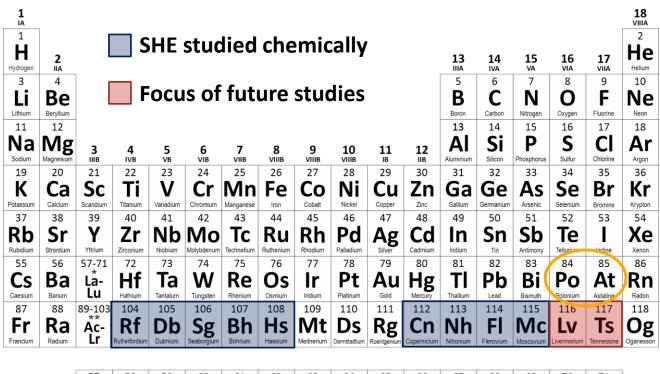


Reactivity of polonium towards quartz surfaces

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Johannes Gutenberg University Mainz

Towards chemistry with livermorium



| | 57 | 58 | 59 | 60 | 61 | 62 | 63 | 64 | 65 | 66 | 67 | 68 | 69 | 70 | 71 |
|--------------|-----------|---------|--------------|-----------|------------|-----------|-----------|------------|-----------|-------------|-------------|---------|-------------|-----------|------------|
| *Lanthanides | La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Tb | Dy | Но | Er | Tm | Yb | Lu |
| | Lanthanum | Cerium | Praseodymium | Neodymium | Promethium | Samarium | Europium | Gadolinium | Terbium | Dysprosium | Holmium | Erbium | Thulium | Ytterbium | Lutetium |
| | 89 | 90 | 91 | 92 | 93 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 | 103 |
| **Actinides | Ac | Th | Pa | U | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No | Lr |
| | Actinium | Thorium | Protactinium | Uranium | Neptunium | Plutonium | Americium | Curium | Berkelium | Californium | Einsteinium | Fermium | Mendelevium | Nobelium | Lawrencium |

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[1] A. Yakushev et al., Front. Chem. 2024, 12, 1474820.

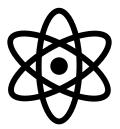
Mc and Nh studied recently^[1]

Studying heavier SHE requires...





