

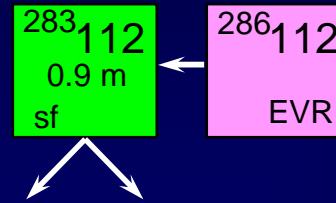
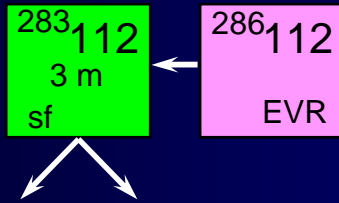
Status of the element 112 experiment

H.W. Gäggeler

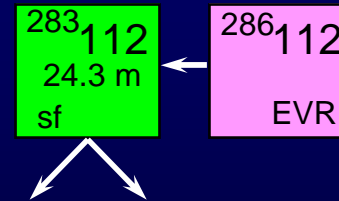
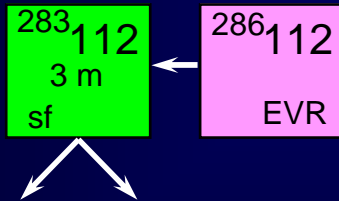
*For a GSI, PSI, Uni Bern, TU Munich,
Uni Mainz, LBNL, FLNR collaboration*

$^{238}\text{U}(^{48}\text{Ca},3n)^{283}\text{112}$

VASSILISSA



$E^* 33 \text{ MeV}$



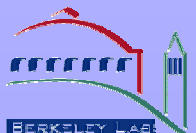
$E^* 35 \text{ MeV}$

Oganessian et al., 1999 & 2004

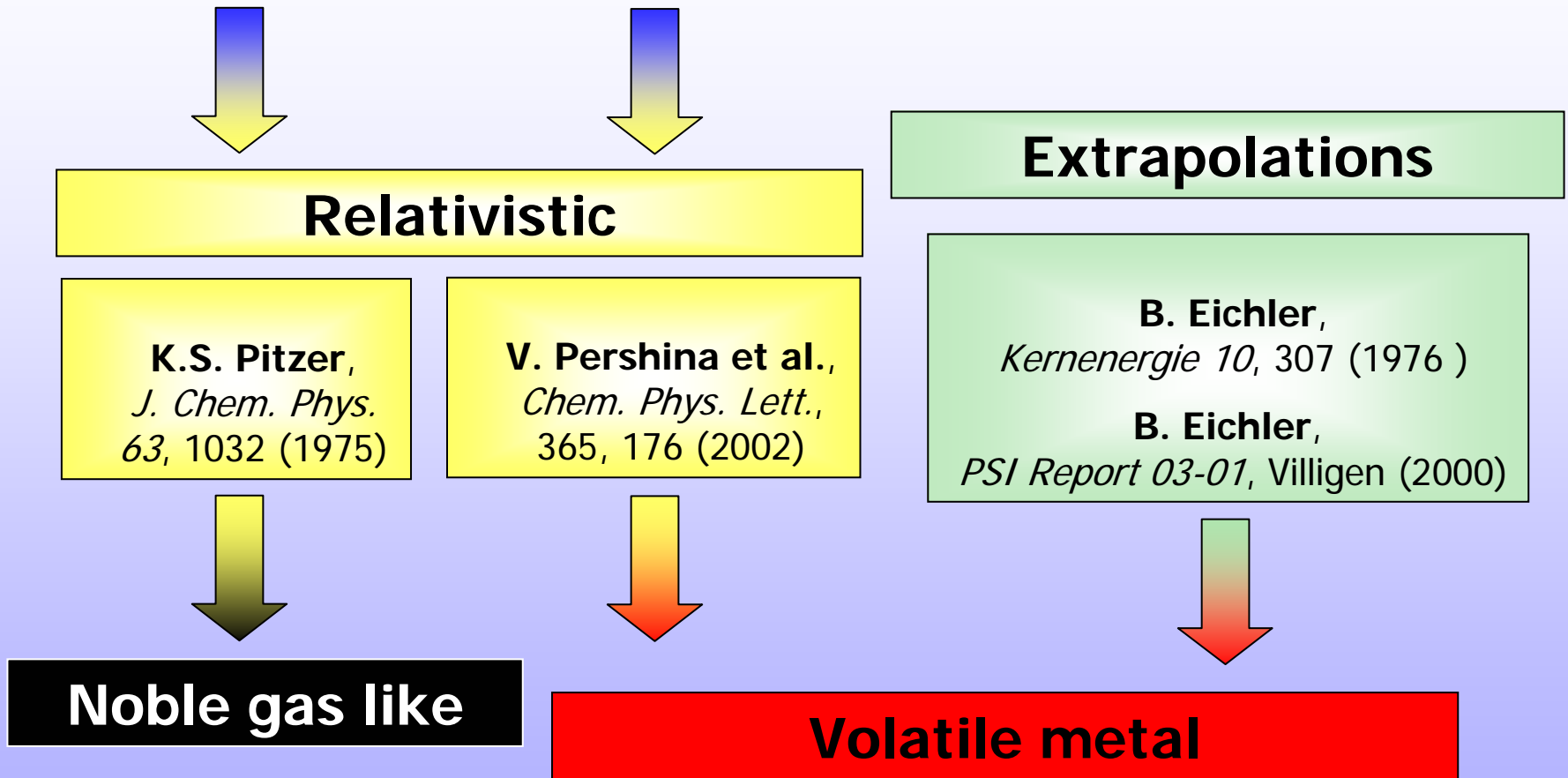
Periodic table of the elements

1																	18
1	2											13	14	15	16	17	2
H	He											B	C	N	O	F	Ne
3	4											5	6	7	8	9	10
Li	Be											B	C	N	O	F	Ne
11	12											13	14	15	16	17	18
Na	Mg	3	4	5	6	7	8	9	10	11	12	Al	Si	P	S	Cl	Ar
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
55	56		72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
87	88		104	105	106	107	108	109	110	111	112	113	114	115	116		
Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Uuu	Uub	Uut	Uuq	Uup	Uuh		

57	58	59	60	61	62	63	64	65	66	67	68	69	70	71
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
89	90	91	92	93	94	95	96	97	98	99	100	101	102	103
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr



Predictions for element 112



PSI

GSI



Berkeley
UNIVERSITY OF CALIFORNIA



univer
sität
mainz

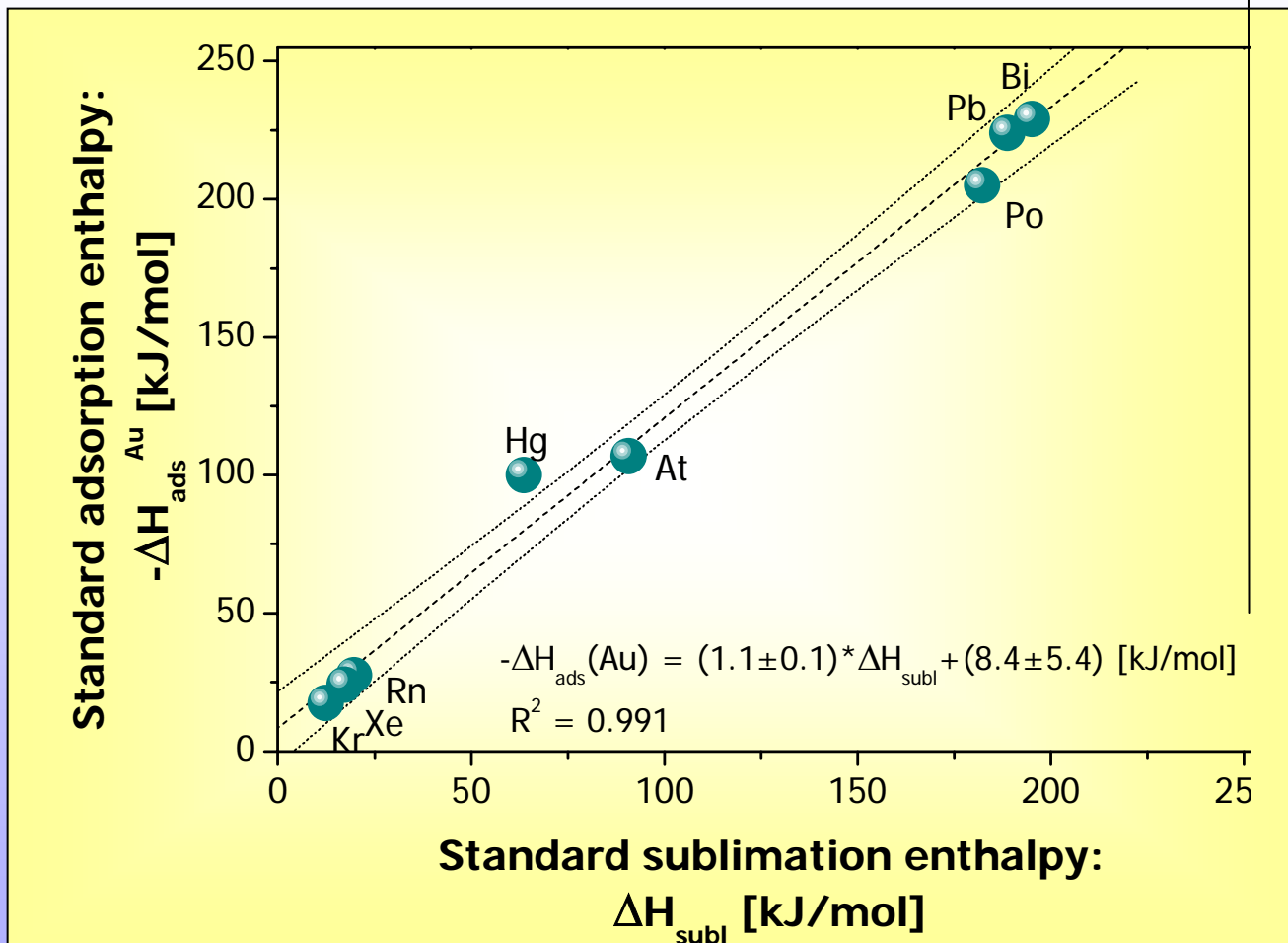


How to experimentally determine a metallic character at a single atom level?

