



How (heavy ion) radiation affects interstellar and solar system ice mantles

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ÉCOLE NATIONALE SUPÉRIEURE D'INGÉNIEURS DE CAEN
& CENTRE DE RECHERCHE



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astrophysical materials:

carbonaceous, silicates, and **Ices**

laboratory simulation:

***solar wind, ions trapped
in magnetospheres,
and cosmic rays***

A

at heavy **ion accelerators**

GANIL Arife, Irrsud, SME, HE
+ GSI Unilac: from **keV** to **GeV**

Infrared absorption spectroscopy FTIR

astrophysics:

Amorphization + Compaction
of Water Ice H_2O

B

Radiolysis of $CO + CO_2$
Origin of gas phase molecules
in interstellar dense clouds ?

C

... cosmic rays versus UV photons

astrochemistry:

Origin of CO_2 and H_2SO_4 on Europa?
endogenic, or exogenic
= implantation of C and S ions ?

D

Formation and radiation resistance
of organic/prebiotic molecules

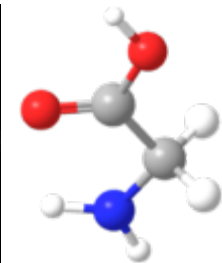
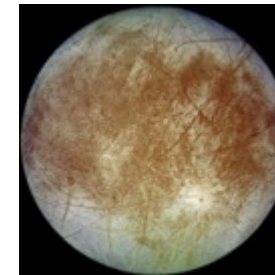
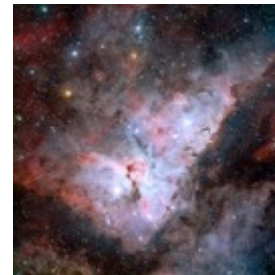
E

1st step: **physics:**

**Compaction, amorphization, sputtering
radiolysis: fragmentation (destruction)**

2nd step: **chemistry:**

formation of new molecules
(radicals, "Implantation")



Astrophysical Ices ...

... exposed to irradiation by UV photons, electrons, and ions:
Cosmic rays, Solar Wind, Giant Planet's Magnetospheres

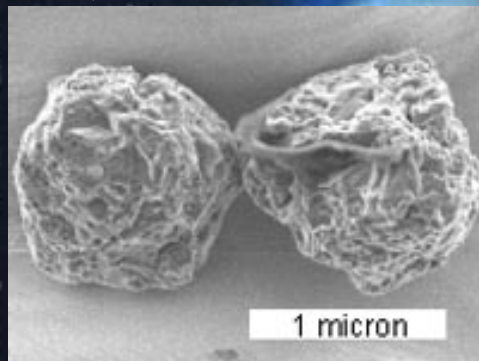
→ Comets



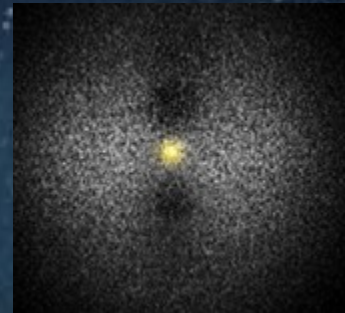
→ Giant Planet's Moons
(Europa, Ganymede, ...)



→ Dust Grains

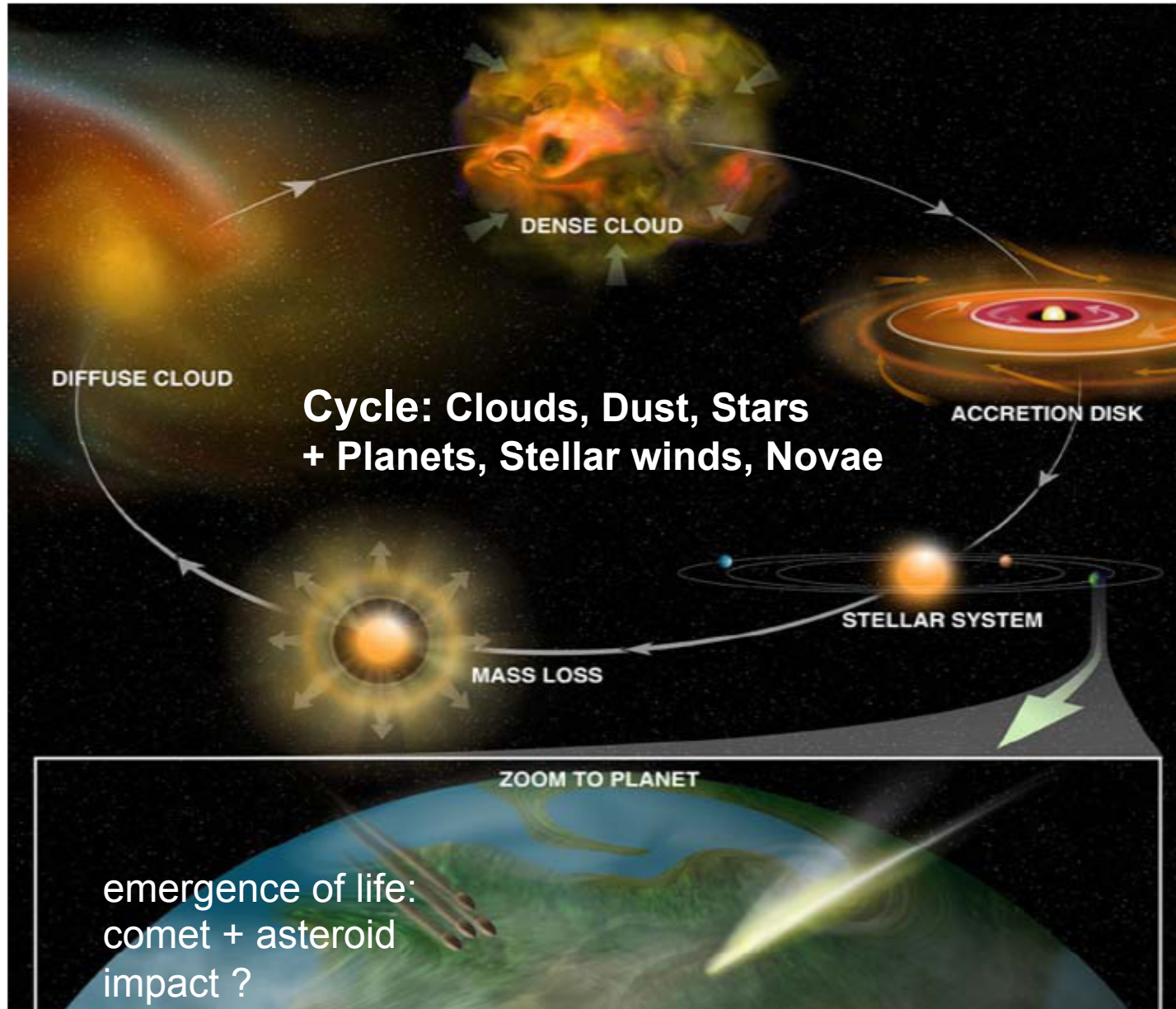


Rings



Dense Interstellar Clouds
(birthplaces of suns and planets)

Astrochemistry and Astrobiology



Interstellar Medium: Dense Molecular Clouds

Density of 10^3 - 10^6 particles cm^{-3} , mainly H_2 , $T \sim 10$ K. Star formation.

Molecules in gas phase and dust grains covered by ice mantles.

Possibly complex organic chemistry in ice mantles due to:

- Surface reactions
- UV and ion processing.

Size: up to some tens of Parsecs

Lifetime: 10^6 - 10^9 years

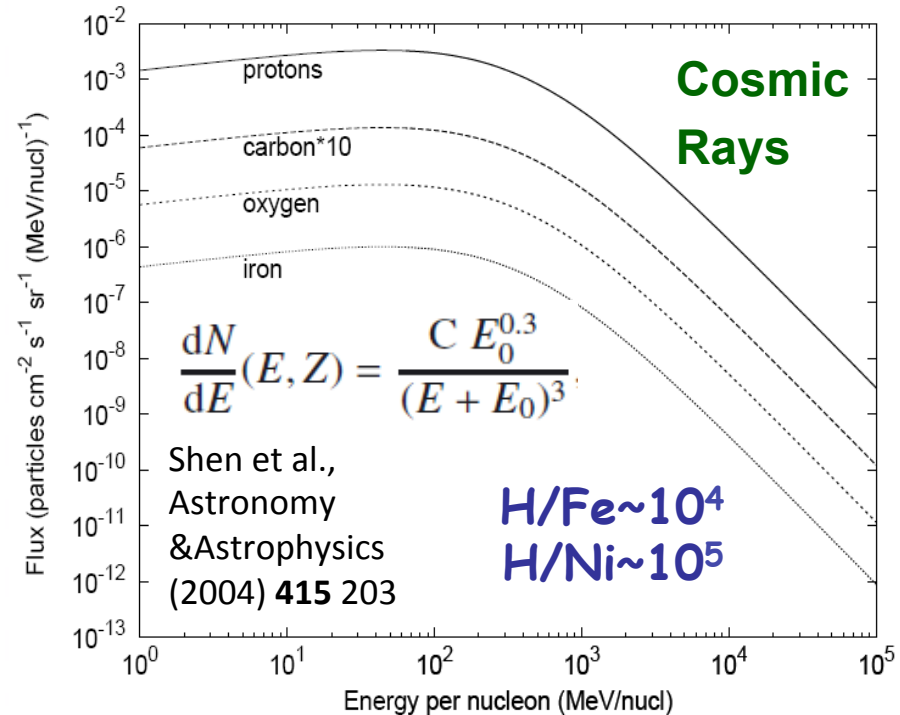
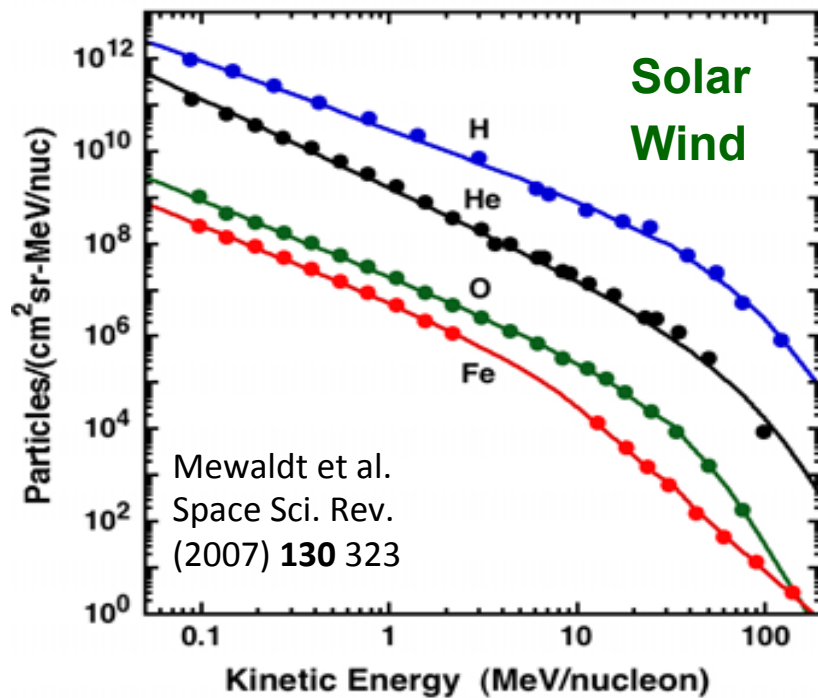
freeze-out time: $10^9/n_{\text{H}}$ (years)



Horsehead Nebula



Radiation Field in Space : complex ! (UV, e-, x-rays, ions)

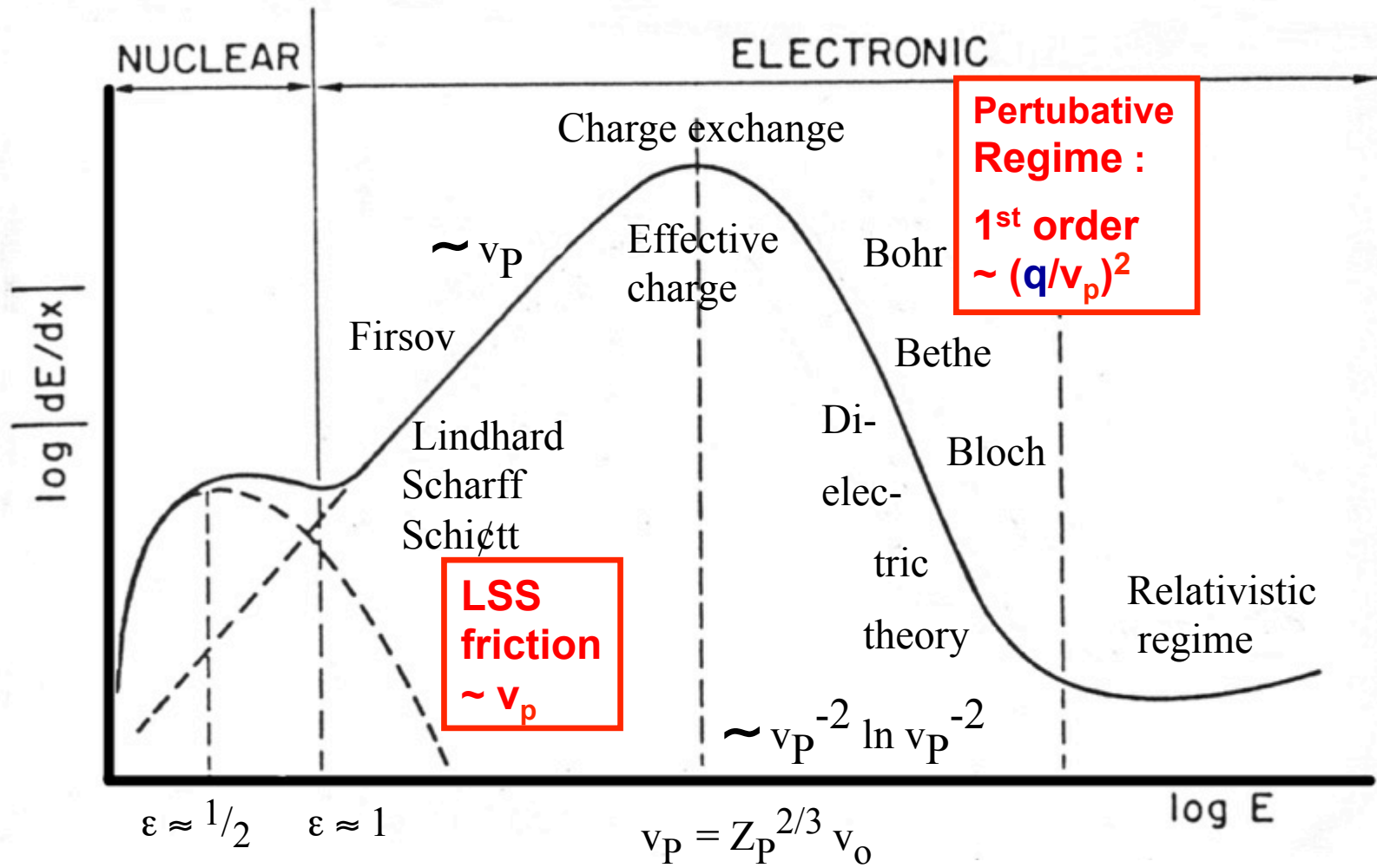


Heavy Ions: why?

- large electronic energy loss S_e
- Scaling laws: S_e^n with $n \approx 1/2, 1, 3/2, 2, \dots 4$
- Unexplained findings (gas phase molecules in dense clouds...), few data
- Astrochemistry: origin of CO₂ and H₂SO₄ on Europe, emergence of life?



Energy loss as a function of projectile energy



solar/stellar wind

cosmic rays



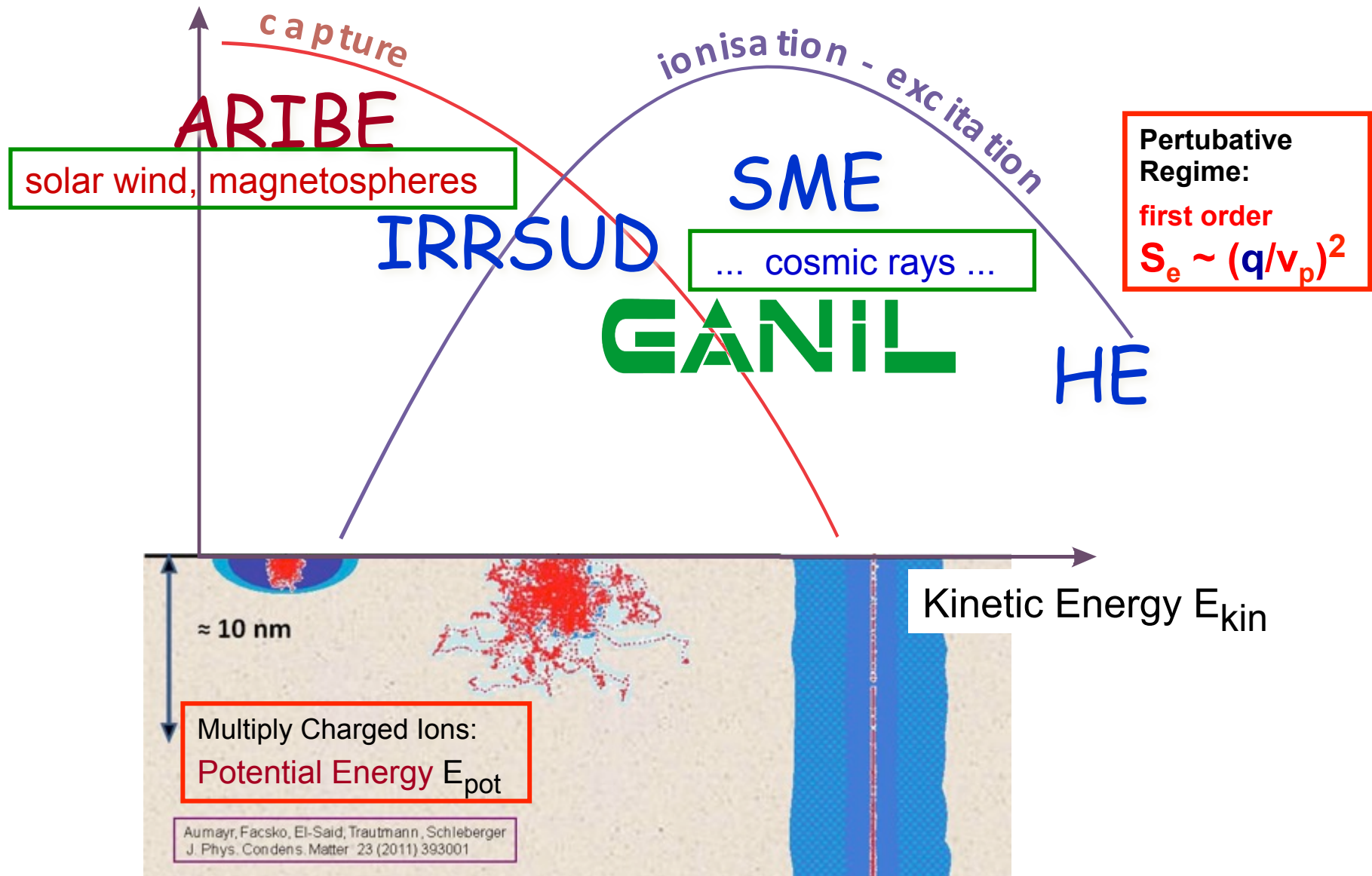
A

**Astrophysics and chemistry
in the laboratory: simulation
of solar wind, cosmic ray
and magnetosphere ion
irradiation of
astrophysical ices**



