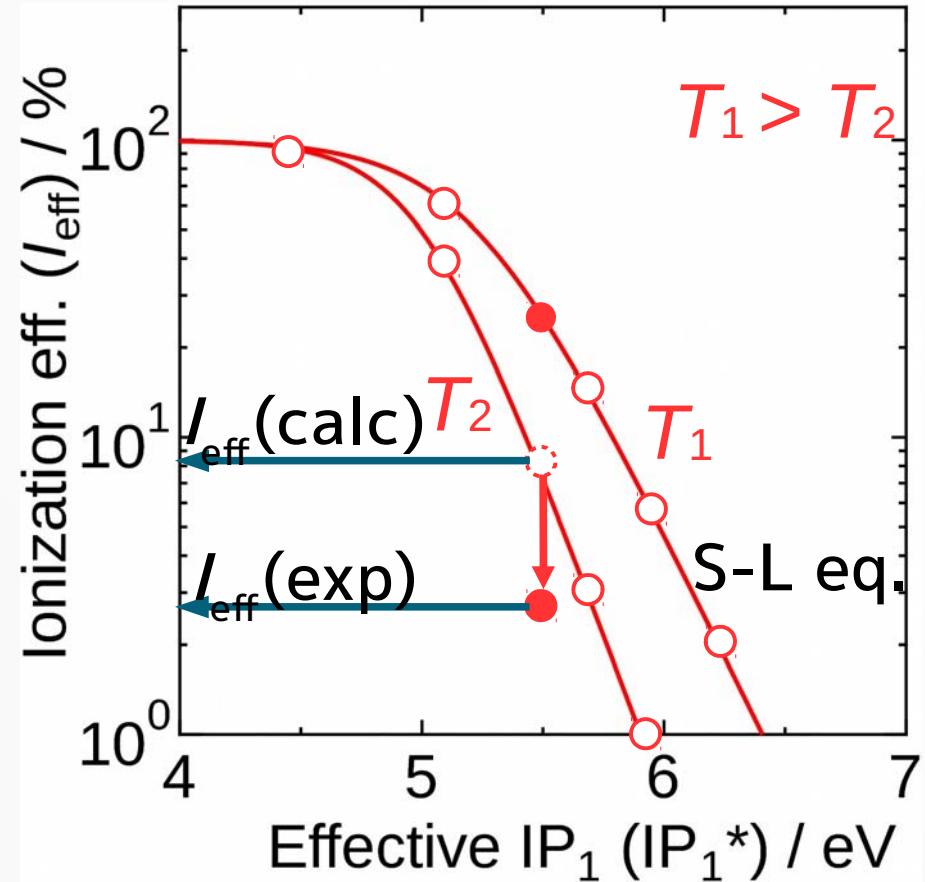
Adsorption of Lr atom



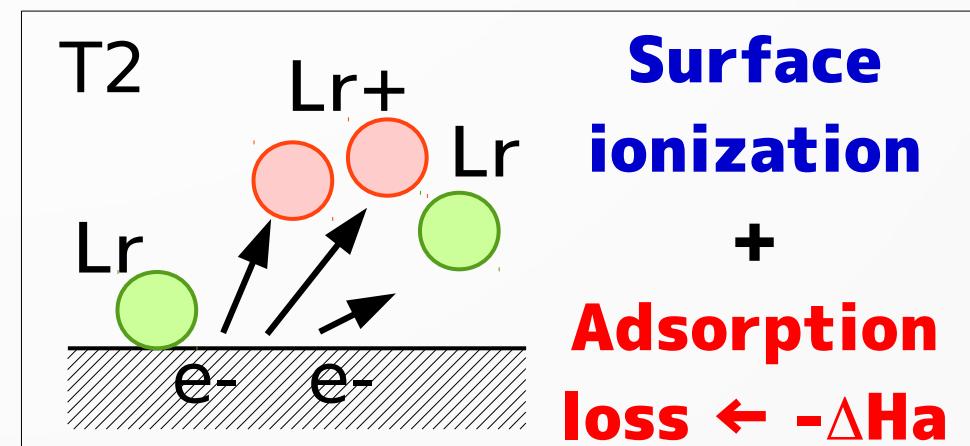
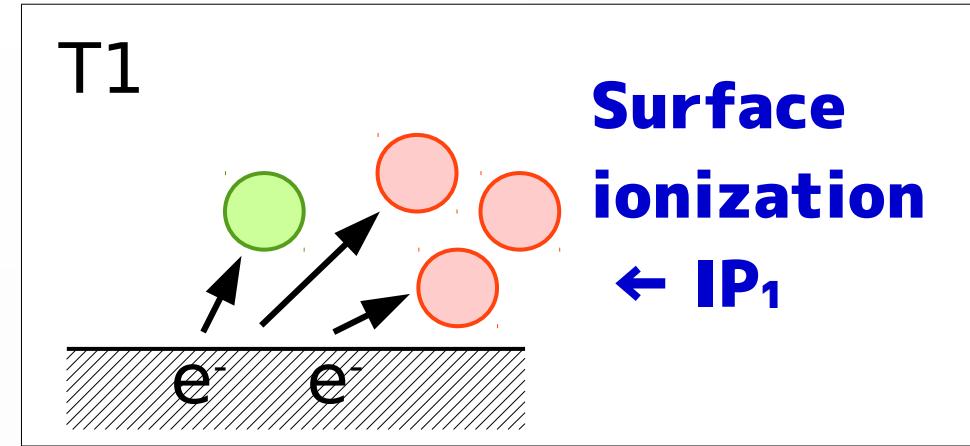
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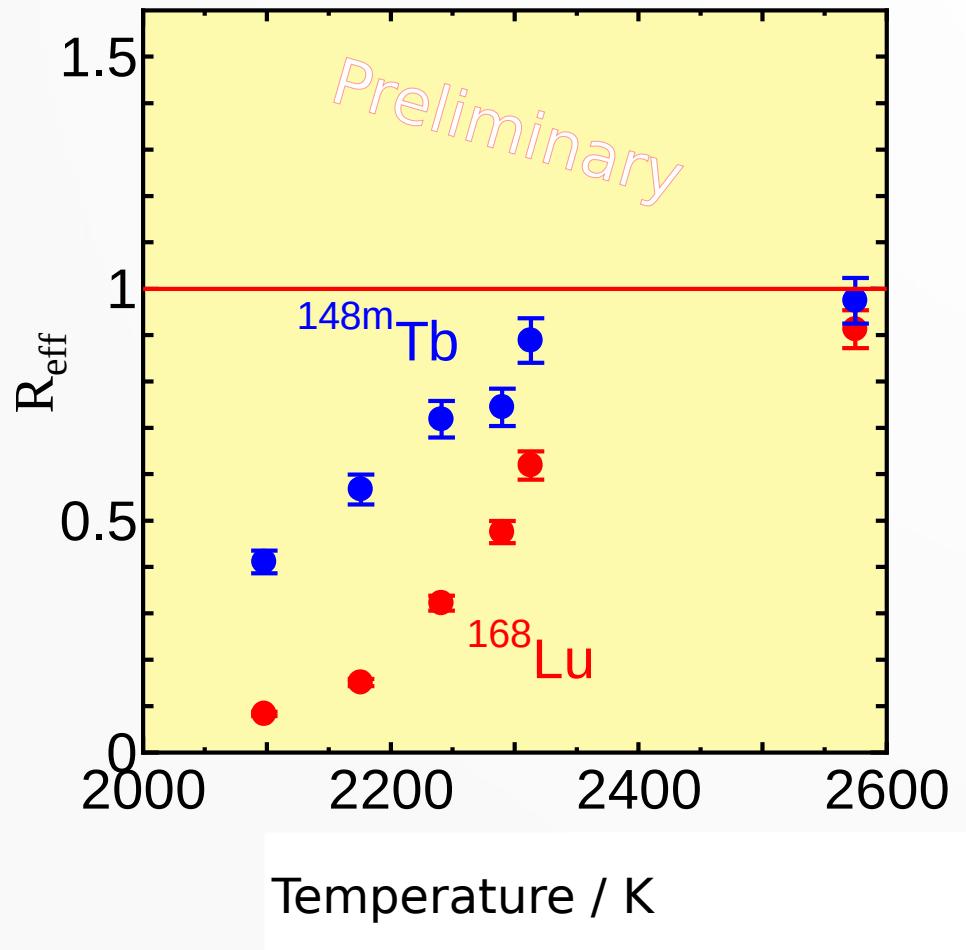
$$R_{\text{eff}} = I_{\text{eff}}(\text{exp}) / I_{\text{eff}}(\text{calc})$$



