

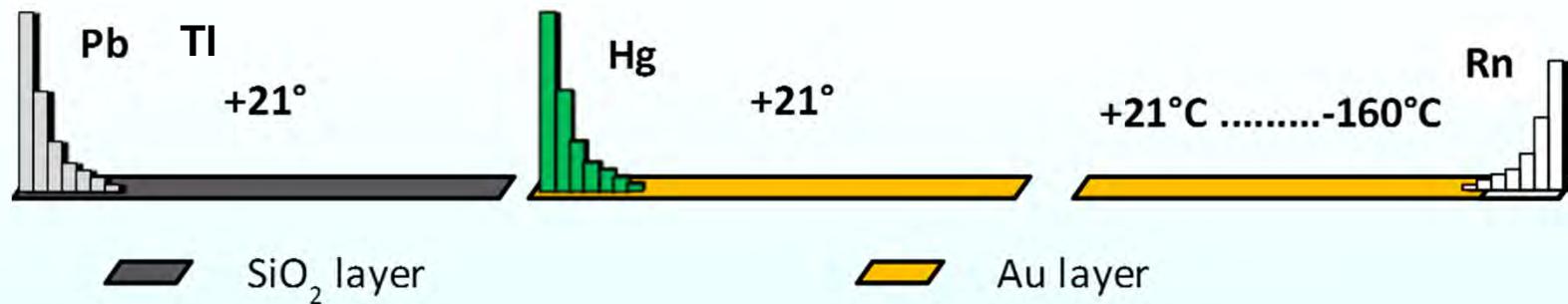
# Theoretical Chemistry of Elements 112, 113 and 114

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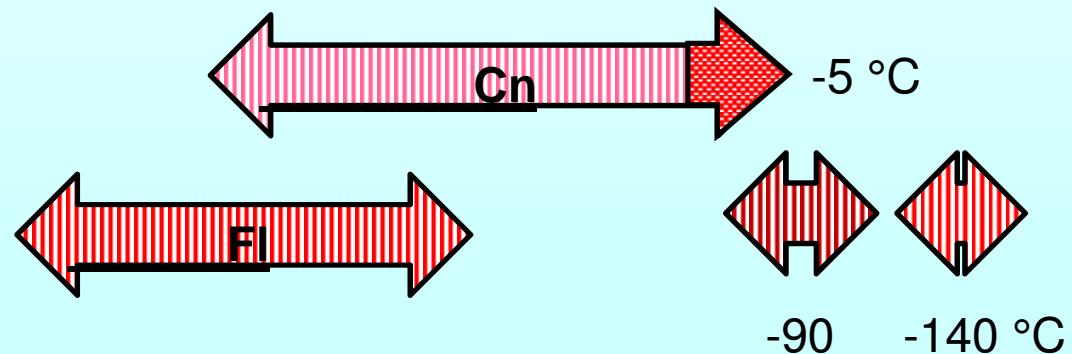
# Experiments on Adsorption of Cn, E113 and Fl on Gold and Quartz



Theory: Cn, E113 and Fl ???

Experiment:

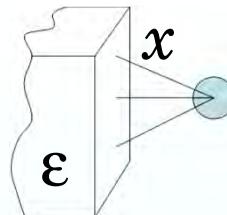
[Eichler, et al. (2007, 2010)  
Yakushev, et al. (2014)]



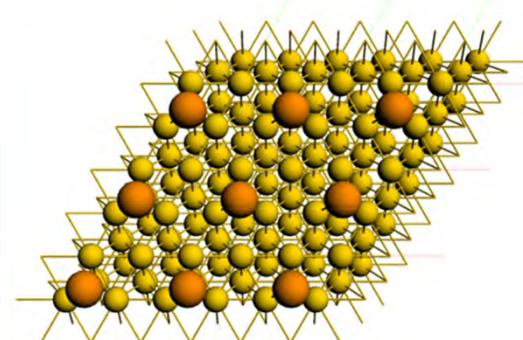
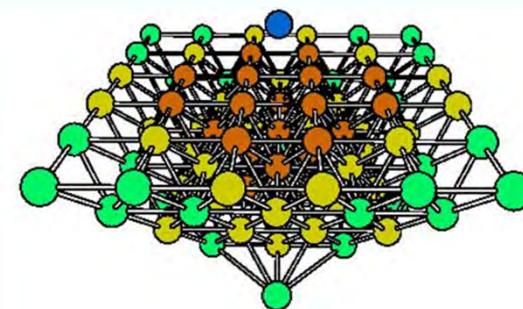
# Calculations of Adsorption Energy