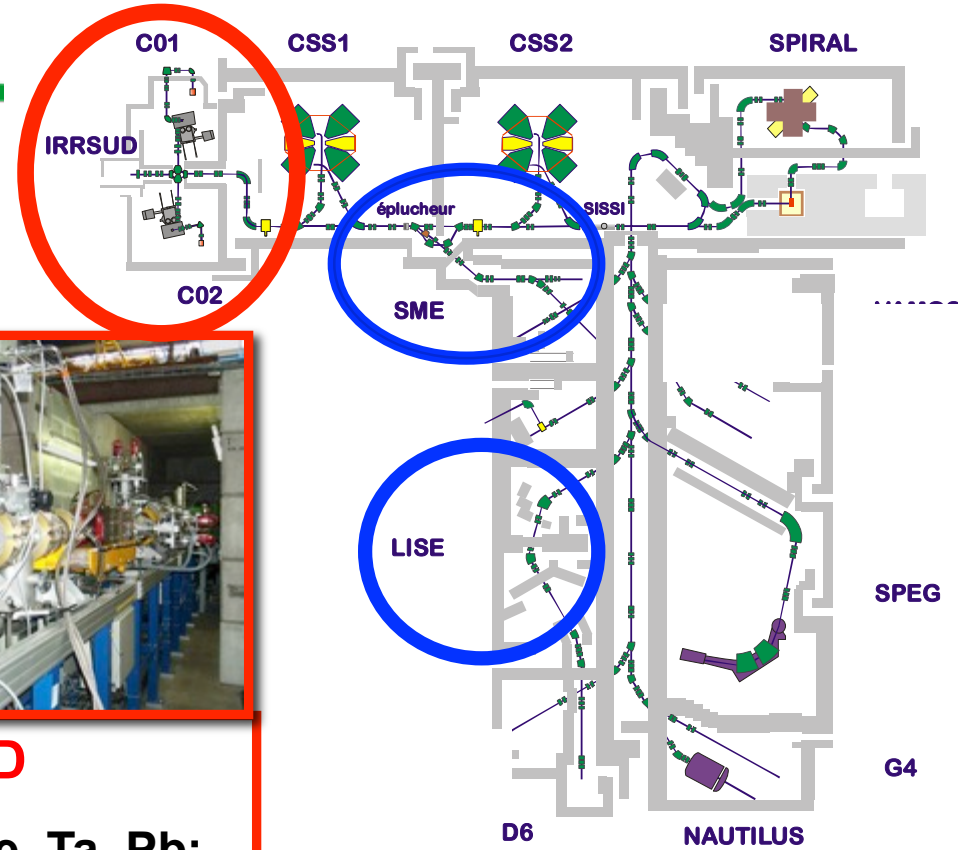
Elastic Collisions:
ion - (screened) nucleus
"nuclear stopping"

Inelastic Collisions: ion - target electron
"electronic stopping" S_e



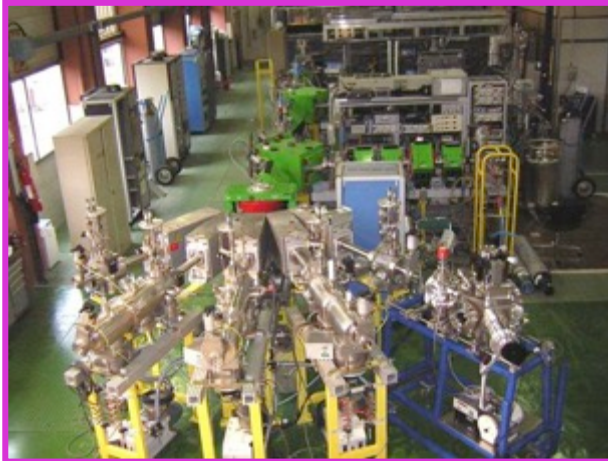
Astromaterials @GANIL

HE, SME, IRRSUD



+ARIBE low energy
multiply charged ions

He, C, O, S, Ar, Xe:
q keV



IRRSUD

O, Ni, Xe, Ta, Pb:
0.5 to 1 MeV/u

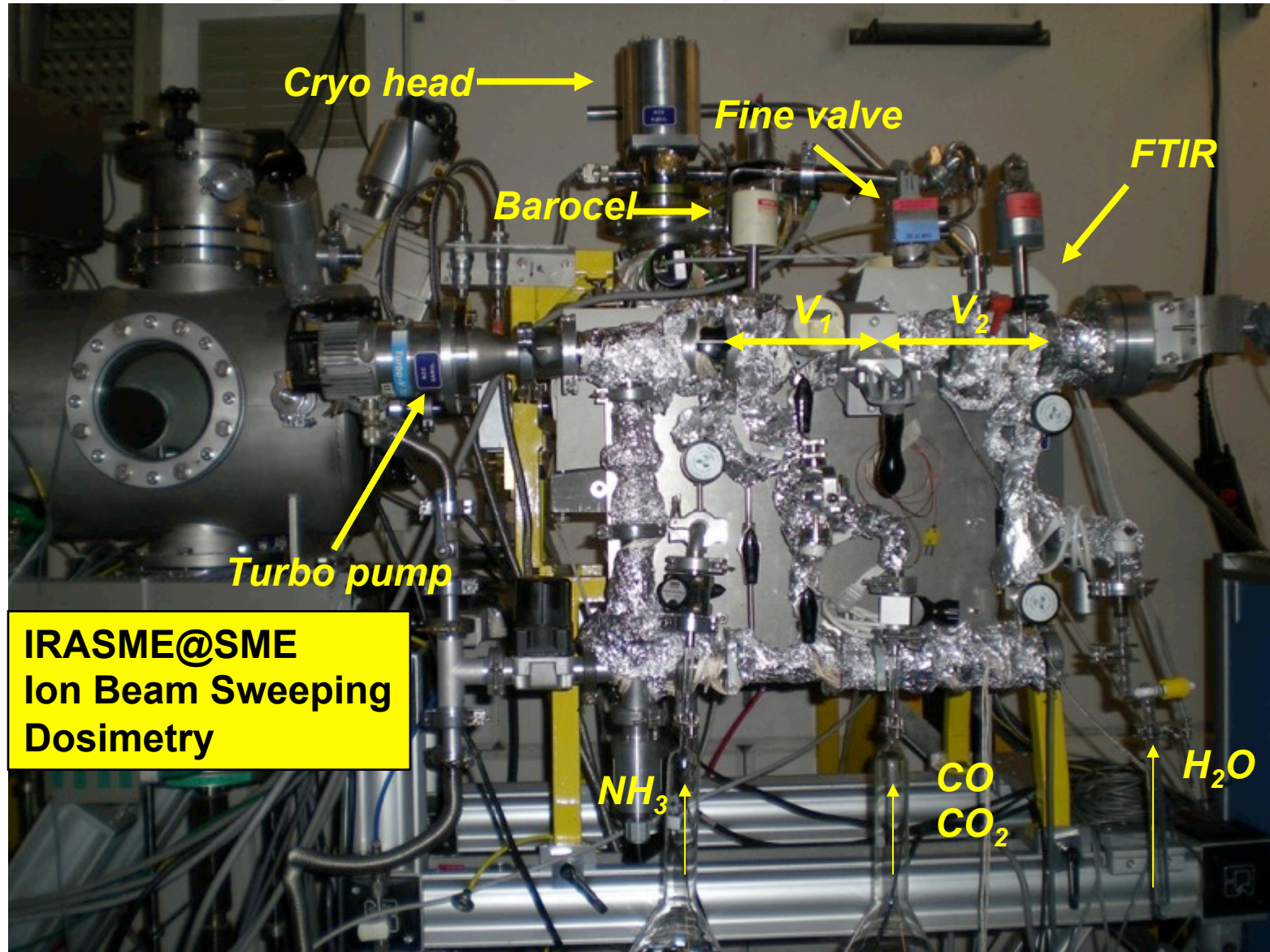
High Energy: **LISE**

Fe: 70 MeV/u

Medium Energy: **SME**

O, Fe, Ni, Kr: 5-13 MeV/u

the "gas mixing and deposition machine"



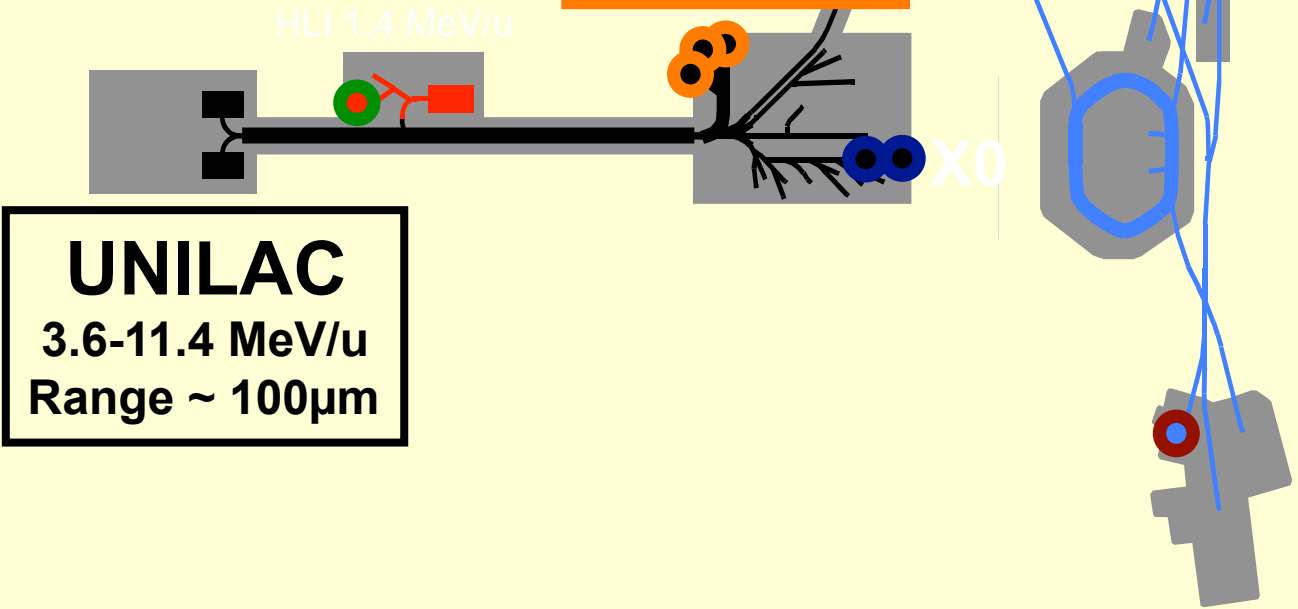


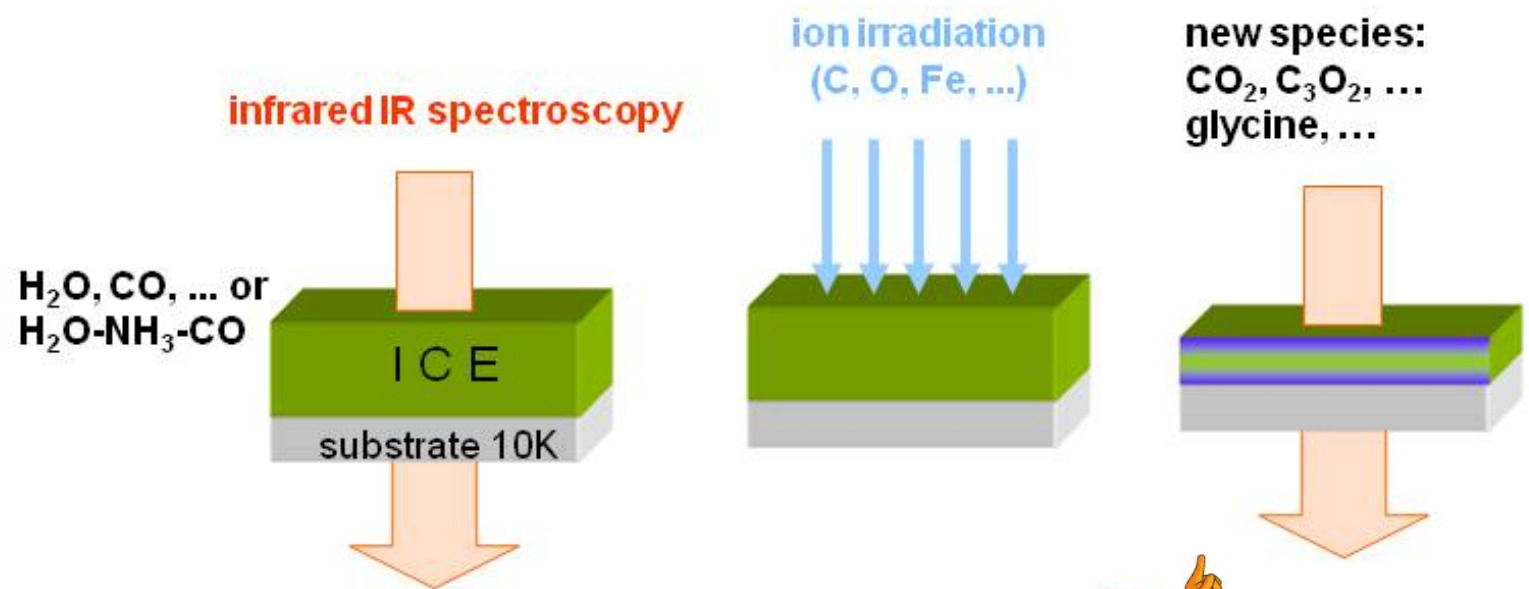
M3 Multi-Analysis Chamber

M-Branch
In-situ and
On-line
Analysis of
Irradiated
Materials

SIS
up to
2 GeV/u

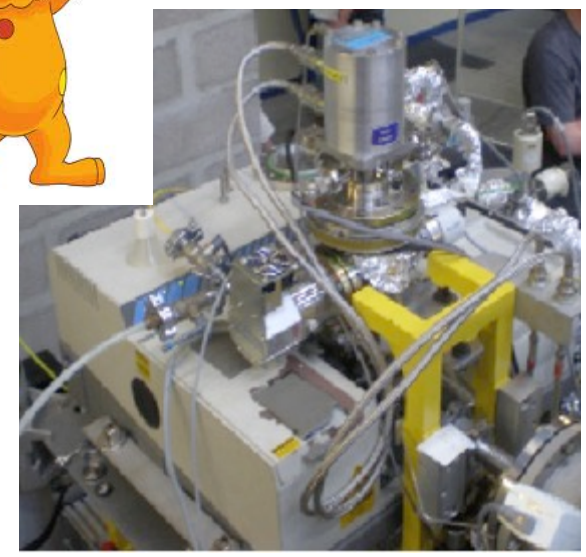
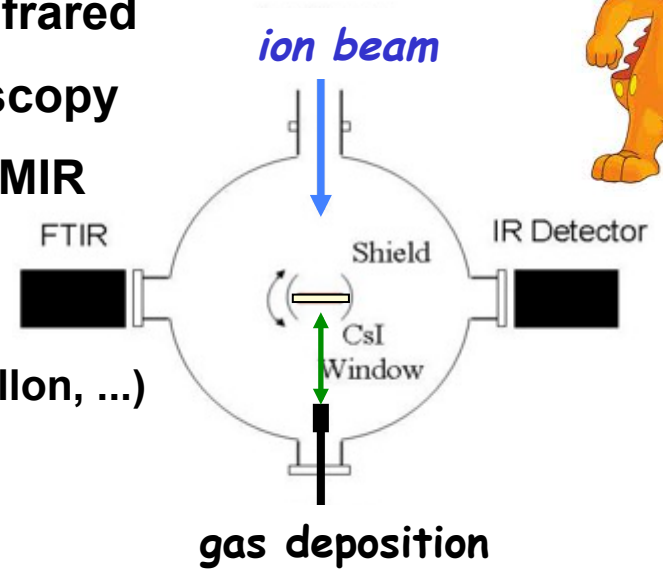
GSI
Materialforschung





**Fourier Transform Infrared
Absorption Spectroscopy
FTIR @CIMAP: CASIMIR**

(E. Balanzat, J.M. Ramillon, ...)



FTIR Fourier Transform Infrared Absorption Spectroscopy: molecular vibrations

plenty of information:

Absorption Line

Position + Shape:

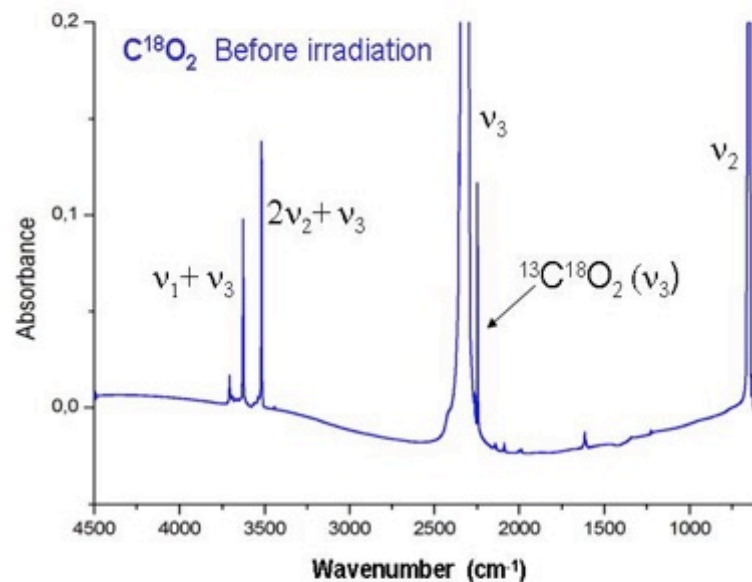
identification of molecules,
environment (“dangling
bonds”: porosity ...)
structure (crystalline,
amorphous)

Integral (Surface)

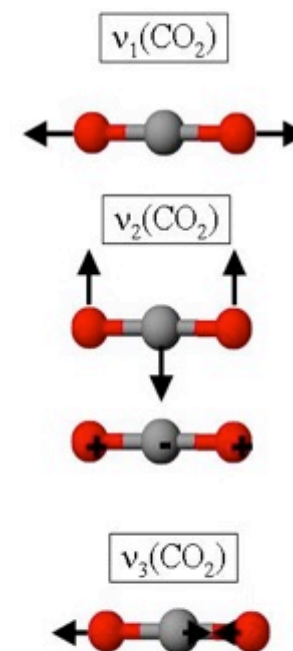
columnar density (thickness)
evolution with projectile
fluence: disappearance and
synthesis of molecules

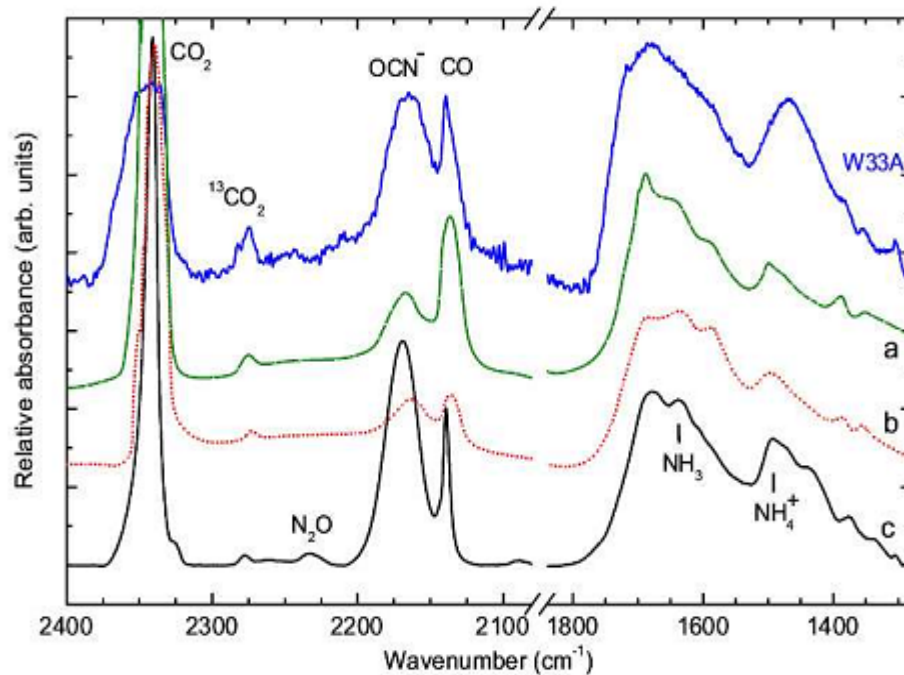
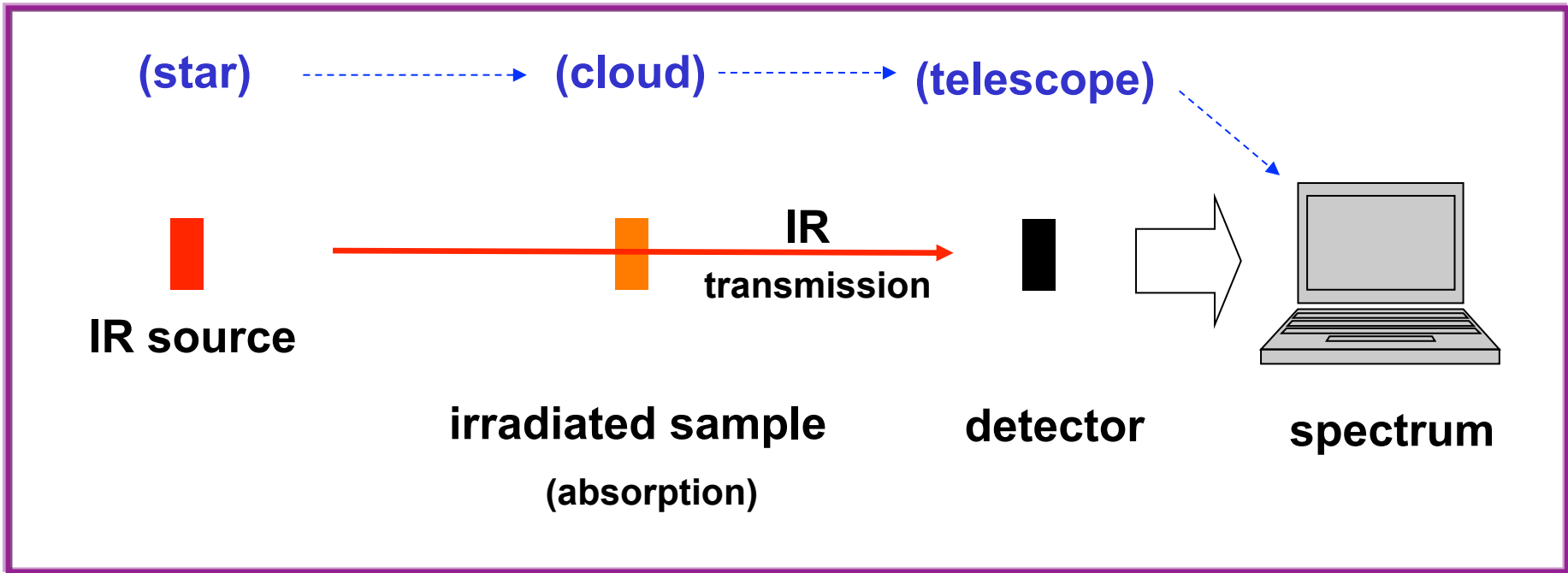
but:

detection of symmetric molecules (O_2 , N_2 ...) difficult



FTIR spectrum of $C^{18}O_2$ ice at 15 K





Space observation:

ISO Infrared Space Observatory,
protostellar source W33a

Laboratory simulation:

UV photons

protons

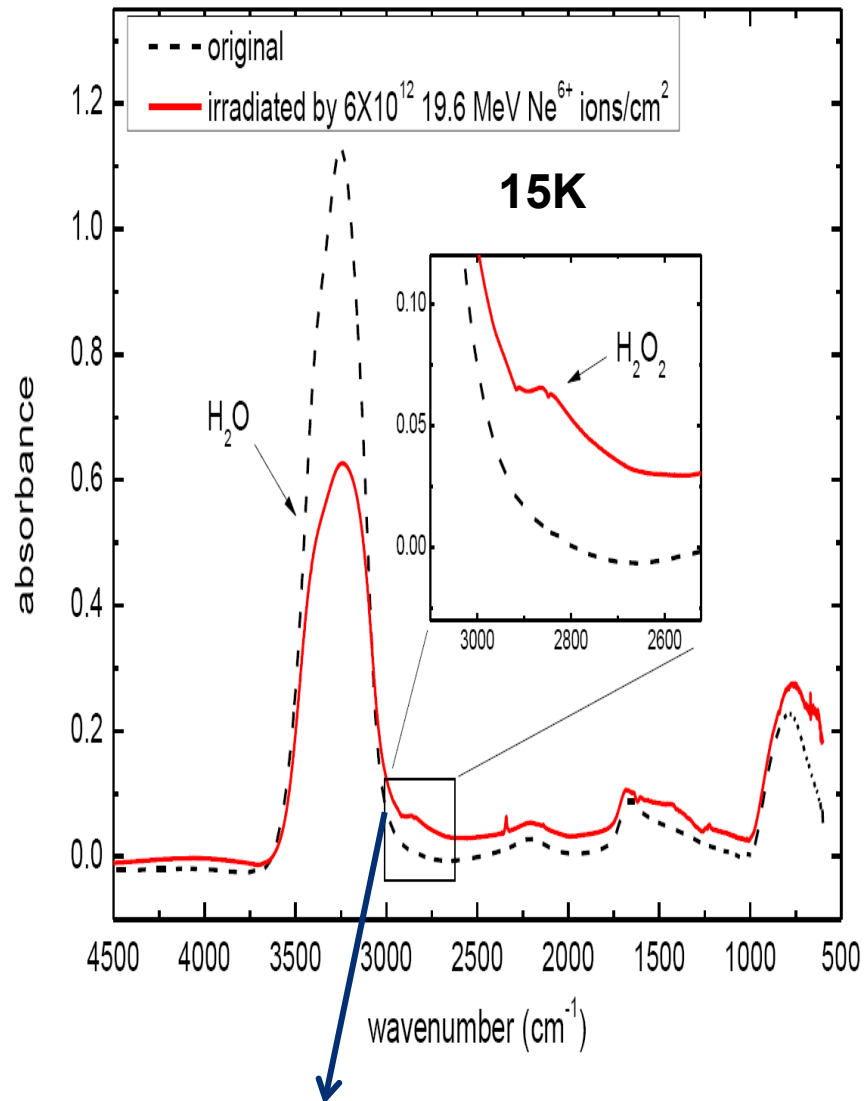
heavy ions

S. Pilling et al.
Astronomy &
Astrophysics
509 (2010) A87

B

**Water ice:
Compaction
and
Amorphization**



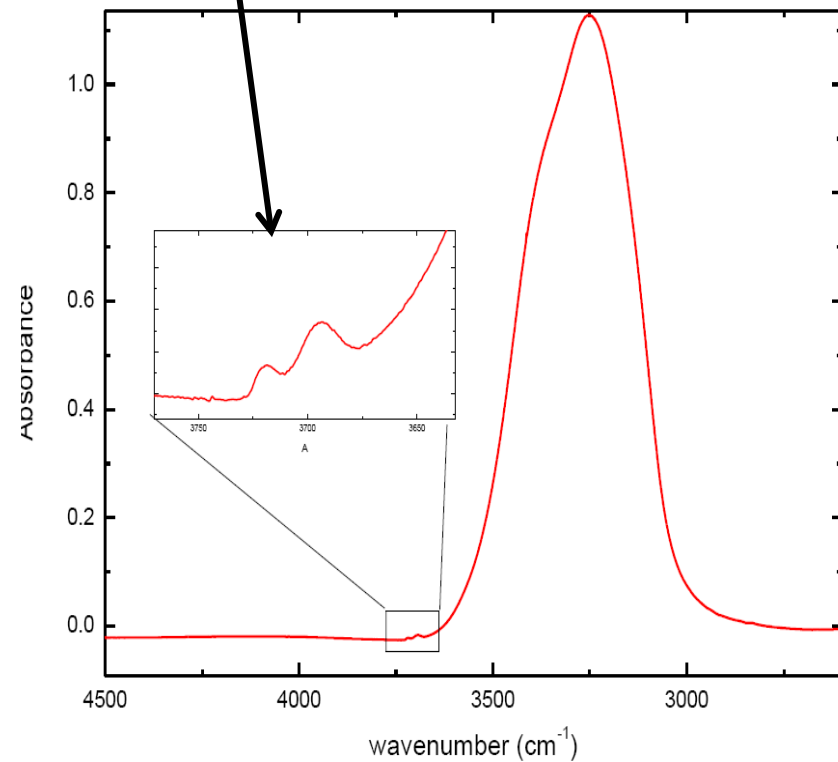


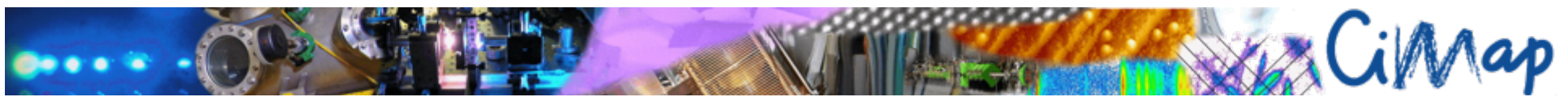
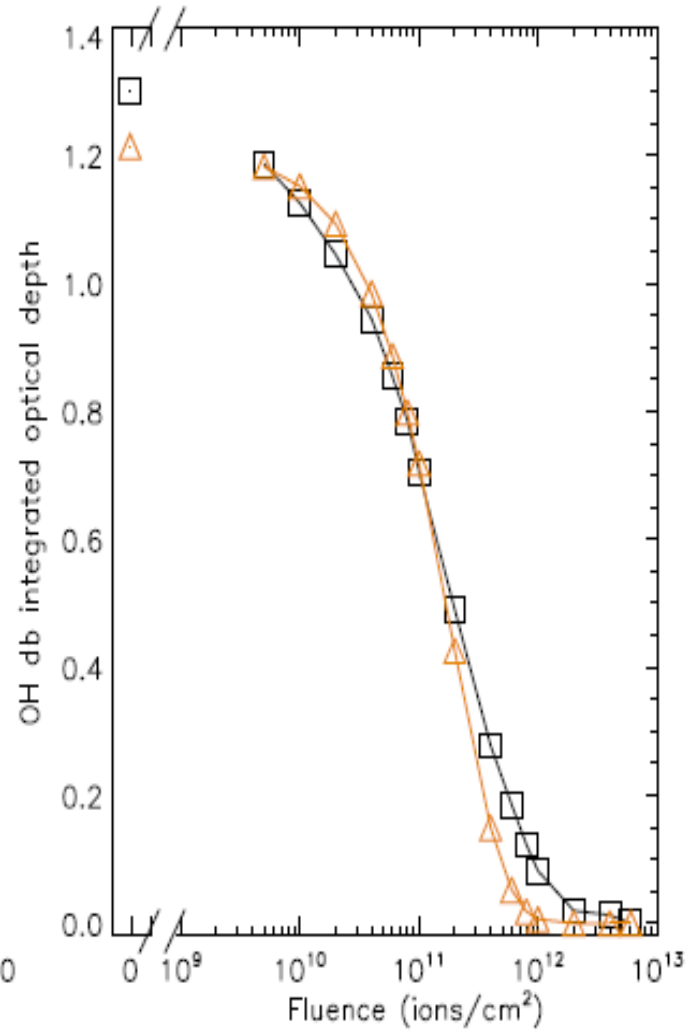
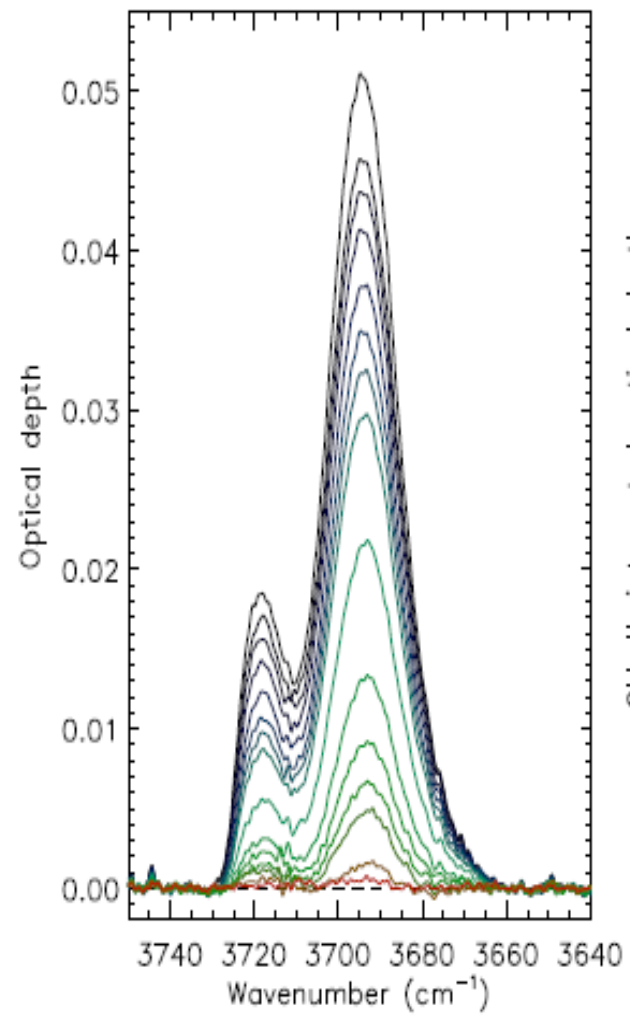
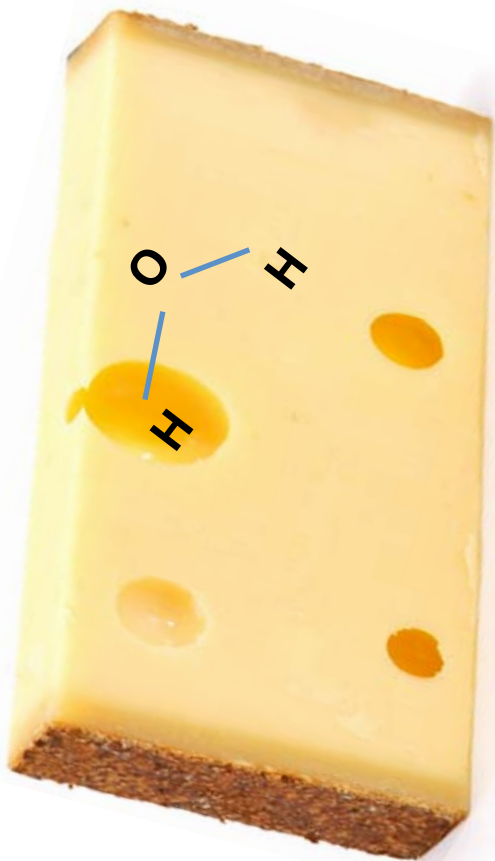
**Irradiation of H_2O ice:
formation of H_2O_2**

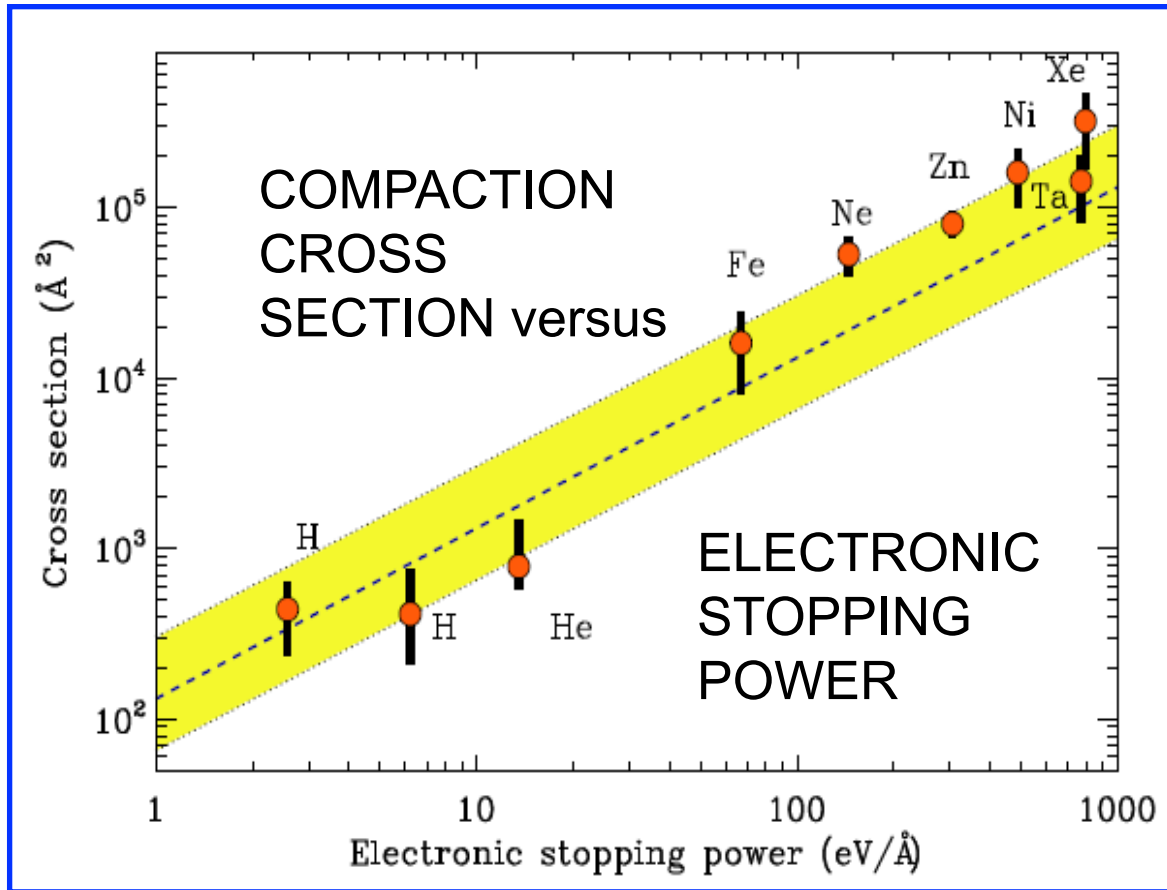
The most abundant molecule
in interstellar ices:

Water H_2O

**Porosity:
OH dangling bonds**







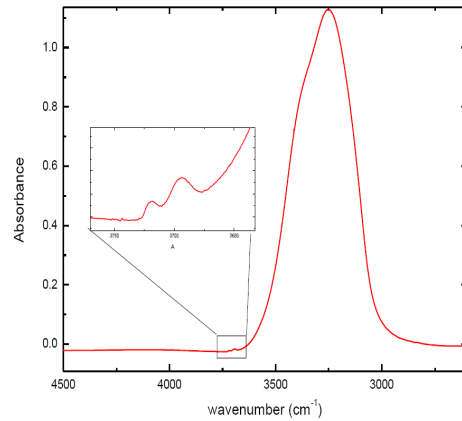
$$t_{comp} = 1 \times 10^5 \text{ to } 2 \times 10^6 \text{ years}$$

Indeed **no**
OH dangling bonds
observed by
ISO in ISM

Compaction of Water Ice by Cosmic Rays: Experiment 2012 GANIL-LISE
E. Dartois, J.J. Ding, A.L.F. de Barros, P. Boduch, R. Brunetto, M. Chabot, A. Domaracka, M. Godard, X.Y. Lv, C.F. Mejia Guaman, T. Pino, H. Rothard, E.F. da Silveira, J.C. Thomas
Swift heavy ion irradiation of water ice at MeV to GeV energies: approaching true cosmic ray compaction
 Astronomy & Astrophysics 557 (2013) A97