Saha-Langmuir eq.

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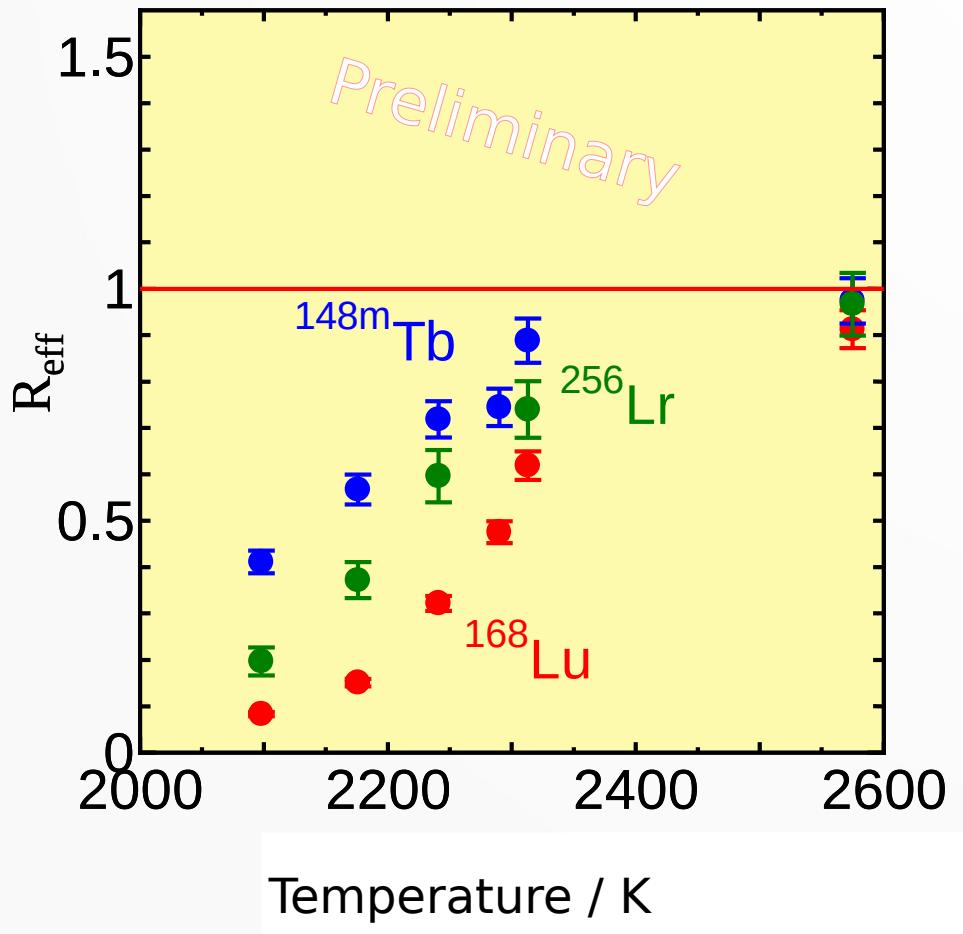
Adsorption of Lr atom



Isotope	$T_{1/2}$	$-\Delta H_a/\text{kJ mol}^{-1}$
^{148m}Tb	2.2 min	578
^{168}Lu	5.5 min	610

[1] R.Eichler, private comm.