- Adatom-slab model (inert surface)

$$E(x) = -\frac{3}{16} \left( \frac{\epsilon - 1}{\epsilon + 2} \right) \frac{\alpha_{mol}}{\left( \frac{1}{IP_{slab}} + \frac{1}{IP_{mol}} \right) x^3}$$



- Cluster model  
(molecular codes,  
convergence with the cluster size)
- Periodic calculations  
(slabs, supercells,  
periodic boundary conditions)

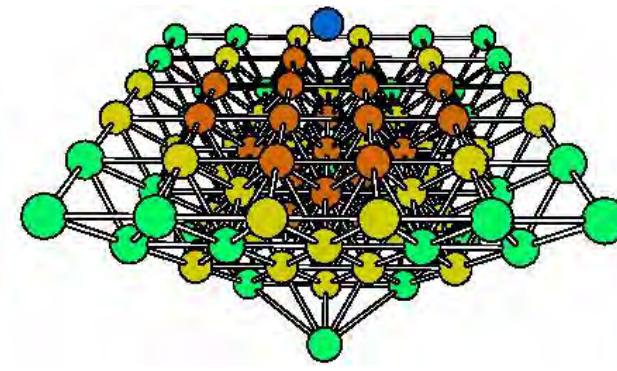
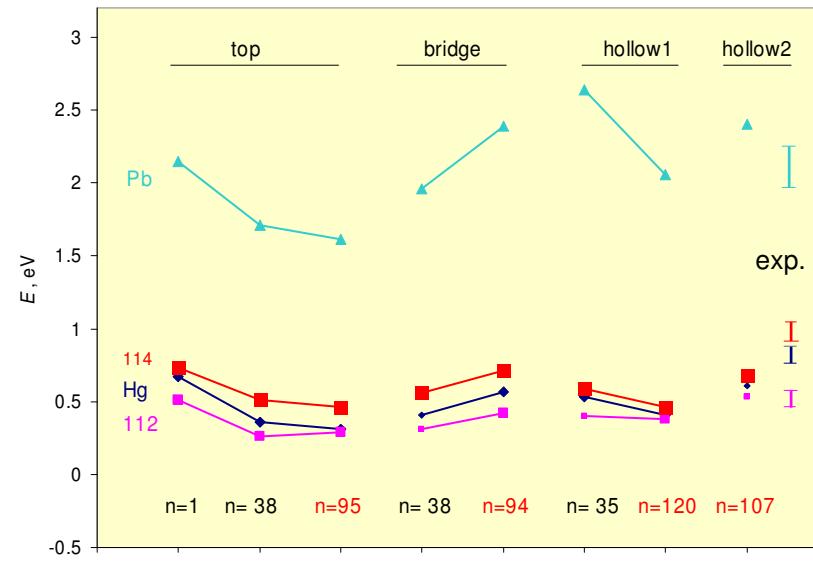


