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Signature of Charge Equilibrium Process in L-Shell Auger-Meitner Electron Emission Spectra of Ar^{+q}-Ar Collisions

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The well-known Auger-Meitner (AM) effect has been used extensively to study electron correlations. In ionatom collisions, the AM process occurs during hard collisions (small impact parameter). It is known that for ions, the energy of AM e^- shifts to the low-energy side, in contrast to X-ray energy, which shifts toward the high-energy side. For ion-atom collisions, understanding this process can be categorized into two regimes: (1) Coulomb ionization (for proton and bare-ion projectiles), and (2) excitation or ionization of a quasi-molecule. When the projectile moves very fast relative to the orbital electron or is lighter than the target, the process can be understood through Coulomb ionization. Excitation or ionization of a quasi-molecule has been studied for X^+ -X collision systems. However, very few studies are available in the literature where the projectile undergoes inner-shell excitation.

In this talk, I shall discuss the double differential cross section (DDCS) for electron emission in Ar^{3+} –Ar and Ar^{6+} –Ar collisions. A significant shift exceeding 40 eV is observed in the characteristic Auger-Meitner (AM) electron energy peak for LMM transitions. For a neutral Ar target, this peak is expected near 197 eV; the shift occurs for both projectile and target after Doppler-shift correction. Although existing literature attributes this shift to binding-energy changes in ions versus neutral atoms, experimental peaks typically result from convolutions of multiple peaks. Here, simply knowing binding-energy changes is insufficient to quantify ionization. We adopt an approach where the probability of AM transitions is calculated using ground-state wave functions of ions. Using these probabilities as weight factors for the energy-peak function, the shift in convoluted peaks may quantify ionization. Comparing calculations with experiments, both target and projectile exhibit equilibrium charge states between 3+ and 4+ before AM emission. I shall also discuss experimental results for C^{q+} (q=1,2)–CH4 and N^{q+} (q=2,3)–CH4 collision systems.

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