Adsorption of Lr atom

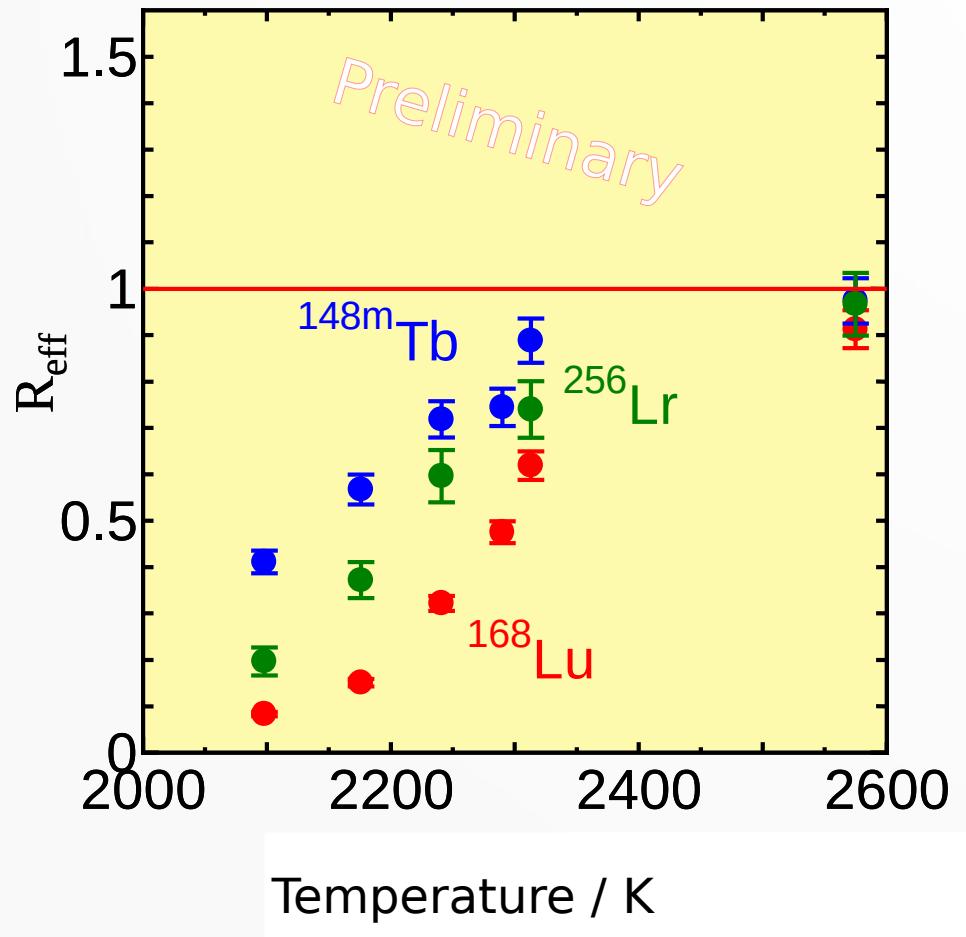


Isotope : ^{256}Lr ($T_{1/2} = 27$ s)
Beam : ^{11}B (67.9 MeV)
Target : ^{249}Cf (260 $\mu\text{g}/\text{cm}^2$)

Isotope	$T_{1/2}$	$-\Delta H_a/\text{kJ mol}^{-1}$	[1]
^{148m}Tb	2.2 min	578	$< ^{256}\text{Lr}$ (27 s)?
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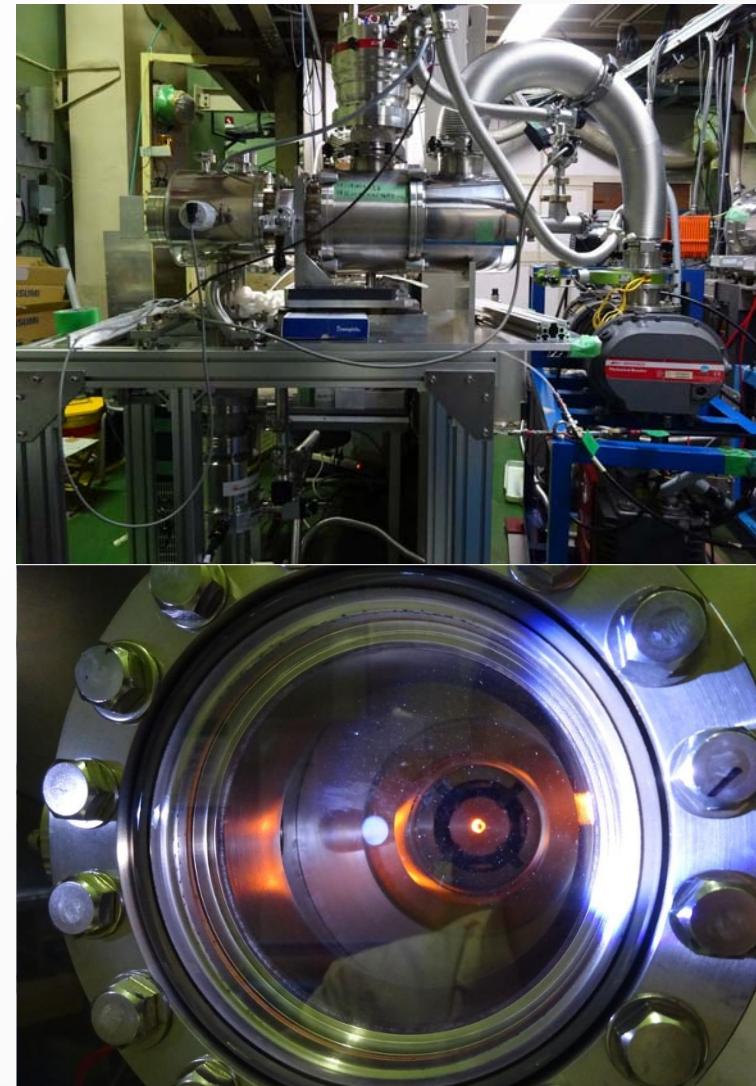
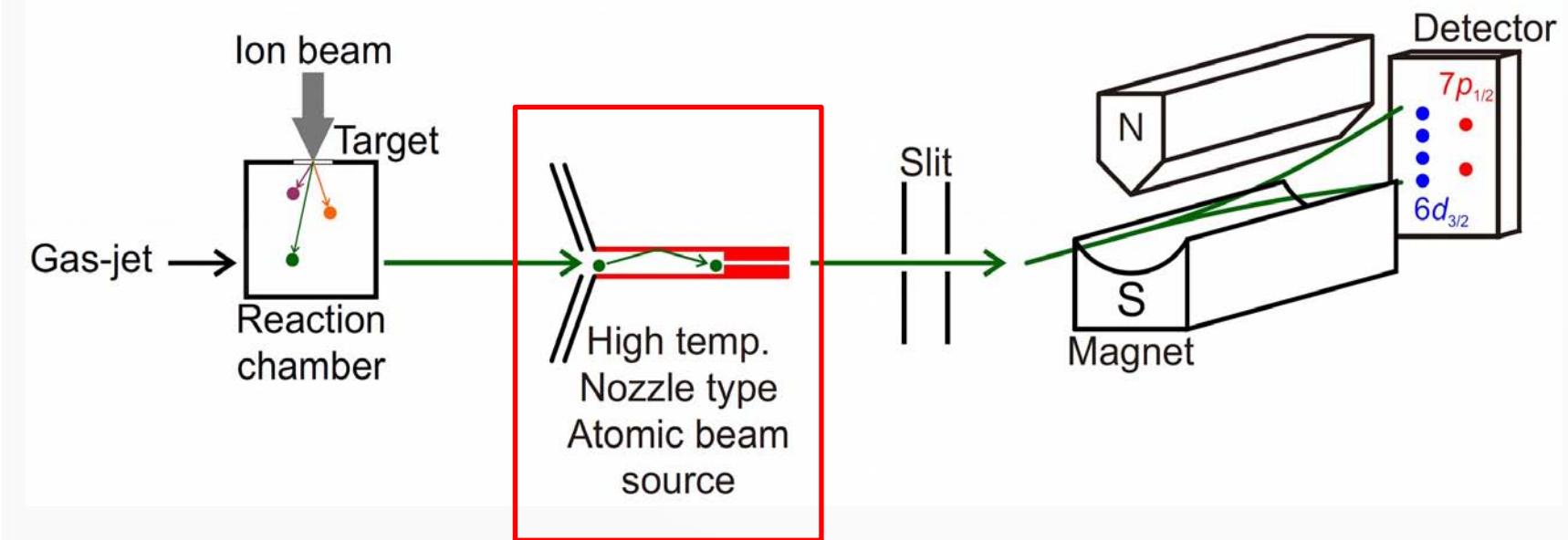
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- $\Delta H_a(\text{Exp.})$: Monte Carlo simulation
(Zvara model + Surface ion.)
- $\Delta H_a(\text{Theory})$: 535 kJ/mol (V. Pershina)