# Periodic Codes

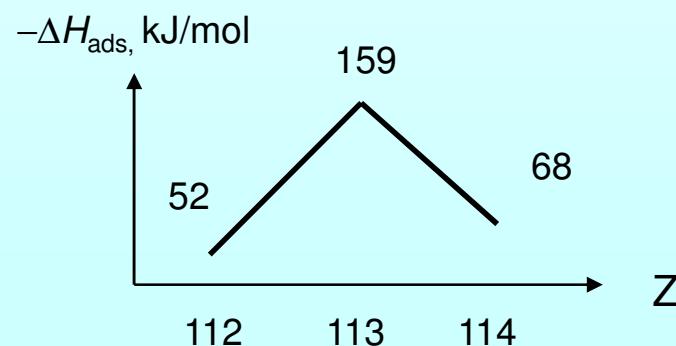
- Available (2c-/4c-DFT)
  - CASTEP (homologs)
  - WASP (PP, homologs)
  - RFPLQ (Cn)
  - **ADF BAND (SHE)**
- **ADF BAND**
  - 2c-ZORA
  - SR, SO iteratively
  - basis sets till Z=120
  - TZP, TZ2P
  - all electron, frozen core
  - revPBE (no dispersive for SHE)
  - full relaxation

# Adsorption of Cn, E113 and Fl on Gold

M-Au<sub>n</sub> binding energies



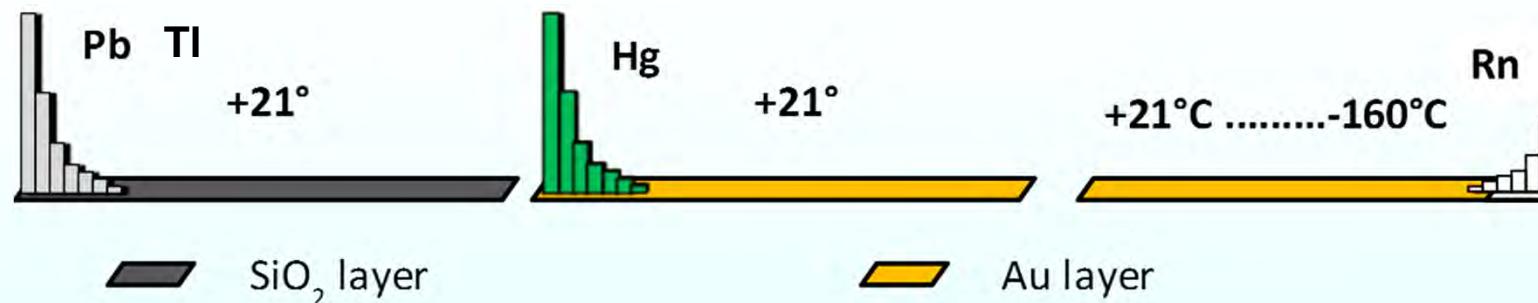
M	position, n	$E_b, \text{ eV}$	$\Delta H_{\text{ads}}, \text{ eV}$	Ref. (exp.)
Hg	bridge n=94	<b>0.56</b>	<b>0.92</b>	Eichler
Cn	hollow n=107	<b>0.46</b>	<b>0.54<sup>+0.4</sup><sub>-0.03</sub></b>	Eichler
Pb	bridge n=94	<b>2.40</b>	<b>2.43</b>	Haennsler
114	bridge n=94	<b>0.71</b>	<b>0.36<sup>+0.5</sup><sub>-0.1</sub></b>	Eichler
			<b><math>\geq 0.5</math></b>	Yakushev
Tl	bridge n=16	<b>2.65</b>	<b>2.48</b>	Serov
113	bridge n=16	<b>1.34</b>	<b>(1.65)</b>	-



like HOAO

[Pershina, Anton, Jacob, JCP, 2009]