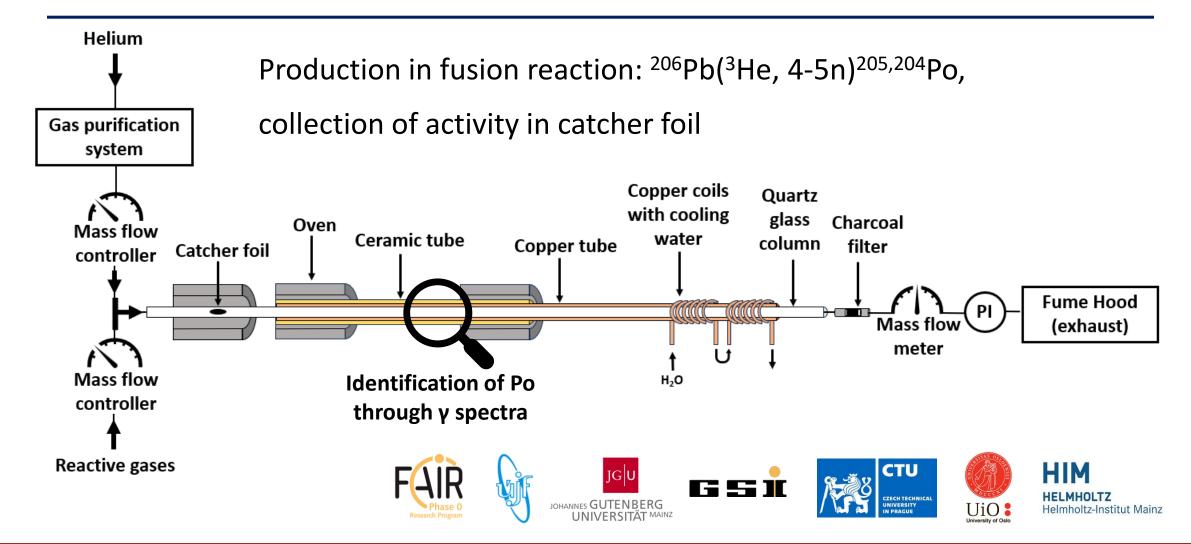
> Unicell



...Studying the chemistry of homologues

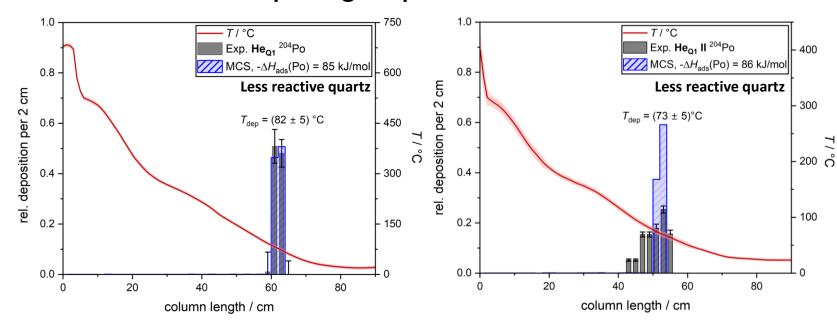
> Polonium, Astatine

Studying Po at the atom-at-a-time regime

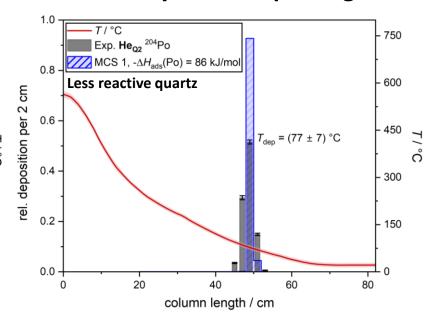


Adsorption enthalpy of elemental polonium confirmed in multiple experiments

On quartz glass produced in vacuum



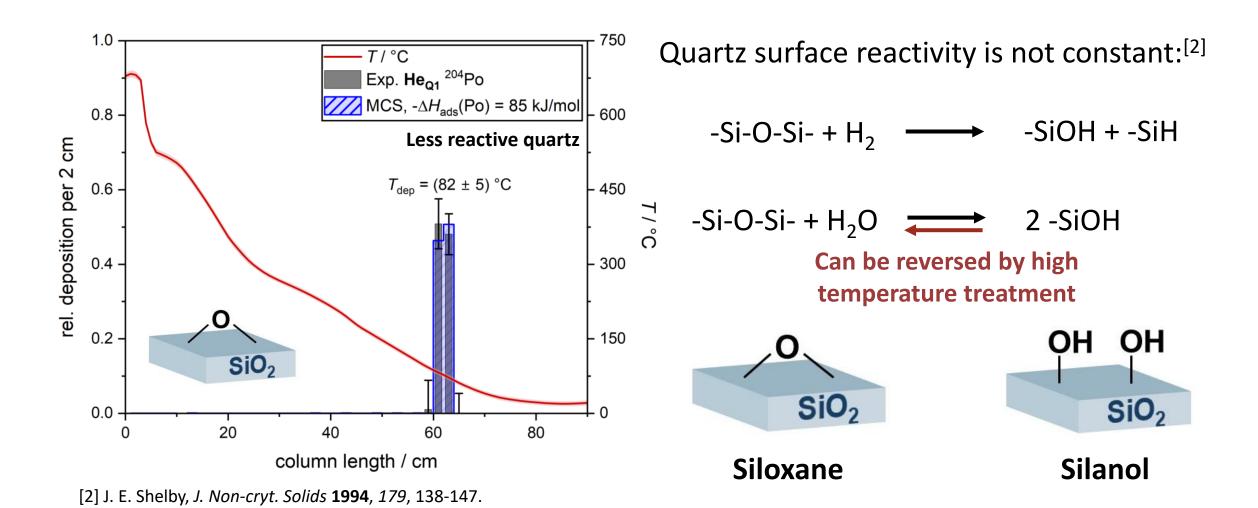