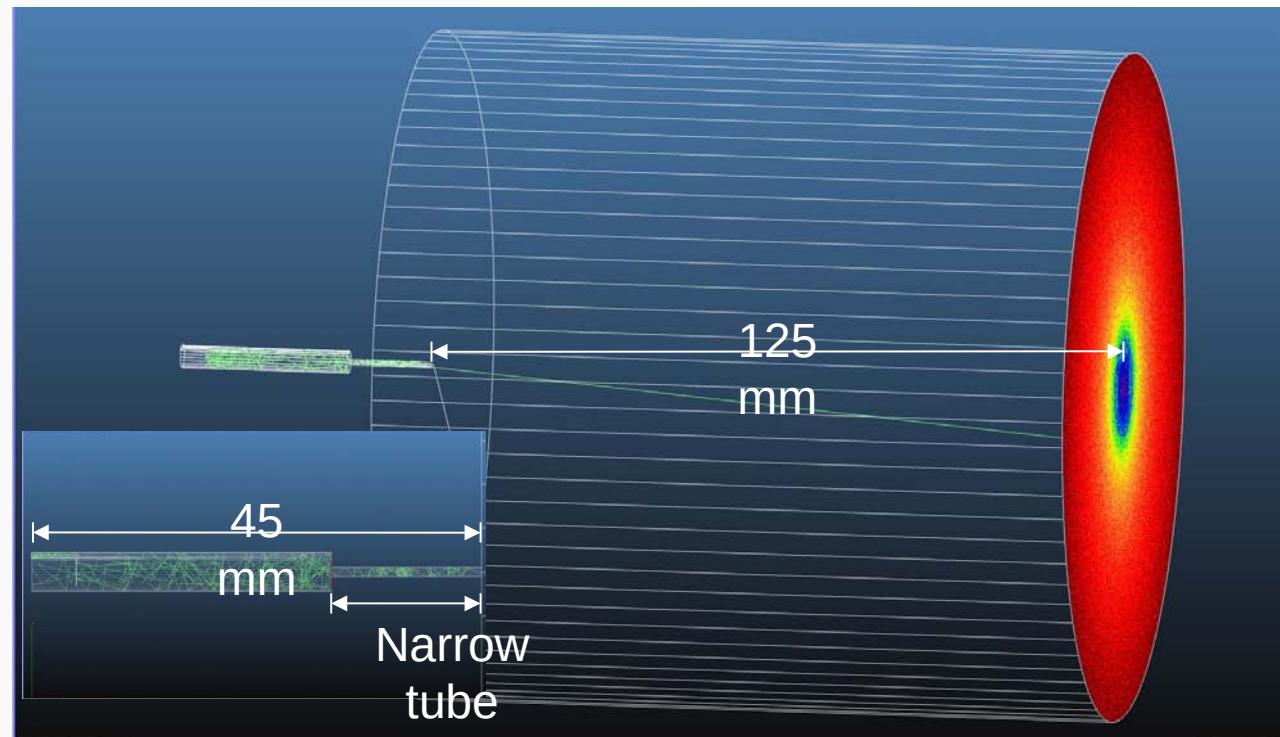
Atomic Beam Source

Atomic spin

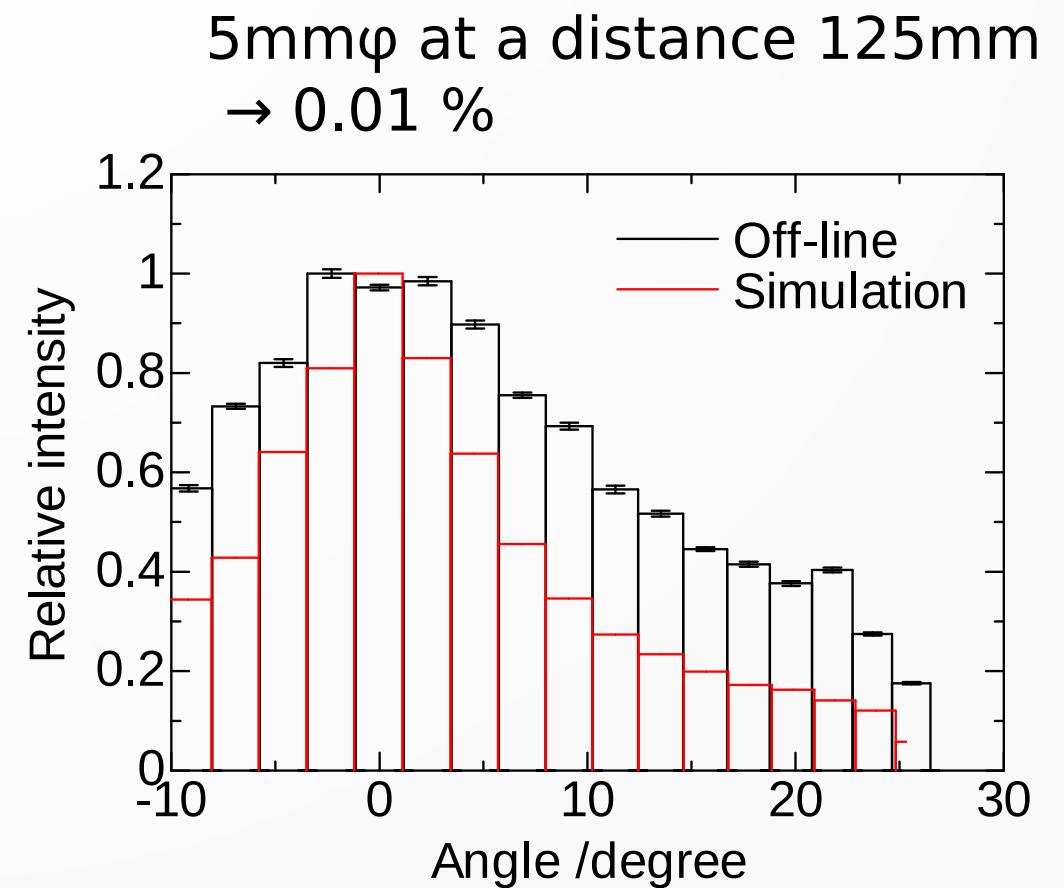
The electronic configuration of Lr atom can be confirmed directly by a determination of the atomic spin.