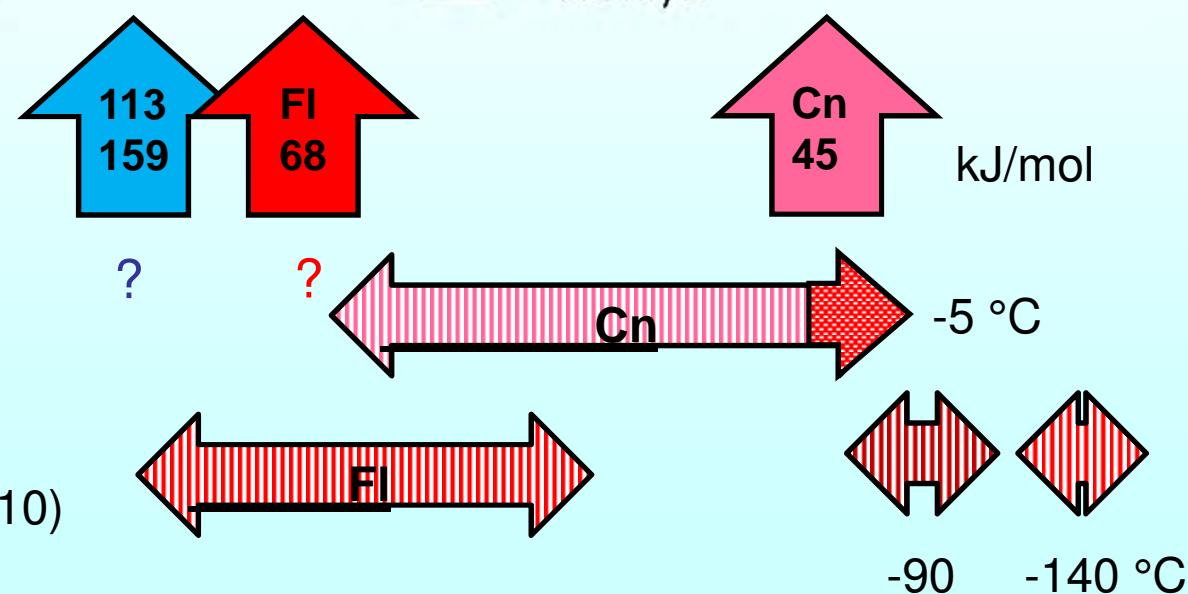
# Predictions of Adsorption of Cn, E113 and Fl on Gold and Quartz



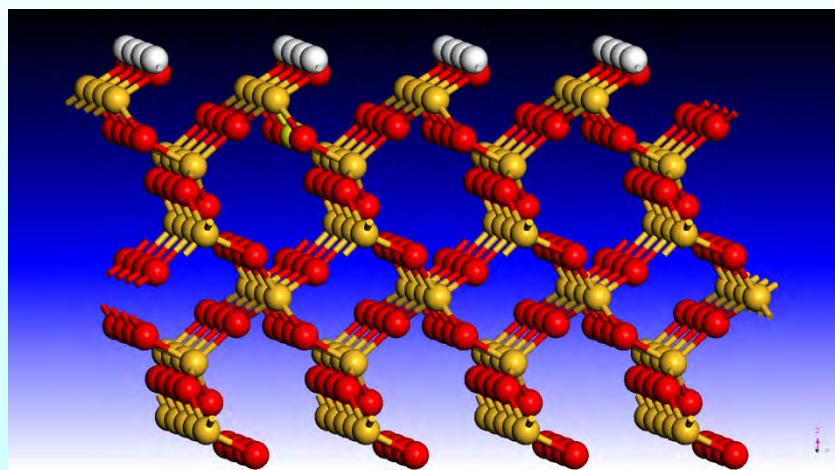
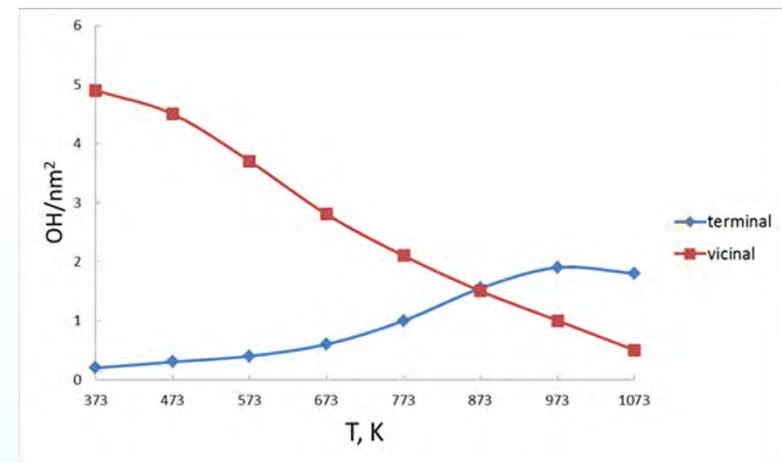
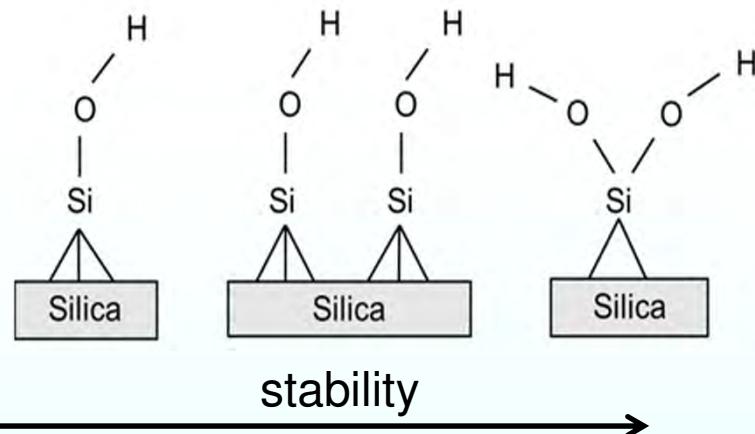
Theory: ???