→ Determine interaction energy (adsorption enthalpy) with (noble*) metals (e.g. Au)

* *Easier to keep clean surface during experiment*

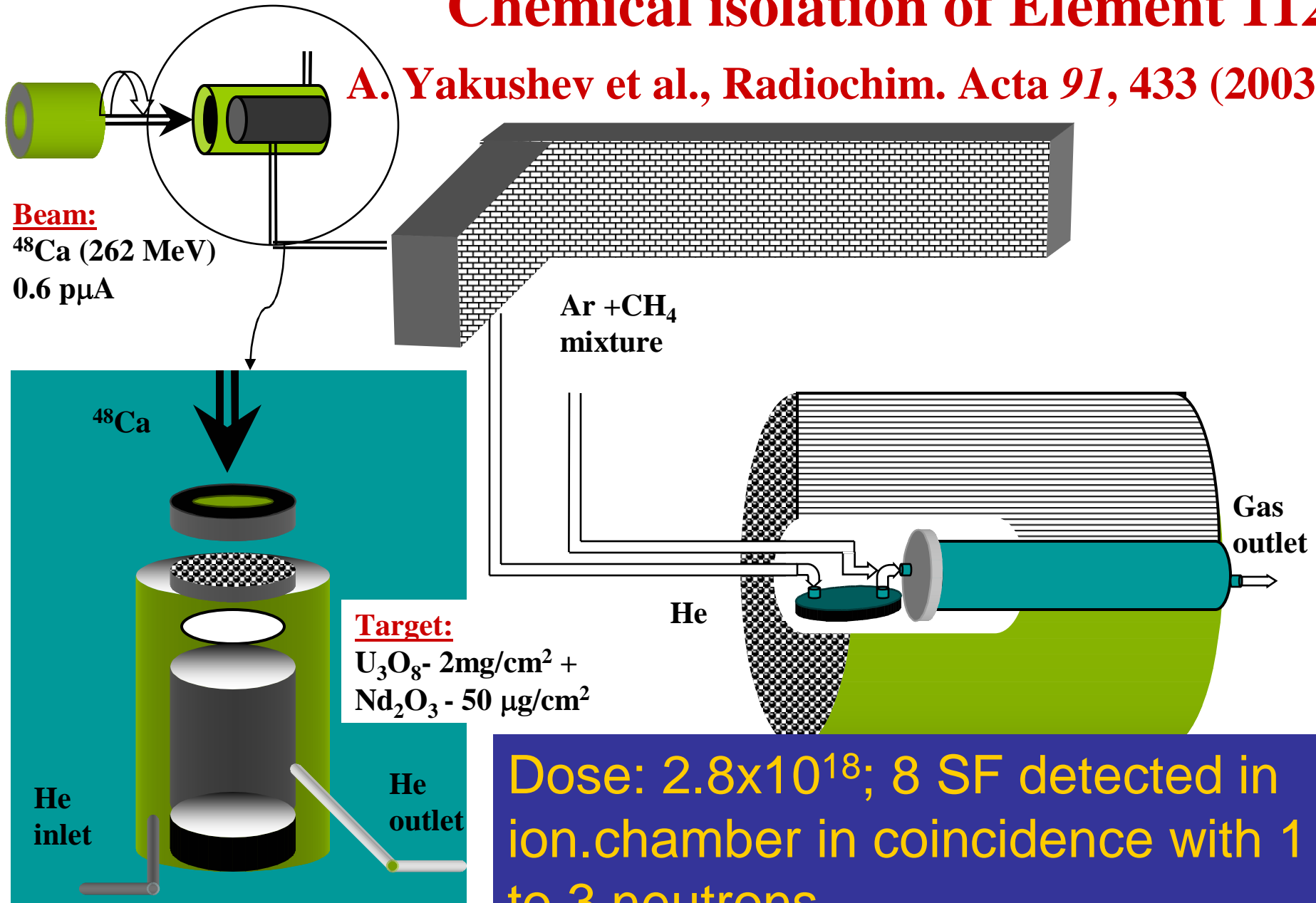
Correlation of $\Delta H_{\text{ads}}^{(\text{Au})}$ with ΔH_{subl}



