

Towards the Lifetime Measurement of the ^{229m}Th³⁺ Nuclear Clock Isomer

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Why ^{229m}Th?



Nuclear vs. Atomic optical clock



²²⁹Th Nuclear Clock Applications

- Expected systematic frequency uncertainty of 1.5 x 10⁻¹⁹ [3]
- fundamentally different operation principle compared to atomic clocks => complementary to atomic: unique quantum sensor beyond timekeeping capability [4, 5]
- theoretical expectation of time variation of fundamental constants -> ²²⁹Th provides largely enhanced sensitivity to variations [4,6,7]
- Search for Dark Matter [4,8]
- Gravity sensor -> clock height shifts in gravitational potential of 1 cm correspond to frequency shifts at 10⁻¹⁸ precision level [8]

What do we know so far?

- Identification via direct decay in internal conversion (IC) decay branch [1]
- Lifetime of the neutral isomer: $\tau_{IC} = 7 \pm 1 \,\mu s$ [9]
- Hyperfine structure measured via collinear laser spectroscopy of ^{229m}Th²⁺ [10]
- First direct and precise measurement of the excitation energy:
- -> via IC: $E_{ex} = 8.28 \pm 0.17 \text{ eV}$ [11]
- -> via magnetic microcalorimetry: $E_{ex} = 8.10 \pm 0.17 \text{ eV}$ [12]

What is still missing?

- Ionic lifetime τ_{γ}
- Theoretical prediction: $\tau_{\gamma} = 10^3 10^4$ s => Natural linewidth $\Gamma = \frac{1}{2\pi \tau} < 0.16 \text{ mHz}$

Confine ^{229m}Th³⁺ ions in a cryogenic Paul trap with long storage times! (Storage time at 273 K is only 1 min)



with optical access ports

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