Experiment:

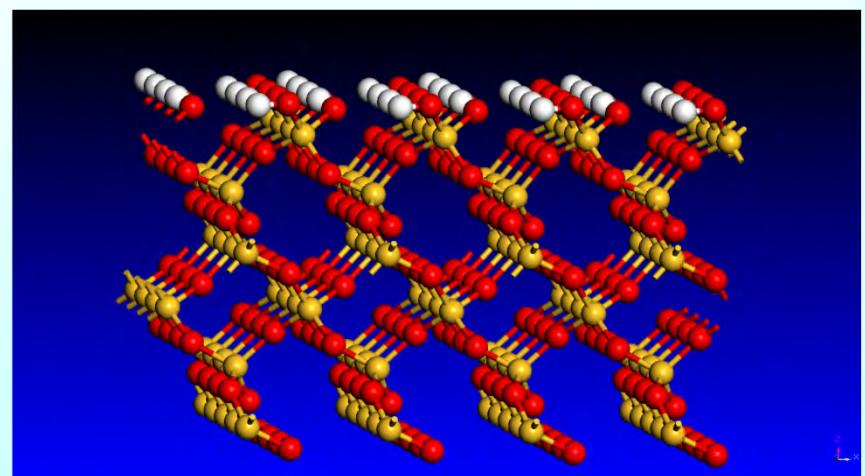
[Eichler, et al. (2007, 2010)  
Yakushev, et al. (2014)]



# Hydroxylated Quartz (Silanols)



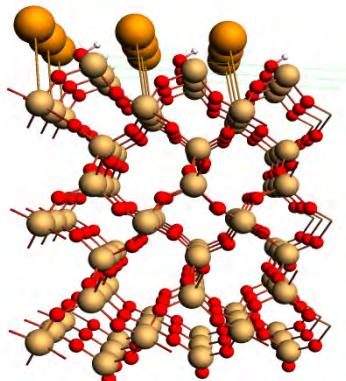
25 °C: vicinal (80%)



geminal (10%)

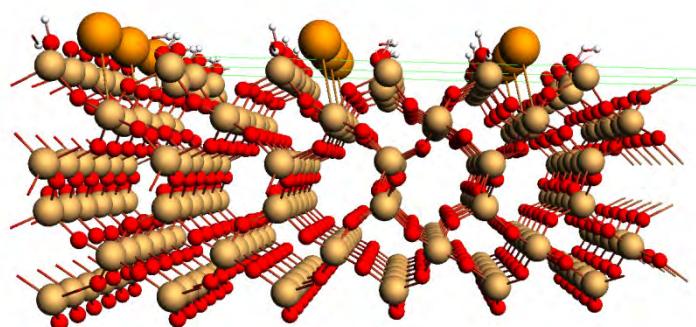
# Influence of Coverage on $E_{\text{ads}}$ of Pb

full (slab)