Porosity: OH dangling bonds



**E. Dartois et al.,
Astronomy &
Astrophysics 557
(2013) A97**

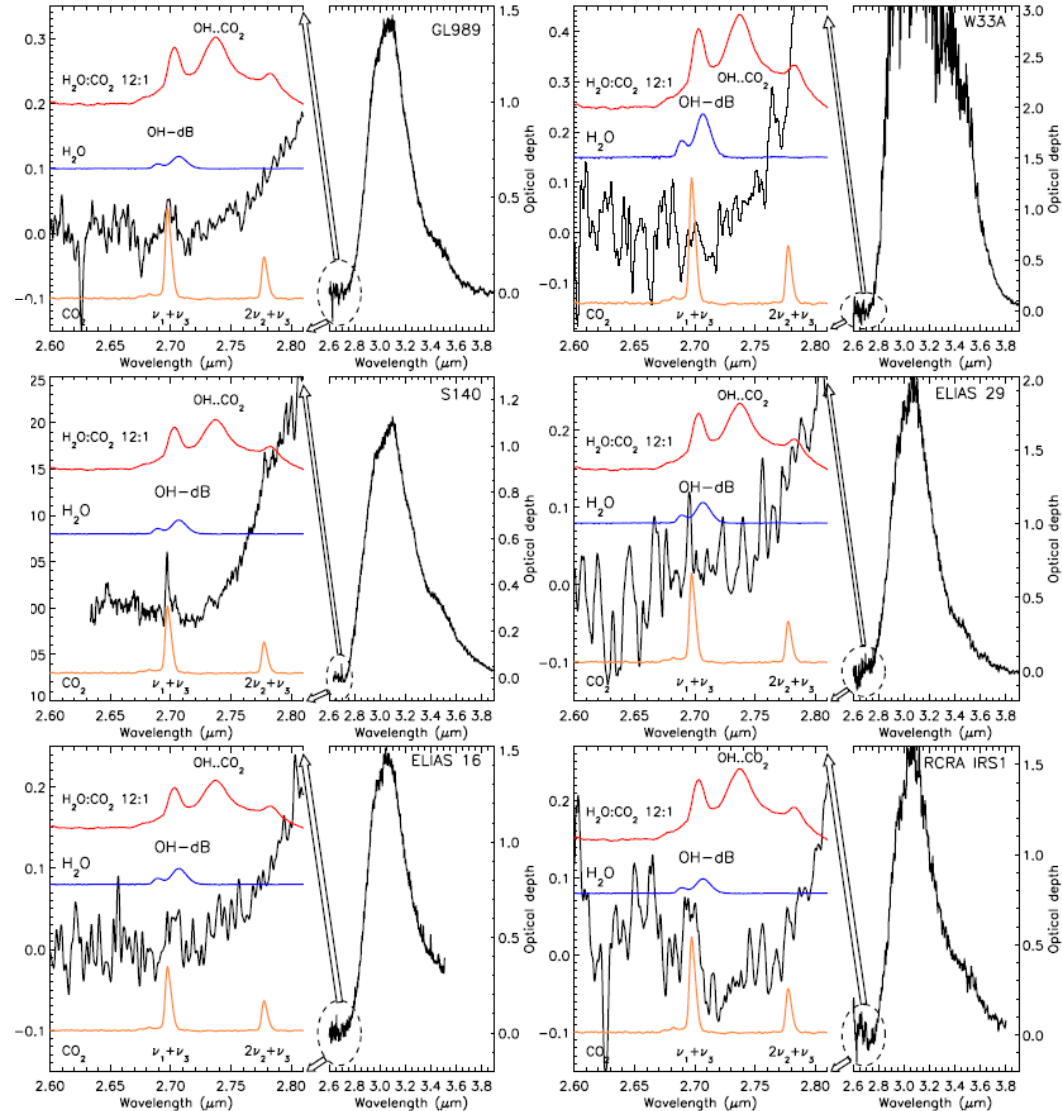
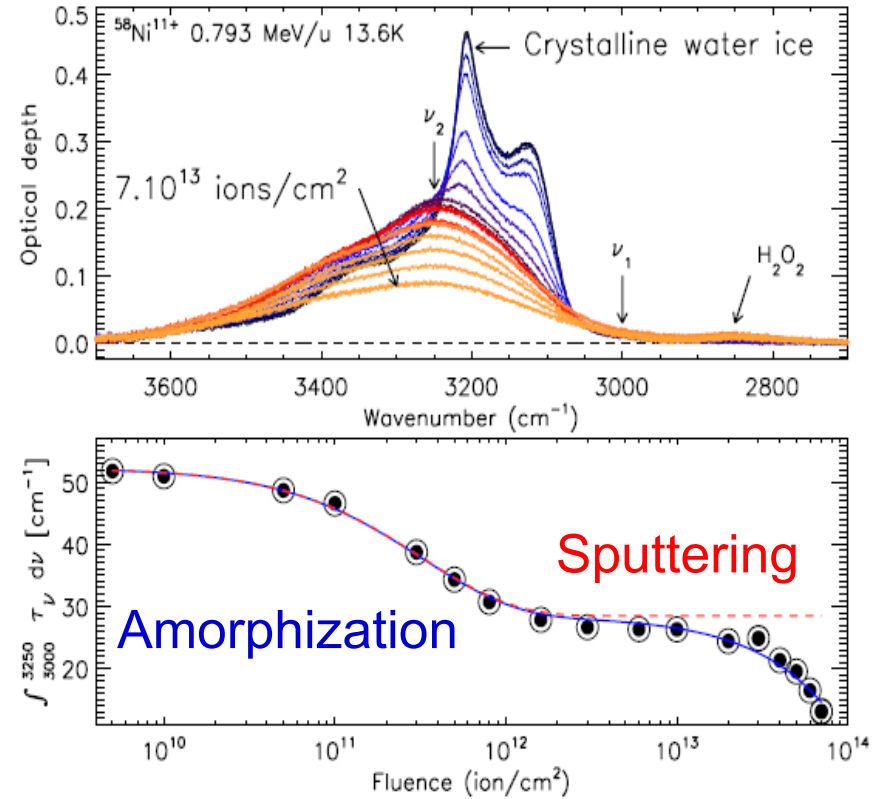
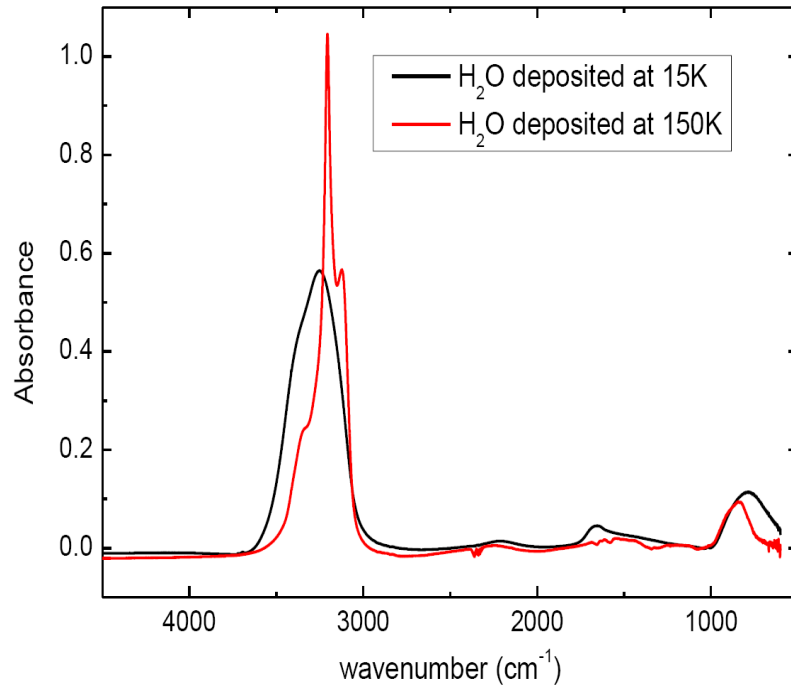


Fig. 4. Young stellar objects spectra measured with the Short Wavelength Spectrometer onboard the Infrared Space Observatory, compared to various laboratory ice mixtures recorded spectra. The optical depth astrophysical spectra are obtained from a continuum baseline subtraction. A close up on the dangling bond region is plotted in the left part of each panel. These spectra are compared to laboratory spectra of pure H₂O and CO₂ ice, and a H₂O:CO₂ 12:1 mixture. The CO₂ and a H₂O:CO₂ laboratory spectra are normalised to the CO₂ column density observed in the mid-IR. The H₂O dangling bond spectrum was normalised with the relation discussed in the preceding section, assuming a porosity of 0.2, and integrating the OH stretching mode. See text for details.



Ion irradiation
3 times more efficient
for compaction than
for amorphization

water ice resistant to
phase transition

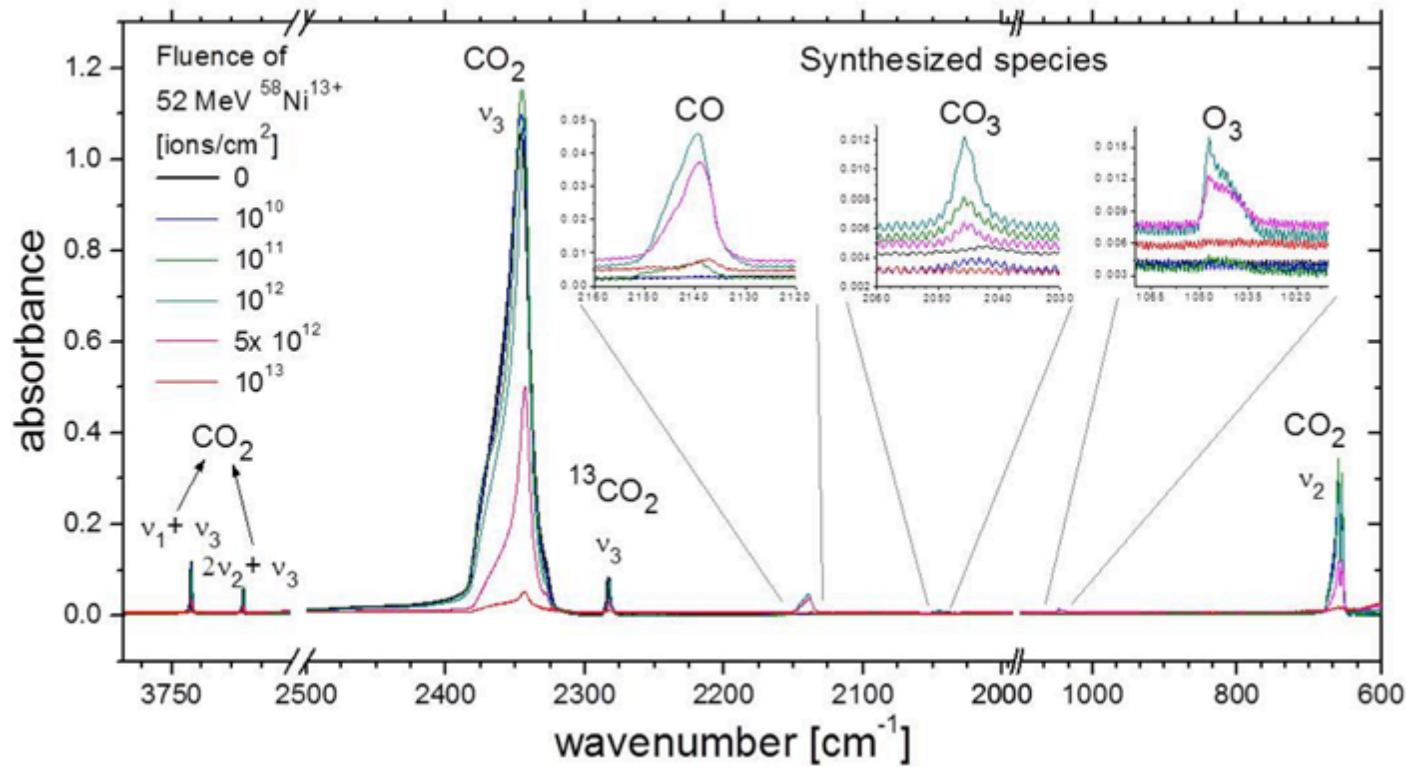
E. Dartois, B. Augé, P. Boduch, R. Brunetto, M. Chabot,
A. Domaracka, J.J. Ding, O. Kamalou, X.Y. Lv,
B. H. Rothard, E.F. da Silveira, J.C. Thomas
**Heavy ion irradiation of crystalline water ice -Cosmic
ray amorphization cross-section and sputtering yield**
Astronomy & Astrophysics 576 (2015) A126



C

**Radiolysis of
CO and CO₂ in
dense molecular Clouds:
UV versus Cosmic Rays.**





CO₂

**@ GSI +
GANIL**

**projectiles:
Ni
Ti
Xe**

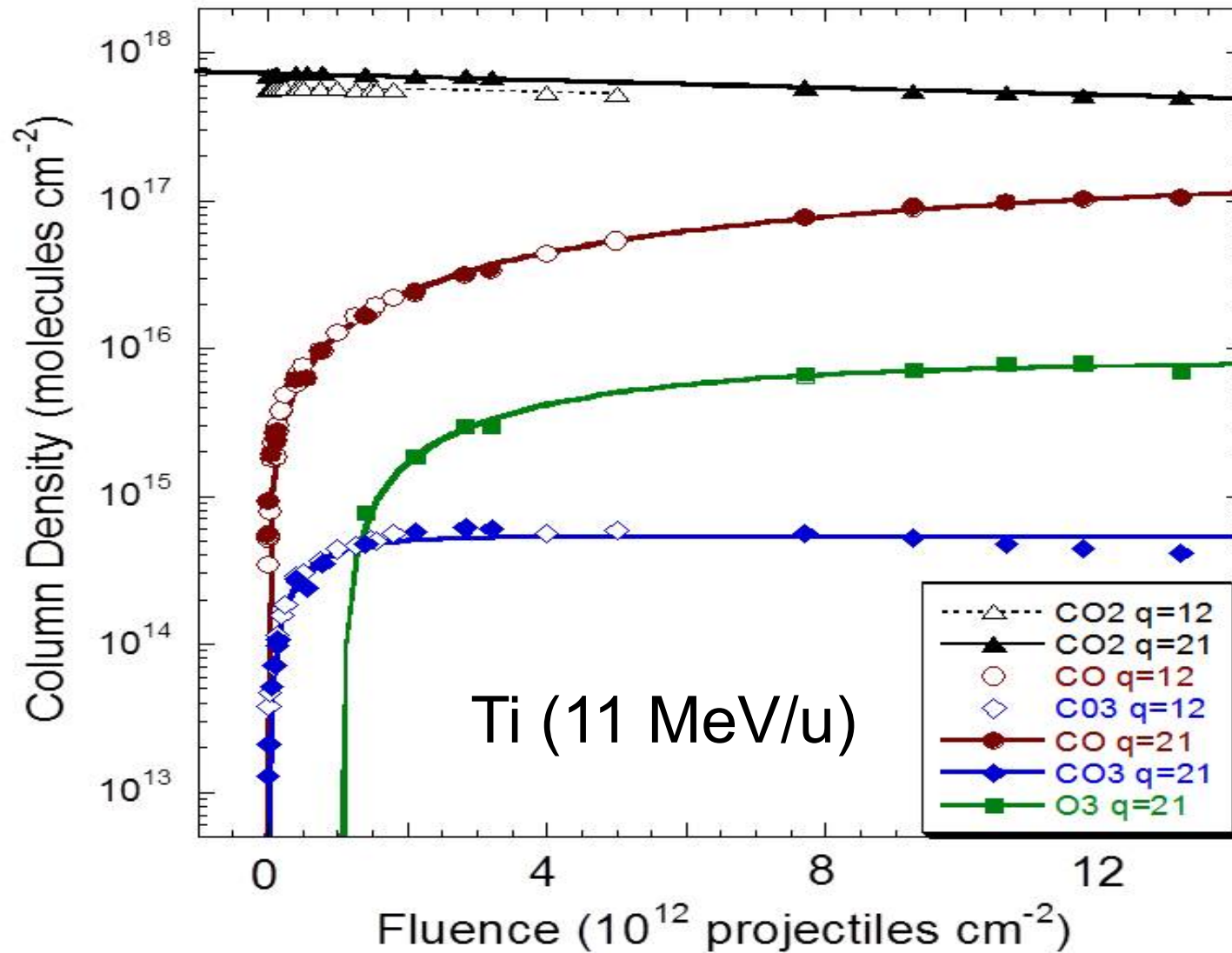
GSI

Materialforschung

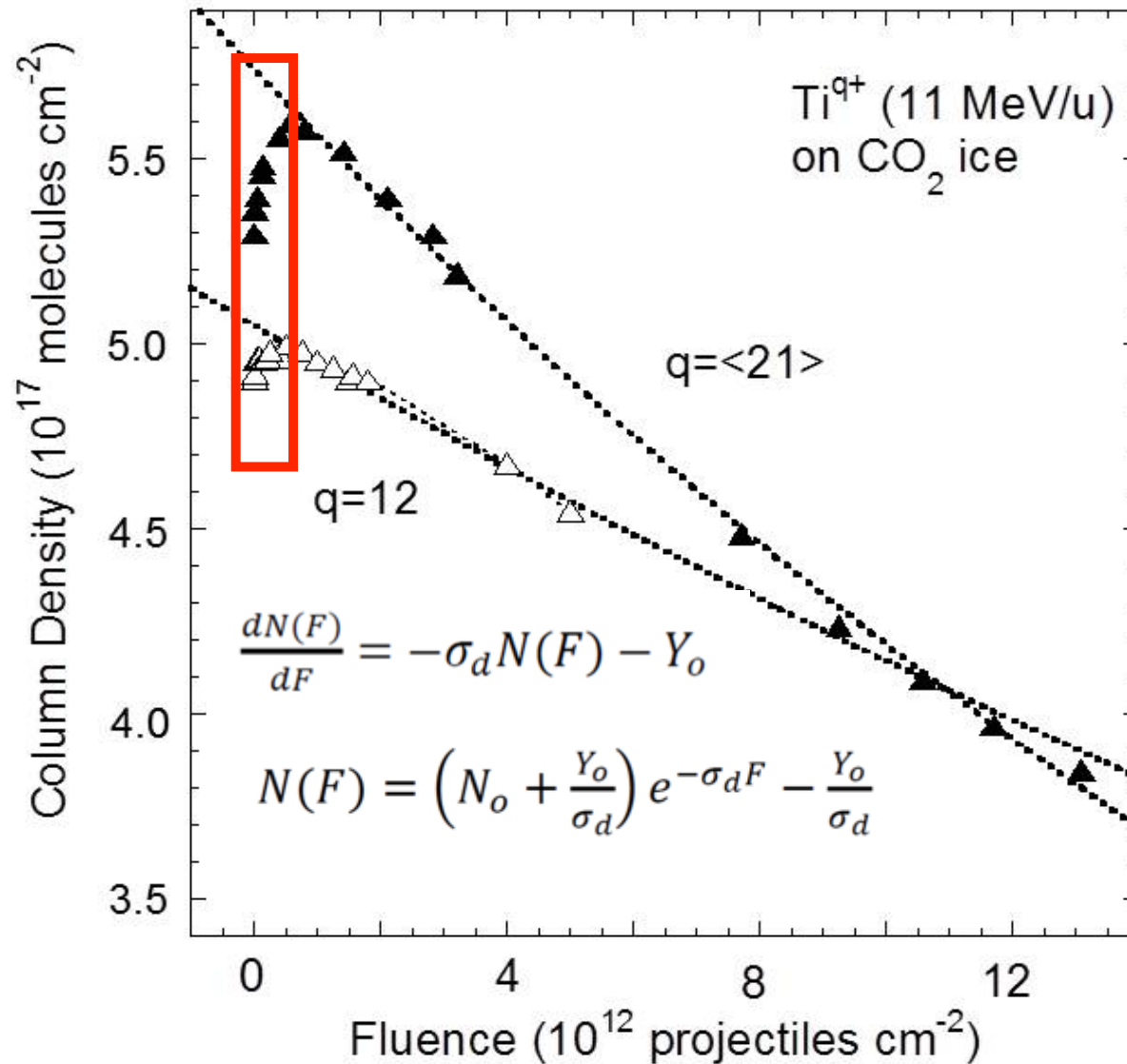
CiMap



Projectile fluence dependence



Projectile fluence dependence



CO₂:
Compaction*
destruction
 (fragmentation)
sputtering

Formation of
CO, **O₃**, **CO₃**

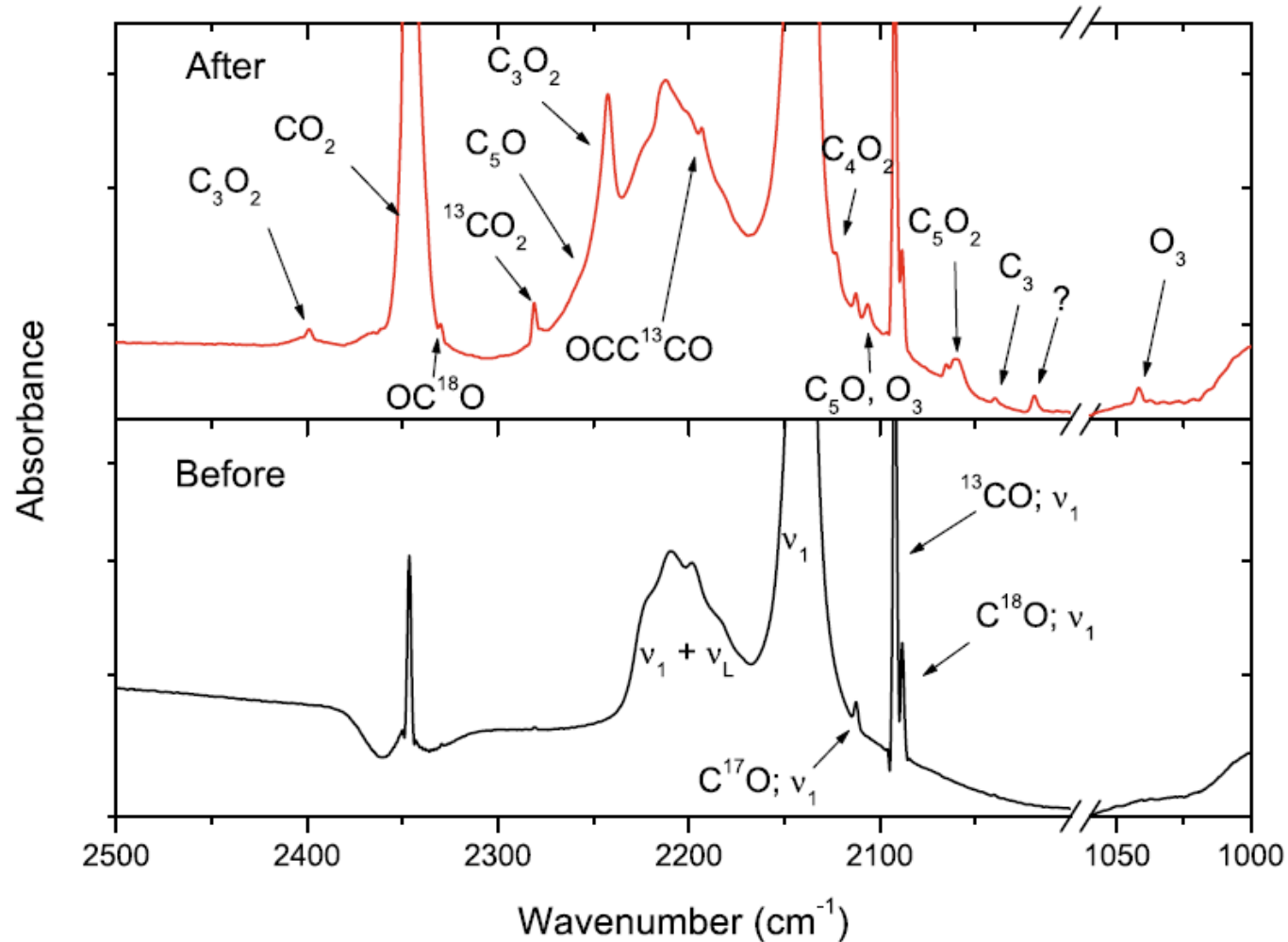
*C.F. Mejía, A.L.F. de Barros,
 E. Seperuelo Duarte,
 E.F. da Silveira, E. Dartois,
 A. Domaracka, H. Rothard,
 P. Boduch,