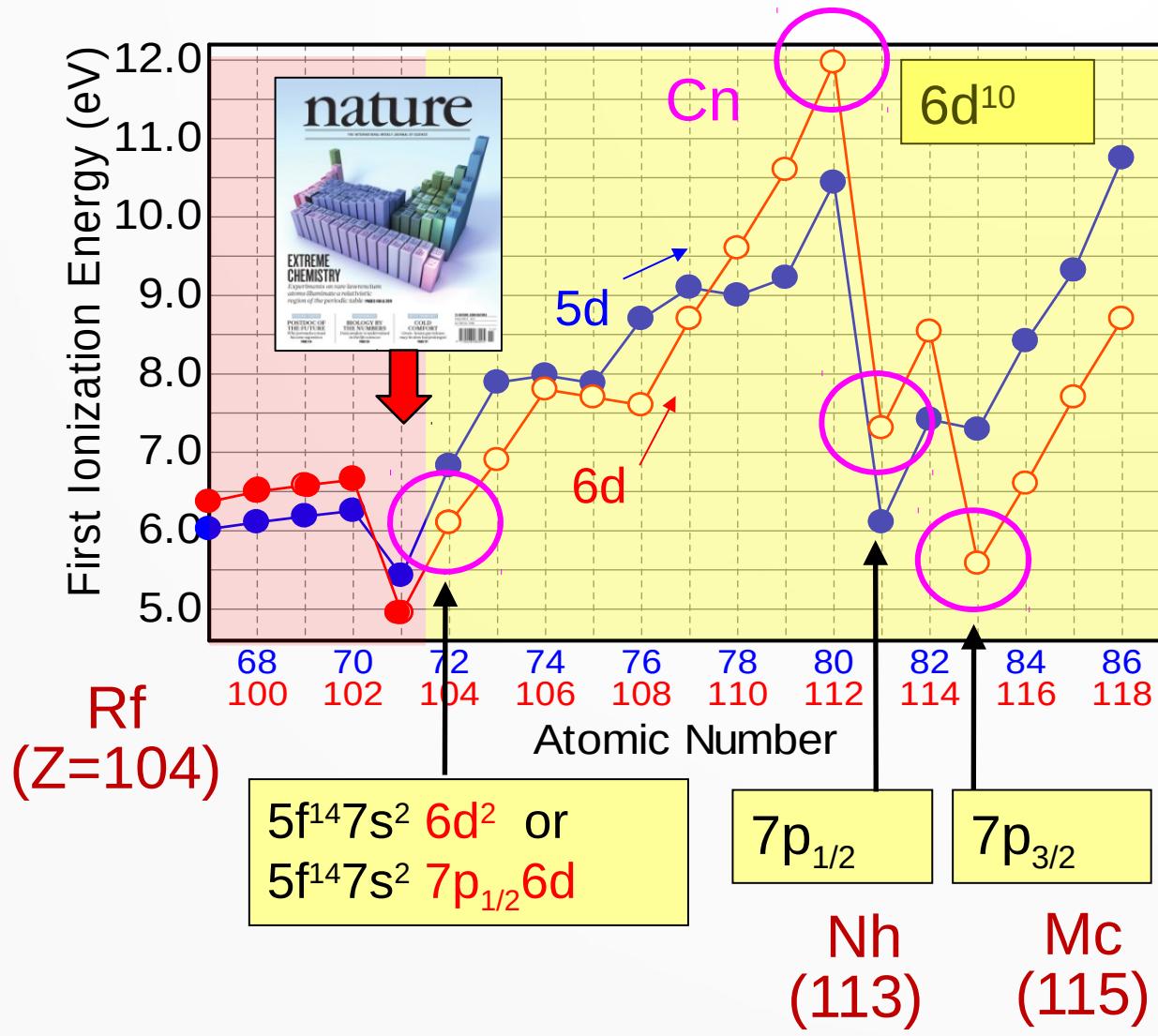
Atomic Beam Source



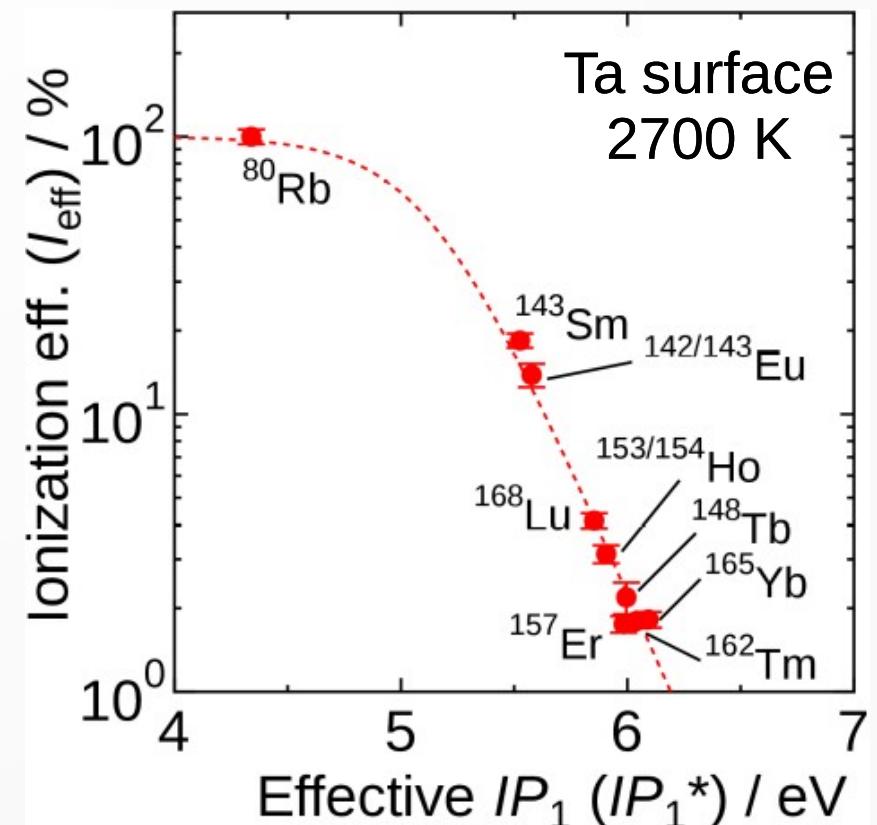