On thermally treated quartz glass



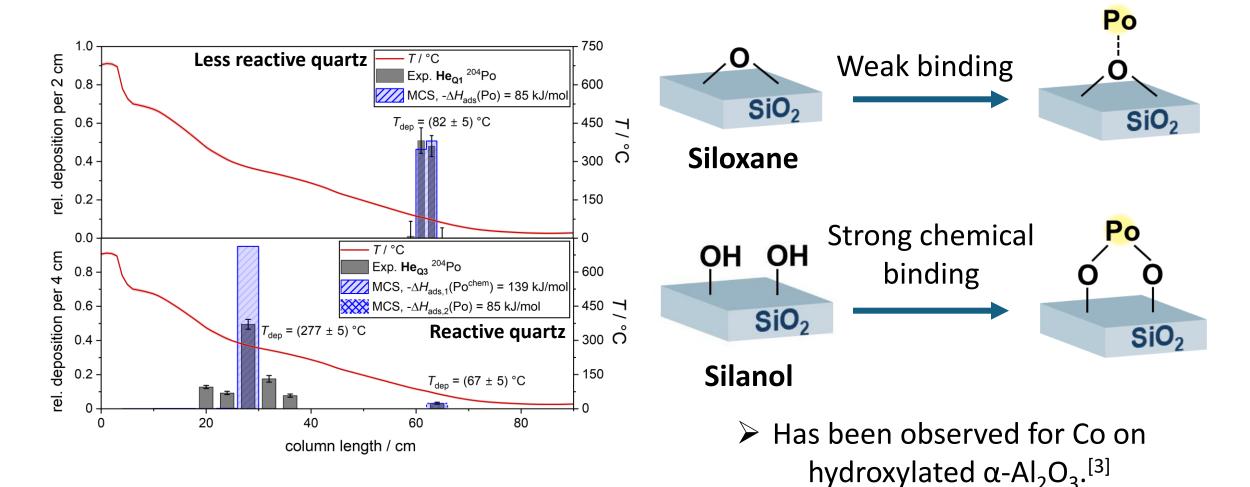
- > All experiments conducted in pure helium.
- > Species is assigned to elemental polonium.

$$-\Delta H_{ads}(Po) = 85^{+3}_{-2} \text{ kJ/mol}$$

Unravelling the influence of the quartz surface



Unravelling the influence of the quartz surface



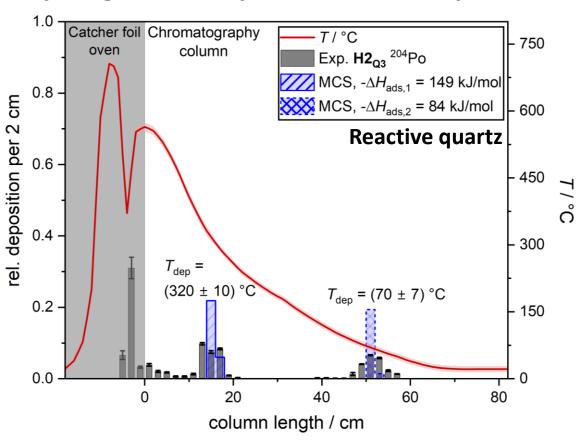
[3] S. A. Chambers, Science 2002, 297, 827-831.