**Compaction of porous ices
 rich in water by swift heavy
 ions**

Icarus 250 (2015) 222

Another example: CO ice

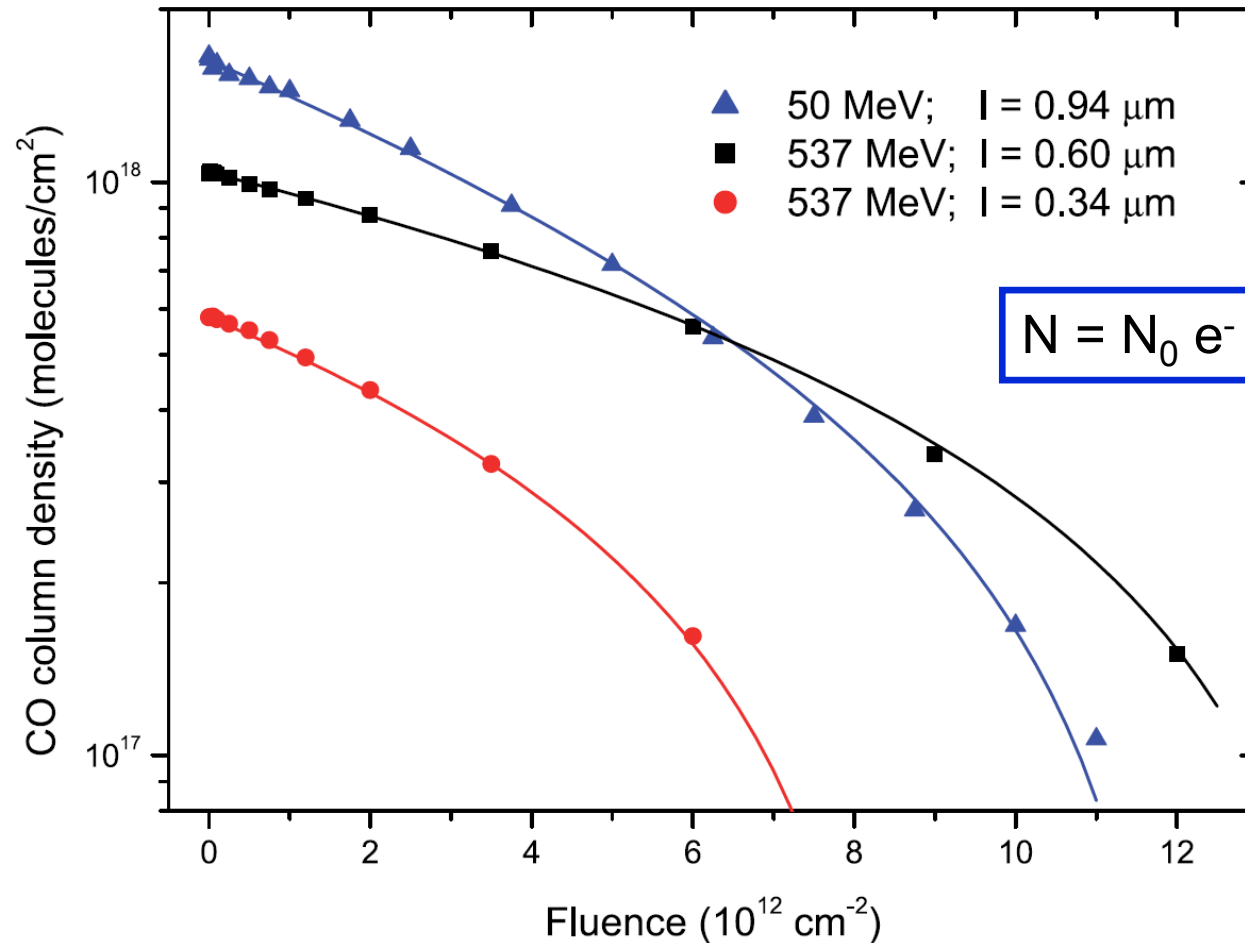
(the second most abundant molecule in space ices after H₂O)



Infrared spectrum of CO ice before and after 50 MeV ⁵⁸Ni¹¹⁺ irradiation with a fluence of $1.0 \times 10^{12} \text{ cm}^{-2}$.



CO ice: disappearance of CO Molecules during Nickel Ion Irradiation:



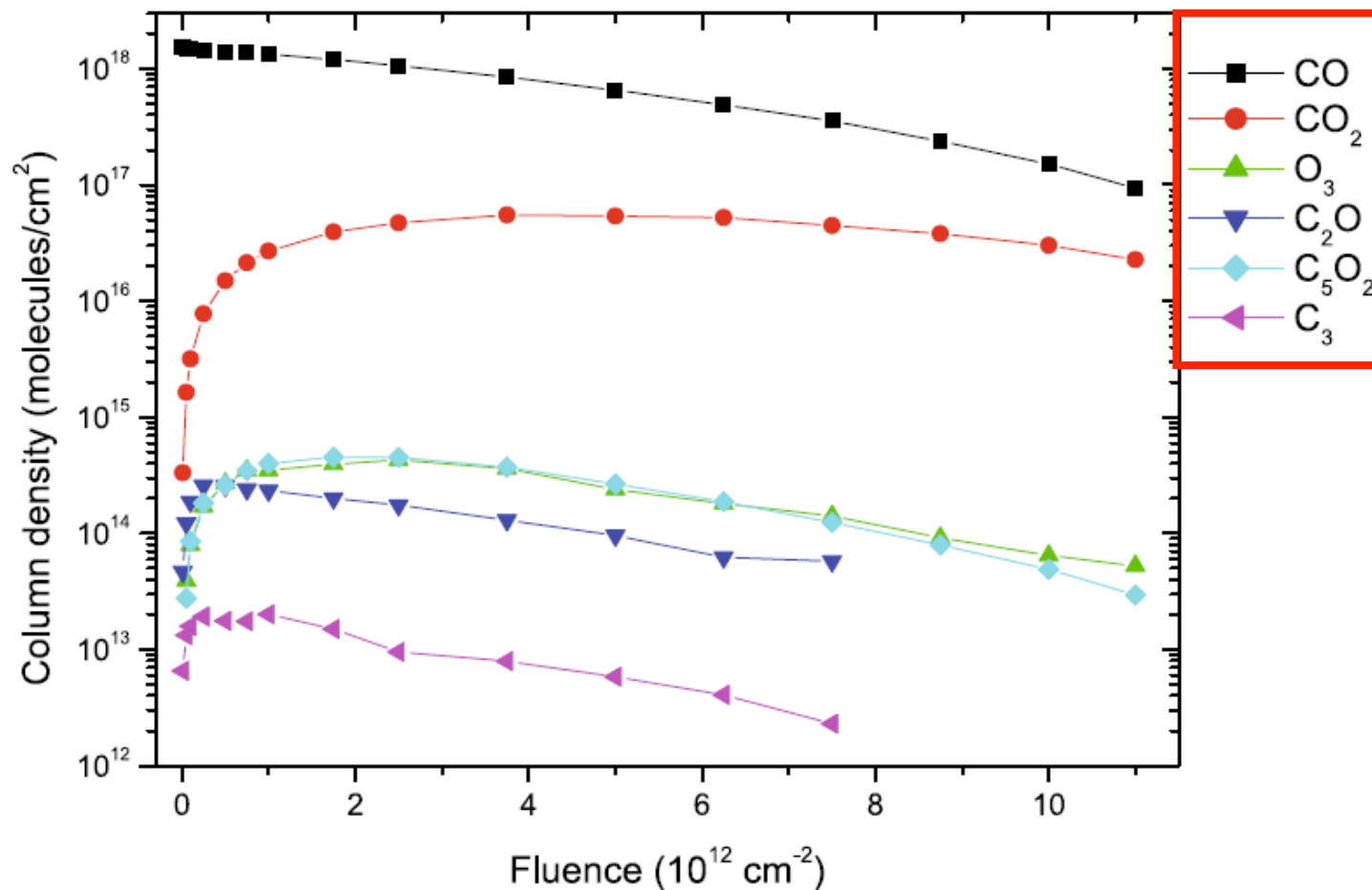
$$N = N_0 e^{-\sigma_d F} - (Y / \sigma_d) (1 - e^{-\sigma_d F})$$

deduced quantities:

- Destruction Cross Section σ_d
- Sputtering Yield Y**
- Formation cross section for daughter molecules σ_f

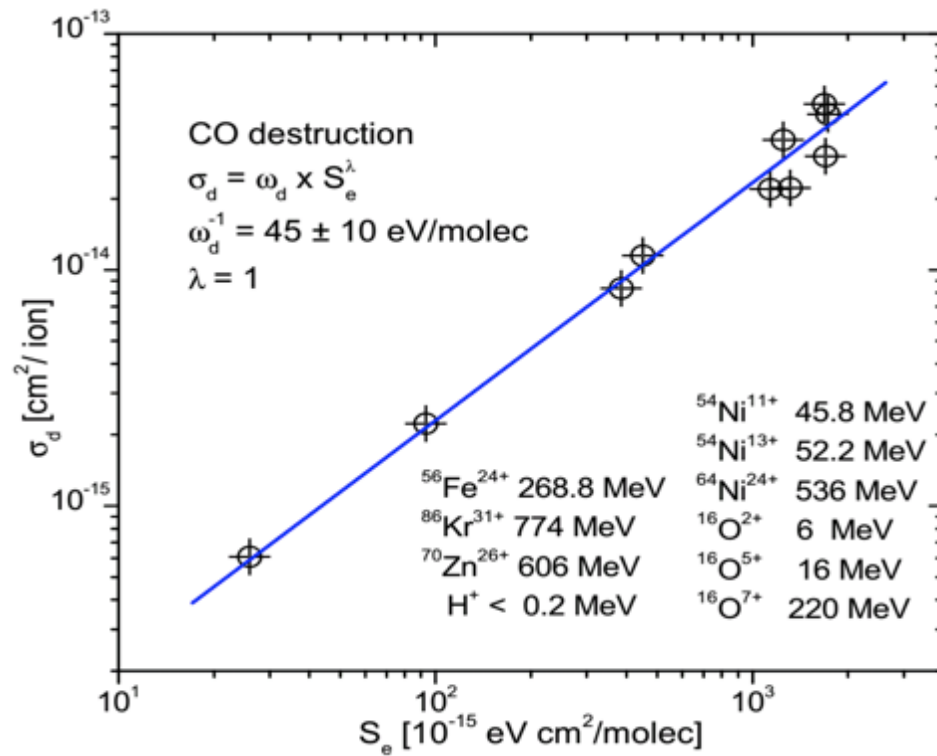


CO ice: formation of **new molecular species**



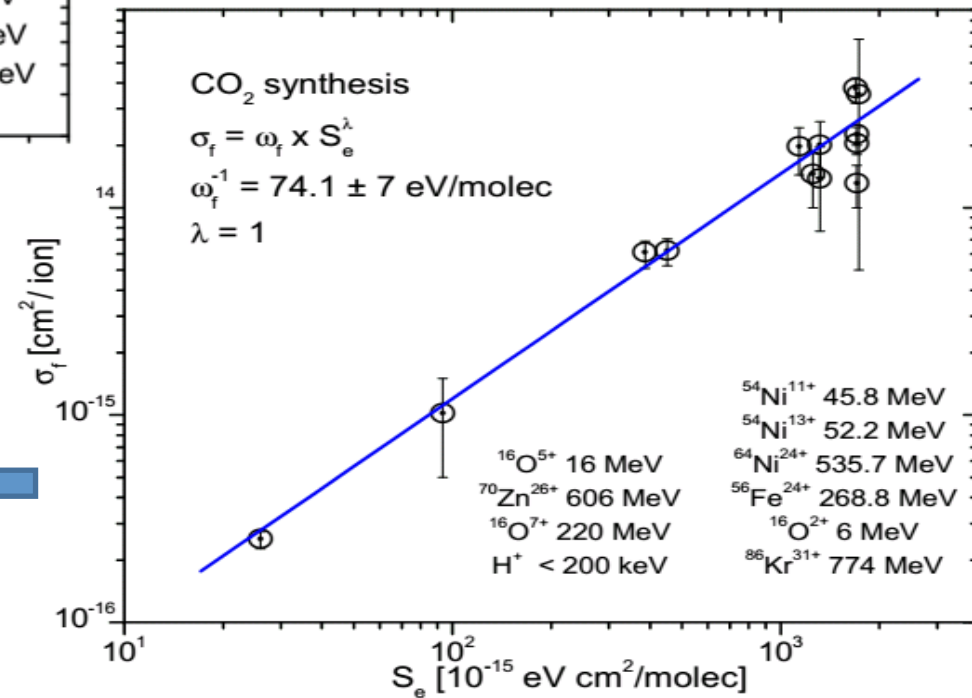
Column density of CO and molecules produced as a function of fluence of 50 MeV Ni ions.