Beam: CdI₂



IP_1 of $Z > 103$

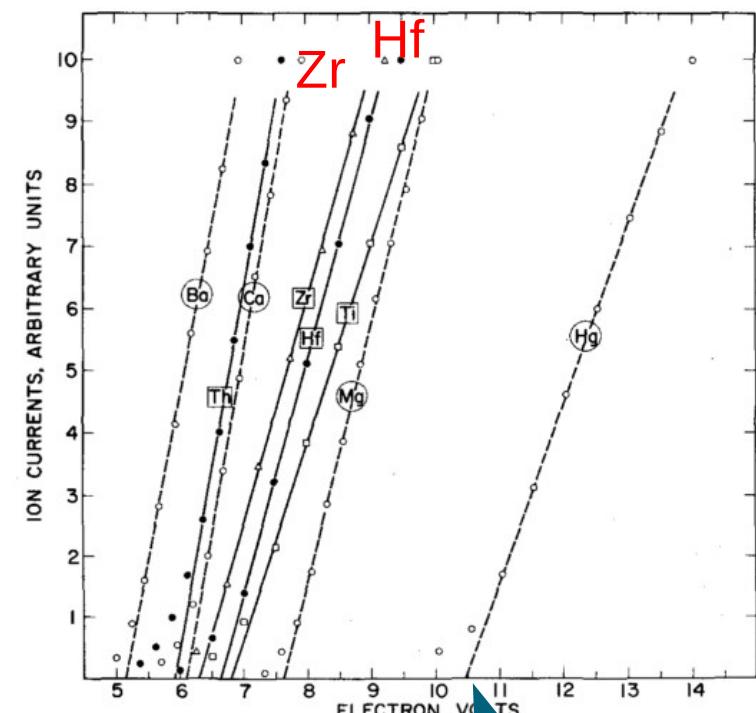
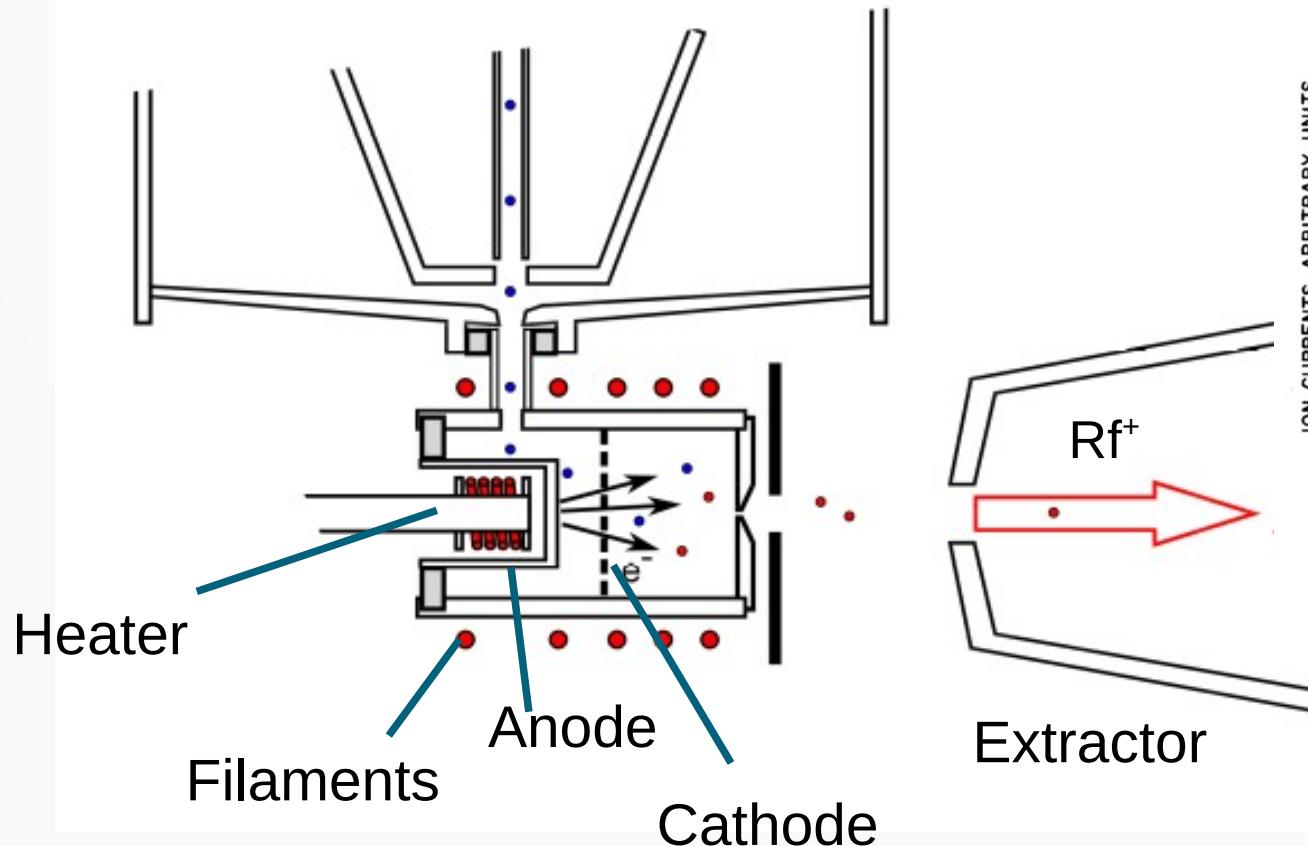