1.28 eV (d = 1.48 Å)

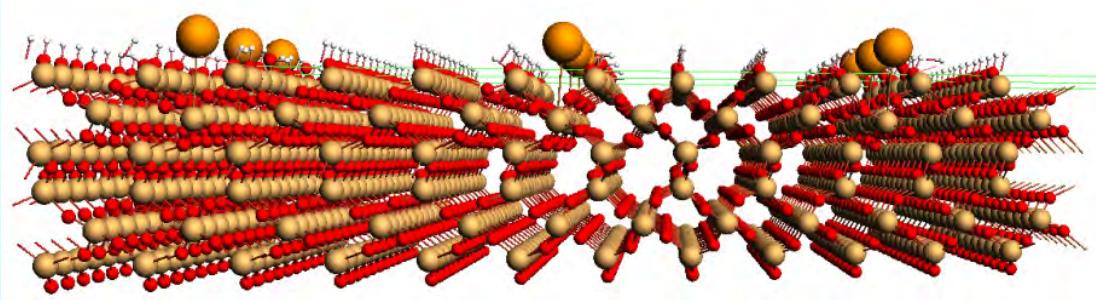
(2 x 2) supercell



2.31 eV (d = 1.07 Å)

Exp:  
2.12 eV

(4 x 4) supercell



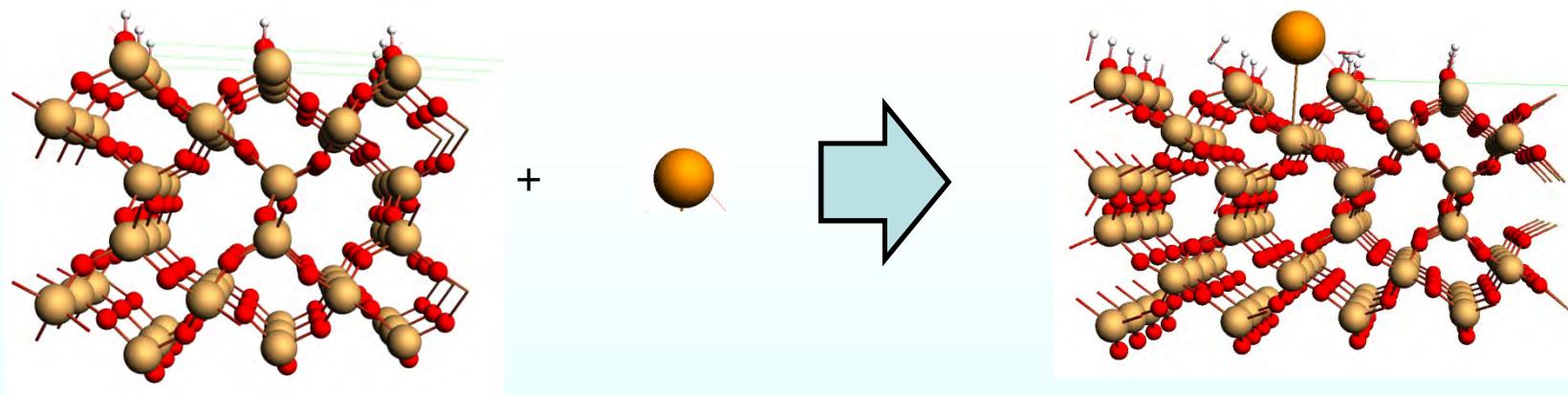
(a uni cell: > 300 atoms)

1.58 eV (d = 1.16 Å)

Exp:  
 $1.66 \pm 0.1$  eV

[V.Pershina, PCCP, 18, 17750 (2016)]

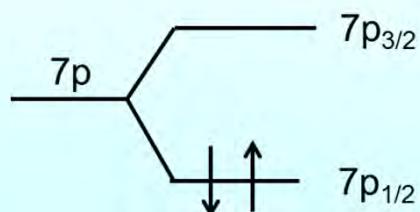
# Adsorption of Pb/Fl on Hydroxylated Quartz: Influence of SO Effects