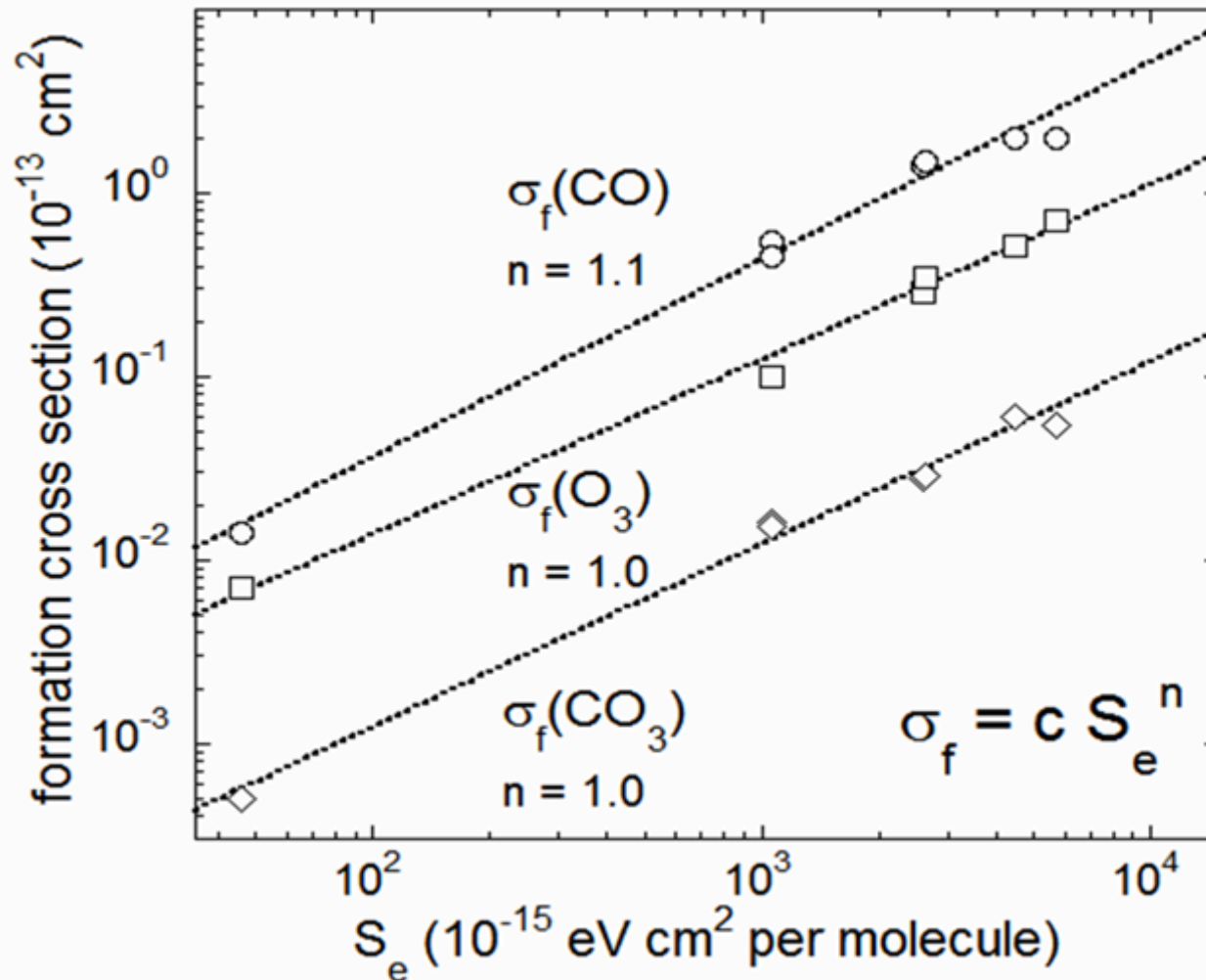


CO destruction cross section proportional to S_e

CO₂ formation cross section proportional to S_e

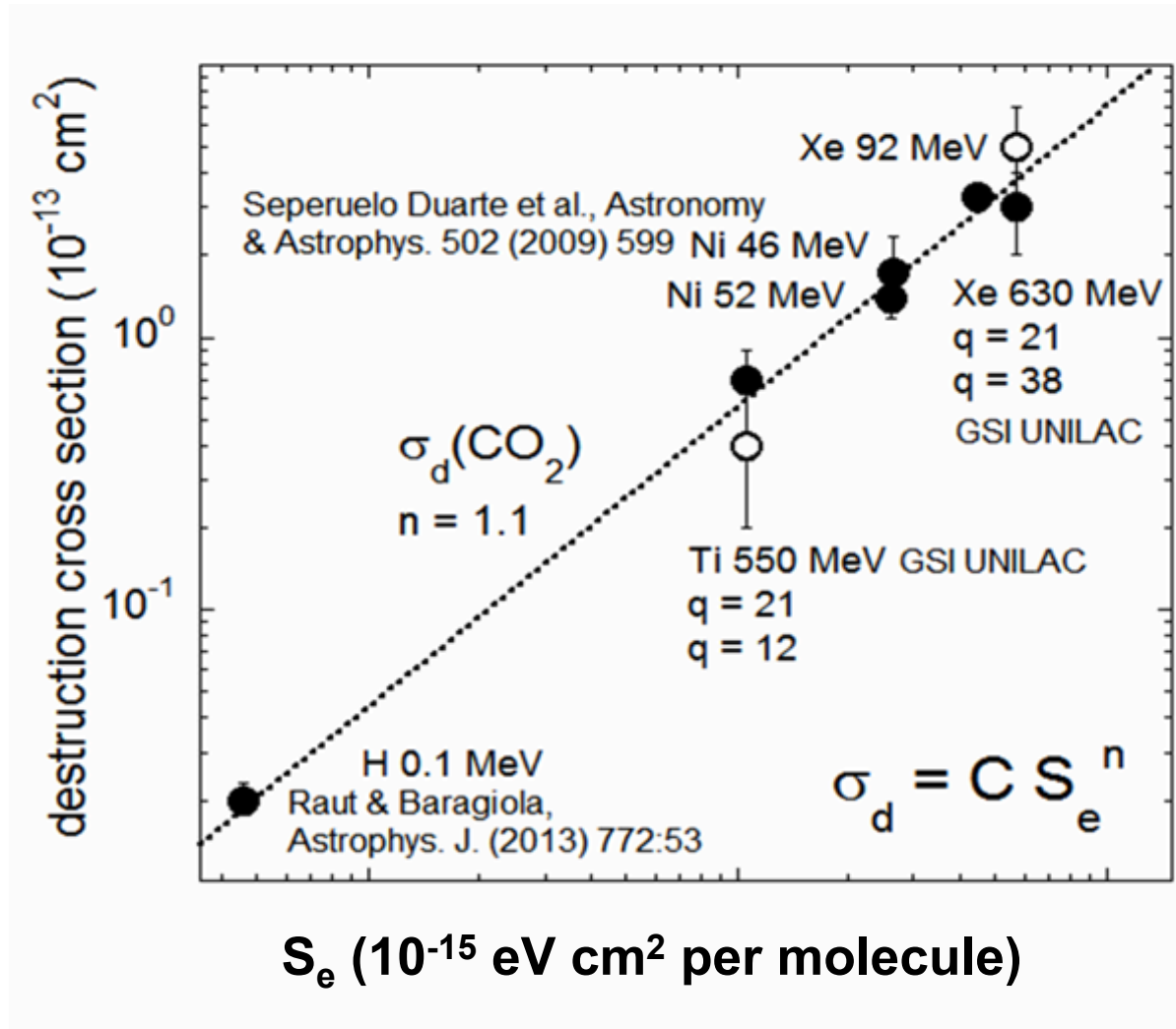


CO₂: Formation cross sections



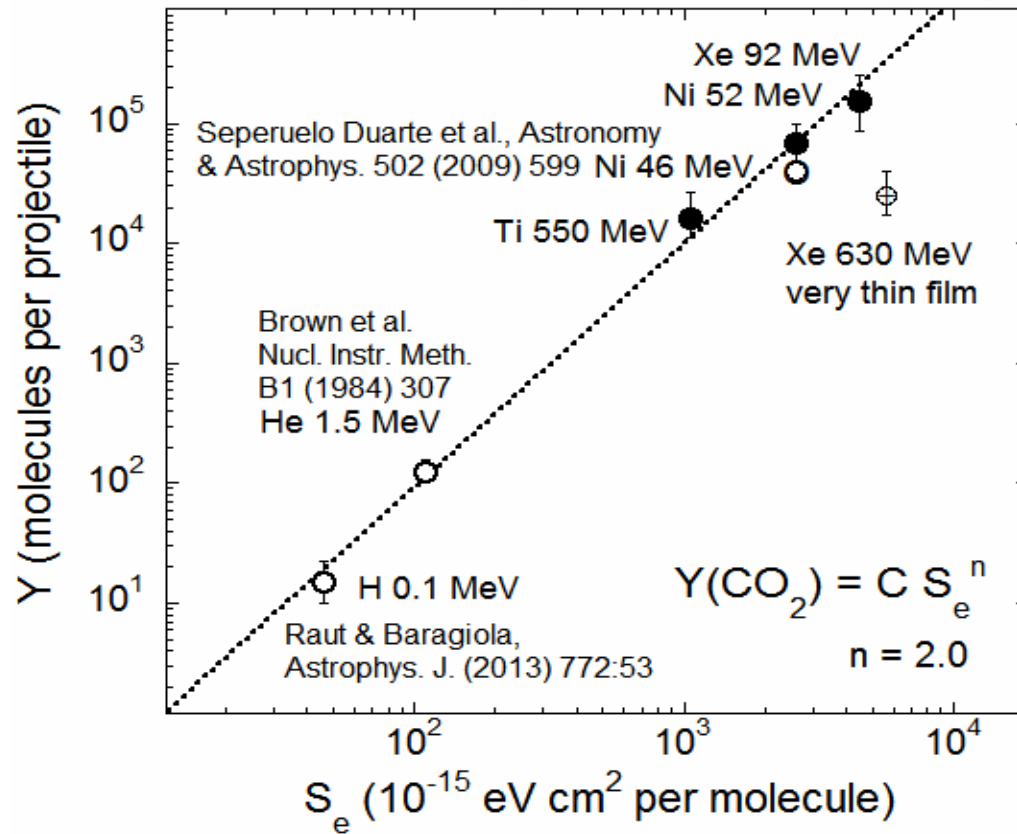
again:
formation
cross
sections
proportional
to S_e

CO₂: Destruction cross sections



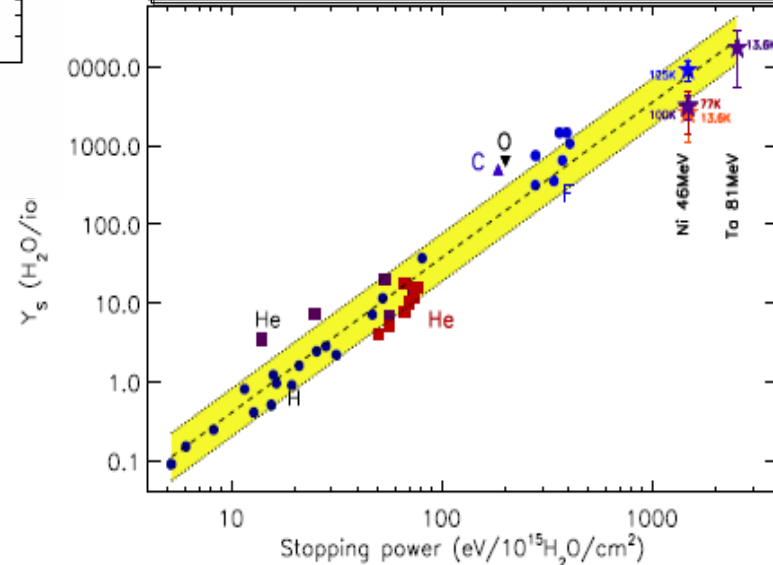
again:
destruction
cross section
proportional
to S_e

CO₂: Sputtering yield

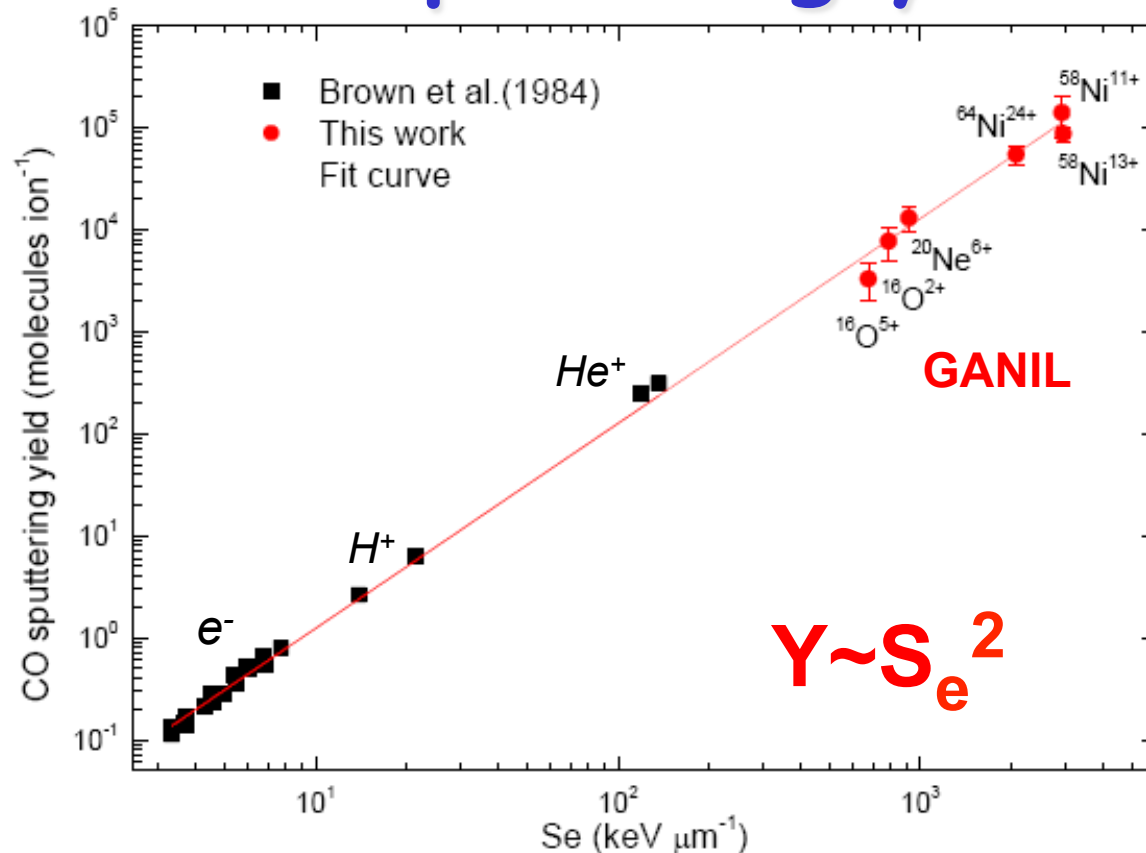


**Y scales
with S_e^2 !**

also true for **H₂O!**
estimation for
galactic cosmic rays:
 $Y_{\text{GCR}}(\text{H}_2\text{O}) \approx 10 \text{ H}_2\text{O}/\text{cm}^2/\text{s}$



CO: Sputtering yield



again:

$$Y \sim S_e^2$$

(universal ...)

$$S_e \sim Z_p^2$$

$$Y \sim Z_p^4$$

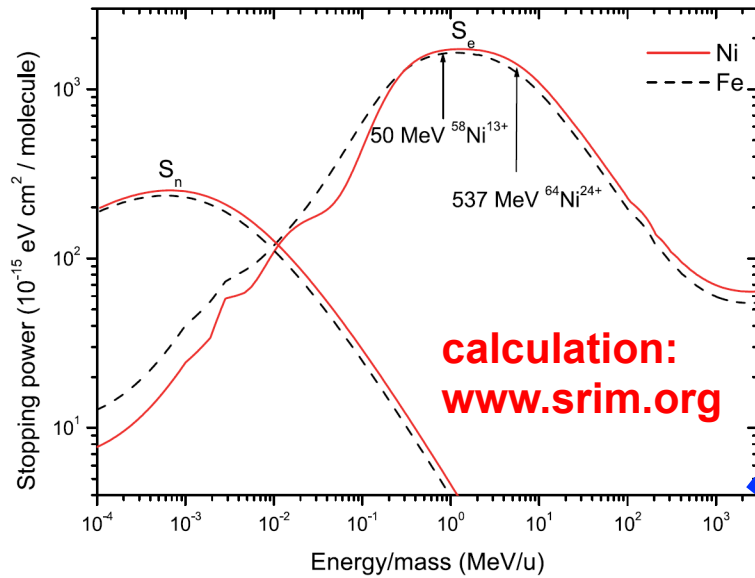
very strong dependence!

W.L. Brown, W.M. Augustyniak, K.J. Marcantonio, E.H. Simmons, J.W. Boring, R.E. Johnson, C.T. Reimann, Nucl. Instr. Meth. B1 (1984) 307

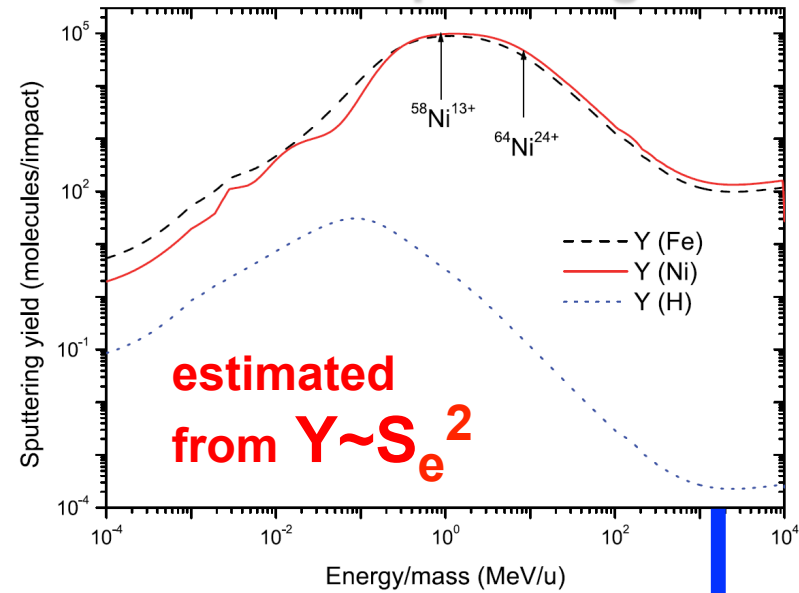
E. Seperuelo Duarte, A. Domaracka, P. Boduch, H. Rothard, E. Dartois, E.F. da Silveira Astronomy & Astrophysics 512 (2010) A71



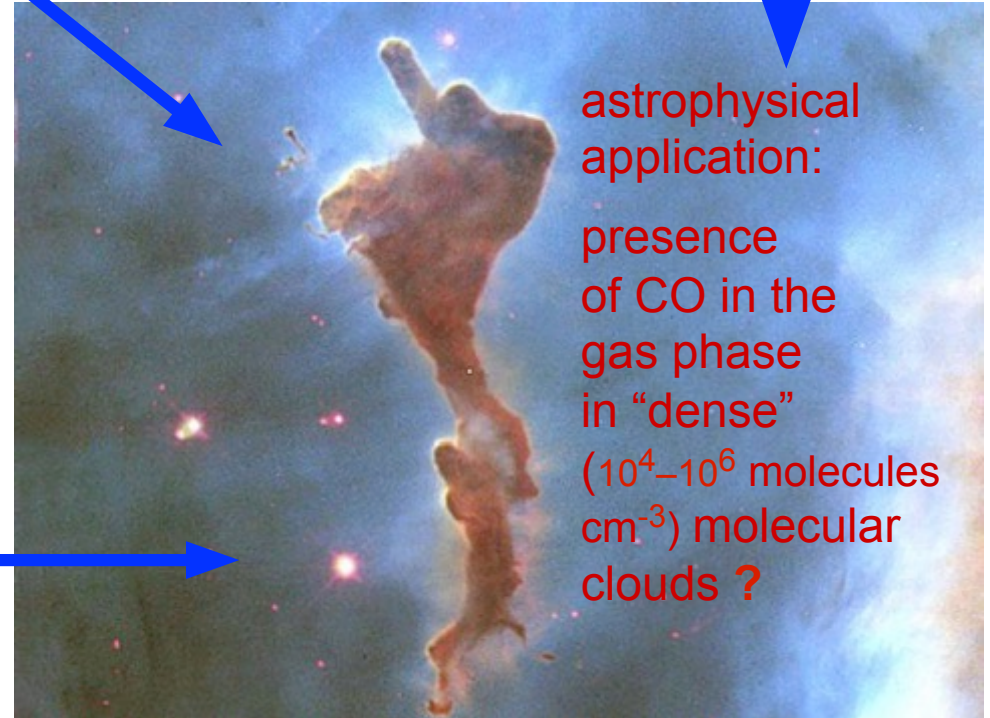
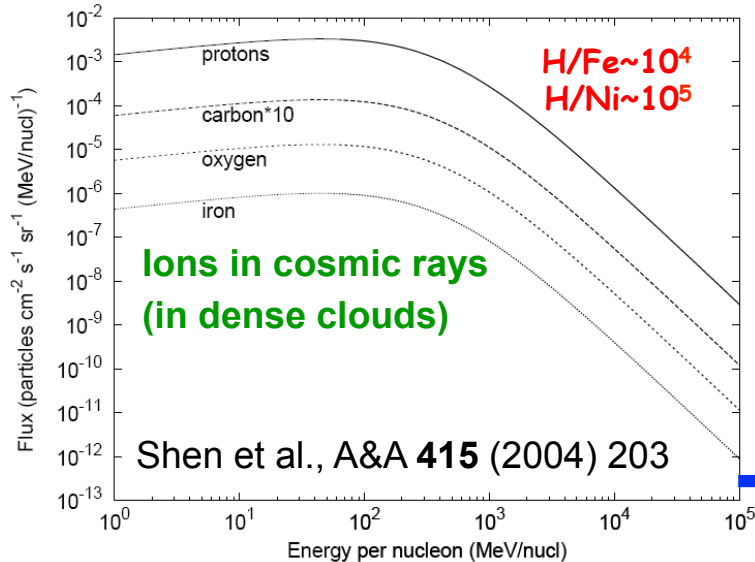
Electronic Energy Loss in CO



CO Sputtering Yield

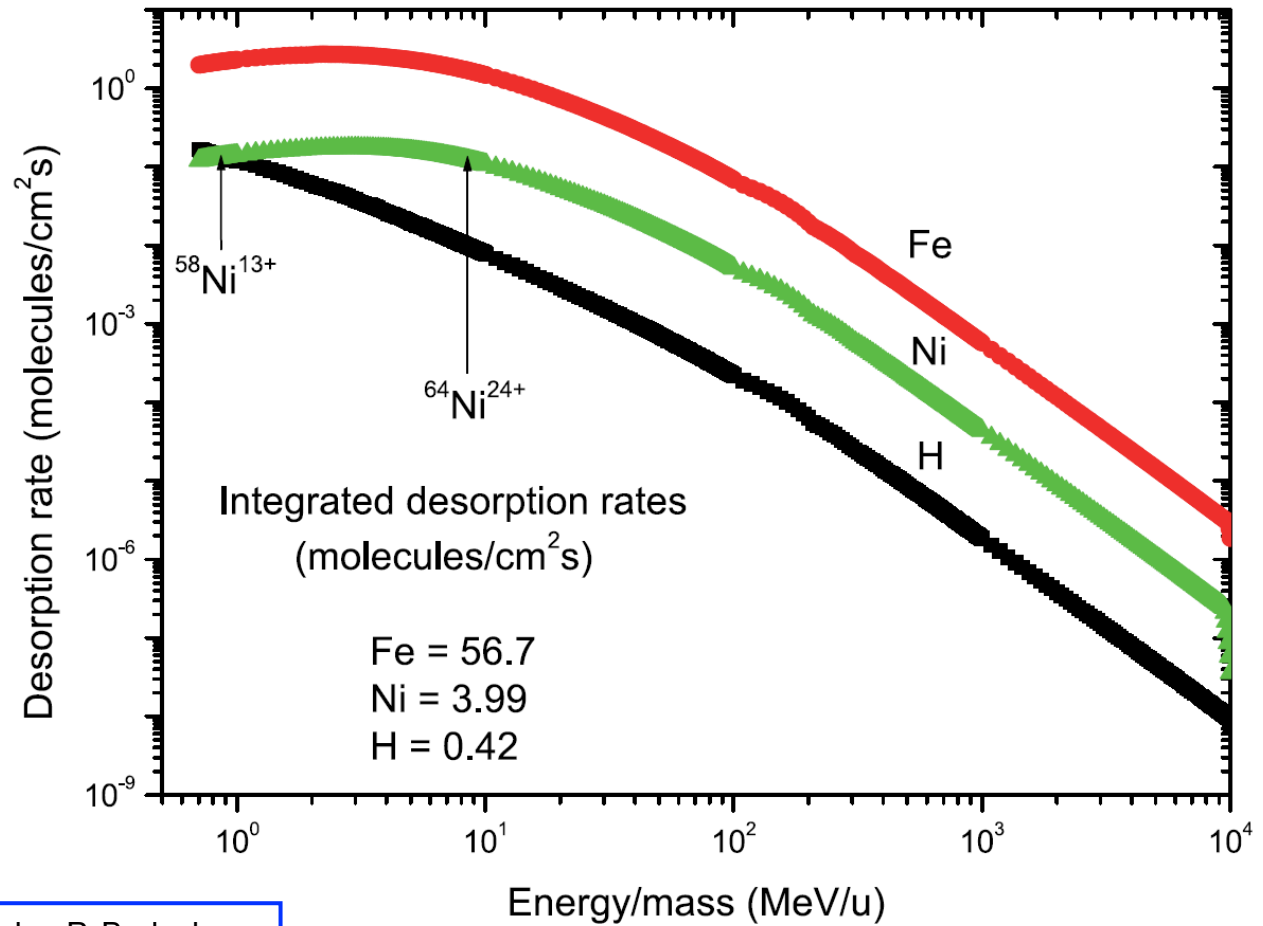
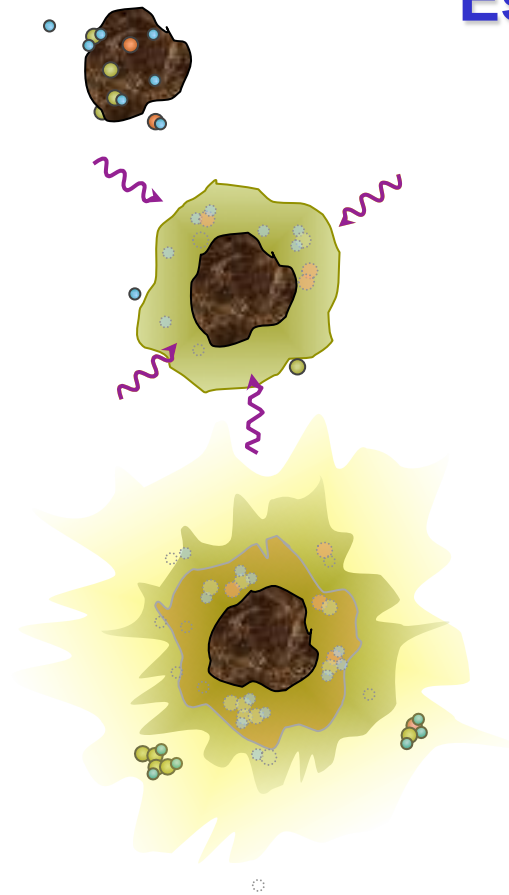