The surface ionization method cannot be applied to elements which $\text{IP}_1 > 6 \text{ eV}$



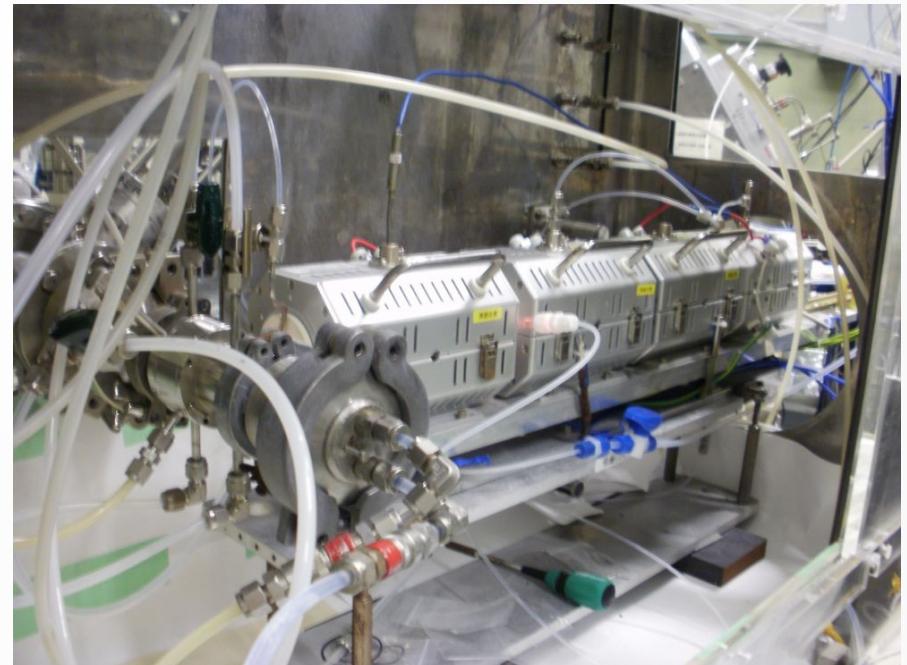
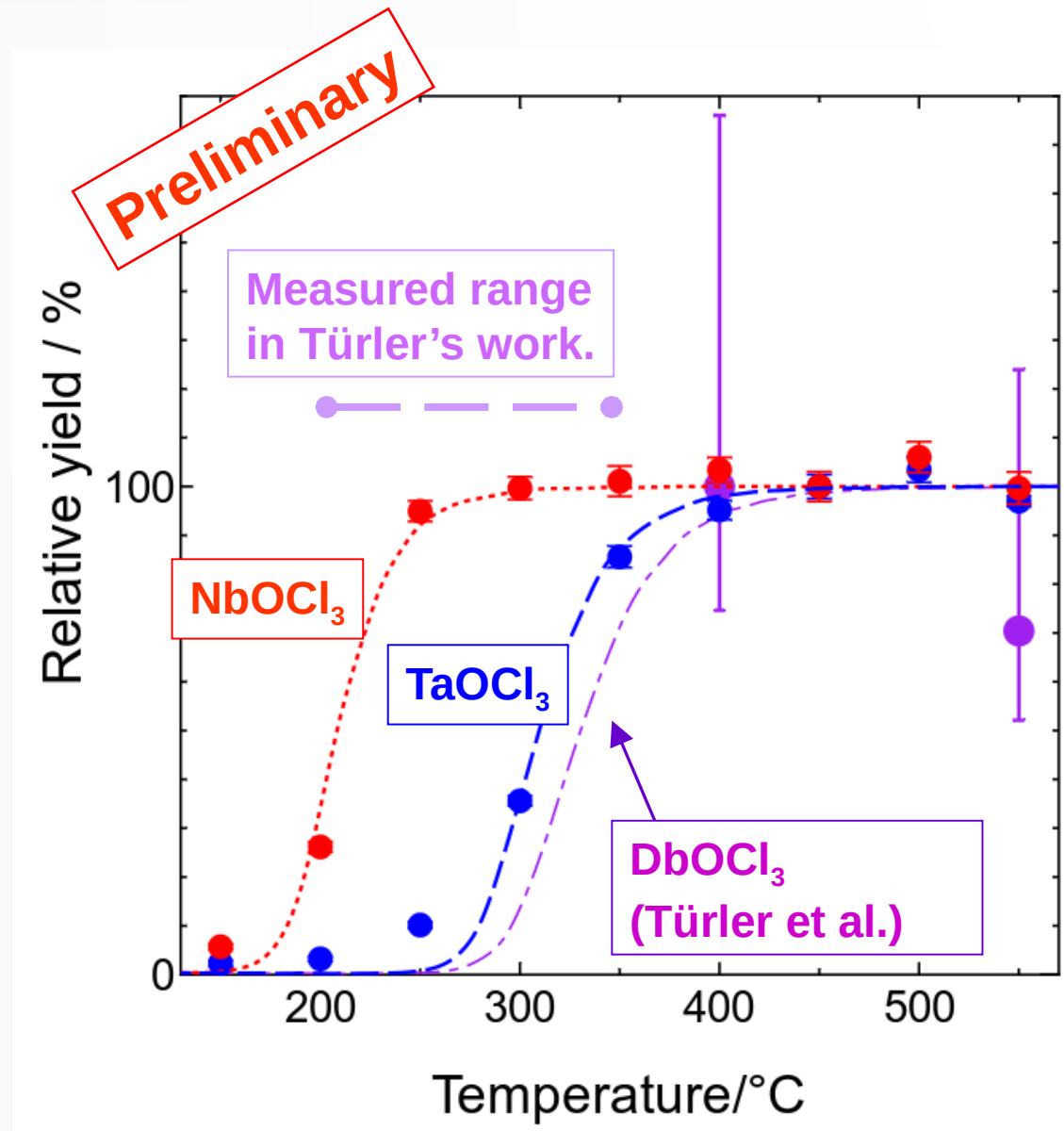
IP_1 of $Z > 103$

IP_1 measurement > 103



E.G.Rauh & J.Ackermann
J. Chem.Phys. 60 (1974) 1396.