R. Eichler, to be published

Chemical isolation of Element 112

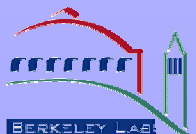
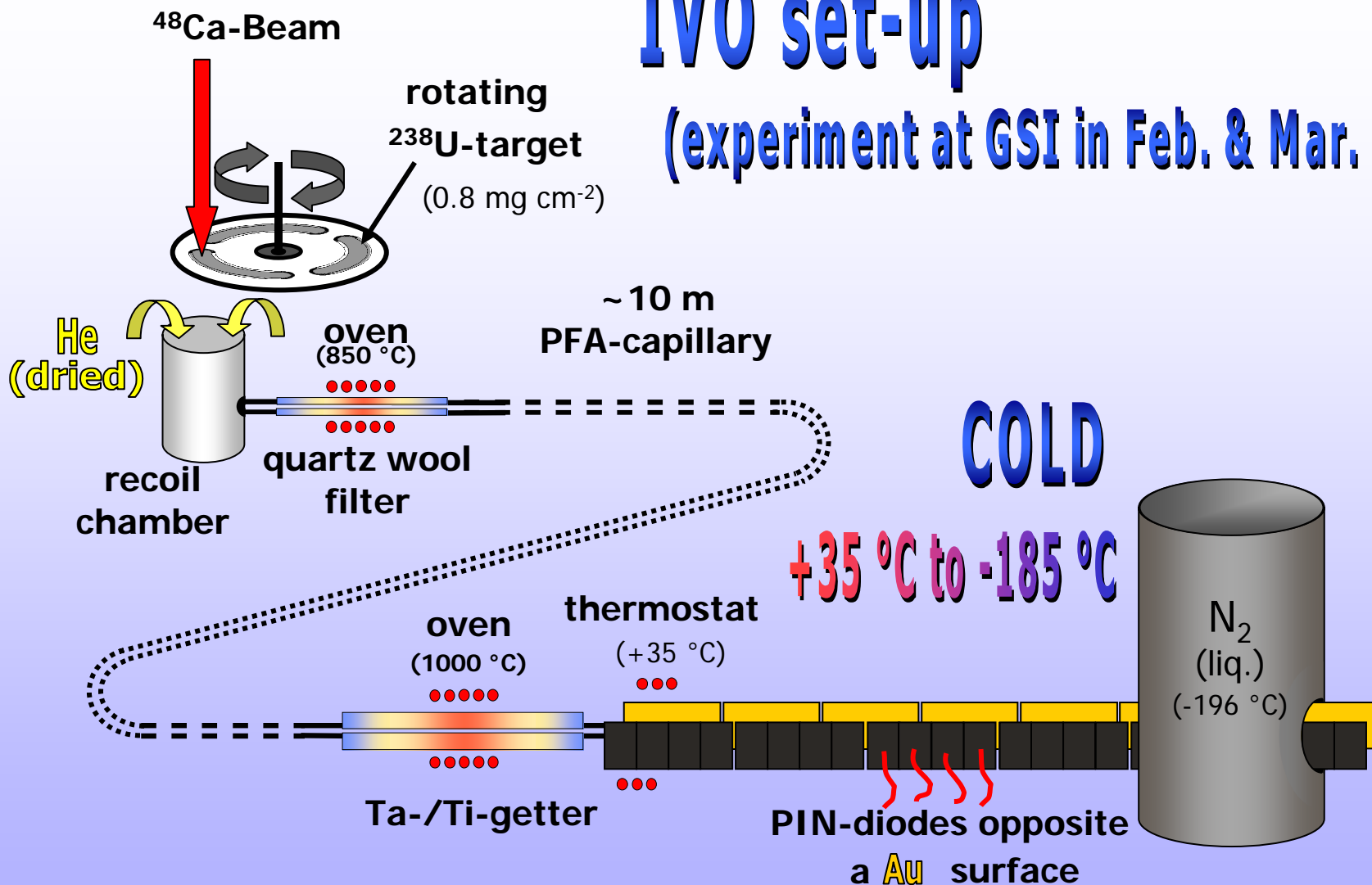
A. Yakushev et al., *Radiochim. Acta* 91, 433 (2003)



Dose: 2.8×10^{18} ; 8 SF detected in ion.chamber in coincidence with 1 to 3 neutrons