Heavy ion Abundance in space



**astrophysical
application:
presence
of CO in the
gas phase
in "dense"
(10^4 – 10^6 molecules
cm⁻³) molecular
clouds ?**

Estimated ion induced CO Desorption Yield

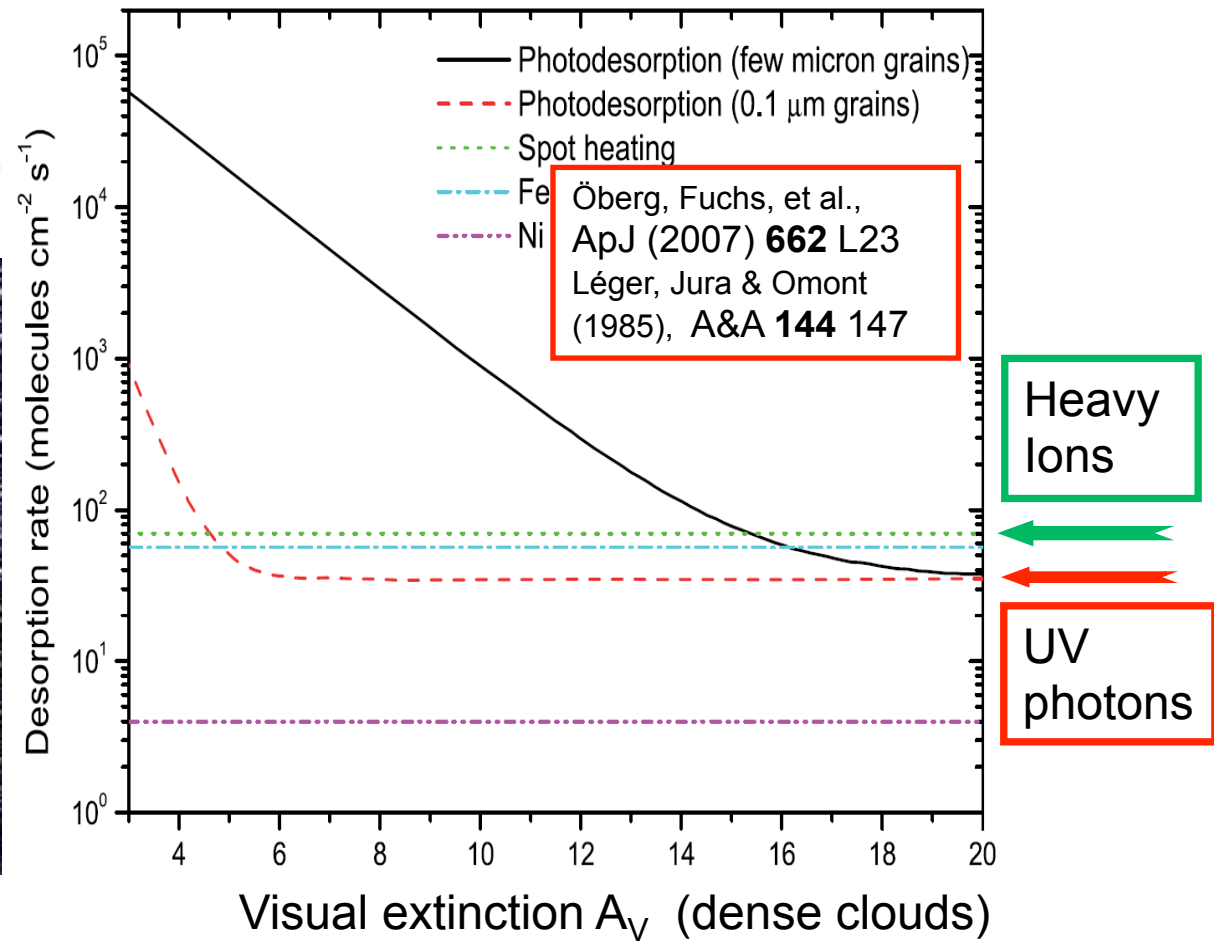


E. Seperuelo Duarte, A. Domaracka, P. Boduch,
H. Rothard, E. Dartois, E.F. da Silveira
**Laboratory simulation of heavy ion cosmic ray
interaction with condensed CO**
Astronomy & Astrophysics 512 (2010) A71



Desorption rate of CO as a function of visual extinction

penetration depth
dependence in
dense molecular clouds



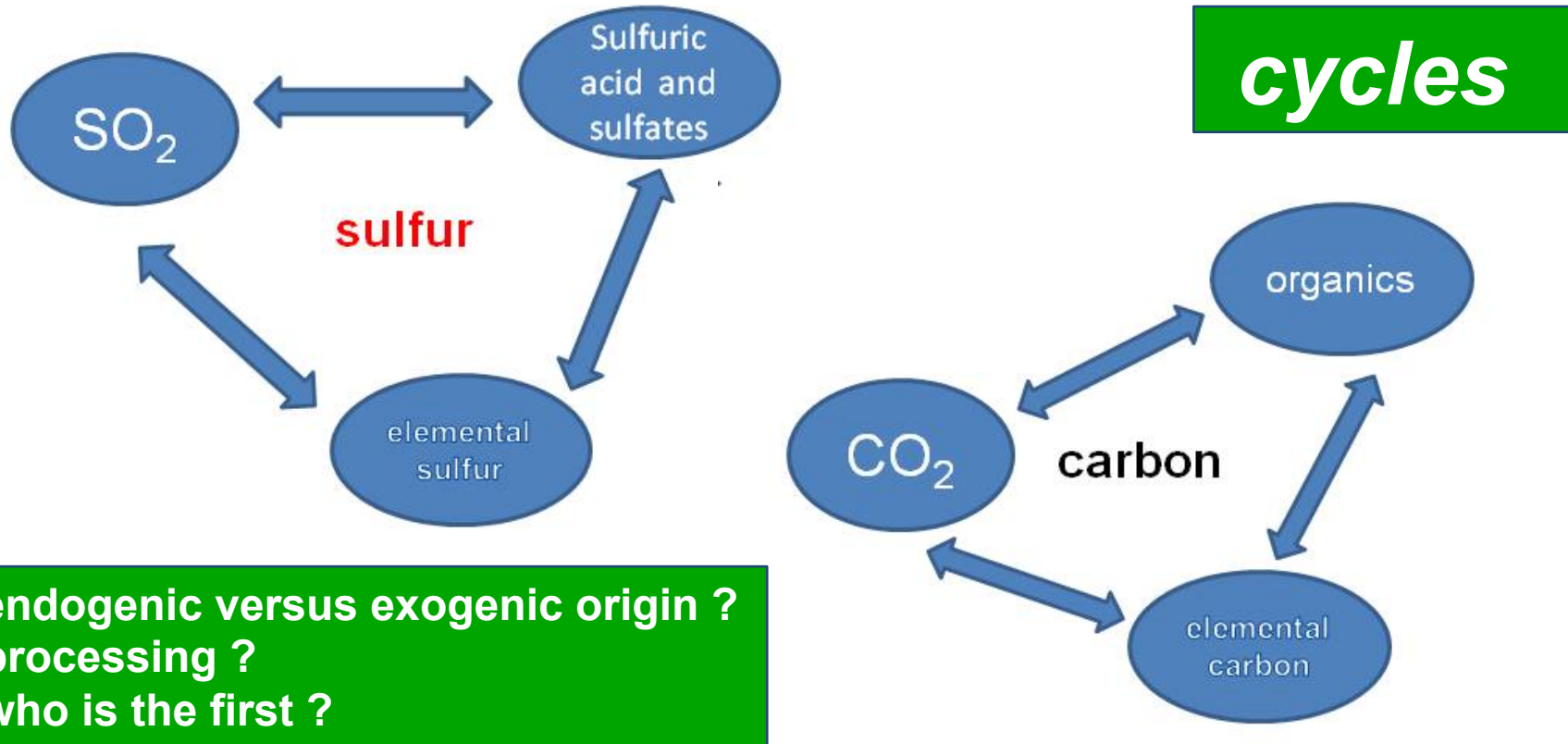
E. Seperuelo Duarte, A. Domaracka, P. Boduch, H. Rothard, E. Dartois, E.F. da Silveira
Astronomy & Astrophysics 512 (2010) A71



D

**Galilean Moons,
Jupiter's Magnetosphere,
and Sulfur and Carbon
cycles.**





Is CO₂ produced by Cⁿ⁺ ion implantation ?

Are SO₂ or H₂SO₄ produced

by Sⁿ⁺ ion implantation ?





Jupiter, NASA's spacecraft *GALILEO*, and the Galilean Moons Io, Europa, Ganymede, Callisto

Io: SO₂ ice dominant

Europa, Callisto, Ganymede: H₂O ice dominant

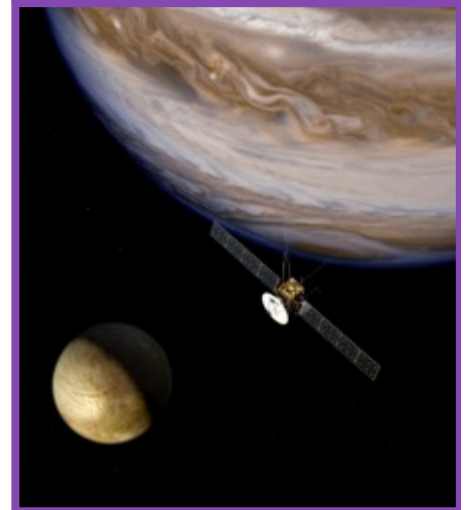
Europa: significant quantities of magnesium, sodium sulfate Na₂SO₄, carbonate hydrates

Other absorption features and prime candidates:

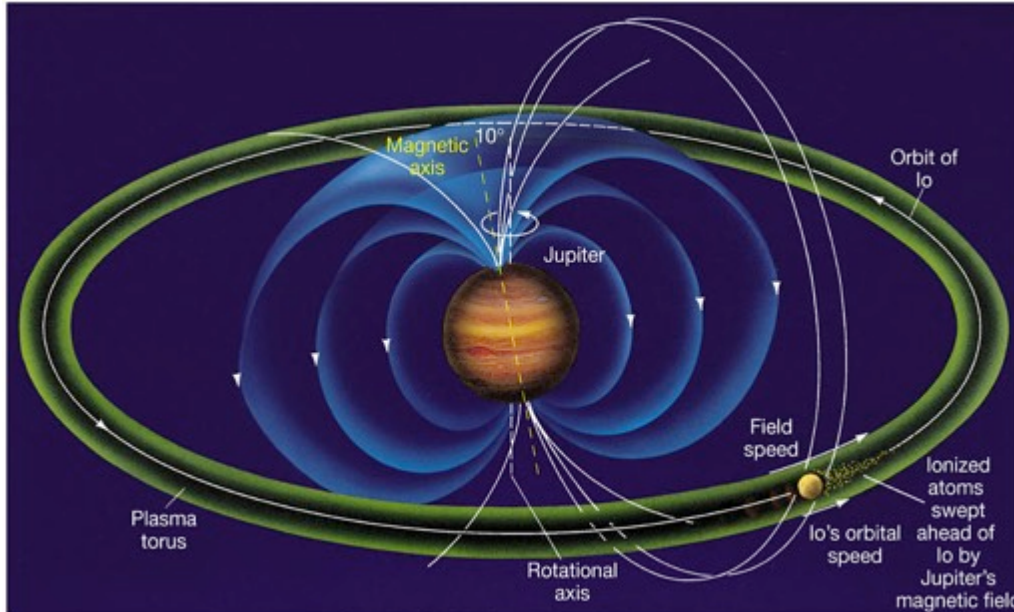
3.4 μm	(~2940 cm ⁻¹)	C-H
3.5 "	(~2857 cm ⁻¹)	H ₂ O ₂
3.88 "	(~2580 cm ⁻¹)	S-H, H ₂ CO ₃
4.05 "	(~2470 cm ⁻¹)	SO ₂
4.25 "	(~2350 cm ⁻¹)	CO ₂
4.57 "	(~2190 cm ⁻¹)	CN

Open question:
are these species **native** from the satellites or **synthesized** by **exogenic processes** e.g. ion implantation ?

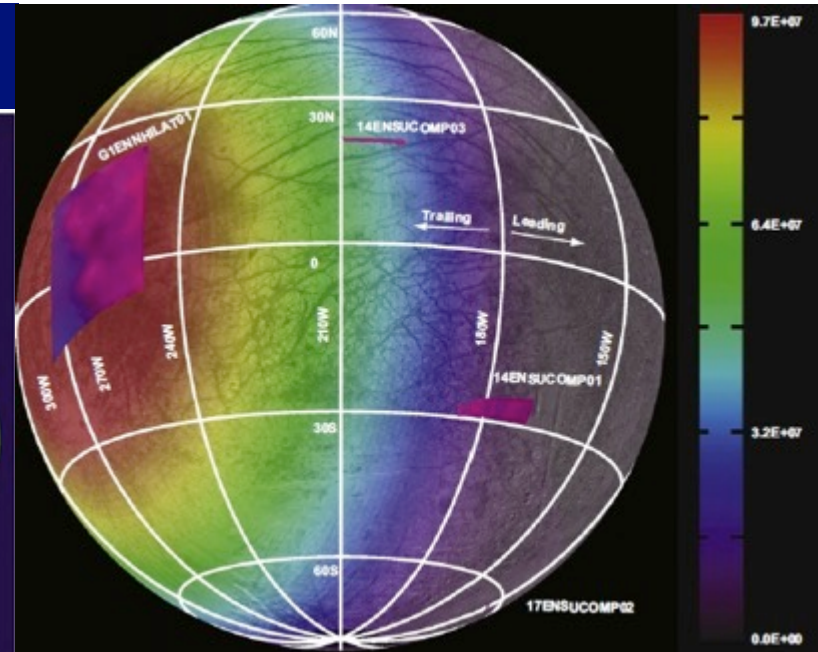
JUICE 2022 - 2033
ESA Cosmic Vision



The Jovian Magnetosphere



Copyright © 2005 Pearson Prentice Hall, Inc.



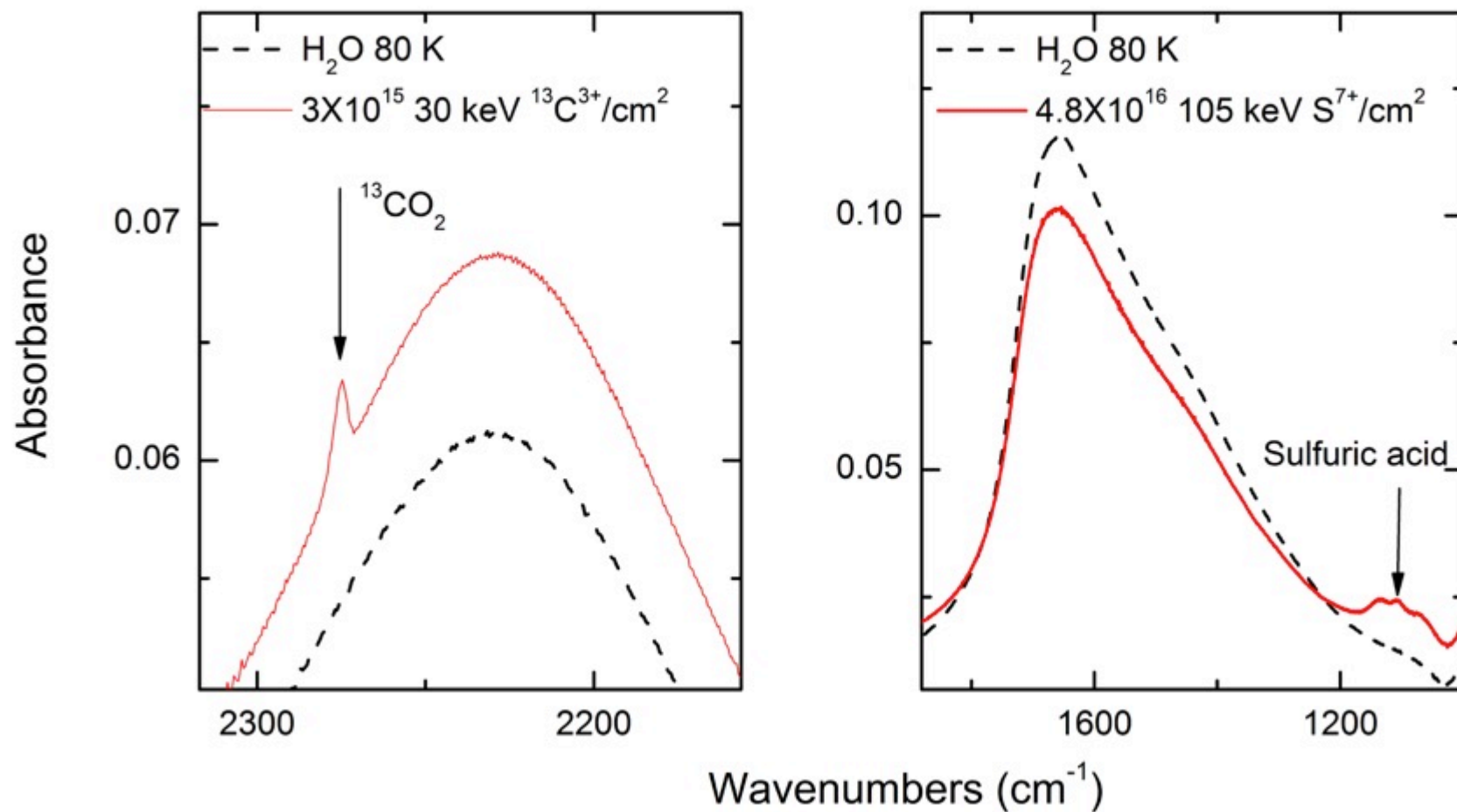
EUROPA: Sulfur Ion Flux

J.B. Dalton III, T. Cassidy, C. Paranicas,
J.H. Shirley, L.M. Prockter, L.W. Kamp
Planetary and Space Science 77 (2013) 45



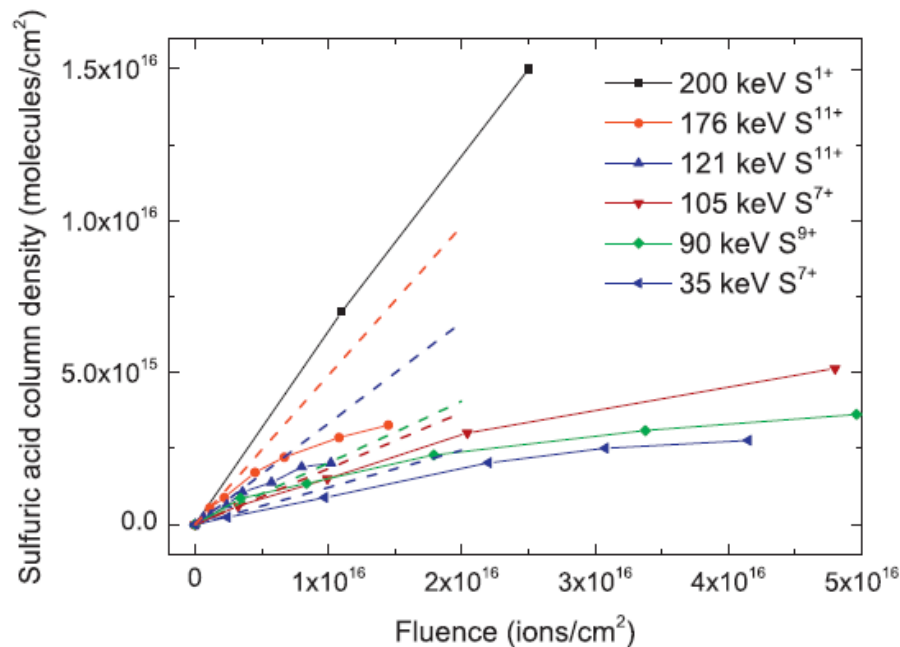
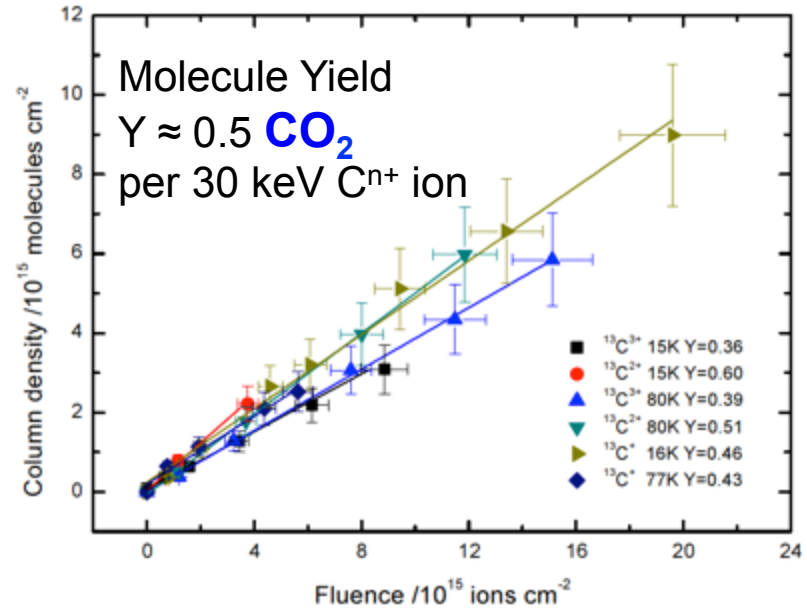
What can we do in the laboratory?

we can measure formation yields of carbon dioxide and sulfuric acid!