$E_f(\text{SO})$ : 2107.4

Pb: -1.87

Fl: -5.08



Pb: -2110.0

Fl: -2112.3

$$-\Delta H_{\text{ads}}(\text{Pb}) = 3.79 \text{ eV (SR)} = 1.57 \text{ eV (SO)} \text{ (exp. 1.66 eV)}$$

$$-\Delta H_{\text{ads}}(\text{Fl}) = 3.86 \text{ eV (SR)} = -0.23 \text{ eV (SO)}$$

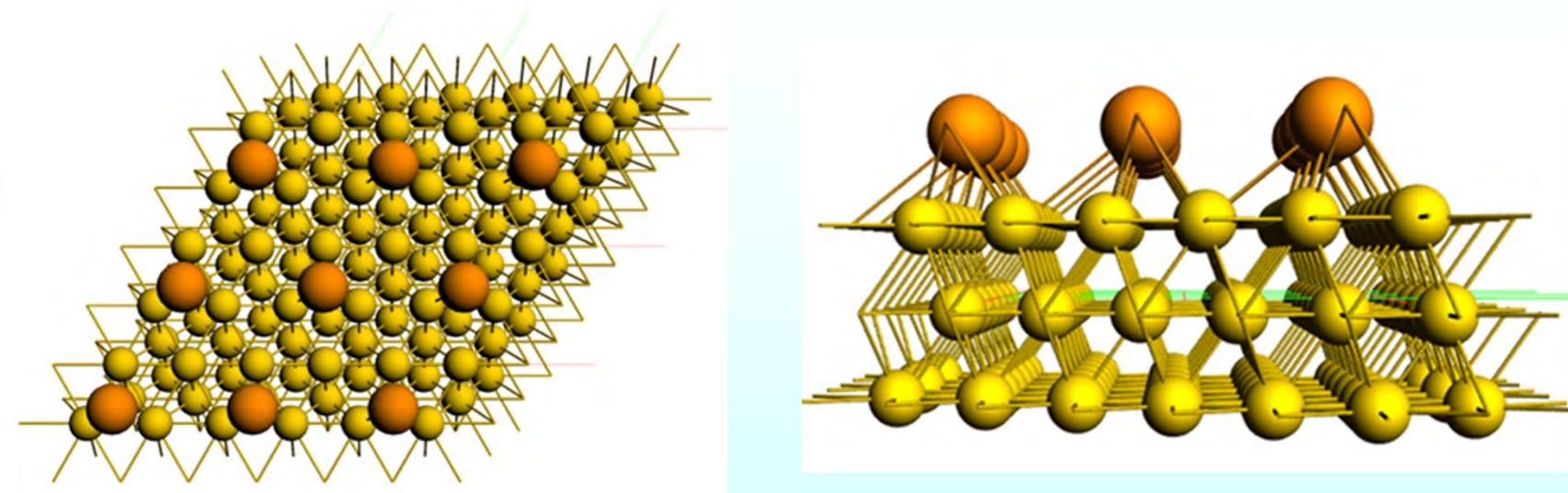
[V.Pershina, PCCP, 18, 17750 (2016)]

# Adsorption of Group 12, 13 and 14 Elements on Quartz and Gold ( $-\Delta H_{\text{ads}}$ , kJ/mol)

Surf.	Hg	Cn	Tl	E113	Pb	Fl	Method
SiO <sub>2</sub>	54	-38	152	58	152	-22	ADF BAND
	$42 \pm 2$	-	$158 \pm 3$	-	$165 \pm 4$	-	Exp.
Au	54	45	(240)	$159 \pm 15$	232	68	4c-DFT
	$98 \pm 3$	$52^{+4}_{-3}$	$240 \pm 5$	-	234	$34^{+54}_{-11}$	Exp. Eichler $\geq 48$ Exp. Yakushev

# Adsorption Energy on Gold (111)

regular Au(111) surface



# Adsorption of Group-12, 13 and 14 Elements on Quartz and Gold