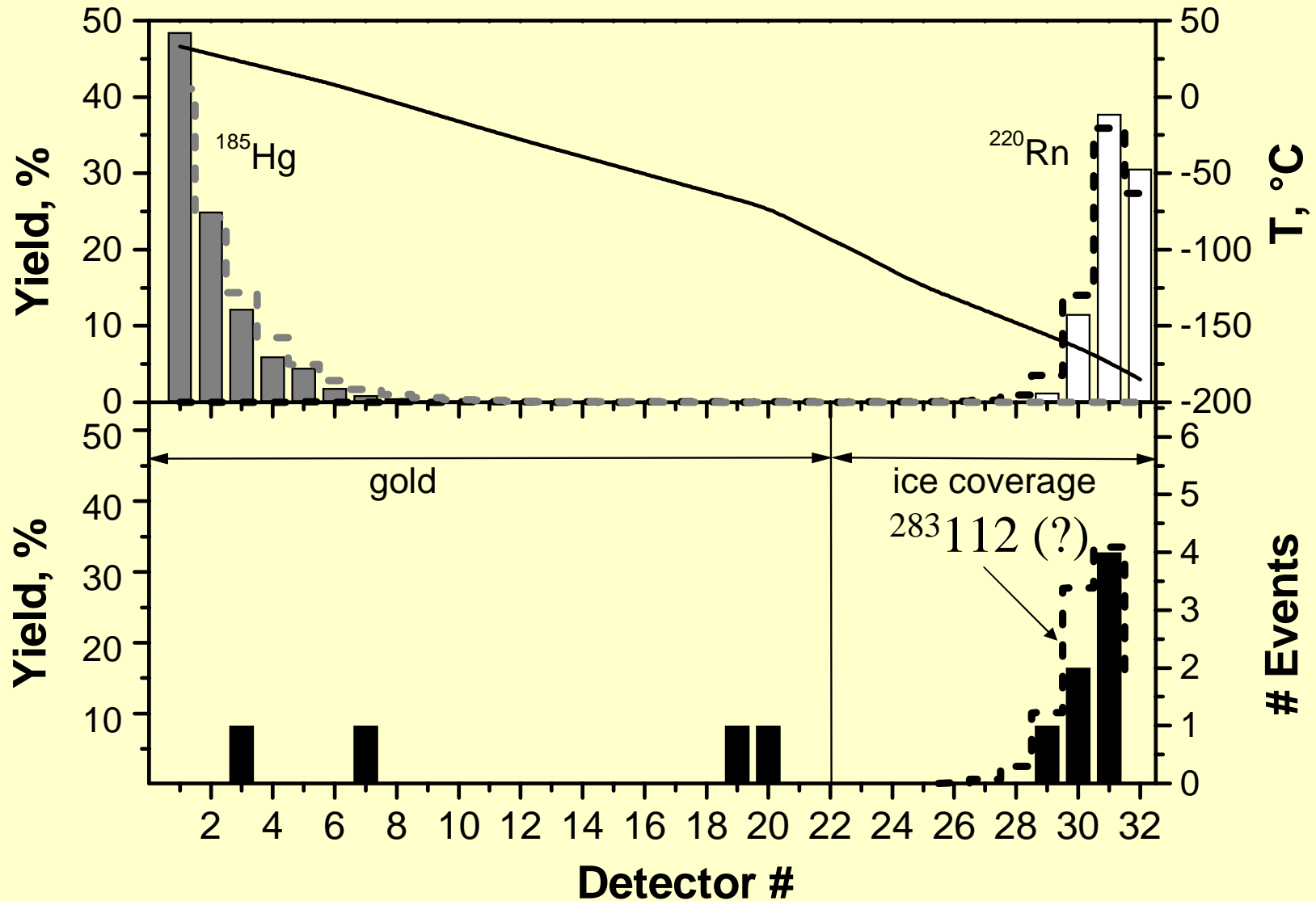
IVO set-up

(experiment at GSI in Feb. & Mar. 2003)



Experiment GSI February-March 2003

$^{238}\text{U}(^{48}\text{Ca}, 3n)^{283}\text{112}$ (SF, 3min); Dose 2.8×10^{18}