IR spectra of water ice before and after implantation of carbon and sulfur ions

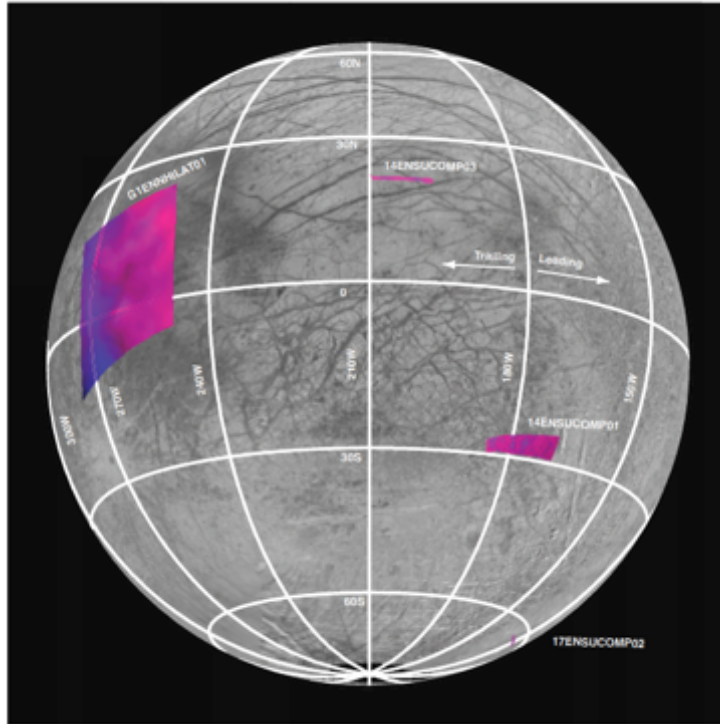
X.Y. Lv, A L F. de Barros, P. Boduch, V. Bordalo,
 E.F. da Silveira, A. Domaracka, D. Fulvio,
 C. A. Hunniford, T. Langlinay, N.J. Mason,
 A.R. W. McCullough, M.E. Palumbo,
 A.S. Pilling, H. Rothard, G. Strazzulla
**Implantation of multiply charged
 Carbon Ions in Water Ice**
Astronomy & Astrophysics 546 (2012) A81



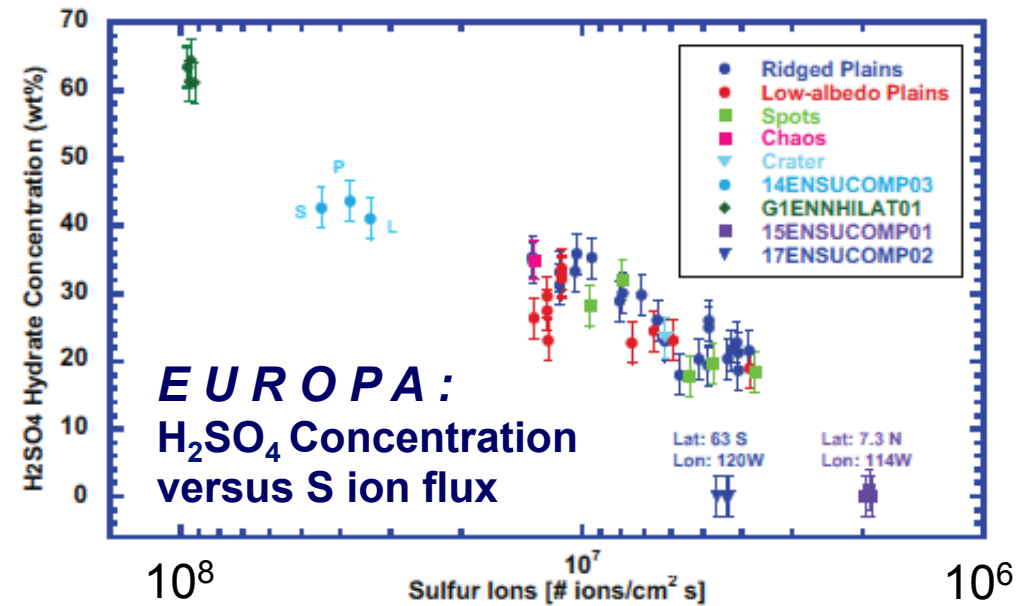
Molecule Yield
 $Y(\text{H}_2\text{SO}_4) \approx 0.12$ (35 keV)
 ≈ 0.64 (200 keV)
 per S^{n+} ion

J. J. Ding, P. Boduch, A. Domaracka,
 S. Guillous, T. Langlinay, X.Y. Lv, M.E. Palumbo,
 H. Rothard, G. Strazzulla
**Implantation of Multiply Charged
 Sulfur Ions in Water Ice**
Icarus 226 (2013) 860–864





*J.B. Dalton III et al.,
Planetary and Space Science 77 (2013) 45:*
Correlation of H₂SO₄ hydrate concentration
with sulfur ion flux



Ding et al., Icarus 336 (2013) 860:
H₂SO₄ Concentration compatible
with measured
Molecule Yield from Implantation!



time needed to obtain
 3×10^{19} molecules/cm² by implantation

Flux of S-ions (cm ⁻² s ⁻¹)	Time (years) Using Y=0.12	Time (years) Using Y=0.64
2×10^6	4×10^6	7×10^5
1×10^8	9×10^4	1.4×10^4

Ding et al., Icarus 336 (2013) 860:

**Concentration compatible with measured
Molecule Yield from Implantation!**



**Exogenic production
of H₂SO₄ probable**

but ... C implantation in water ice does **not** explain observed **CO₂ concentration**.

and ... no evidence (yet) for production of SO₂ or H₂S in *water ice* ...



Sulfur implantation in CO and CO₂ ices

X. Y. Lv^{1,2*}, P. Boduch², J. J. Ding², A. Domaracka², T. Langlinay², M. E. Palumbo³,
H. Rothard² and G. Strazzulla^{3†}

¹School of Nuclear Science and Technology, Lanzhou University, Lanzhou 730000, China

²Centre de Recherche sur les Ions, les Matériaux et la Photonique (CEA/CNRS/ENSICAEN/UCBN), CIMAP-CIRIL-GANIL,
Boulevard Henri Becquerel, BP 5133, F-14070 Caen Cedex 05, France

³INAF-Osservatorio Astrofisico di Catania, Catania, Italy

Monthly Notices of the Royal Astronomical Society MNRAS 438 (2014) 922

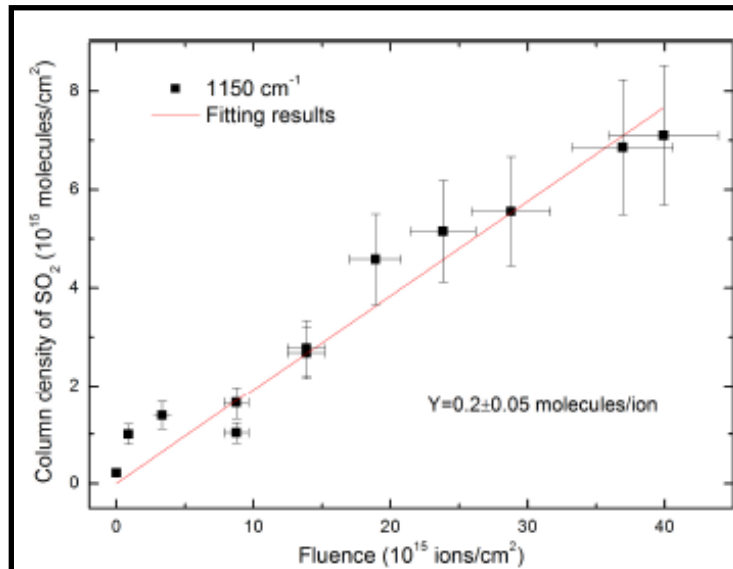


Figure 4. Column density of SO₂ produced after 176-keV S¹¹⁺ implantation in CO ice, as a function of ion fluence. The error bar is about 10 per cent for fluence and 20 per cent for column density.

S implantation in CO and CO₂:

$Y(\text{SO}_2) = 0.20 \pm 0.05$ molec./ion at 176 keV in **CO**

$Y(\text{SO}_2) = 0.38 \pm 0.02$ molec./ion at 90 keV in **CO₂**

and ... **CS₂** produced in **CO₂** and **OCS** in **CO**

Europa: time to produce observed amount of **SO₂** depends strongly on CO₂ concentration: 200 years ... up to 20000 years

Star forming regions: strong flux of stellar wind from young stars (T-Tauri phase) interacts with CO₂ rich dust, later incorporated in comets



E

**Formation of
prebiotic molecules**



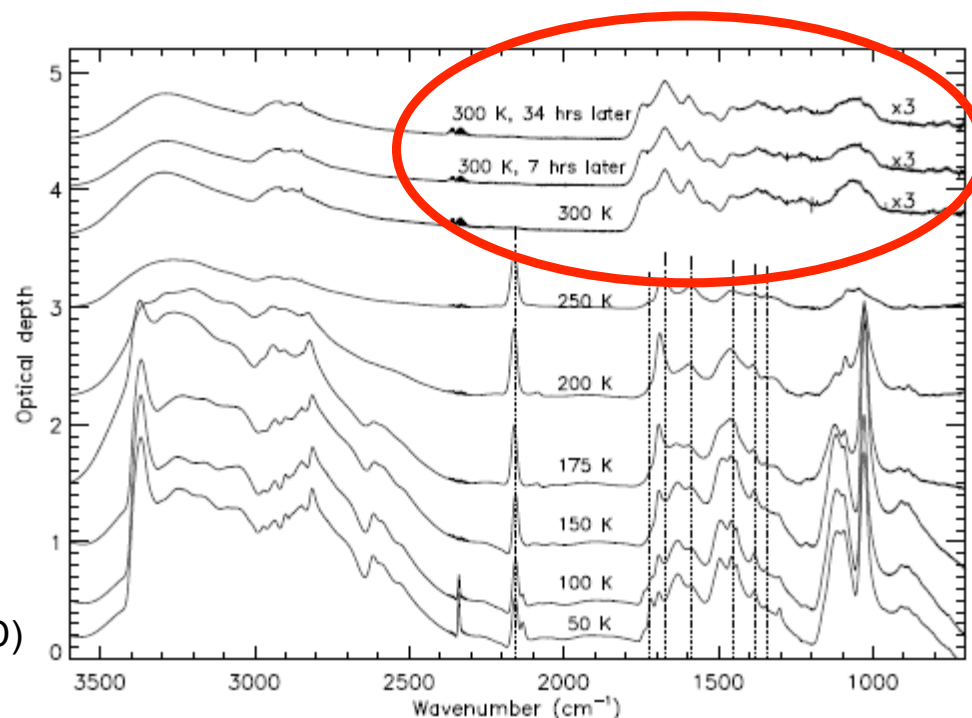
Radiolysis: formation of prebiotic molecules ?

G. M. Muñoz Caro, E. Dartois,
P. Boduch, H. Rothard,
A. Domaracka, A. Jiménez-Escobar
Comparison of UV and high-energy ion irradiation of methanol:ammonia ice,
Astron. & Astrophys. 566 (2014) A93

$\text{NH}_3:\text{CH}_3\text{OH}$ ice

CASIMIR@GANIL: Zn (SME), Ne (IRRSUD)

New bands attributed to irradiation products



position ^a (cm^{-1})	Assignment	vibration mode
2340	CO_2	CO str.
2160	OCN^-	CN str.
2138	CO	CO str.
1740	C=O ester/aldehyde	CO str.
1720	H_2CO	CO str.
1694	HCONH_2 ?	CO str.
1587	COO^- in carb. ac. salts ^{b,c}	COO^- asym. str.
1498	H_2CO	CH_2 scis.
1385	CH_3 groups	CH_3 sym. def.
1347	COO^- in carb. ac. salts ^{b,c}	COO^- sym. str.
1303	CH_4	def.

**at 300K:
stable organic
Residues!**

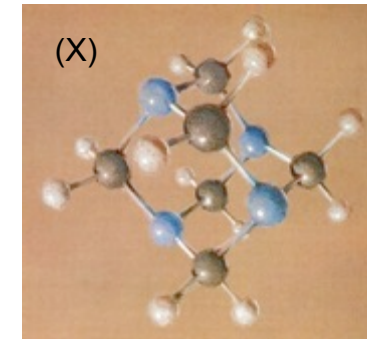
UV vs. heavy ions:
same products



Frequency (cm ⁻¹)	Wavelength (μm)	Temp. (K)	Molecule
2233	4.48	13	N ₂ O
2218–2200	4.51–4.54	300	nitriles [†]
2168	4.61	13, 300	OCN ⁻
2147	4.66	300	aliph. isocyanide [†]
~2112	4.73	300	NCO ₂ [†]
1725	5.80	300	ester [†]
1683	5.94	300	amides [†]
1652	6.05	300	asym-N ₂ O ₃ [†]
1637	6.11	13	?
1593	6.28	300	NH ₃ ⁺ CH ₂ COO ^{-†}
1558	6.42	300	?
1533	6.52	300	?
1506	6.64	300	NH ₃ ⁺ CH ₂ COO ^{-†}
~1490	6.71	13	NH ₄ [†]
1474	6.78	13	NO ₃ [†]
1440	6.94	13	NH ₃ ⁺ CH ₂ COO ^{-†}
1415	7.07	300	NH ₃ ⁺ CH ₂ COO ^{-†}
~1370	7.30	13, 300	HMT [†] HCOO ⁻
~1338	7.47	13, 300	NH ₃ ⁺ CH ₂ COO ^{-†} NH ₂ CH ₂ COO ^{-†} HCOO ⁻
1305	7.66	13	N ₂ O ₃ [†] ; N ₂ O ₄ [†]
1283	7.80	300	N ₂ O [†]

H₂O - CO - NH₃ ice

⇒ glycine (amino acid)



hexamethylene-
tetramine HMT

Donia Baklouti, R. Brunetto, E. Dartois, IAS
(Institut d'Astrophysique Spatiale, Orsay), et al.

Analysis of the **Residues** by Chromatography

H₂O- NH₃-CH₃OH ice

(experiment 2014)



S. Pilling, E. Seperuelo Duarte, E. F. da Silveira,
E. Balanzat, H. Rothard, A. Domaracka, P. Boduch
*Radiolysis of ammonia-containing ices by
energetic, heavy and highly charged ions
inside dense astrophysical environments,*
Astronomy & Astrophysics 509 (2010) A87

Williamary Portugal, Sergio Pilling, Philippe Boduch, Hermann Rothard, Diana Andrade
*Radiolysis of Amino Acids by Heavy and Energetic Cosmic Ray Analogs
in Simulated Space Environments: α-Glycine Zwitterion Form*
Monthly Notices of the Royal Astronomical Society 441 (2014) 3209–3225

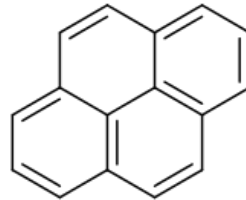
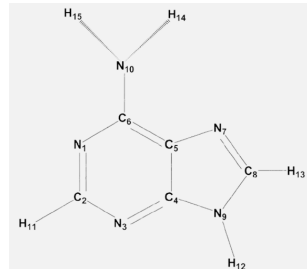
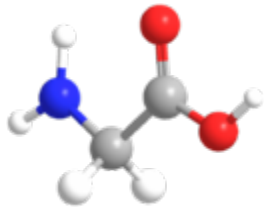
(X) <http://osulibrary.oregonstate.edu/specialcollections/coll/pauling/bond/index.html>



Radiation resistance of organic molecules

⇒ Irradiation of ices containing complexe molecules

e.g. glycine, adenine, PAH (Polycyclic aromatic hydrocarbons)



other materials:
carbonaceous, silicates
chemistry at interfaces?

complex radiation field in space: synergy effects?



Emergence of life: Present + future Space Missions

Origin of organic matter in ...

Comets

Rosetta + Philae ESA

Micrometeorites e.g. UCAMMs:
(Ultracarbonaceous Antarctic
Micrometeorites)

Asteroids

Hayabusa JAXA

Transneptunian Objects
(e.g. Pluto + Charon)
+ Kuiperbelt Objects

New Horizons NASA

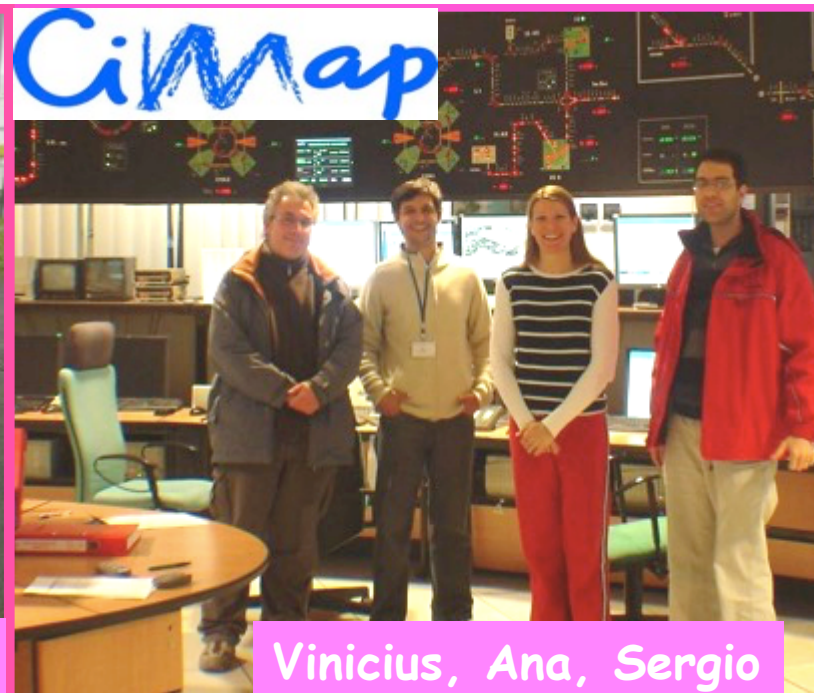
Jovian Moons

JUICE ESA Cosmic Vision launch 2022
(Jupiter Icy Moons Explorer)





Jingjie, Xueyang, Gianni, Thomas,
Hermann, Philippe, Stéphane



Vinicius, Ana, Sergio



Hussein+Enio



Eduardo



Emmanuel



Alicja