No experimental evidence for reaction of polonium with hydrogen

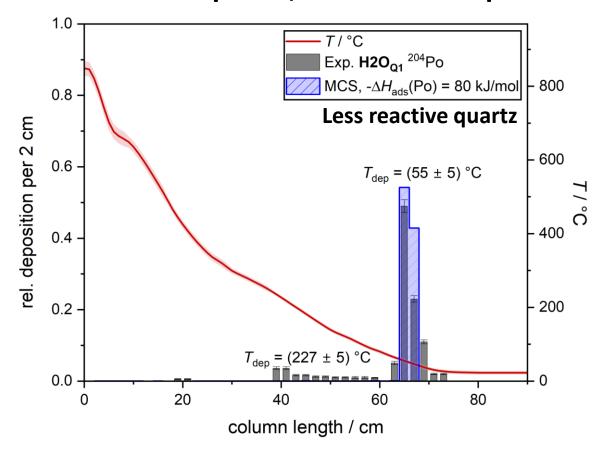
Hydrogen atmosphere, reactive quartz Chemical reaction with the surface Catcher foil Chromatography 750 oven column Exp. **H2_{O3}** ²⁰⁴Po $-Si-O-Si- + H_2$ -SiOH + -SiH MCS, $-\Delta H_{ads.1} = 149 \text{ kJ/mol}$ 8.0 rel. deposition per 2 cm \times MCS, $-\Delta H_{\text{ads},2}$ = 84 kJ/mol IL 600 OH OH **Reactive quartz** 450 SiO2 - 300 $T_{\rm dep} =$ $(320 \pm 10) ^{\circ}C$ Deposition of elemental polonium $= (70 \pm 7) ^{\circ}$ 0.2 0.0 > No additional species observed in hydrogen column length / cm

No experimental evidence for reaction of polonium with hydrogen and water

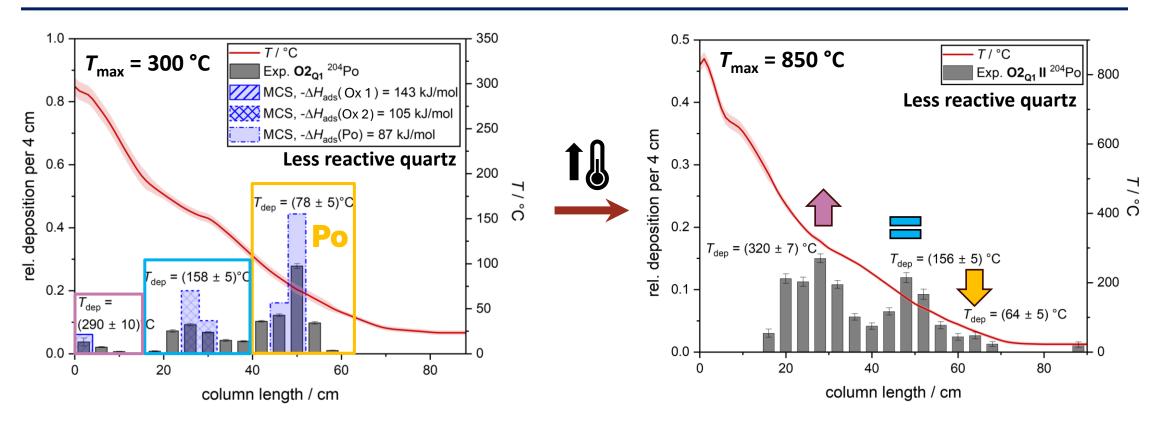
Hydrogen atmosphere, reactive quartz



Moist atmosphere, less reactive quartz



Reactions of polonium with oxygen



> Two oxidized species involved.

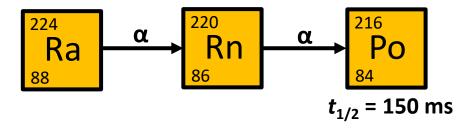
> Signs of chemical reaction!

Summary and outlook

- Elemental polonium (- ΔH_{ads} (Po) = 85⁺³₋₂ kJ/mol) was the most volatile species observed.
- The reactivity of the quartz surface is decisive.
- No observation of a reaction of polonium with hydrogen or water.
- Two oxygen-containing polonium species detected in oxygen atmosphere.

Experiments suggest to study livermorium in the elemental state, in agreement with theory

Transition to experiments with short-lived ²¹⁶Po





Thank you for your attention!