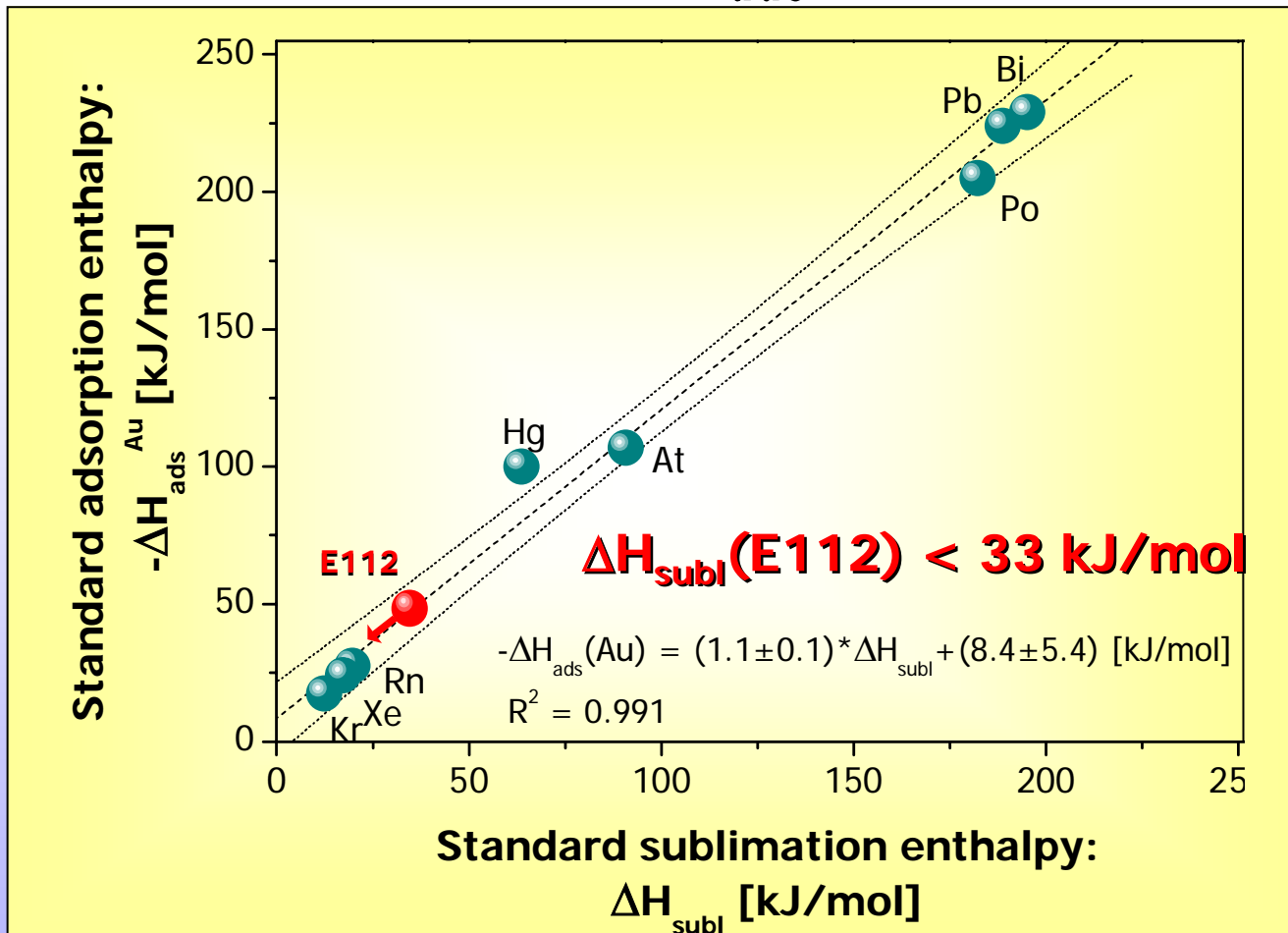
Current status on element 112 chemistry: Indication for

At FLNR: Isothermal chromatography on Au:

E112 does not adsorb at room temp. $\Delta H_a < 60$
kJ/mol (A. Yakushev et al.)

At GSI: Thermochromatography: E112 does not
deposit on Au down to $-90\text{ }^\circ\text{C}$ $\Delta H_a < 48$ **kJ/mol**
(S. Soverna et al.)

Correlation of $\Delta H_{\text{ads}}^{(\text{Au})}$ with ΔH_{subl}



If high energy events due to element 112, then it is most likely gaseous under standard conditions

Both experiments not conclusive, because

- Detection of SF activity not specific to assign it to a given (isotope of an) element.
- Fission fragment energies too low (FLNR: ion. chamber; GSI: PIN-diodes)

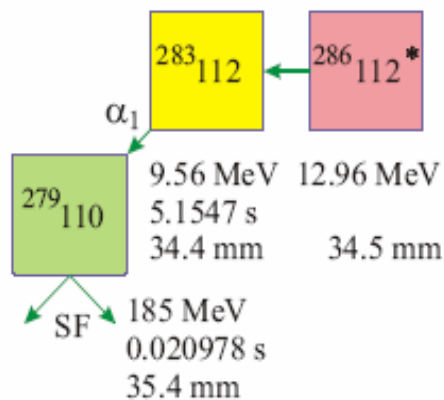
Even more concern:

- Several attempts to confirm Dubna findings failed at LBL (K.E. Gregorich et al): No observation of any evaporation residues using BGS at cross section limits being a factor of four below the claimed Dubna values
- Recent attempt to confirm the VASSILISSA result with DGFRS at FLNR failed also: no sf-activity was found but an α -decaying isotope discovered (Utyonkov, priv. comm.)