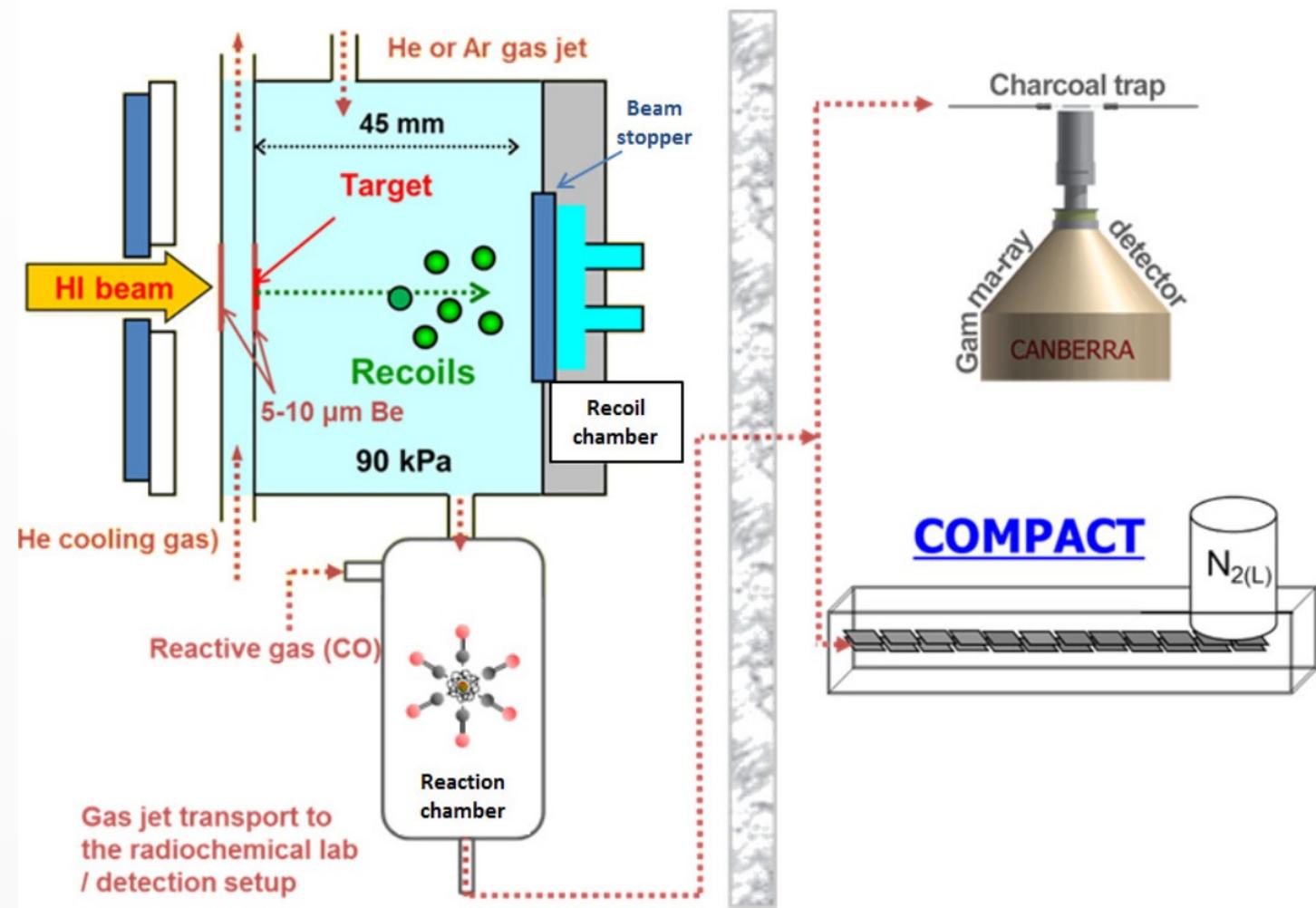
Volatility of Oxychlorides of Group 5



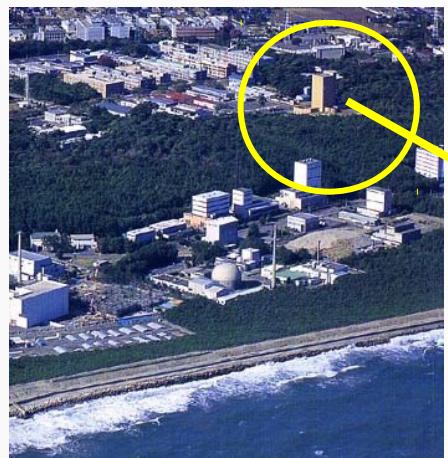
Adsorption enthalpy :
-ΔHa (this work)
Nb : 105±3 kJ/mol
Ta : 126±2 kJ/mol

International collaboration

- In-situ production of M(CO)x / GSI
 - REIMEI budget
 - M(CO)x production w/o a separator
 - → Group 6~10



JAEA Tandem Acc.(1982~)



The JAEA Tandem accelerator is one of the largest electrostatic accelerators in the world

Terminal voltage	2.5~18 MV
Pressure vessel	diameter, 8.3m; height, 26.6m (Insulating gas: 6.7 kg/cm ² of pure SF ₆ gas)
Ion sources	3 Negative ion-source ECR ion-source (in high voltage terminal)
Typical beam current	3 pμA (H), -1.0 pμA (Li – F), 0.1 pμA (Ar, Xe)
Target room	Five rooms, 10 beam lines

PAC

- 2 times / year (May & Dec.)
- If you apply to next PAC (Dec.2018) and the application is approved, you can start the experiment after Feb. 2019.

More detailed information

→ <https://ttandem.jaea.go.jp/index-eng.html>

Or google “JAEA tandem”



Summary

- IP₁ of heavy actinides : finished
- Adsorption behavior of Lr atom : almost finished
- Development of atomic beam source : on-going
- IP₁ of Element (Z>103) : planning
- Oxychlorides of Group 5 elements : on-going

Thank you very much for your kind attention!!



Grant-in-Aid for Scientific Research