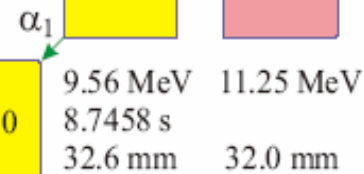
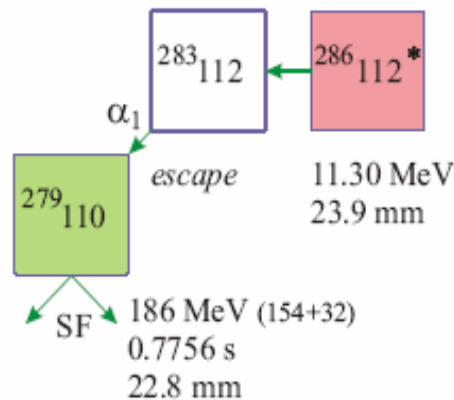
$E^*=35.0$ MeV

December 26, 2003
6:05 Strip 5

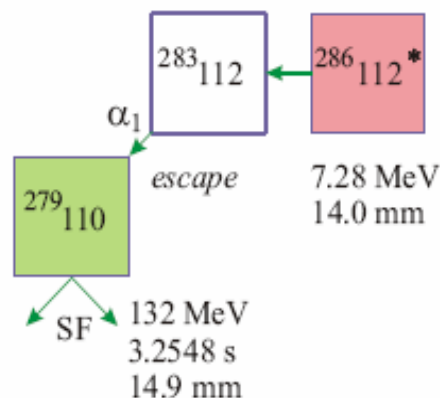
December 24, 2003
4:37 Strip 5



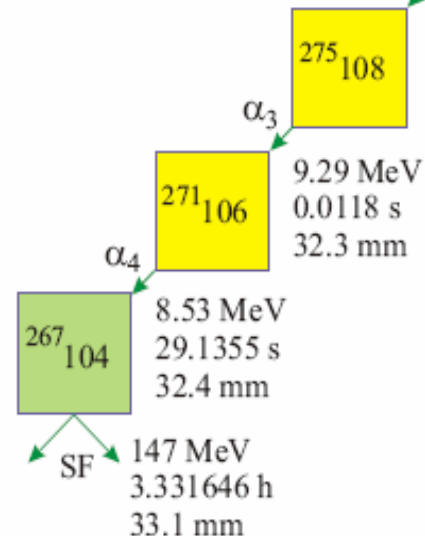
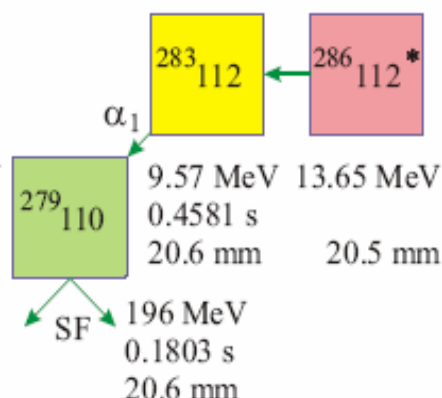
December 26, 2003
2:45 Strip 4



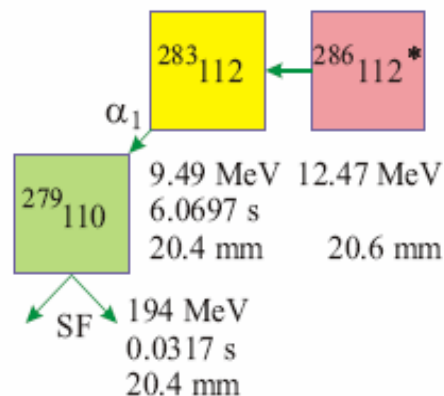
January 26, 2004
19:54 Strip 5



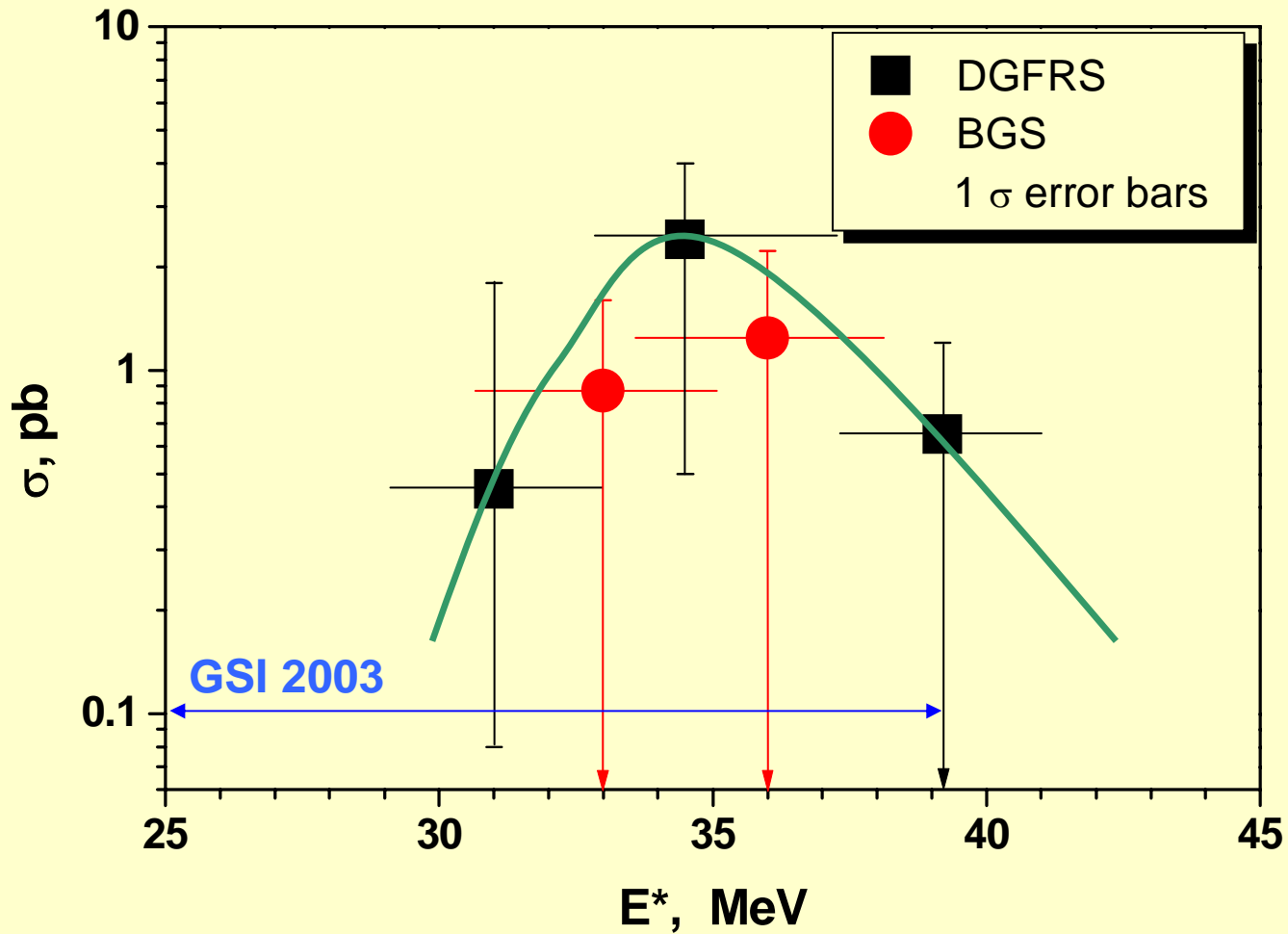
January 27, 2004
12:10 Strip 5



February 2, 2004
11:51 Strip 5



$^{238}\text{U}(^{48}\text{Ca},3n)^{283}112$



What's next ?

Repeat GSI experiment with improved set-up

September 2004

- 4π COLD
higher detection efficiency: single α -decay: 88 %
SF-decay: 76 % (coinc.)
- Faster transport time ≈ 2 s (sensitive to long-lived sf and short-lived α -emitters)
- Closed loop for the carrier gas (less water vapor)



3RD INTERNATIONAL CONFERENCE ON THE CHEMISTRY AND PHYSICS OF THE TRANSACTINIDE ELEMENTS

<http://tan07.web.psi.ch>

SEPTEMBER 23-28, 2007

DAVOS, SWITZERLAND

August 2004

The Third International Conference of the Chemistry and Physics of the Transactinide Elements (TAN 07) will be held September 23-28, 2007 in Davos, Switzerland. This conference is the continuation of TAN 03, the Second International Conference on the Chemistry and Physics of the Transactinide Elements that was held in Napa, California, USA during November 2003. Davos is located in a picturesque mountainous area in eastern Switzerland at an altitude of 1560 m asl.



Preliminary Schedule:

First Circular	September 2006
Deadline for Questionnaire	October 31, 2006
Second Circular and Call for Abstracts	December 2006
Deadline for Abstracts	February 28, 2007
Authors Notification and Deadline for Registration	April 30, 2007
Third Circular and Final Program	June 2007